

# ALPHA ANALYTICAL, INC.

## Quality Assurance Manual

### Volume II

Rev. 15

January, 2007



#### Corporate Offices & Laboratory

255 Glendale Ave.  
Suite 21  
Sparks, NV 89431  
Phone: (800) 283-1183  
Phone: (775) 355-1044  
Fax: (775) 355-0406  
Email: [info@alpha-analytical.com](mailto:info@alpha-analytical.com)  
<http://www.alpha-analytical.com>

#### Client Services Office

3117 Fite Cr.  
Suite 103  
Sacramento, CA 95827  
Phone: (916) 366-9089  
Fax: (919) 366-9138  
Cell: (916) 730-6027  
Cell: (916) 705-6199

#### Client Services Office

6255 McLeod Dr.  
Suite 24  
Las Vegas, NV 89120  
Phone: (702) 736-7522  
Fax: (702) 736-7523  
Cell: (702) 281-4848

# **Appendix A**

## **P**ersonnel and Technical Resumes

**APPENDIX A**  
**PERSONNEL**

## **1.0 Personnel**

### **1.1 General Requirements for Laboratory Staff**

Technical staff positions are filled with personnel that fulfill the necessary requirements of education, training, technical knowledge, and experience for their assigned functions. This includes a general knowledge of laboratory operations, test methods, quality assurance/quality control procedures and records management.

In accordance with our training policies and procedures, Alpha maintains documentation that certifies technical personnel have the appropriate educational and/or technical background to perform all accredited test procedures.

### **1.2 Staffing Policies**

1.2.1 It is Alpha's policy to have a laboratory organized with sufficient managerial staff with the authority and resources needed to discharge their duties.

1.2.2 It is Alpha's policy to hire personnel which have appropriate education and/or On-the-Job-Training (OJT) adequate to perform their job duties.

1.2.3 It is Alpha's policy to conduct a training program that includes initial and continuing training of laboratory personnel.

1.2.4 It is Alpha's policy to ensure the competence of technical staff personnel who operate analytical equipment, evaluate results, and sign test reports.

### **1.3 Personnel Responsibilities**

1.3.1 It is the responsibility of the trainee to ensure they have received adequate initial and continuing training and the documentation of that training to achieve and maintain skills commensurate with their responsibilities.

1.3.2 It is the responsibility of all staff personnel to comply with all quality assurance/quality control requirements that pertain to their technical function in the laboratory.

## **2.0 Personnel Qualifications**

### **2.1 Technical Director**

#### **2.1.1 Definition**

The technical director means a full-time member of staff who exercises actual day-to-day supervision of the laboratory operations for the associated



fields of accreditation and reporting of results.

#### 2.1.2 Working Status

Technical directors who are absent for a period of time exceeding 15 consecutive calendar days must designate another full-time staff member meeting the qualifications of the technical director to temporarily perform this function.

If this absence exceeds 65 consecutive calendar days, the primary accrediting authority is notified in writing.

#### 2.1.3 Titles

The title of such person(s) may include but is not limited to:

- a) Laboratory Director;
- b) Technical Director;
- c) Laboratory Supervisor
- d) Laboratory Manager; and
- e) Quality Assurance Officer.

#### 2.1.4 Duties

Technical director(s) minimum duties include:

- a) monitoring standards of performance in quality control and quality assurance;
- b) monitoring the validity of the analyses performed; and
- c) monitoring the data generated to assure reliable data.

#### 2.1.5 Qualifications of Technical Directors

##### 2.1.5.1 Laboratory Director

The Laboratory Director must have a bachelors degree in chemical, environmental, biological sciences, physical sciences or engineering, with at least 24 college semester credit hours in chemistry and at least 2 years of experience in environmental analysis.

A masters or doctoral degree may be substituted for 1 year of experience or a BS degree in a science or science related field plus 10 years of analytical experience may be substituted for the chemistry credit requirement.

#### 2.1.5.2 Laboratory Manager

The Laboratory Manager must have a bachelors degree in chemical, environmental, biological sciences, physical sciences or engineering and at least one year of experience in environmental analysis. The Laboratory Manager must have a working knowledge of quality assurance principals.

#### 2.1.5.3 Qualifications of Quality Assurance Officer

The Quality Assurance Officer must have a bachelors degree in chemical, environmental, biological sciences, physical sciences or engineering with at least 24 college semester credit hours in chemistry and at least 2 years experience in environmental analysis.

A masters or doctoral degree may be substituted for 1 year of experience or a BS degree in a science or science related field plus 10 years of analytical experience may be substituted for the chemistry credit requirement.

### 2.3 Qualifications of Technical Staff

It is the responsibility of Alpha's management to formulate the goals with respect to education, training and skills of the technical and non-technical staff members.

Alpha's training program specifies the training policies and procedures for identifying the training needs and providing training of personnel. The qualification of all staff members is a combination of education, experience and training and are critical element in maintaining the qualifications of our staff.

#### 2.3.1 Laboratory Analyst

The Laboratory Analyst must have a bachelors degree in science or a science related field and at least 1 year of experience in environmental analysis. If the analyst is responsible for the operation of analytical instrumentation, they must complete specialized training offered by the manufacturer or another qualified training facility or served a minimum of a six month apprenticeship under an experienced analyst.

#### 2.3.2 Laboratory Technicians

The Laboratory Technician must have at least a high school diploma or equivalent, completed the in-house training program under an experienced analyst/technician and must have served a minimum of a six month apprenticeship under an experienced analyst/technician.

### 2.3.3 Laboratory Apprentice

The Laboratory Apprentice is a laboratory analyst or technician who has not fulfilled either the educational or experience requirements and is performing the job duties of one of those positions. The apprenticeship requirements are judged on a case-by-case basis. However, all employees performing job duties as an Apprentice are required to be under the direct tutelage of a senior Analyst or Technician who reviews all data or work produced until such time that the educational and experience requirements can be met.

**APPENDIX A**  
**TECHNICAL PERSONNEL RESUMES**

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 7 of 50

## **Laboratory Management**

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 8 of 50

Personnel Information

Name: Roger Scholl, Ph.D.  
Title: Laboratory Director  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Las Lomas Walnut Creek, CA	20 credit hours	1962	H.S. Diploma
Technical School or College	University of California Davis Davis, CA	55 credit hours	1966	B.S. Chemistry
Graduate Studies	University of California Davis Davis, CA	80 credit hours	1971	Ph.D. Chemistry

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1987	Present	Laboratory Director - oversee operations of environmental laboratory with two satellite customer service offices. Special expertise in GC/MS analyses and methods.
University of Nevada at Reno Reno, NV	1981	1987	Research faculty position in the Biochemistry Department, including a contract with the Nevada Division of Environmental Protection to conduct VOC analyses by GC/MS.

Date: 01/07 Signature: Roger Scholl

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 9 of 50

Personnel Information

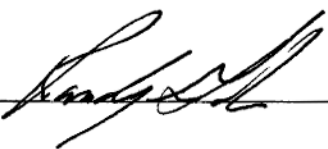
Name: Randy Gardner  
Title: Laboratory Manager  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Kapaun Mt Carmel Wichita, KS		1977	H.S. Diploma
Technical School or College	Angelo State Univ. San Angelo, TX	2 semesters	1981	B.S. Business
Graduate Studies	Wichita State University Wichita, KS	2 semesters	1985	M.S. Geology

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	1987 Present	Laboratory Manager and Administrator  Analyst, including EPA methods TPH-E by 8015B-DRO/DHS-LUFT and TOC by EPA method 415.1/9060  Semi-volatile and TPH-E sample preparation

Date: 1/07/ Signature: 



Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 10 of 50

### Personnel Information

Name: Walter Hinchman  
 Title: QA Officer  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Worthington-Jefferson H.S. Worthington, IN		1976	H.S. Diploma
Technical School or College	Indiana University Bloomington, IN	38 credit hour chemistry 8 credit hours biology 48 credit hours geology	1982	B.S. Geology (minor in chemistry)
Graduate Studies	Wichita State University Wichita, KS	9 credit hour chemistry 32 credit hours geology	1987	M.S. Geochemistry (core requirements)

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, NV	1998	Present	QA Officer - Manage the QA program
	1990	1998	QA Officer/Analyst/Extraction chemist for methods 504.1/505/515.1/531.1/547/548/549.2/524.2/525.2/551/552.1/608/615/ 8151/625/8270/624/8260/8011/300/314/TPH-P/TPH-E/8310/8330/8318 and various other analytical methods and extraction/clean-up procedures.
	1987	1990	Analyst/Extraction chemist- methods 608/8081/8082/625/8270

Date: 1/07

Signature: Walter Hinchman

**GC/MS**  
**Analysts**

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 12 of 50

Personnel Information

Name: Guanzhen Ji  
 Title: GC/MS analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	China			H.S. Diploma
Technical School or College	Fudan University Shanghai, China	45 credit hours	1984	B.S. Chemistry
Graduate Studies	Fudan University Shanghai, China Univ. of Nevada Reno, Reno, Nevada	50 credit hours	1987	M.S. Chemistry
		50 credit hours	1998	M.S. Chemistry

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1998	Present	Analyst, including EPA methods 8260B, TPH-P by 8015B-GRO/DHS-LUFT/NWTPH-Gx
Department of Chemistry University of Nevada, Reno	1994	1998	Research of copper chemistry by NMR, GC/MS, IR, UV-VIS
Shanghai Import and Export Commodity Inspection Bureau Shanghai, China	1987	1993	Inorganic analysis for minerals by volumetric, gravimetric, ICP, AA and x-ray fluorescence methods

Date: 1/07 Signature: Guanzhen Ji

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 13 of 50

### Personnel Information

Name: John Vasquez  
 Title: GC/MS analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks                      State: Nevada                      Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Othello H.S. Othello, WA		1992	H.S. Diploma
Technical School or College	Eastern Washington Univ. Cheney, WA	72	1998	B.S. Chemistry
Graduate Studies	University of Nevada, Reno	18		Core Class Requirement Fulfilled

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From                      To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	06/01/06 – Present 09/13/04 – Present	GC/MS Analyst to include 625/827C and PNA SIMS GC/HPLC Analyst to include methods 608/8081, 8082, Organic Acids
Florin Analytical Services	06/03 – 09/04	Sample digestion, Analyst for the following instrumentation: AA, CVAA, ICP and final reports.
Polyglass USA	08/02 – 06/03	Testing physical properties of raw and modified bitumens. Research information of new resins, and QC final product.

Date: 01/07

Signature: \_\_\_\_\_

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 14 of 50

### Personnel Information

Name: Melanie Wickham  
 Title: GC/MS analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Horizon High School Thornton, CO		1993	H.S. Diploma
Technical School or College	University of Tulsa Tulsa, OK	33 credit hours	1998	B.S. Geology
Graduate Studies	Univ. of Nevada Reno, Reno, Nevada	6 credit hours		geo-chemistry

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	2001	Present	Analyst, including EPA methods 8260B, TPH-P by 8015B-GRO/DHS-LUFT/NWTPH-Gx
Alpha Analytical, Inc. Reno, Nevada	1999	2001	VOC/TPH-P sample preparation
University of Tulsa	1995	1997	Prepared samples for XRD, SEM, FTIR, etc.

Date: 1/07 Signature: Melanie Wickham

Personnel Information

Name: Thomas Wickham  
 Title: GC/MS analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	La Cueva High School Albuquerque, NM		1989	H.S. Diploma
Technical School or College	University of New Mexico Albuquerque, NM	9 credit hours chemistry	1993	B.S. Geology
Graduate Studies	UNLV Las Vegas, NV	3 credit hours chemistry		geo-chemistry volcanology

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	1998 Present	Analyst, including EPA methods 524.2, 8260B and TPH-P

Date: 1-07 Signature: Thomas A. Wickham

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 16 of 50

### Personnel Information

Name: Wei Wu  
 Title: GC/MS analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks                      State: Nevada                      Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
<b>High School</b>	Xuejun High School	18 credit hours	1986	H.S. Diploma
<b>Technical School or College</b>	Zhejiang University of Technology, Hangzhou, China	36 credit hours	1990	B.S. Biochemical Engineering
<b>Graduate Studies</b>	University of Illinois, Champaign-Urbana, IL	18 credit hours	1995	M.S. Food Science

**EXPERIENCE:** Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1999	Present	Analyst, including EPA methods 8260B, TPH-P by 8015B-GRO/DHS-LUFT/NWTPH-Gx
Department of Chemical & Metallurgical Engineering, University of Nevada, Reno	1997	1999	Metals analysis for gold bio-leaching process by AA and ICP
Chemax Laboratories	1996	1997	Gold and silver analysis by AA
Bio-process laboratory, University of Illinois, Champaign-Urbana, IL	1994	1996	Analysis of proteins and starches by HPLC & TLC
Dept of Food Science University of Illinois, Champaign-Urbana, IL	1992	1995	Research Assistant - NMR analysis of proteins
Zhejiang Electric Power Research Institute, Hangzhou, China	1990	1991	Chemical and biochemical analyses for power plant waste streams

Date:

1-7-07

Signature:





## **GC/ECD and IC Analysts**

Personnel Information

Name: Joe Blaylock

Title: Analyst

Laboratory: Alpha Analytical, Inc.

Street: 255 Glendale Ave., Ste. 21

City: Sparks

State: Nevada

Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Bishop Gorman Las Vegas, NV		1981	H.S. Diploma
Technical School or College	UNLV Las Vegas, NV	11 credit hours chemistry	1992	B.S. Geology
Graduate Studies	UNLV Las Vegas, NV	3 credit hours chemistry		Geology

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Analyses now being done: Pesticides/PCBs by EPA method 608/8081/8082; Anions by EPA method 300.0; perchlorate by EPA method 314.0

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	9/2005	Present	Analyst, including EPA methods 608/8081/8082, 300, 314
Alpha Analytical, Inc. Reno, Nevada	2001	4/2004	Analyst, including EPA methods 608/8081/8082, 8151, 300, 314
UNLV Las Vegas, Nevada	1990	1994	X-ray fluorescence spectrometry for hard-rock geochemistry

Date: JAN, 07

Signature: \_\_\_\_\_

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 19 of 50

Personnel Information

Name: Matt Weikel  
 Title: IC, GC/ECD and HPLC analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431


Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Bishop Manogue H.S. Reno, NV		1996	H.S. Diploma
Technical School or College	University of Nevada, Reno	10 Biochemistry 18 Biology 20 Chemistry	NA	Major Biochemistry
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. 255 Glendale Ave., Ste. 21 Sparks, NV 89431	April 2004 - Present	Analyst for EPA method 300.0, 314.0, 608/8081, 8082 and organic acids
ALS Chemex Reno, NV Fairbanks, AK	April 2003 - April 2004	Reno Laboratory - Fire assay for precious metals Fairbanks Laboratory - branch manager of sample preparation facility
American Assay Laboratories 1500 Glendale Ave. Sparks, NV 89431	July 98 - Oct. 2002	Environmental Laboratory Manager - managed daily operations of the inorganic environmental laboratory at AAL Wet chemistry - pH, EC, TDS, turbidity, alkalinity, acidity, ammonia, total-P. Etc Extractions - TCLP, SPLP, STLC etc. Soil tests - MWMP, ABA, humidity cells, speciality columns, etc. Instrumentation - IC (HPLC) EPA method 300.0 ICP-AES by EPA method 200.7 Cold Vapor Hg by EPA method 245.2 FIA for CN by EPA method OIA1677 LECO for C & S determination

Date: 01/07

Signature: 

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 20 of 50

Personnel Information

Name: Ami Awano  
Title: Ion Chromatography and Wetchemistry analyst  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Toko High School Kawasaki, Kanagawa Japan	2 years	1996	Graduated
Technical School or College	University of Nevada, Reno Reno, NV	32 Credit hours	2001	B.S. Biochemistry
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	11/2003 - Present	Analyst for EPA Methods 300.0, 314 Inorganic/wetchemistry Lab Bench Analyst (pH, alkalinity, conductivity, etc.)
Various classes at UNR	1998 - 2000	

Date: 1/07 Signature: Ami Awano

**GC/FID  
&  
TOC  
Analysts**

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 22 of 50

Personnel Information

Name: Jennifer Webster  
 Title: GC analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Ponderosa H.S. Shingle Springs, CA		1993	H.S. Diploma
Technical School or College	Cosumnes River College University of Nevada, Reno	5 Biology 4 Biology 13 Chemistry	1998	B.S. Environmental & Natural Resource Science
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. 255 Glendale Ave., Ste. 21 Sparks, NV 89431	May 2002 - Present	TPH-E/Metals Analyst and extraction/digest chemist, Wet Chemistry-ph, conductivity, titrations, turbidity, ammonia, gravimetric procedures
American Assay Laboratories 1500 Glendale Ave. Sparks, NV 89431	March 99 - May 2002	Wet Chemistry - pH, conductivity, titrations, turbidity, phosphorous, TDS, etc. Extractions - TCLP, SPLP, STLC, MWMT, CN, etc. ICP-OES Analyst EPA Method 200.7, 6010B ICP-MS Analyst EPA Method 200.8, 6020 Metals Digestion EPA Method 3015, 3051A Inorganic QA Officer - QC review of all environmental inorganic data produced by AAL to include: 200.7, 200.8, 6010B, 6020, 300.0, 245.1, STM 4500-CN, and wet chemistry procedures Laboratory Manager - managed daily operations of the inorganic environmental laboratory at AAL

Date: 1/07

Signature: Jennifer Webster

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 23 of 50

Personnel Information

Name: Darin Hussey  
Title: GC analyst/laboratory technician  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	De Anza El Sobrante, CA		1988	H.S. Diploma
Technical School or College	Emergency Medical Technician		1992	EMT-1A #92-138
Graduate Studies	U.S. Navy USS Comstock LSD-45		1989-1991 Active Duty 1991-1993 Reserve Duty	

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	2001 Present	Analyst, including EPA methods TPH-E by 8015B-DRO/DHS-LUFT and TOC by EPA method 415.1
Alpha Analytical, Inc. Reno, Nevada	1995 2001	Semi-volatile and TPH-E sample preparation

Date: 01/07

Signature: \_\_\_\_\_

*Darin Hussey*



Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 24 of 50

## **Metals Analyst Wet Chemistry**

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 25 of 50

Personnel Information

Name: Jeff Yoshimoto  
Title: Metals analyst  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Saratoga H.S. Saratoga, CA	40	1982	H.S. Diploma
Technical School or College	U.C. Davis Davis, CA	80	1987	B.S. Environmental Toxicology
Graduate Studies	U.C. Davis Davis, CA	50	1992	M.S. Water Science

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. 255 Glendale Ave., Ste. 21 Sparks, NV 89431	Nov 2002 - Present	Metals Analyst - EPA method 200.8/6020/6020A  Metals Digestion chemist EPA method 200.2/3015/3051  Inorganic/Wetchemistry Lab Bench Analyst
NEL Laboratories 4750 Longley Ln. Ste., 106 Reno, NV 89502	March 1993 - November 2002	Metals Analyst/Inorganic Supervisor - EPA method 200.8/200.7/245.1/6020/6010/7470/7471 Metals Digestion chemist EPA method 200.2/3010/3020 pH Analyst EPA method 150.1/9045C Turbidity Analyst EPA method 180.1 Silica Analysis SM4500-Si-D Organic Lead DHS-method

Date: 1/2007 Signature: 

Personnel Information

Name: Megan Leslie  
Title: Inorganic/Wetchemistry analyst  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Herlong H.S. Herlong, CA 96113	1 year 2 semesters	1994	H.S. General Diploma
Technical School or College	University of Nevada, Reno Reno, NV 89557	37	2001	B.S. Geology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	06/05/03 - Present	Inorganic/Wetchemistry Lab Bench Analyst
Biostratigraphy, UNR LMR Building Reno, NV	02/1998 - 05/1998	Extracted conodonts and other microfossils from prepared rock samples.
Various classes at UNR Reno, NV	1994 - 2001	<p><b>Invertebrate paleontology</b> - overview and ID of invertebrate fossils</p> <p><b>Chemistry 101 &amp; 102</b> - general chemistry labs</p> <p><b>Geochemistry</b> - interaction between elements to produce reaction in rocks</p> <p><b>Ore Deposits</b> - application of all geology courses in rock ID, geochemistry structural and physical analyses, chemical compositions of rocks, rock-forming processes, interactions and implications.</p> <p><b>Field Methods</b> - 5 weeks working in field school to practice principles of rock ID, formation, interaction, metamorphosis, etc.</p>

Date: 1/07 Signature: Megan Leslie

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 27 of 50

### Personnel Information

Name: JT Simonsen  
 Title: Inorganic Analyst  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Lowry High School Winnemucca, NV		2001	
Technical School or College	University of Nevada, Reno (UNR) Reno, NV 89577	6 credits - chemistry 5 credits - biology 25 credits -biochemistry	2005	B.S. - Biochemistry
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	August 2005 Present	<p>Analytical Procedures -</p> <p>Percent Moisture, (120.1-Conductivity), (150.1 - pH) , (160.1-TDS), (16.2 - TSS), (160.3 - TS), (180.1 - Turbidity), (305.1 - Acidity), (310.1 - Alkalinity), (330.5-Res. Chlorine), (350.3 - Ammonia), (351.4 -TKN), (365.2 - Total and ortho-Phosphorus), (376.2 - Sulfide), (410.4-COD), (SM3500-Cr - Cr 6), (SM3500 Fe D - ferrous iron), (1664A -oil and grease), (3015 -Aqueous Microwave acid digestion ), (3051 -Solid Microwave acid digestion)</p> <p>Other duties -</p> <p>Cleaning of laboratory equipment          Maintaining paperwork and appropriate documentation.</p>
University of Nevada, Reno Biochemistry Lab 148	July 2002 August 2005	<p>Polyacrylamide gel formation,          Solution production,          Sterile procedures for tissue culture,          mini-maxi preparation,          cleaning of lab equipment,          pH meter,          lab maintenance</p>

Date: Jan 2007 Signature: JT Simonsen

## **Extraction Chemists**

Personnel Information

Name: Rick Overton

Title: VOC Analytical Manager

Laboratory: Alpha Analytical, Inc.

Street: 255 Glendale Ave., Ste. 21

City: Sparks

State: Nevada

Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Ranum High School Denver, CO		1968	H.S. Diploma
Technical School or College	Colorado St. University Denver, CO Metropolitan State College, Denver, CO	10 credit hours	1968 - 1970 1972 - 1978	B.S. Electrical Engineering (Core Requirements)
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1992	Present	GC/MS VOC/SVOC Section Manager
Bondar Clegg, Inc. Reno, Nevada	1984	1992	Laboratory Manager for mine assay laboratory. Supervised fire assays and analytical instruments AA, ICP and XRF.
Bondar Clegg, Inc. Denver, CO	1972	1984	Inorganic analyst - AA, ICP, XRF and fire assays

Date: 1/07Signature: Rick Overton

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 30 of 50

Personnel Information

Name: Zhen Jin  
 Title: VOC Sample Preparation Chemist  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks                      State: Nevada                      Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	High School #1 Sizhiwang qi, China		1973	H.S. Diploma
Technical School or College	Northeast University Shenyang, China	60 credit hours	1982	B.S. Chemical Engineering
Graduate Studies	University of Idaho Moscow, ID	20 credit hours 30 credit hours	1988 1991	M.S. Metallurgical Eng. Ph.D. Metallurgical Eng.

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1999	Present	GC/MS VOC sample preparation, screen analysis
Kappes, Cassidy and Assoc. Reno, NV	1993	1999	Analyst - AA, ICP, IC, LC & HPLC
University of Idaho Metallurgical Department Moscow, ID	1987	1992	Analyst - AA, ICP, IC, LC, HPLC & XRF
Shenyang Mining and Metallurgy Institute Shenyang, China	1982	1987	Analyst - metals

Date: 01-07 Signature: Zhen Jin



Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 31 of 50

Personnel Information

Name: Behrooz Aryainejad  
Title: SVOC Sample Preparation Chemist  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Bloomington North H. S. Bloomington, IN	4 Semesters	1988	H.S. Diploma
Technical School or College	Truckee Meadow Community College Reno, Nevada University of NV Reno Reno, Nevada	4 Semesters	1993	B.S. Fine Arts Minor in biology, chemistry
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	1991 Present	SVOC sample preparation for the following analysis: 515.1, 548, 549.2, 525.2, 552.1, 608, 615, 625, 827C, 8081, 8082, 8151, 8270, 8015B-DRO/ORO, 1311(TCLP), 1312(SPLP), and various other extraction and clean-up procedures for both soil and water matrices.

Date: 1/07

Signature: \_\_\_\_\_

*Behrooz Aryainejad*

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 32 of 50

### Personnel Information

Name: David Maestas  
 Title: Laboratory Apprentice/Sample Disposal  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Elko H.S.  Elko, NV		1970 (GED)	H.S. Diploma
Technical School or College	Truckee Meadow Community College (TMCC) Reno, NV	Mathematics (10)	Continuing Studies	General Studies
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	2002	Present	Collect and bulk samples for waste disposal and sample pick-up.  SVOC sample preparation for the following analysis: 625, 8270, 608, 8081, 8082, and various other extraction and clean-up procedures for both soil and water matrices.
NEL 4750 Longley, LN, Ste., 106 Reno, NV 89502	1999	2001	Organic Ext Supervisor - Supervised extraction of the following drinking water methods: 515.1, 507, 508, 525.2, 548.1, 549.2 Soil/water methods 625/8270, 608/8081, 8082, TPH
Barringer Laboratories, Inc. 5301 Longley Ln Bldg E. Reno, NV 89502	1992	1998	Acid digestion of rock samples for preparation of mineral/precious metals assay.

Date: 1-07

Signature: \_\_\_\_\_

*David Maestas*

Alpha Analytical, Inc.  
 Section No.: A.1  
 Revision No.: 15.0  
 Date: January, 2007  
 Page: 33 of 50

### Personnel Information

Name: Autumn Durkin  
 Title: Soil Extraction Chemist  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

### Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Reno H.S. Reno, NV		5/98	
Technical School or College	UNR Reno, NV	8 credits - chemistry 66 credits - geology	12/03	B.S. Geology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	11/05 - Present	SVOC soil extraction (Method 3545, ASE)  Inorganic/wetchemistry procedures  Percent Moisture, (120.1-Conductivity), (150.1 - pH), (160.1-TDS), (16.2 - TSS), (160.3 - TS), (180.1 - Turbidity), (SM3500 Fe D - ferrous iron), (3015 -Aqueous Microwave acid digestion ), (3051 -Solid Microwave acid digestion)

Date: 01/07

Signature: Autumn Durkin

## **Sample Custodians**

Personnel Information

Name: Latricia Edrosa  
 Title: Sample Custodian, QC/Report Writer  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Campbell County H.S. Gillette, Wy	4	1994	General H.S. Diploma
Technical School or College	University of Dubuque 2000 University Ave. Dubuque, IA 52001	37	1998	B.S. Environmental Science and English
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	2005 - Present 2003 - 2004	Sample Custodian, QC/Report Writer and Project Manager GC/MS VOC sample preparation, screen analysis

Date: 01/07 Signature: Latricia Edrosa

Personnel Information

Name: Tara Dickinson  
 Title: Sample Custodian, QC/Report Writer  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Carson High 1111 Salimon St Carson City, NV 89706		2000	
Technical School or College	University of Nevada, Reno (UNR) Reno, NV 89577	6 credits - chemistry 3 credits - biology 3 credits - microbiology	2004	B.S. - Health Ecology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	2005 Present	Sample Custodian, QC/Final report writing and uploading data into LIMs

Date: 01/07 Signature: Tara Dickinson

Personnel Information

Name: Tasha Pascal  
 Title: Sample Custodian  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Las Vegas Academy		1997	
Technical School or College	University Nevada, Reno Reno, NV	9 - Chemistry 18 - Biology	In attendance	Environmental Science
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	Jan 2006 - Present	Sample Custodian
DRI	Jun 2004 - Oct 2005	Lab technician - GC/FID laboratory analyst, cleaned sample media and shipping/receiving
UNR	Jan. 2003 - Sept 2003	Lab technician - Soil chemistry laboratory (prepared soil samples for analysis)

Date: 01/01/07 Signature: Tasha Pascal

Personnel Information

Name: Elizabeth Sauvageau  
 Title: Sample Custodian  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Tonapah H.S. Tonapah NV		2001	
Technical School or College	University Nevada, Reno Reno, NV	9 - Chemistry		B.S. Environmental Policy Analysis
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	5/2006 - Present	Sample Custodian

Date: 01/07 Signature: Elizabeth Sauvageau



**Document Control Officers  
&  
Quality Control / Report Writers**

Personnel Information

Name: Tracy DeGiovanni  
 Title: Report Writer  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Reed High School Sparks, NV		1991	H.S. Diploma
Technical School or College	University of Nevada, Reno (UNR) Reno, NV	27 credit hours	1999	B.S. Biology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	2000 Present	Report Writing - QC and Analytical report writer. Review QC and analytical data entry for LIMs upload accuracy used for final QC and analytical data report generation.
	2000 2002	Sample Custodian - ( sample log-in/sample receipt of specific client samples shipped through a courier service, phone reception, data filing, customer service, bottle orders, general secretarial).  Project Manager - Manage and coordinate specific environmental projects as an acting liaison between the field engineers, laboratory managers, analysts, QC officer and report writers to ensure projects requirements are full-filled.
University of Nevada, Reno Mammal Laboratory Reno, NV	1998 2000	Behavioral studies and captive breeding of <u>Dipodmys meramii</u>

Date: 01/07 Signature: Tracy DeGiovanni

Personnel Information

Name: Reyna Vallejo  
 Title: QC/Report Writer  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Washoe High School		1997	H.S. Diploma
Technical School or College	CET		1998	Office Skills
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To		Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, NV	2000	Present	QC/Analytical Report Writer - Review analytical QC and data entry into the LIMS for final analytical/QC report generation.
Nevada Environmental Labs Reno, NV	1997	2000	Sample Custodian - ( sample log-in/sample receipt of drop-in clients and samples shipped through a courier service, phone reception, data filing, customer service, bottle order, data entry and general secretarial.

Date: 1/07 Signature: Reyna Vallejo

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 42 of 50

Personnel Information

Name: Tara Manzini  
Title: Document Control  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Sparks H.S. Sparks NV		2004	Honors Diploma
Technical School or College	TMCC Reno, NV		In attendance	
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, Nevada	July 2006 - Present	Document Control - duties to include data filing, faxing, scanning, e-mailing, data entry and sub-laboratory follow up.

Date: 01/07 Signature: Tara Manzini

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 43 of 50

Personnel Information

Name: Emily Steele  
Title: Shipping and File Clerk  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Alexandria Monroe Township H.S. Indiana		1966	H.S. Diploma
Technical School or College	Anderson College Anderson, IN	Attended 1 year		Studied elementary education
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, NV	1999	Present	Shipping/File Clerk - ( prepare sample and bottle orders for field engineers/samplers, phone reception, data filing, and general secretarial.
Nevada Environmental Labs Reno, NV	1998	1999	Clean and sterilize laboratory glassware
Chemax Laboratories Sparks, NV	1990	1998	Clean and sterilize laboratory glassware, Laboratory Technician for MBAS.
Legend Metallurgical Labs Reno, NV	1987	1989	Sample digestion, Laboratory Technician/analyst for metals analysis by AA.

Date: 01/07 Signature: Emily Steele

Personnel Information

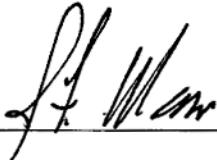
Name: Joseph Mauro  
 Title: Shipping and File Clerk  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Regis H.S. Denver, CO		1975	H.S. Diploma
Technical School or College	Doane College Crete, NE		1979	B.S. Accounting
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, NV	8/2006 Present	Shipping/File Clerk - data scanning and general office duties.

Date: 01/11/07 Signature: 

## **Computer Specialists**

Personnel Information

Name: Michael Aseltine  
Title: LIMS Administrator/Analyst  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Marysville High School Marysville, CA		1978	H.S. Diploma
Technical School or College	University of the Pacific Stockton, CA	60 credit hours	1982	B.A. Biology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1997	Present	Analyst 8260B/624, 8270C/625, TPH-P, 525.2 & 548.1 LIMS Administrator
Columbia Analytical, Inc. Anchorage, AK	1994	1997	Analyst - 8260B, 8270C & 8015Mod.
ALS Stafford Australia	1993	1994	GC/MS Trainer for 8240/8270
Enseco Sacramento, CA	1991	1993	GC/GCMS/HPLC analytical instrument maintenance
ECOVA/PNELI Redmond, WA	1988	1990	Analyst - 8010, 8020, 8015, 8080, 8240 & 8150
Canonie Environmental Stockton, CA	1986	1988	GC Supervisor - 8010, 8020 & 8015
California Water Labs Modesto, CA	1984	1986	Analyst - 8010, 8020 & 8015

Date: Jan 2007 Signature: Michael L Aseltine



Personnel Information

Name: Jason Herrmann  
Title: Systems Administrator  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431


Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Franklin High School Michigan		1989	H.S. Diploma
Technical School or College	University of Nevada at Las Vegas (UNLV) Las Vegas , Nevada	6 credits hour chemistry 4 credit hours geo-chem	N/A	Geology
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, Nevada	1998	Present	Systems Administrator, Safety Officer, Waste Management
Secor, Inc.	1998	1999	Assistant Environmental Consultant
Southern NV Water Authority	1997	1998	Environmental Technician, took GPS coordinates of wells in the Las Vegas Valley

Date: 01/07

Signature: 

**Accounts Payable and Receivable  
&  
Employee Benefits**

Alpha Analytical, Inc.  
Section No.: A.1  
Revision No.: 15.0  
Date: January, 2007  
Page: 49 of 50

Personnel Information

Name: Tammy Brace  
Title: Accounts Receivable/Billing Department  
Laboratory: Alpha Analytical, Inc.  
Street: 255 Glendale Ave., Ste. 21  
City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Reno High Booth St., Reno, NV		1984	H.S. Diploma
Technical School or College	Monterey Peninsula United School (Clerical)		1990	Business/Office
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed		Duties (including specific analyses performed)
	From	To	
Alpha Analytical, Inc. Reno, NV	1991	Present	Invoices/Monthly Statements/Quotes/Collections/Mailing of Final Reports

Date: 01/07 Signature: Tammy Brace

Personnel Information

Name: Corinne Miner  
 Title: Accounting/Employee Benefits  
 Laboratory: Alpha Analytical, Inc.  
 Street: 255 Glendale Ave., Ste. 21  
 City: Sparks State: Nevada Zip Code: 89431

Education

	Name and Address	Semester Credit Hours in Chemistry & Biology	Year Graduated	Degree and Major Area of Study
High School	Blackfoot High School		1974	General HS Diploma
Technical School or College				
Graduate Studies				

EXPERIENCE: Include all experience involving chemical, microbiological or radiochemical analyses.

Laboratory & Address	Date Employed From To	Duties (including specific analyses performed)
Alpha Analytical, Inc. Reno, NV	1992 Present	Accounts payable, payroll, payroll taxes, quarterly taxes, reconciling bank accounts, workers' compensation insurance, company liability insurance and certificates, general ledger, and monthly financial statements. Corinne is also responsible for coordinating the employee benefits which include the medical and disability insurance plans and the Alpha 401 (K) Plan.

Date: 1/07 Signature: Corinne Miner

# **Appendix B**

## **Field Sampling Plan**

## 1.0 FIELD SAMPLING PLAN (FSP)

### 1.1 Introduction

Generalized field procedures used in the FSP are described by SOP's and should be followed while performing field work. For most sites, specific work plans are generally developed and specific SOP's are designed for each unique site.

The FSP has been written to assure uniformity and consistency of the sampling procedures. The FSP should be implemented at the sampling site when no other field sampling plan has been developed by the field engineers. These procedures should be performed during all phases of the sampling plan.

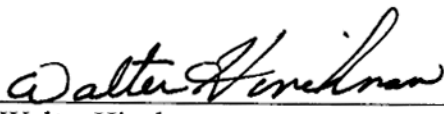
The FSP includes descriptions of sampling documentation, sampling procedures, decontamination procedures and field QC procedures. Discussions of data quality objectives, laboratory quality control, sample labeling, shipment and custody records, and quality assurance oversight are provided in the Quality Assurance Manual.

### 1.2 Management Review of Potential New Work

Before sampling is performed at any site, the Project Manager should meet with a representative of Alpha Analytical to establish the sampling methodology to be employed, and the tests which will be performed on the samples. A sample collection plan should be determined, areas of responsibility delineated and a logistical plan developed.

After the planning meeting, the appropriate laboratory personnel is alerted of the need for specific laboratory analysis. Laboratory personnel will be provided with the information required to satisfy all specifications outlined in the Statement of Work (SOW). In addition, the logistical criteria such as sample containers, sample volume, preservatives, transportation and the appropriate numbers and types of QC samples needed is finalized, and finally the laboratory is assessed to ensure we have the appropriate facilities and resources to complete the project.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

**1.0 TITLE: FIELD SAMPLING DOCUMENTATION**

Description: Sample integrity begins from the time the sample is actually taken. A bound logbook is used to record all pertinent information regarding the sampling trip to ensure proper and complete documentation.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 The logbook must contain information to distinguish samples and sample locations from any other. Entries into the logbook should include the following:

- Client/Project name for which the sampling is being conducted,
- Matrix sampled (i.e. groundwater, soil, etc.),
- Specific sampling location in sufficient detail to allow re-sampling at the same location,
- Sampling date and time,
- Sample depth,
- Method of sampling,
- Volume of water removed during well purge,
- Sample type taken (e.g., duplicate, split or field blank),
- Analytes of interest,
- Preservation techniques,
- Field Sample ID,
- Significant observations made during the sampling process,
- Field measurements taken,
- Decontamination procedures; and,
- Printed name and signature of the person performing the sampling.

2.2 Sampling situations vary widely and no general rules can specify the extent of information that must be entered in a logbook or standardized form. However, records should contain sufficient information so that someone can reconstruct the sampling activity without relying on the collector's memory.

Prior to the collection of any samples, the sampling location should be verified using site maps if available. If discrepancies are noted, sampling locations should be verified with the Project Coordinator before sampling.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

## **1.0 TITLE: QUALITY CONTROL FIELD SAMPLES**

Description: The following quality control samples should be collected to ensure the reliability and integrity of field and analytical data for water and soil samples.

## **2.0 STANDARD OPERATING PROCEDURE**

### **3.0 WATER QC SAMPLES**

- 3.1 The same type of QC samples should be collected for groundwater and surface water samples. Field duplicates and three types of quality control blanks are typically used (trip blanks, field blanks and equipment blanks).
- 3.2 Trip Blanks consist of a sample bottle filled with organic free water. Trip blanks are prepared in the laboratory and sent out in the field. Trip blanks are transported to the site in the same ice chest used in the field and transported back to Alpha along with samples obtained in the field.
- 3.3 Equipment/Rinsate Blanks are used to ensure non-dedicated sampling devices (bailers, filtering equipment, pumps, etc.) have been effectively decontaminated. Field equipment are rinsed with reagent free water, collected and transferred to a sample bottle for future analysis.
- 3.4 Field blanks consist of reagent free water placed in a sample container at Alpha and treated as a sample in all respects, including exposure to sampling site conditions, storage, preservation and all analytical activities. The purpose of the field blank is to determine if method analytes or other interferences are present in the field environment.
- 3.5 Duplicates should be collected and analyzed on a ten percent basis. The relative percent difference between duplicate measurements should provide an estimate of sampling precision.

Table B.3-1 lists the number of field QC samples that should be collected for each sampling event. Section 11 of the QA plan addresses the laboratory QC program with respect to matrix spike/matrix spike duplicates.

Matrix spike/matrix spike duplicates are the only samples which requires additional sample volume. When determining sampling logistics make sure this is taken into consideration.



TABLE B.3 - 1 SUMMARY OF QUALITY CONTROL SAMPLES		
	Field QC	
Sample Type	Target Frequency %	Comment
Trip Blank		1 per cooler for VOC's / Analyze as needed
Equipment/Rinsate Blanks	10%	1 per analysis per day / analyze as needed
Field Blanks		1 per source per event analyzed for VOC's as needed
Field Duplicates	10%	ensure adequate volume
	Laboratory QC	
Sample Type	Target Frequency %	Comment
Duplicate	5%	1 per analytical method
Blank		1 per analytical or extraction batch
Matrix Spike/Matrix Spike Duplicate	5%	1 per analytical method
Surrogate Spike	Method Specific	Method Specific
Laboratory Control Spike	5%	1 per analytical or extraction batch

#### 4.0 SOIL/SOLID QC SAMPLES

- 4.1 Trip Blanks should be sent with each cooler when volatile organic analysis is required. The trip blanks should be analyzed for volatile organics only.
- 4.2 Equipment/Rinsate Blanks should be used to ensure non-dedicated sampling equipment has been effectively decontaminated. Samples should be collected each day soil sampling is conducted. Only a representative number should be analyzed. Additional blanks may be analyzed if a problem is perceived.
- 4.3 Field Duplicates should be collected at a frequency of ten percent of the total number of soil samples collected. Field duplicates are collected and analyzed to determine the precision of field sampling. Table B.3-2 lists the number and types of field QC samples that should be collected for each sampling event.

**TABLE B.3 - 2**  
**SUMMARY OF QUALITY CONTROL SAMPLES**

<b>Field QC</b>		
<b>Sample Type</b>	<b>Target Frequency %</b>	<b>Comment</b>
Trip Blank		1 per cooler for VOC's / Analyze as needed
Equipment/Rinsate Blanks	10%	1 per analysis per day / analyze as needed
Field Duplicates	10%	ensure adequate volume
<b>Laboratory QC</b>		
<b>Sample Type</b>	<b>Target Frequency %</b>	<b>Comment</b>
Duplicate	5%	1 per analytical method
Blank		1 per analytical or extraction batch
Matrix Spike/Matrix Spike Duplicate	5%	1 per analytical method
Surrogate Spike	Method Specific	Method Specific
Laboratory Control Spike	5%	1 per analytical or extraction batch

Approved By: Walter Hinchman  
 Walter Hinchman  
 QA Officer

Date: January, 2007

**1.0 TITLE: VOLATILE SAMPLING TECHNIQUES - 524.2**

Reference: Methods for the Determination of Organic Compounds in Drinking Water, USEPA, EMSL, Cincinnati OH, Supplement III; EPA-600/R-95/131, Method 524.2, Revision 4.1, 1995.

Description: This sampling procedure is written specifically for those samples requiring analysis by method 524.2 under the Safe Drinking Water Act.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Collect samples in triplicate using 40mL clear VOA vials. If samples are suspected to contain residual chlorine add approximately 25mg of ascorbic acid per 40ml of sample to the sample bottle before filling, (pre-preserved in the laboratory).
- 2.2 Adjust the pH of all samples to  $< 2$  at the time of collection (field preserve), but after dechlorination, by carefully adding two drops of 1:1 HCL for each 40mL VOA vial. Seal the sample vial and mix for 1 minute. Do not mix the ascorbic acid with HCL in the sample bottle prior to sampling.
- 2.3 When sampling for Trihalomethanes (THM's) analysis only, acidification may be omitted if sodium thiosulfate (3mg per 40ml sample) is used to dechlorinate the sample. This exception to acidification does not apply if ascorbic acid is used for dechlorination.
- 2.4 If a sample foams vigorously when HCL is added, discard the sample. Collect a set of samples but do not acidify them. These samples must be flagged as "not acidified" and must be stored at 4°C. These samples must be analyzed within 24 hours of collection time if they are to be analyzed for any compounds other than THM's.

**Table B.4 - 1**  
**Sample Preservation and Holding time Table**  
**Method 524.2**

<b>Description</b>	<b>Sample Volume</b>	<b>Dechlorination</b>	<b>Sample Preservation</b>	<b>Analysis Holding time</b>
Full List Compounds	3 x 40mL	25mg ascorbic acid per 40mL sample	pH<2, 2 drops 1:1 HCL Field preserved cool 4°C	14 days
Full List Compounds sample foams when HCL is added carbonaceous waters	3 x 40mL	25mg ascorbic acid per 40mL sample	No acid	Analyze within 24 hours
THM's only	3 x 40mL	25mg ascorbic acid per 40mL VOA vial	pH<2, 2 drops 1:1 HCL Field preserved cool 4°C	14 days
THM's only	3 x 40mL	Sodium Thiosulfate 3mg/40mL sample	No acid	14 days
THM's only sample foams when HCL is added carbonaceous waters	3 x 40mL	Sodium Thiosulfate 3mg/40mL sample	No acid	14 days

- 2.5 Place the septums, Teflon side down, on the sample, and screw on the cap with out dislodging the Teflon liner.
- 2.6 Invert the sample and lightly tap the lid to ensure the absence of entrapped air bubbles. If air bubbles are trapped in the vial, add additional sample until sample container is free of air bubbles.
- 2.7 As each VOA vial is filled, enter the applicable information on the label and pack the vial in the shipping container at 4 degrees C. Samples must be refrigerated at the time of collection and maintained at that temperature, analyze within 14 days of collection. Samples that will not be received at the laboratory on the day of collection must be packaged for shipment with sufficient ice to ensure they will be at 4 degrees C on arrival at the laboratory.
- 2.8 Never allow the sample to freeze during transportation. If samples are refrigerated with ice, pack the vials to ensure contact between the ice and sample vials are minimized to avoid potential freezing.

- 2.9 Never filter VOC samples.
- 2.10 Never sample for volatile organics near a running motor or any type of exhaust system because discharged fumes and vapors may contaminate the sample.
- 2.11 VOC samples may also be contaminated by diffusion of volatile organics through the septum during shipment and storage. To monitor possible contamination, a trip blank prepared from reagent grade water, is carried throughout the sampling storage, shipping and analytical process. Additional QC samples may be collected (i.e., field blanks, rinse blanks, field duplicates, etc.), but should be considered on a case-by-case basis.
- 2.12 When sampling from a water tap, open the tap and allow the system to flush until the water temperature has stabilized usually about 5 to 10 minutes). Adjust the flow to about 500 ml/min and collect the samples from the flowing stream.
- 2.13 When sampling from an open body of water, fill a wide mouth bottle or a beaker with sample from a representative area, and carefully fill sample bottles from the container.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

**1.0 TITLE: VOLATILE SAMPLING TECHNIQUES-WATER (624/8260B)**

Reference: Guidelines establishing test procedures for the Analysis of Pollutants, 40 CFR Part 136, Appendix A, SW846, Final Update III, June 1997.

Description: There are several requirements common to most VOC sampling events. Overall care must be taken in regard to equipment and, container handling, storage, decontamination, and record keeping.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Collect all samples in triplicate. Slowly fill each container to overflowing avoiding aeration or agitation of the sample to the greatest possible extent.
- 2.2 Adjust the pH of the samples to <2 by carefully adding 1 drop of 1:1 HCL for each 20 ml of sample volume. Samples are collected in 40 ml VOA vials and generally require 2 drops of 1:1 HCL per vial.
- 2.3 Place the septums, Teflon side down, on the sample, and screw on the cap with out dislodging the Teflon liner.
- 2.4 Invert the sample and lightly tap the lid to ensure the absence of entrapped air bubbles. If air bubbles are trapped in the vial, add additional sample until sample plus the duplicate vials are free of air bubbles.
- 2.5 As each VOA vial is filled, enter the applicable information on the label and pack the vial in the shipping container at 4 degrees C. Samples must be refrigerated at the time of collection and maintained at that temperature until analysis. Samples that will not be received at the laboratory on the day of collection must be packaged for shipment with sufficient ice to ensure a temperature of 4 degrees C will be maintained on arrival at the laboratory.
- 2.6 Never allow the samples to freeze during transportation. If samples are refrigerated with ice, pack the vials to ensure contact between the ice and sample vials are minimized to avoid potential freezing.
- 2.7 Never filter VOC samples.
- 2.8 Never sample for volatile organics near a running motor or any type of exhaust system because discharged fumes and vapors may contaminate the sample.
- 2.9 VOC samples may also be contaminated by diffusion of volatile organics through the septum during shipment and storage. To monitor possible contamination, a trip blank prepared from reagent grade water, is carried throughout the sampling storage,

shipping and analytical process. Additional QC samples may be collected (i.e., field blanks, rinse blanks, field duplicates, etc.), and should be considered on a case-by-case basis.

- 2.10 When sampling from a water tap, open the tap and allow the system to flush until the water temperature has stabilized usually about 5 to 10 minutes). Adjust the flow to about 500 ml/min and collect the samples from the flowing stream.
- 2.11 When sampling from an open body of water, fill a wide mouth bottle or a beaker with sample from a representative area, and carefully fill sample bottles from the container.
- 2.12 If protective gloves are used, they should be clean, new, disposable, and Nitrile. Gloves should be changed between sampling events to prevent the possibility of cross-contamination. Care should be taken to prevent the sample from touching the gloves while filling the containers or touching the inside of the container with the gloves.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: SOIL AND SEDIMENT SAMPLING TECHNIQUES**

Description: Soil or sediment samples are typically collected in 8oz/250 ml or 4oz/125mL wide-mouth glass bottles according to the prescribed protocol below.

**2.0 STANDARD OPERATING PROCEDURES**

- 2.1 Fill each container, as a grab sample or by using a sampling device, with the soil sample. Avoid aeration of the sample to minimize the loss of volatile organic compounds if VOC methods are to be taken from the same sampling containers.
- 2.2 As each container is filled, enter the applicable information on the label and pack the bottle in the shipping container at 4 degrees C. Samples must be refrigerated at the time of collection and maintained at that temperature until analysis. Samples not received at the laboratory on the day of collection must be packaged for shipment with sufficient ice to ensure a temperature of 4°C on arrival at the laboratory.
- 2.3 Never sample near a running motor or any type of exhaust system because discharged fumes and vapors may contaminate the sample.
- 2.4 Samples may also be contaminated by the use of non-dedicated sampling equipment such as a scoop or split sampling device. To monitor possible contamination a rinse blank should be collected each day soil sampling is conducted and analyzed for the compounds of interest.
- 2.5 If protective gloves are used they should be clean, new, disposable, and Nitrile. Gloves should be changed between sampling events to prevent the possibility of cross-contamination. Care must be taken to prevent the sample from touching the gloves while filling the containers or touching the inside of the container with the gloves.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer



**1.0 TITLE: SEMI-VOLATILE SAMPLING TECHNIQUE - WATER**

Description: When sampling water for semi-volatile organic compounds, samples should be collected in 1 liter amber bottles according to the prescribed protocol below. In addition to the following procedures, overall care must be taken in regard to equipment and container handling, storage, decontamination, and record keeping.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Collect all samples in duplicate by slowly filling each container, minimizing the amount of sediment collected, so that the head space is no greater than the threaded portion of the neck.
- 2.2 Add the appropriate preservatives to the samples as described in the QA manual. It is important the right amounts of preservative be added by checking the pH, residual chlorine etc. for this verification.
- 2.3 Cap the bottles with a Teflon lined cap and invert the sample to ensure the preservatives are well dispersed in the sample.
- 2.4 As each bottle is filled, enter the applicable information on the label and pack the bottle in the shipping container at 4 degrees C. Samples should be refrigerated at the time of collection and maintain refrigeration until extraction. Samples not received at the laboratory on the day of collection must be packaged for shipment with sufficient ice to ensure a temperature of 4°C is maintained on arrival at the laboratory. It is important for samples to be shipped as soon as possible because most Semi-volatile water samples need to be extracted within seven days of sampling.
- 2.5 Never allow the sample to freeze during transportation. If samples are refrigerated with ice, pack samples such that contact between the ice and sample bottles are minimized to avoid potential freezing.
- 2.6 Precautions must be taken to limit the contamination of samples from outside sources. Hands should be washed and gloves worn prior to sampling. The order of sampling should be from the least contaminated well to the most contaminated well.
- 2.7 When sampling from a water tap, open the tap and allow the system to flush until the water temperature has stabilized (usually about 5 to 10 minutes). Adjust the flow to about 500 ml/min and collect the samples from the flowing stream.
- 2.8 Samples from springs, surface water or other open bodies of water should be collected as grab samples. These samples should be carefully collected to minimize the turbulence and amount of sediment collected with the water samples.

- 2.9 If protective gloves are used, they should be clean, new, disposable, and Nitrile. Gloves should be changed between sampling events to prevent the possibility of cross-contamination. Care must be taken to prevent the sample from touching the gloves while filling the containers or touching the inside of the container with the gloves.

Approved By:  Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: GROUNDWATER SAMPLING - MONITORING WELLS**

Description: Monitoring wells are particularly important to understanding the hydrology and remediation efforts at a project site. Well information is used to define the geochemical baseline and it is therefore extremely important to sample each well exactly the same. The following procedures incorporate the necessary aspects of sampling QA and should be used each time a monitoring well is sampled.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Whenever feasible, wells not expected to be contaminated should be sampled first, followed by wells with increasing levels of contamination.
- 2.2 In most wells, a submersible pump, if available, should be used to facilitate purging. The pump is generally powered by a portable generator. The generator must be operated down wind and as far away as possible from the actual sampling location. Actual samples to be submitted for analysis should be taken by a bailer. In low-yield wells, purging and sampling should be conducted by bailing.
- 2.3 Immediately prior to well purging and sample collection activities, the static water level below the top of the well casing should be measured with an electronic sounder and recorded in the logbook. Water levels should be recorded to the nearest 0.01 foot.
- 2.4 Measure and record the depth from the top of the casing to the bottom of the sediment/water interface. Subtract the depth to top of the water from the depth to the bottom of the sediment/water interface and determine the height of standing water in the casing.
- 2.5 Measure the well diameter and determine the water volume using the following equation:

$$\text{well volume} = (\pi) (r^2) (h)$$

Where;

r = radius measured in feet

h = height of water measured in feet

$\pi = 3.141$

Well volume calculated in cubic feet. The conversion of cubic feet to gallons is:

$$1 \text{ ft}^3 = 7.48 \text{ gallons}$$

- 2.6 Prior to sample collection purge three well volumes from the well. Purging and sampling should be performed in a manner that minimize the agitation of sediments in the water column and to reduce the potential of organic chemical volatilization.

- 2.7 If the well goes dry during purging, it is assured of removing all water which has prolonged well casing or air contact. If the recovery rate is quick, allow the well to recover to its original level before subsequent purging of the well. If water recovery is very slow, samples should be taken when sufficient water is available.
- 2.8 Samples for volatile organics should be collected first and immediately sealed in 40mL VOA vials with no head space. All samples should be checked for the presence of bubbles that may bias the analytical data. Samples for volatile organics should not be homogenized, composited or filtered.
- 2.9 All samples should be taken in pre-cleaned containers. Semi-volatile samples should be placed in 1 liter amber glass bottles in duplicate, volatile organic samples should be sampled in three 40 ml VOA vials and metals or inorganic parameters should be sampled in 125 to 1000 ml polyethylene containers. Add the appropriate preservatives and cap the sample containers for storage and transportation to the laboratory.
- 2.10 Refrigerate all samples in an ice chest at 4 degrees C immediately after sampling and deliver to the laboratory as soon as possible.
- 2.11 In addition to the record keeping requirements of the QA plan, the following information should be recorded each time a well is purged and sampled:
- Depth to water before and after purging,
  - Well casing volume calculations,
  - Condition of each well,
  - Apparent thickness of any floatable hydrocarbon layer; and
  - Any required field parameter (i.e., pH, EC, temperature).
- 2.12 All non-dedicated purging and sampling equipment should be decontaminated between wells. In addition, non-dedicated bailers should be rinsed once with well water prior to collecting a sample.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

**1.0 TITLE: FIELD DECONTAMINATION AND WASTE DISPOSAL**

Description: Sampling equipment must be cleaned prior to and after each use to minimize cross contamination of samples. Good house-keeping practices are reflected in the analysis of clean equipment/rinsate blanks.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Equipment Decontamination**

2.1.1 Sampling equipment should be cleaned prior to sample collection, in a controlled environment, preferably at the laboratory and transported to the field pre-cleaned and ready to use. Sampling equipment should be cleaned between sample locations, and at the end of sampling activities. Large HDPE drums should be used to hold wash and rinse solutions.

- o Clean all sampling equipment with tap water and a non phosphate detergent (Alconox or Detergent-8). Brush if necessary to remove particulate matter or surface film.
- o Rinse thoroughly with tap water.
- o Rinse thoroughly with deionized or distilled water. Enough water should be used to insure all surfaces are flushed with water.
- o Rinse twice with hexane or methanol. One rinse may be used as long as all surfaces are thoroughly wetted with free flowing solvent.
- o Rinse thoroughly with deionized or distilled water.
- o Rinse with organic free water and allow to air dry.
- o Clean sampling equipment should be wrapped in aluminum foil or protected by other means to prevent contamination during storage or transportation to the field.

**2.2 Disposal of Field Waste Material**

All discarded material or other objects should be handled in a way to minimize waste and to preclude the potential for spreading contamination or causing litter to be left on-site.

The major waste material generated during field activity is purged well water and fluids collected during decontamination.

2.2.1 Well Purge Water

All water collected from monitoring wells should be collected in approved containers. The containerized liquids should be handled according to the analytical results of the well water. Those waters purged from existing wells showing the presence of contaminants, should also be containerized and handled accordingly.

2.2.2 Decontamination Water

Similar to purged well water, water generated from decontamination activities should also be containerized, sampled and handled according to approved disposal procedures.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: SAMPLE PACKING AND TRANSPORTATION**

Description: Samples taken in the field must be packaged in a manner to prevent breakage, or cross-contamination during transportation to the laboratory. Samples must be refrigerated with sufficient ice to ensure a temperature of 4°C on arrival at the laboratory.


**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Samples should be segregated by site, sampling location, or by sample analysis type during the packing of coolers. Sample segregation may follow this scheme or any other scheme that is sensible and well thought out. These schemes are dependent on the levels of contamination present, the number of bottles to be transported, the size of the bottles etc.
- 2.2 VOC samples from different sources may be placed into the same cooler to reduce the number of required trip blanks.
- 2.3 Shipping containers should be insulated for cooling.
- 2.4 Samples in breakable containers should be packed with material ( bubble wrap, vermiculite, foam etc.) to prevent breakage.
- 2.5 Sample coolers should be sealed with strapping tape or other means to prevent tampering. Custody seals may also be placed on the container lids.
- 2.6 All samples requiring thermal preservation should be packed in thoroughly insulated coolers with wet ice.
- 2.7 Packed samples should be delivered to the laboratory by the sampling team or common carrier. If sent by a common carrier, all documentation (transmitted forms, bills-of-lading, COC's etc.) should be sealed and placed inside the shipping containers prior to sealing it. It is recommended to place all forms inside a plastic bag and tape it to the underside of the cooler lid.
- 2.8 Secondary Containment
  - 2.8.1 To prevent melted ice from leaking out of the sample cooler during transportation, it is recommended to encapsulate all samples and cooler ice in a garbage bag. This is accomplished by double bagging the ice-chest and unfolding the garbage bags lining the bag in the cooler. Place all samples, packing material etc. in the first of the two garbage bags. This first or inner garbage bag is then pigtailed and tied.
  - 2.8.2 If a temperature blank is to be used, place the blank in the second garbage

bag. To prepare a temperature blank Place a temperature blank which contains a thermometer in a 40 ml VOA vial into a larger 500 ml HDPE (plastic) container. The plastic container filled with water acts as secondary containment as well as providing a much larger thermal mass to help stabilize the temperature readings. Finally, pre-cool the temperature blanks prior to placement into the coolers. Reducing the thermal mass of a volume of water from ambient temperature to 4°C takes approximately 4 to 6 hours. Therefore pre-cooling the temperature blank will help expedite the thermal transfer.

- 2.8.3 Place crushed ice not block ice in the second garbage bag, filling the remaining space of the cooler to produce contact with the ice and the samples that are in the first garbage bag.
- 2.8.4 Tie the second garbage bag which now contains the first garbage bag with samples, the temperature blank and the loose ice and prepare to shipment to the laboratory.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007



# **Appendix C**

## **Sample Tracking Plan**

## 1.0 SAMPLE TRACKING PLAN (STP)

- 1.1 The Sample Tracking Plan is a set of SOPs designed and written specifically for maintaining the integrity of a sample throughout its life at Alpha. These SOPs are strictly followed during all phases of the sample's stay at Alpha.
- 1.2 The STP includes procedures for the transportation, receipt, handling, protection storage, and disposal of samples, including the procedures necessary to protect the integrity of the sample, and to protect the interests of the laboratory and client.
- 1.3 Sample tracking and the maintenance of custody starts with the sample collection activities in the field and continues throughout the sample's progress in the lab. Sample collection is the responsibility of the field sampler; however, once the samples arrive at the laboratory, sample custody is turned over to our laboratory.
- 1.4 We cannot control the custody or the maintenance of custody until it arrives at our laboratory. Therefore, standard chain of custody procedures are typically used for all samples unless otherwise required.
- 1.5 Sample integrity is an equally important factor in all samples, regardless of sample type.
- 1.6 The Sample Custody Officers (SCO) is responsible for the implementation of the STP. This responsibility includes assuring that the proper handling and documentation of all samples are performed according to the SOPs described in this plan. Occasionally, a QAPP or SOW will require additional or different procedures to be followed for their samples.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: SAMPLE IDENTIFICATION**

Description: Alpha Analytical, Inc. has developed a system for uniquely identifying samples to ensure traceability of samples while in the possession of Alpha Analytical, Inc. and to maintain sample identity while in-house. Each sample received is assigned a unique lab ID number.

**2.0 STANDARD OPERATING PROCEDURE:**

2.1 The sample identification assigned to a sample is retained throughout the life of the sample at the laboratory.

2.2 The sample identification system is specifically designed and operated to ensure that samples cannot be confused physically or when referred to in records or other documents.

Once samples have been assigned a sample identification number, this identification is retained for the sample, sub-samples, subsequent extracts, and/or digestates related to that original sample.

2.3 The sample identification system is specifically designed to accommodate the grouping of sample (e.g., SVOC grouping from TPH-P) and the transfer of samples within and from the laboratory.

2.4 The Alpha Analytical Identification (AAI) is generated to assign a unique identification code to each sample container received by the laboratory.

1) Clarification: NELAC notes the use of container shape, size, or other physical characteristics, such as amber glass, or purple top is an unacceptable practice. Since we use a unique ID system this is not an issue but is stated only to make the point.

2.5 The AAI number assigned to each sample container is also the same ID used to unequivocally link the sample containers, extracts, etc. to the ID assigned in the field by the sampling team.

This link is unequivocally established using the Chain of Custody where the relationship between the field ID and laboratory ID are defined.

2.6 The laboratory ID is placed on the sample container using a durable sample label.

- 2.7 Sample containers and extract vials are labeled with a lab ID number in the following format:

XXXYYMMDDSC- ZZ

Where:

XXX represents a 3-letter prefix unique to each individual client,

YY refers to the last two digits of the year,

MM refers to the month,

DD refers to the day,

SC refers to the sample custodian who logged in the sample, and

ZZ refers to the sample number

For example: JDI07011030-05 would be a sample belonging to John Doe Incorporated. The sample was received on January 10, 2007 and logged in by the assistant SCO assigned to ID number 30. The -05 means this is the fifth sample in the sample set.

- 2.8 This lab ID number is used by Alpha Analytical for continuous identification of the sample from receipt to completion of analysis. Samples which are received as a fraction or subsequently extracted in the laboratory are identified using the lab ID number and a suffix identifying the sample type, whether its a matrix spike, matrix spike duplicate, etc.

Approved By:



Walter Hinchman  
QA Officer

Date: January, 2007

**1.0 TITLE: LABELING FIELD SAMPLES**

Description: To ensure sample integrity, Alpha uses company labeled tags on samples. This tag is designed to make all entries visible on a white background, and is used to discourage the use of non-waterproof labeling material.

**2.0 STANDARD OPERATING PROCEDURE:**

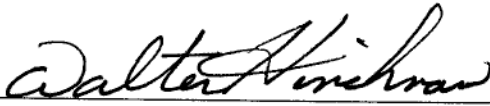
2.1 Alpha provides customized waterproof labels to be used in the field. This is the preferred procedure; however, not all clients use our containers or sampling materials. These labels should be affixed to the sample containers and completed using waterproof ink. If necessary, clear tape can be placed over the label. Alpha encourages the use of these labels to help our staff visually check and verify labels through continuity and standardization. Each label can be completed with the following information:

- a) Analysis requested,
- b) Preservation type,
- c) Sample location,
- d) Clients identification,
- e) Date and time sampled, and
- f) Alpha's sample identification.

2.2 Not all information is essential on the label. However, the following information is strongly recommended:

- Preservation Type,
- Client's Identification, and,
- Date and Time sampled.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

Alpha Analytical, Inc.  
Section No.: C.3  
Revision No.: 15.0  
Date: January, 2007  
Page: 2 of 2



Alpha Analytical, Inc.

255 Glendale, Suite 21  
Sparks, Nevada 89431-5778  
(775) 355-1044

Analysis		Preservatives
8081 _____	8082 _____	HCl _____
TPH Ext. _____	8270 _____	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> _____
TPH Gas _____	8260 _____	H <sub>2</sub> SO <sub>4</sub> _____
BTXE _____	MTBE _____	HNO <sub>3</sub> _____
Other _____	Other _____	Other _____
Client _____		
Time/Date _____		
Sample Description _____		
LAB ID _____		
(lab use only)		

Figure C.3 - 1

**1.0 TITLE: SAMPLE RECEIVING AND PROJECT/CLIENT COMMUNICATION**

**2.0 STANDARD OPERATING PROCEDURE**

2.1 Sample receiving and project/client communication go hand-in-hand and is difficult to separate the two activities since one is associated with the other. The following synopsis discusses, in general terms, the primary activities associated with these procedures. The procedural order may vary somewhat, but this is generally the proper sequential order used to receive in-coming samples.

2.2 Background

2.2.1 Client Information

Specific project information is communicated throughout the laboratory using the "Client Information" menu in the Omega database. Whenever a change/policy is made by the client, it is recorded in this menu. The specific policy changes are entered and the name of the person making the request is recorded. The person typing the requests initials and dates the entry to indicate who typed the change and when the change was made.

2.2.2 Work Order Information

All client contacts for a particular work order are typed in this menu of the Omega database, initialed and dated by the person typing the request. The work order information is then printed and placed in the main file folder. This menu is similar to the Client Information screen; however, this screen is used for a particular work order only and the Client Information screen is used for policies that apply to entire projects.

2.2.3 Sample Receipt Checklist

A sample receipt checklist is automatically created for every work order/chain-of-custody that comes into the laboratory. After logging the samples into the Omega database, the sample receipt checklist is printed and faxed or e-mailed to the client immediately. The checklist identifies any sample integrity issues associated with a particular group of samples or work order and gives the client instructions on communicating with the laboratory how to resolve those abnormalities/nonconformities.

2.3 Sample Receiving Procedure

2.3.1 Alpha's manual chain-of-custody is completed and signed by both the relinquishing and receiving parties.

- 2.3.2 Samples are given a unique sample identification number.
- 2.3.3 Information described on the manual chain-of-custody is entered into the Omega LIM System.
- 2.3.4 Samples are labeled.
- 2.3.5 Documentation of Sample Integrity and Compliance

Upon receipt of the sample, the condition, including any abnormalities or departures from normal or specified conditions in the environmental test method, is recorded on the sample receipt checklist (Fig C.4-1). The client is consulted and the conversations documented by the use of the sample receipt checklist for the following general items:

- When there is doubt as to the suitability of a sample for a particular environmental test,
- When a sample does not conform to the description provided, or
- When the test required is not specified in sufficient detail.

If doubt exist, the sample receipt personnel, consults the client for additional instructions and records those discussions before completing the sample receipt protocols.

- 2.3.6 A master file and method specific colored file folders are created which include a LIMS generated sample report indicating the target analyte list.
- 2.3.7 If the client has specific requirements associated with a work order, the client's comments are typed into the LIMS "Work Order" or "Comments" information section and this is printed out and placed into all associated files.
- 2.3.8 Samples and file folders are disseminated throughout the laboratory.
- 2.3.9 A copy of the chain-of-custody and the sample receipt checklist is faxed or e-mailed as a pdf file to the client.
- 2.3.10 The faxed or e-mailed confirmation report is placed into the master file.
- 2.3.11 If the chain-of-custody and/or sample receipt checklist faxed or e-mailed to the client does not initiate changes to the chain-of-custody within one day, that is confirmation the chain-of-custody and associated information is correct without error.



If the fax or e-mail of the chain-of-custody and/or sample receipt checklist initiates a response, then those phone conversations, amendments to the chain-of-custody or other communications are annotated to the master and associated method files.

## 2.4 Subcontract Laboratories

It is Alpha's policy to subcontract out analytical services not performed by Alpha, to laboratories that have been certified by the appropriate state agencies, methods, and programs to the best of our ability. See QAM, section 3 for a complete listing of Alpha's subcontracting policies.

2.4.1 If samples can not be analyzed in-house, then the sample custodian must determine if the client has specified a subcontract laboratory.

2.4.2 If a subcontract laboratory has not been identified by the client, then the sample custodian must review the subcontract laboratory register and determine which laboratory is most suitable to subcontract the work to.

This determination should take into account such things as; appropriate laboratory certification, methods of analysis, capabilities, is the subcontract laboratory capable of receiving the samples and reporting the data with the specified data quality objectives etc.

1) Clarification: The DoD and Navy requires subcontract laboratory work to be approved by the Navy. Since the Navy and DoD does not produce a listing of certified/approved laboratories, it is of most importance to ensure when subcontracting samples, that the laboratory has the required approval.

2.4.3 During the sample receipt process, samples are identified and segregated to be subcontracted. All sample receipt and COC protocols are followed for subcontracted sample analysis.

2.4.4 After the subcontract laboratory has been chosen, a subcontract chain-of-custody and a subcontract sample receipt checklist (Fig C.4-2) is sent along with the samples.

2.4.5 Once the samples have been received in the subcontract laboratory's facility, a return copy of the sample receipt checklist is requested.

Alpha request this checklist be returned to document and identify any potential abnormalities/nonconformities with the samples and gives the subcontract laboratory instructions on communicating with our SCO if the samples are acceptable or not.

Thus we have confirmation regarding the subcontracted samples' receipt, preservation, holding time and any other potential problems prior to the commencement of sample analysis by the subcontractor.

## 2.5 Service to Clients

The environmental testing business is a service oriented business, requiring a large amount of interaction with our clients. It is in our best interest, to emphasis the importance of conducting client communication in an environment that is professional, informational and confidential.

2.5.1 It is Alpha's policy to cooperate with our clients or their representatives to clarify the client's request and to monitor the analytical performance in relation to the work performed on their project, and to provide this service in a climate that ensures confidentiality to other clients.

2.5.2 Service to clients is a proactive engagement with our clients require staff to notify clients of problem situations such as:

- a) incorrect or improper method requests;
- b) the need to optimize methods to ensure data quality objectives are met for difficult matrix or poor performing analytes;
- c) lack of project guidance; and
- d) problems with sampling or analysis that may impact results (e.g., improper preservation of sample).

## 2.6 Customer Complaints

It is Alpha's policy to respond to complaints and/or problems in a reasonable time frame and in a courtesy manner that is both polite and professional to the customer.

2.6.1 Customer complaints are directed to Alpha's Sample Coordination Supervisor. These complaints are documented in the Customer Complaint logbook. Customer complaint documentation includes information such as:

- a) client name,
- b) date,
- c) complaint,
- d) information on who received the complaint,
- e) a remedy of those complaints, and
- f) initials of the Sample Coordinator or person receiving that complaint.

2.6.2 If complaints can be solved immediately than the remedy for that particular complaint is documented. Conversely, if the complaint can not be solved

Coordinator relinquishes the duty of finding a remedy to the complaint to the Laboratory Manager.

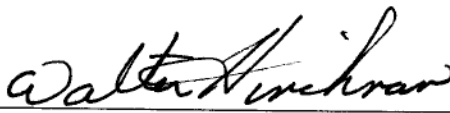
- 2.6.3 The documentation of customer complaints, the response to these complaints, and their resolution is useful information to improving the quality of our client service. This information, as part of our quality system, helps identify patterns of problems and is important in formulating a corrective response to those problems.

2.7 Document Confidentiality

- 2.7.1 All sample documents to include phone conversations, Electronic Data Deliverables, Faxes or Work Order information is strictly confidential and will only be released to the client or Principal Investigator (PI) who originally requested the sample analysis.

Persons or organizations requesting such information may only receive the information upon approval to release the data. If there are any doubts concerning the identity of the organization or authority, then they must show proof of identification before releasing information. This is documented by the SCO, see Fig C.4-3. A full description of our policy concerning physical security and document confidentiality is found in section 7.4 of the QA Manual.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

# Alpha Analytical, Inc.

Phone : (775) 355-1044 FAX : (775) 355-0406

## Sample Receipt Checklist

Date Report is due to Client : 11/29/2006

Date of Notice : 11/21/2006 3:29:25

Please take note of any NO check marks. If we receive no response concerning these items within 24 hours of the date of this notice, all of the samples will be analyzed as requested.

Client Name: **McGinley & Associates, Inc.**

Project ID : **CVO001**

Project Manager: **Tracy Johnston**

Client's EMail: **tjohnston@mcgin.com**

Work Order Number: **MGA06112147**

Client's Phone: **(775) 829-2245**

Date Received: **11/21/2006**

Client's FAX: **(775) 829-2213**

Received by: **Latricia Edrosa**

### Chain of Custody (COC) Information

Carrier name Client

Chain of custody present ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Custody seals intact on shipping container/cooler ?	Yes <input type="checkbox"/>	<input type="checkbox"/> No	Not Present <input checked="" type="checkbox"/>
Custody seals intact on sample bottles ?	Yes <input type="checkbox"/>	<input type="checkbox"/> No	Not Present <input checked="" type="checkbox"/>
Chain of custody signed when relinquished and received ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Chain of custody agrees with sample labels ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Sample ID noted by Client on COC ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Date and time of collection noted by Client on COC ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Samplers's name noted on COC ?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Internal Chain of Custody (COC) requested ?	Yes <input type="checkbox"/>	<input checked="" type="checkbox"/> No	
Sub Contract Lab Used :	None <input checked="" type="checkbox"/>	<input type="checkbox"/> See Comments	

### Sample Receipt Information

Shipping container/cooler in good condition?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	Not Present <input type="checkbox"/>
Samples in proper container/bottle?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Sample containers intact?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Sufficient sample volume for indicated test?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	

### Sample Preservation and Hold Time (HT) Information

All samples received within holding time?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
Container/Temp Blank temperature in compliance (0-6°C)?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	Cooler Temperature 4 °C
Water - VOA vials have zero headspace / no bubbles?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	No VOA vials submitted <input type="checkbox"/>
Sample labels checked for correct preservation?	Yes <input checked="" type="checkbox"/>	<input type="checkbox"/> No	
TOC Water - pH acceptable upon receipt (H2SO4 pH<2)?	Yes <input type="checkbox"/>	<input type="checkbox"/> No	N/A <input checked="" type="checkbox"/>

### Analytical Requirement Information

Are non-Standard or Modified methods requested ?	Yes <input type="checkbox"/>	<input checked="" type="checkbox"/> No	
Are there client specific Project requirements ?	Yes <input type="checkbox"/>	<input checked="" type="checkbox"/> No	If YES : see the Chain of Custody (COC)

Comments :

# Alpha Analytical, Inc.

Phone : (775) 355-1044 FAX : (775) 355-0406

## Subcontract Sample Receipt Checklist

Date Report is due at Alpha : 01-Dec-06

Date of Notice : 11/21/2006 3:31:09

If any items are checkmarked NO or are non-compliant, a phone call back to Alpha Analytical is required immediately. If all items are acceptable, a faxed copy of the signed sub chain of custody (COC) and the completed sample receipt check list is required within 24 hours of sample receipt.

Alpha's Work Order Number : GGI06112142

SubContract Work Order Number :

Date Received :

### Chain of Custody (COC) Information

Carrier name \_\_\_\_\_

Chain of custody present ?

Yes ☐

☐ No

Custody seals intact on shipping container/cooler ?

Yes ☐

☐ No

Not Presen ☐

Custody seals intact on sample bottles ?

Yes ☐

☐ No

Not Presen ☐

Chain of custody signed when relinquished and received ?

Yes ☐

☐ No

Chain of custody agrees with sample labels ?

Yes ☐

☐ Non-Compliant

Internal Chain of Custody (COC) requested ?

Yes ☐

☐ No

### Sample Receipt Information

Shipping container/cooler in good condition?

Yes ☐

☐ No

Not Presen ☐

Samples in proper container/bottle?

Yes ☐

☐ Non-Compliant

Sample containers intact?

Yes ☐

☐ No

Sufficient sample volume for indicated test?

Yes ☐

☐ No

### Sample Preservation and Hold Time (HT) Information

All samples received within holding time?

Yes ☐

☐ Non-Compliant

Cooler Temperature : Is Wet Ice present in Cooler ?

Yes ☐

If YES, then temperature is 4°C.

No ☐

If NO, then actual cooler temperature is : \_\_\_\_\_ °C

### Analytical Requirement Information

Are non-Standard or Modified methods requested ?

Yes ☐

☐ No

SubContract Lab NV STATE certified?

Yes ☐

☐ No

SubContract Lab NELAP certified?

Yes ☐

☐ No

SubContract Lab CERTIFIED for the various methods requested

Yes ☐

☐ No

Will the SubContract Lab be able to meet the turn-around time (TAT) requirements ?

Yes ☐

☐ No

Comments :

**REPORT RELEASE REQUEST**

**To:** \_\_\_\_\_ **Fax:** \_\_\_\_\_

**From:** \_\_\_\_\_ **Date:** \_\_\_\_\_

**RE:** \_\_\_\_\_

**Please respond**

**Work Order Number:**

\_\_\_\_\_ is requesting the above listed report/s. Due to client confidentiality we are unable to release these report/s without your written permission. Please sign below and fax back to us at (775) 355-0406 if we are allowed to fax the above named report/s requested.

Thank you,  
Alpha Analytical

\_\_\_\_\_  
your signature

**1.0 TITLE: SAMPLE CONTAINERS, PRESERVATION, HOLDING TIMES AND GENERAL SAMPLE RECEIPT PROTOCOLS**

Description: Once samples are received it is important to verify and document the sample has maintained its integrity (condition) and is in compliance with the requested test method. This SOP is used in conjunction with the Sample Receipt Checklist to document sample integrity.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Introduction**

Sample integrity issues are most often written into the methods of analysis. This would include such issues as sample collection, preservation and holding time. These items may be critical to the final data results and are important factors that must be addressed by the field sampler and verified in the laboratory.

For instance, if samples are collected and thermally cooled to 4°C in the field, but had an inadequate amount of wet ice added and are received at the laboratory without ice and are warm; then sample integrity has been lost and the sample is noncompliant with the particular test method as data results may now be compromised.

**2.2 Documentation of Sample Integrity and Method Compliance**

Upon receipt of the sample, the condition, including any abnormalities or departures from normal or specified conditions in the environmental test method, is thoroughly checked and documented on the sample receipt checklist.

Sample containers, preservation, holding times, field and laboratory generated COC's, etc. are all reviewed and checked to verify sampling integrity and method compliance.

**2.3 Corrective Actions for Non Compliant Samples**

2.3.1 If the sample does not meet the sample receipt acceptance criteria, the SCO must retain any correspondence records of conversations concerning the final disposition of the rejected sample; or document the decision to proceed with the analysis of samples not meeting the acceptance criteria.

- i. The conditions of these abnormalities must be noted on the COC and/or on the sample receipt checklist.
- ii. The sample analysis is appropriately footnoted on the final report.

2.3.2 If sample information errors are discovered, then these errors or discrepancies are recorded on the sample receipt checklist.

2.3.3 The client is faxed a copy or an e-mail of the completed COC and the sample receipt checklist. Therefore, an unequivocal accurate record, which documents all laboratory activities is produced and maintained.

## 2.4 Sample Receipt Review Items to Verify Sample Integrity and Method Compliance

### 2.4.1 Chain of Custody Information

The following items are reviewed by the SCO to verify the specific COC items are correct and documented.

2.4.1.1 SCO should review and verify the carrier name and ensure accompanying transportation documents are retained.

2.4.1.2 SCO should review and verify the field completed COC is present.

2.4.1.3 SCO should review and verify if custody seals are used and intact on the shipping container.

2.4.1.4 SCO should review and verify if custody seals are used and intact on individual sample bottles.

2.4.1.5 SCO should review and verify if COC was signed by the relinquishing party and signed by Alpha, the receiver.

2.4.1.6 SCO should review and verify if the COC agrees with the sample labels.

2.4.1.7 SCO should review and verify if the date and time of collection is noted by the client on the COC.

2.4.1.8 SCO should review and verify if an internal or evidentiary COC is requested or required.

2.4.1.9 SCO should review and verify if a subcontract laboratory is required and if so was a subcontract laboratory specified.

### 2.4.2 Sample Containers

Sample containers are normally borosilicate glass or high density polyethylene (HDPE). Sample containers must be clean and free of the target analytes of interest.



#### 2.4.2.1 General Information

- 2.4.2.1.1 All sample containers used by Alpha are pre-cleaned according to EPA established protocols and require no further cleaning prior to sample collection.
- 2.4.2.1.2 Alpha does not clean or reuse sample containers.
- 2.4.2.1.3 Alpha maintains a sequestered supply of sample containers to eliminate the possibility of contamination of the sample from the container and to track sample containers by lot numbers.
- 2.4.2.1.4 Containers are purchased and sequestered by lot in order to minimize the tracking associated with a group of sample containers.
- 2.4.2.1.5 VOC containers are checked by Alpha to insure that the container is not contaminated and is not adding low level analytes to the sample.

2.4.2.2 SCO should review and verify that the cooler containing the individual samples are in good condition.

2.4.2.3 SCO should review and verify sample container type and size are appropriate for the requested method of analysis.

2.4.2.4 SCO should review and verify that the individual sample containers are intact and that the COC completed in the field matches the sample containers for number of containers, types, field identification etc.

2.4.2.5 SCO should review and verify that the number of sample containers or sample volume is appropriate for the requested method of analysis.

#### 2.4.3 Thermal Preservation

Sample temperature can vary greatly from vial to vial or sample to sample within the same ice chest. Therefore, it is important to obtain an accurate temperature measurement.

##### 2.4.3.1 General Information

- 2.4.3.1.1 When applicable, temperature measurements are taken through the use of temperature blanks or by using an IR gun. If the IR gun is used, then the

temperature of 3 samples are taken and their average is recorded.

2.4.3.2 SCO should review and verify that the samples are received at the correct temperature.

The following description details temperature verification procedures.

2.4.3.2.1 Samples which require thermal preservation are considered acceptable if the arrival temperature is either within 2°C of the required temperature or the method specified range.

2.4.3.2.2 For samples with a specified temperature of 4°C, samples with a temperature range from just above freezing temperature of water to 6°C is acceptable.

2.4.3.2.3 If samples are cooled with wet ice and any remaining ice is left, then the temperature recorded is 4°C.

Note: The use of blue ice or reusable cold packs is not an acceptable alternative to wet ice.

2.4.3.2.4 Samples that are hand delivered to the laboratory on the same day they are collected may not meet these criteria. In these cases, the samples are considered acceptable if there is evidence that the chilling process has begun such as arrival on ice.

#### 2.4.4 Chemical Preservation

Preservatives are used to maintain the chemical integrity of the sample. Most solid samples have only cooling as a preservative. Water samples are subject to a variety of specific preservation techniques, depending on the target analytes. Samples requiring chemical preservation means that the pH or removal of residual chlorine of the sample should be checked in the field and verified in the laboratory.

##### 2.4.4.1 General Information

2.4.4.1.1 Chemical preservation can be generally divided into two basic categories:

- a) acids and bases added to control pH and minimize or stop microbial degradation; and

- b) dechlorination reagents, such as ascorbic acid or sodium thiosulfate, added to reduce the effect of residual chlorine or other oxidizers found in some samples.

#### 2.4.4.2 Techniques for Verifying Sample Preservation

Samples that are preserved in the field are also checked at the laboratory. Essentially all samples are checked and verified prior to sample analysis or extraction/digestion. There are separate techniques for verify this without contaminating the samples. For instance:

- 2.4.4.2.1 There are multiple vials collected for VOC analysis. The first vial is used to verify and record the sample pH with wide range (0-14) pH paper. Since this vial is not used for finals sample analysis, the pH strip can be dipped directly into the sample vial.
- 2.4.4.2.2 Methods which require a sample extraction have the pre-extraction sample pH checked using wide range (0-14) pH paper and recorded on the sample preparation logs.
- 2.4.4.2.3 TOC samples are the only universal exception to checking and verify sample preservation at the lab bench. Sample preservation for TOC is verified using wide range (0-14) pH paper at the time of sample receipt.

#### 2.4.4.3 Prevention of Sample Contamination when Verifying Preservatives

Concern over possible contamination of the sample resulting from dipping a probe or a test strip into the sample suggests that a slight modification in how samples are obtained may be in order.

- 2.4.4.3.1 The most common suggestion to field samplers is to take an identical sample, and to use that sample to determine the preservation requirements. For example, if it is found that 2.3 ml of nitric acid is required to lower the sample to a pH <2, then the addition of the same amount of acid to the other container of sample will achieve the correct preservation, without contaminating the sample to be sent to the laboratory.

- 2.4.4.3.2 When only a limited number of sample bottles are taken, and the sample pH needs to be verified with pH paper, a disposable pipette or stir rod is used to take a small aliquot of sample. This small sample aliquot is used to place a drip onto the pH paper. The pH paper is not dipped into the sample container in an effort to limit the possibility of contamination by the pH paper.

2.4.4.4 SCO should review and verify that the samples are received with the correct preservation.

## 2.4.5 VOC Methods Requiring No Head Space

Aqueous VOC samples should be collected in 40 mL (nominally 43 mL) VOA vials in triplicate.

### 2.4.5.1 General Information

- 2.4.5.1.1 Method protocols require the sample vials be completely filled such that they form a meniscus at the top of the vial taking care not to flush out the preserving agents.

- 2.4.5.1.2 Sample vials should not contain any bubbles exceeding 5 - 6mm (pea sized) as they may cause significant degassing and loss of volatiles. Conversely, sample vials containing bubbles smaller than 5-6 mm in size (total head-space) is acceptable.

2.4.5.2 SCO should review and verify that the VOC sample vials were received with zero head space.

## 2.4.6 Sample Holding Time

The holding time before analysis is of critical practical and regulatory importance. Analytes will degrade and be lost from the sample over time, even when correctly preserved and stored. All analytes have required holding times, from immediate analysis for dissolved oxygen up to 6 months for heavy metals.

### 2.4.6.1 General Information

- 2.4.6.1.1 The holding time clock starts with the moment of sampling and ends with the beginning of the

analytical procedure.

In other words, holding times do not start from receipt of the sample at the laboratory.

2.4.6.1.2 Holding times were originally conceived with the idea of providing guidelines for performing analyses within a sensible time frame to minimize sample degradation. Now holding times have taken on a legal life of their own, often times with no connection to scientific reality.

For instance the misconception of how a sample with a holding time of 7 days prior to extraction, is acceptable if the extraction is begun 6 days, 23 hours, 59 minutes and 59 seconds after the moment of sampling, but magically turns to garbage one second later if not in a separatory funnel!

2.4.6.2 SCO should review and verify that the sample were received within the method specified holding time.

2.4.6.3 SCO should review and verify that the samples have a sufficient amount of holding time left to conduct the sample analysis.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: SAMPLE ACCEPTANCE POLICY**

Description: The sample acceptance policy is an SOP which outlines the circumstances under which samples are accepted or rejected.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 Data from any samples which do not meet the following criteria is flagged clearly defining the nature and substance of the variation.

This policy is an important centerpiece for correctly recording the sample receipt checklist and COC documents.

2.2 Chain of Custody

2.2.1 The COC must be properly, fully and completely documented, which includes:

- i. sample identification,
- ii. sample or project location, i.e. state sample was taken,
- iii. date and time of sample collection,
- iv. collectors name,
- v. preservation type,
- vi. sample type, and
- vii. any special remarks concerning the sample.

2.2.2 See the following documents for additional details:

- QAM, Vol II, Appendix C, SOP C.7, Manual COC Procedures,
- QAM, Vol II, Appendix C, SOP C.8, LIMS Generated COC Procedures,
- QAM, Vol II, Appendix C, SOP C.9, Internal COC Procedures.

2.3 Sample Labeling

2.3.1 Sample labels are completed using Alpha's sample identification scheme to uniquely identify and label all sample containers.

2.3.2 Sample labels used in the field should be water resistant and field samplers are encouraged to use indelible ink.

2.3.3 See the following documents for additional details:

- QAM, Vol II, Appendix C, SOP, C.2, Sample Identification, and
- QAM, Vol II, Appendix C, SOP, C.3, Labeling Field Samples.

2.4 Sample Containers - See the following documents for details:

- QAM, Vol I, Section 6.4 and Table 6-1 through Table 6-8.
- QAM, Vol, II, Appendix C, SOP C.5, Sample Containers, Preservation, Holding Time and General Sample Receipt Protocols.

2.5 Sample Holding Times - See the following documents for details:

- QAM, Vol I, Section 6.6 and Table 6-1 through Table 6-8.
- QAM, Vol, II, Appendix C, SOP C.5, Sample Containers, Preservation, Holding Time and General Sample Receipt Protocols.

2.6 Sample Preservation - See the following documents for details

- QAM, Vol I, Sections 6.3 and 6.5, and Table 6-1 through 6-8.
- QAM, Vol, II, Appendix C, SOP C.5, Sample Containers, Preservation, Holding Time and General Sample Receipt Protocols.

2.7 Adequate Sample Volume - See the following documents for details:

- QAM, Vol I, Section 6.7 through 6.8 and Table 6-1 through Table 6-8.

1) Clarification: Since Alpha does not take field samples, all sampling supplies are furnished by Alpha to meet the sample acceptance policy criteria.

2.8 New Work

All new work (i.e. new clients) is approved for sample acceptance by either the Laboratory Director or Laboratory Manager.

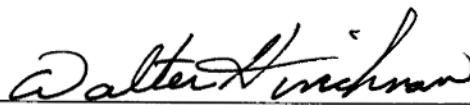
This is a case-by-case approach and is not entirely a QA/QC issue. Items that affect the acceptance of new work include the following:

- a) is the work in a state or under an agency which the laboratory has approval,
- b) are the requested methods of analysis, methods we are approved for,
- c) are the target analytes, compounds we are approved for and analyze on a normal basis,
- d) what are the reporting limits or LOQs requirements,
- e) what are the QC requirements,
- f) what are the turn around requirements,

- f) what are the turn around requirements,
- g) what is the sample matrix,
- h) what are the number of samples to be analyzed, and over what time period will they arrive,
- i) do we have the appropriate facilities and resources before commencing the work, and
- j) can these samples be analyzed without overly stressing the current staff.

This is not a comprehensive list of questions to be determined, but is a general starting point for all new projects. In addition, these do not answer the business questions regarding payment, pricing etc which are always a factor when evaluating new work.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007



## **1.0 TITLE: MANUAL CHAIN-OF-CUSTODY PROCEDURES**

Description: The chain-of-custody can be regarded as a legal document in some situations and should be completely filled out and as error free as possible. All samples received by Alpha are entered into our Sample Tracking System to enhance the legal defensability of all data produced at Alpha. Samples are documented on a chain-of-custody form and signed by both the client and laboratory. This document formalizes the sample transaction and is critical to the maintenance of sample custody.

## **2.0 STANDARD OPERATING PROCEDURE**

2.1 The Sample Custodian handles and/or processes samples dropped off at the laboratory or sample shipments, including pickup of samples at Reno-Tahoe International Airport, bus station, Federal Express, UPS, or other carrier service within Alpha Analytical's geographic area. The Sample Custodian is available to receive sample shipments at any time the delivery service is operating, including weekends.

2.1.1 Alpha's manual Chain-of-Custody record is completed with the following information:

- a) Client name, address, phone/FAX number,
- b) Who the report should be addressed to,
- c) Sampler,
- d) Date Sampled,
- e) Time of collection,
- f) Matrix,
- g) Lab identification number,
- h) Client identification/sample description,
- i) Number of containers,
- j) Analysis requested, and
- k) All necessary signatures/dates/times.
- l) Billing Information
- m) PO #, PWS #, Job # or any additional information

2.1.2 Examine the condition of the sample and note in the comments section:

- a) The integrity of the sample container,
- b) The amount of sample for the analysis,
- c) The presence of air bubbles in VOA vials,
- d) Whether the sample was preserved according to the method prescribed preservation, and
- e) Note any other irregularities with sample condition.

- 2.1.3 Compare all documents (client vs. laboratory) to verify information. In the event errors are discovered, record the discrepancies in the remarks section.
- 2.1.4 Issue the yellow copy to the client, and retain the white copy as the laboratory's copy.
- 2.1.5 All entries made on this document must be filled out in ink.
- 2.1.6 Samples received in the absence of the Sample Custodian or other designated recipients are placed in their appropriate refrigerator until the correct personnel are available to process the samples according to established protocol.
- 2.1.7 An example of Alpha's chain-of-custody is in the following Figure C.7-1.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007



**1.0 TITLE: LIMS GENERATED CHAIN-OF-CUSTODY PROCEDURES**

Description: Once samples have been manually logged-in to the laboratory they are also entered into the LIM system. Essentially, the same information used in the original manual chain-of-custody is used to complete the log-in procedures for the LIM system.

**2.0 STANDARD OPERATING PROCEDURE:**

2.1 The Omega system is entered from the main computer menu by double clicking the Omega icon. When the log-in prompt appears, type "System" and click the "O.K." button. This procedure will take you to the main Omega IV menu. From this menu, double click "Work Orders."

2.2 To add a sample, double click the "Add" key in the lower left-hand corner of the screen and amend the order numbers to correspond with the correct sequence:

Lead SCO: 1-19  
1<sup>st</sup> Assistant SCO: 20-39  
2<sup>nd</sup> Assistant SCO: 40-59  
3<sup>rd</sup> Assistant SCO: 60-79  
4<sup>th</sup> Assistant SCO: 80-89  
5<sup>th</sup> Assistant SCO: 90-99

**2.3 Sample Log-in Procedure**

2.3.1 Alpha's LIMS generated Chain-of-Custody record is completed with the following information:

- a) Client name, address, phone/FAX number;
- b) Who the report should be addressed to;
- c) Sampler;
- d) Date Sampled;
- e) Time of collection;
- f) Matrix;
- g) Lab identification number;
- h) Client identification/sample description;
- i) Number of containers;
- j) Analysis requested; and
- k) All necessary signatures/dates/times.
- l) Billing Information
- m) PO #, PWS #, DWR #, Job # or any additional information

2.3.2 Click once in the "Client Sample ID" box and enter the clients' three letter code. The company name, point-of-contact, address, phone and fax numbers

will automatically appear.

- 2.3.3 The sample Turn-Around-Time (TAT) is automatically indexed to 10 days for all samples. If a TAT is shorter than 10 days is required, amend the TAT box as needed.
- 2.3.4 If a project name is required click the “Order Name” box and type the project name.
- 2.3.5 Double click the “Received” box and a calender will appear with the current date. Double click the date and the system will automatically enter the date. Enter the sample state from the pull-down menu.
- 2.3.6 Click on the “Login” key and enter the sample ID for the first sample.
- 2.3.7 Double click the “Date Due” box and the system will count ahead the number of days selected in the TAT box (excluding weekends) and automatically enter the information.
- 2.3.8 Open the “Collection Date” box and enter the appropriate information. This information should be written on the sample label - otherwise, ask the client.
- 2.3.9 If a sample time is provided, double space after the date sampled and enter the correct information converted to military time.
- 2.3.10 Open the sample “Matrix” box and enter the appropriate information. This information may be typed by hand or entered by clicking the down arrow key adjacent to the matrix box, and selecting the proper matrix.
- 2.3.11 Enter the number of sample containers currently being logged-in. There is a tab labeled “Bottle Information”. Enter the type and number of bottles provided for each sample.
- 2.3.12 Click once in the “Test Groups” box. Enter the appropriate test and click the down arrow key next to the box to view the list of possible choices. Tests are segregated by state and test method.
- 2.3.13 Trip Blanks
  - 2.3.13.1 Many sample coolers will have associated Trip Blanks. Trip Blanks are treated as separate samples.
  - 2.3.13.2 For “Client Sample ID” type in Trip Blank.
  - 2.3.13.3 Enter a TAT that corresponds to the sample being logged-in.

- 2.3.13.3 Enter a TAT that corresponds to the sample being logged-in.
- 2.3.13.4 Enter the collection date which corresponds to the earliest sample date on the chain-of-custody.
- Trip Blank collection dates are entered in this fashion to prevent the LIM System from creating an artificial holding time problem. The true Trip Blank preparation date should be listed on the vial label.
- 2.3.13.5 TB original preparation date information and the place of origin are then entered into the comments section.
- 2.3.13.6 Trip Blanks are normally not analyzed unless requested by the client. Therefore, under "Test Groups" type "Hold". This will tell the analyst that the sample should not be analyzed.
- 2.3.14 For each new sample, click the "Add Sample" box and repeat the sample log-in procedure starting with a new client ID.
- 2.3.15 Add any additional comments in the "Comments" box on the "Main" Screen by clicking on the box and typing the relevant information. Information such as rush TAT, California samples, special QA/QC, etc. are entered into this section.
- 2.3.16 If a client has a Purchase Order (PO) number, click the "Invoice Info" box at the top of the screen and enter the PO number.
- 2.3.17 Print the chain-of-custody by clicking the "WO COC" button at the top of the screen.
- 2.3.18 The analyte list then needs to be generated. Click on "Print Test" and put in the sample number to be printed. Staple the analyte list on the inside cover of the appropriate test folder.
- 2.3.19 Laboratory ID labels are then generated for the sample containers. Click on the "Labels" button, enter the number of bottles present for each sample and print the labels.
- 2.3.20 When all relevant information has been entered, click the "out-the-door" box at the lower right hand corner of the screen and exit.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January 2007

## CHAIN-OF-CUSTODY RECORD

NV

## Alpha Analytical, Inc.

255 Glendale Avenue, Suite 21 Sparks, Nevada 89431-5778

TEL: (775) 355-1044 FAX: (775) 355-0406

## Client:

McGinley & Associates, Inc.  
425 Maestro Drive  
Suite 202  
Reno, NV 89511

Tracy Johnston

TEL: (775) 829-2245 x  
FAX: (775) 829-2213  
Email tjohnston@mcgin.com

WorkOrder : MGA06112147

Report Due By : 5:00 PM On : 29-Nov-06

EDD Required : Yes

Sampled by : Max

Report Attention : Tracy Johnston

Job : CVO001

PO :

Client's COC # : 16425

Cooler Temp

4 °C

Samples Received

21-Nov-06

Date Printed

21-Nov-06

QC Level : 1 = Final Rpt Only

Alpha Sample ID	Client Sample ID	Matrix	Collection Date	No. of Bottles			Requested Tests				Sample Remarks
				ORG	SUB	TAT	PWS #	VOC_W	BTXEM_N	BTXEM_N	
MGA06112147-01A	CVO001-MW11-WS3	AQ	11/16/06 12:50	3	0	5					
MGA06112147-02A	CVO001-MW6-WS10	AQ	11/16/06 13:22	3	0	5					
MGA06112147-03A	CVO001-MW7-WS11	AQ	11/16/06 14:00	3	0	5					

Comments: Samples brought in by client. Frozen ice.

Signature	Print Name	Company	Date/Time
		Alpha Analytical, Inc.	

Logged in by:

NOTE: Samples are discarded 60 days after results are reported unless other arrangements are made. Hazardous samples will be returned to client or disposed of at client expense.

The report for the analysis of the above samples is applicable only to those samples received by the laboratory with this COC. The liability of the laboratory is limited to the amount paid for the report.

Matrix Type : AQ(Aqueous) AR(Air) SO(Soil) WS(Waste) DW(Drinking Water) OT(Other) Bottle Type: L-Liter V-Voa S-Soil Jar O-Orbo T-Tedlar B-Brass P-Plastic OT-Other

**1.0 TITLE: INTERNAL CHAIN-OF-CUSTODY PROCEDURES**

Description: Occasionally the use of internal COC protocols may be required by some clients and/or programs. This SOP is designed to supplement the record keeping system as discussed in various sections of the QAM.

**2.0 Standard Operating Procedure**

**2.1 Internal COC Client Management**

It must be stated and understood in the strongest terms that the use of this SOP will be implemented only when specifically requested by the client. This request must be documented on the Sample Receipt Checklist and acknowledged by the client. If the client does not request or acknowledge that this requirement is needed, then the normal record keeping system has precedence and is the system of choice.

1) Clarification: A complete evidentiary COC begins at sample collection, unless otherwise specified by the client, and ends after laboratory analysis of the sample is completed and the sample is ready for disposal. Samples are disposed of in accordance with the normal sample disposal procedures.

**2.2 Internal Chain-of-Custody**

The internal C-O-C record is designed to produce an intact, continuous record of the physical possession, storage and disposal of samples, sample aliquots and extracts. For ease of use, samples, sample aliquots and extracts are referred to as samples.

**2.2.1 A sample is in someone's custody if:**

- a) it is in one's actual physical possession;
- b) it is in one's view, after being in one's physical possession;
- c) it is one's physical possession and then locked up so that no one can tamper with it; or
- d) it is kept in a secured area, restricted to authorized personnel only.

**2.2.2 The internal COC record is filled out and should account for all time periods associated with the sample. See Fig C.9-1 and Fig C.9-2.**

**2.2.3 The internal COC is also initialed and dated each time an individual physically handles the sample.**

**2.2.4 Clients usually provide water samples for each analysis (TPH and VOCs) in**



triplicate and occasionally semi-volatile analysis will have more than one bottle. Therefore, when any of the vial/bottles are consumed, the analyst or extraction technician must document the consumption of that vial/bottle (e.g. vial 1/3 consumed) on the internal COC. The vial consumption documentation must also include the date of consumption and the initials of the person that used that vial.

- 2.2.5 The internal COC contains a “master work order sheet” which is kept in a master logbook. This sheet provides information on when the samples were moved into the primary refrigerator, secondary storage, and the date that the entire sample set was disposed. The same information (except when samples were moved into the primary refrigerator) is recorded for the sample extracts. (See figure on following page).

This master work order sheet along with the internal COC and the client COC details the custody trail of the samples from the sampling event to final disposal.

- 2.2.6 The internal COC includes a sign-off sheet for each different test. These sheets are stapled to the inside rear cover of each file folder. The analyst/extraction technician that handles the sample must maintain internal custody by initialing and dating the COC when logging samples in and out of the refrigerator.

## 2.3 Evidentiary Chain-of-Custody Procedures

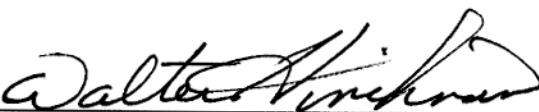
- 2.3.1 The COC procedures as described in Section C.7 and C.8 are followed completely and error free as possible.
- 2.3.2 It is important not to break the continual record of physical possession of the samples. Therefore, it is critical to ensure the sampler has signed off or relinquished the samples to either a second party or to the laboratory.
- 2.3.3 If samples are mailed, then they should be registered with return receipt requested. If samples are sent by common courier, receipts should be retained as part of the permanent COC documentation.
- 2.3.4 Once samples are received by Alpha, then the responsibilities for the care and custody of the samples are in our hands.

## 2.4 Sample Disposal

- 2.4.1 Disposal of the sample will occur 60 days after sample receipt as outlined in Section E.10.
- 2.4.2 All conditions of disposal and correspondence between parties concerning

- 2.4.2 All conditions of disposal and correspondence between parties concerning the final disposition of the sample is recorded and retained as part of the permanent record.
- 2.4.3 Internal COC records indicate the date of disposal, the nature of disposal (i.e. sample depleted, sample disposed of in hazardous waste facility or sample returned to client) and the name of the person who performed the task.
- 2.4.4 All other disposal practices are described in Section E.10.

Approved By:



Walter Hinchman  
QA Officer

Date: January, 2007

# Alpha Analytical, Inc.

Refrigerator # SAR-14A

## DOD Sample Refrigerator Login / Logout Sheet

Alpha's Sample ID	Client's Sample ID	Date		Matrix	Requested Test	Date / Time / Initials		Date / Time / Initials	
		Sampled	Received			OUT	IN	OUT	IN
BMI06112022-01A	1657-SB-5-5	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-02A	1657-SB-5-10	11/17/2006	11/17/2006	Soil	TPH/E_S				
BMI06112022-03A	1657-SB-5-15	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-04A	1657-SB-5-20	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-05A	1657-SB-5-25	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-06A	1657-SB-5-30	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-07A	1657-SB-5-35	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-08A	1657-SB-6-5	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-09A	1657-SB-6-10	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-10A	1657-SB-6-15	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-11A	1657-SB-6-20	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-12A	1657-SB-6-25	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-13A	1657-SB-6-30	11/17/2006	11/20/2006	Soil	TPH/E_S				
BMI06112022-14A	1657-SB-6-35	11/17/2006	11/20/2006	Soil	TPH/E_S				

COMMENTS :

# Alpha Analytical, Inc.

WorkOrder : BMI06112022

Refrigerator # SAR-14A

## DOD Sample Refrigerator Login / Storage / Disposal Sheet

<u>Date Samples Moved Into Primary Refrigerator</u>	<u>Date Samples Moved Into Secondary Storage</u>	<u>Date Entire Sample Set was Disposed</u>

COMMENTS :

## DOD Extract Refrigerator Storage / Disposal Sheet

<u>Date Samples Moved Into Secondary Storage</u>	<u>Date Entire Sample Set was Disposed</u>

COMMENTS :

**1.0 TITLE: SAMPLE LOG-IN LEDGER**

Description: Upon arrival at the laboratory, the Sample Custodian completes a LIMS generated COC. From the LIMS generated chain-of-custody a macro routine is used to parse information to be placed in a 3-ring binder called the Sample Log-In Ledger. This is a notebook used to keep a running total of all samples received by Alpha. This ledger is used in conjunction with other Log-in documents to rectify sample receipt problems.

**2.0 STANDARD OPERATING PROCEDURE:**

**2.1 NELAC Requirements**

2.1.1 NELAC requires the Sample Log-In Ledger to record the following minimum information:

- i. client name,
- ii. date and time of sample receipt,
- iii. laboratory sample identification, and
- iv. signature or initials of the person making the entries.

2.1.2 NELAC also requires the following additional information to be unequivocally linked between the COC and Sample Log-In Ledger.

- i. The client/field identification on each container is linked to the laboratory identification.
- ii. The date and time of sample collection is linked to the sample container and to the date and time of sample receipt.
- iii. The requested analyses is linked to the laboratory identification.
- iv. Any comments resulting from inspection for sample rejection is linked to the laboratory identification.

2.2 Since the Sample Log-In Ledger is parsed from the LIMS generated COC essentially any information contained on the COC can be unequivocally linked to COC log in records.

2.2.1 All COC and sample log-in records/information are entered into our LIM system and stored as permanent archived records. These records are easily retrieved upon request and readily available to all staff members who will process the samples.

2.2.2 The Sample Log-In Ledger records the following information:

- a) initials,
- b) date of sample receipt,
- c) laboratory's sample identification,
- d) client's sample identification,
- e) matrix type,
- f) analysis requested,
- g) Turn-Around-Time (TAT),
- h) date sampled,
- i) temperature upon sample receipt,
- j) work order comments, and
- k) any additional comments.

2.3 Approximately every 2 weeks the SCO or assistant SCO activates the Sample Log-In Ledger macro and initiates a computer parsing routine which identifies the requested information as stated above.

All samples and associated sample information entered into the LIMS, regardless of the person whom actually logged-in the sample, will be printed out in chronological order.

Once this information is printed, the SCO or Assistant SCO checks the information for accuracy, paginates and initials the ledger (Fig C.10 - 1).

Approved By: \_\_\_\_\_

Walter Hinchman  
QA Officer

Date: January, 2007

ERS	09/29	LVF06100201-01A	720 Skyway	AQ	524_DW (524/MTBE/ETBE/TAME), ALCOHOL_W (Low Level MeOH / ETOH), TPH_E_W (TPHE_C), TPH_P_W (GAS-C), VOC_W (BTX/TBA/DIPE_C)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100202-01A	2615 Hegan	AQ	524_DW (524/MTBE/ETBE/TAME), ALCOHOL_W (Low Level MeOH / ETOH), TPH_E_W (TPHE_C), TPH_P_W (GAS-C), VOC_W (BTX/TBA/DIPE_C)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100203-01A	721 Skyway	AQ	524_DW (524/MTBE/ETBE/TAME), ALCOHOL_W (Low Level MeOH / ETOH), TPH_E_W (TPHE_C), TPH_P_W (GAS-C), VOC_W (BTX/TBA/DIPE_C)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100204-01A	SW-1	AQ	524_DW (524/MTBE/ETBE/TAME), ALCOHOL_W (Low Level MeOH / ETOH), TPH_E_W (TPHE_C), TPH_P_W (GAS-C), VOC_W (BTX/TBA/DIPE_C)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100204-02A	FB-1	AQ	HOLD (Hold)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100204-03A	TB-1	AQ	HOLD (Hold)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	09/29	LVF06100204-04A	TB-2	AQ	HOLD (Hold)	10	09/29	(4°C)	Samples brought in by client. Frozen ice. Samples received 9/29/06 kept cold and secure until login on 10/02/06. Samples received after 4:30 pm cutoff time therefore one day added to TAT. Per Roger's phone conversation with Steve Osborne on 3/13/01, report TPH/P with three different numbers: 1) Total 2) Total minus MTBE 3) Total minus all olys.	
ERS	10/02	POR06100205-01A	TBH-1 @ 2 fl.	SO	HOLD (Hold)	10	09/28	(20°C)	Samples brought in by client. No ice. Total Xylenes. No sample times recorded on client COC or sample containers.	
ERS	10/02	POR06100205-02A	TBH-1 @ 5.5	SO	TPHE_S (TPHE_C), TPHIP_S (GAS-C), VOC_S (BTX/OXY_C)	10	09/28	(20°C)	Samples brought in by client. No ice. Total Xylenes. No sample times recorded on client COC or sample containers.	
ERS	10/02	POR06100205-03A	TBH-1 @ 6.5	SO	TPHE_S (TPHE_C), TPHIP_S (GAS-C), VOC_S (BTX/OXY_C)	10	09/28	(20°C)	Samples brought in by client. No ice. Total Xylenes. No sample times recorded on client COC or sample containers.	
ERS	10/02	POR06100205-04A	TBH-2 @ 3.0	SO	TPHE_S (TPHE_C), TPHIP_S (GAS-C), VOC_S (BTX/OXY_C)	10	09/28	(20°C)	Samples brought in by client. No ice. Total Xylenes. No sample times recorded on client COC or sample containers.	
ERS	10/02	POR06100205-05A	TBH-2 @ 5.5	SO	TPHE_S (TPHE_C), TPHIP_S (GAS-C), VOC_S (BTX/OXY_C)	10	09/28	(20°C)	Samples brought in by client. No ice. Total Xylenes. No sample times recorded on client COC or sample containers.	

NO 0 6 2006

DA

1389

## **1.0 TITLE: SAMPLE RECEIPT STORAGE**

Description: Alpha ensures sample integrity through disciplined sample processing and storage. All samples received by the lab are placed in the appropriate storage area immediately after log in. If the sample is removed from the storage area for extraction or analysis, this activity is documented in the technician's extraction or the analyst's instrument logbooks. Therefore, samples can be informally tracked in the laboratory by reconstructing historical data.

## **2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Samples are maintained under frozen or refrigerated conditions as required to guarantee sample integrity.
- 2.2 The samples removed from the shipping container are stored in their original containers unless damaged.
- 2.3 Damaged samples are disposed of in an appropriate manner and documented.
- 2.4 Storage temperatures for all samples are maintained at 4 degrees C for refrigerated samples and -20°C for frozen samples, unless otherwise requested by the client.
- 2.5 Alpha maintains a large inventory of refrigerators and freezers in order to facilitate proper segregation of samples, to minimize cross-contamination. This segregation is based primarily on the following sets of parameters:
  - 1) Class of analytes to be measured;
  - 2) methods of analysis;
  - 3) samples vs. their respective extracts; and
  - 4) segregation of standards.
- 2.6 All samples are placed in cold storage until they have been analyzed, and then held for a minimum of sixty days. The following table identifies the refrigerators and the types of samples that are stored in them.



**TABLE C.11 -1**  
**SAMPLE AND EXTRACT STORAGE AREAS**

Analysis Requested	Matrix	Sample Storage Refrigerator
624 / 8260	Water	SAR- 1B
"Suspect high contamination"	Water/ Soil	SAR- 2C
8270 / 8081 / 8082	Water/Soil	SAR- 3A
8015-DRO (2 <sup>nd</sup> and 3 <sup>rd</sup> vials) / 415.1	Water	SAR- 4B
VOC Trip Blanks	Water	SAR- 5D
Purgeables in Air	Air	SAF - 6B
8015-DRO	Water/Soil	SAR- 7A
624 / 8260	Water	SAR- 8A
Inorganic	Water/Soil	SAR- 9B
524.2	Water	SAR- 10A
VOC	Soils	SAR- 11A
624 / 8260	Water	SAR- 12A
Inorganic	Water/Soil	SAR- 13A
DOD	Water/Soil	SAR- 14A
Incoming samples (sample receipt)	Water/Soil	SAR- 15A
Dissolved Gases	Water/Soil	SAR- 16A
Organic Acids	Water/Soil	SAR- 17A

Note: Refrigerator/Freezer temperature requirements and numbering system can be found in the appropriate SOP's.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

## **1.0 TITLE: SAMPLE TRACKING**

Description: Sample tracking is a fundamental responsibility of the SCO. This person must, at any given time, know where a particular sample or client file is located and have the ability to track the samples analytical progression through the laboratory. The large number of samples in-house requires a procedure for tracking and documenting this activity.

## **2.0 STANDARD OPERATING PROCEDURE**

- 2.1 The SCO must properly fill out a chain-of-custody, identify, label and store all samples upon receipt, according to established SOPs.
- 2.2 Once this activity has taken place, multiple copies of the chain-of-custody are made and placed in various locations.
- 2.3 A master client-file is then generated for the purpose of tracking analytical data produced by the various analyses. Each sample or group of samples from an individual client is given a client file with the client and sample identification written on the folder's tab. Each method of analysis has its own folder and is discriminated between methods by the file folder color. Table C-2 is a listing of file folder colors associated with a particular method of analysis.
- 2.4 Bin World
  - 2.4.1 Bin World is a part of the physical tracking system used in conjunction with the master files to carry out the sample tracking procedure. Bin World is a set of various discrete bins associated with a given day used to segregate primary sample streams. The various discrete bins include:
    - Amended COC,
    - Other lab data only,
    - Alpha + other lab,
    - Alpha only, and
    - ASAP 24 or 48hr rush.
  - 2.4.2 Master files are placed in the Bin World according to the discrete primary sample stream. Secondly, master files are placed in Bin World according to their "due date". This enables client data to be tracked effectively according to due date.
  - 2.4.3 If the analytical data is not produced in a manner reflective of the client requested due date, or if QA/QC data is to be sent at a later date, then the master file is relocated to another set of discrete bins. There are four primary areas which may delay final data production and therefore have an associated bin. They are as follows:

- Delayed Alpha reports,
- Delayed Inorganic reports, and
- Needs QC Alpha / Inorganic.

## 2.5 Master Files

2.5.1 A master file is generated for each group of samples represented by a single chain-of-custody. A manilla file folder is used for this master file. If the client is requesting a single method of analysis, then the master file is generated using the same file folder color as the color used for the requested method of analysis. The master file is labeled with the sample identification number attached to the folders tab. The master file initially contains the following items:

- The manually produced, hand written COC with original signatures,
- The LIMS generated COC,
- If samples were received using courier service, then any information such as air bills, bills-of-landing or other courier information,
- Nonconformance letter / Sample Receipt Checklist,
- Faxed confirmations that the sample submitting party has received, and was notified of, Sample Receipt Checklist/Nonconformance letter,
- Client communications in regards to special requests or changed orders, and
- Any other relevant information.

## 2.6 Analytical Files

2.6.1 Once the analytical file folders are generated they are placed in the extraction or analytical lab depending upon the matrix and analysis requested. This gives the extraction chemist or the analyst two places to check and verify samples information; 1) the client file; and 2) the LIMS System.

2.6.2 The extraction chemist or analyst will then perform the required work and place the relevant data in the client folder. If the data can be uploaded, it is then uploaded into the LIM System. This will indicate in the LIM System that the work has been completed. The sample extract and/or original sample will be placed back in storage after analysis for a minimum of sixty days.

2.6.3 Once the analysis has been completed and the data calculated, the client file is reviewed by several staff members. This is a rigorous review by four staff members for the purpose of reducing, evaluating and reporting the analytical data generated. This procedure is described in the Data Reductions Validation and Reporting sections of the QA plan.

- 2.6.4 Once the analytical report has been signed, the method files and all of their associated analytical data is then collected into the master file.
- 2.6.5 The DCO and SCO both monitor the master file to determine and physically track the status of analytical data.
- 2.6.6 This procedure continues until all data has been produced and physically collected in the master file.
- 2.6.7 Final analytical data is then faxed and/or sent to the client.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

Table C.12-1 File Folder Tracking Table	
Analytical Method	File Folder Color
524.2	Yellow
8260 (Water) / (Soil)	Brown/Lavender
624	Blue
8010 / 601 / (Water) / (Soil)	Blue
8081/8082 /608 / (Water)/ (Soil)	Orange
8151 / 615 (Water) / (Soil)	Orange
8270 / 625 / (Water) / (Soil)	Red
TPH-Purgeable (Water) / (Soil)	Purple
TPH-Extractable (Water) / (Soil)	Green
TCLP Volatiles	Gray
TCLP Pest / Herb/ Sv	Gray
S / O 11 Regulated	Blue
TOC	Pink
Methanol	White
Metals	Black
Wet Chemistry	Teal/Pink
Gravimetric	Maroon
COD/Phosphorus	Navy Blue
Anions	Light Blue
Perchlorate	Orange
Methane	White
Organic Acids	Yellow

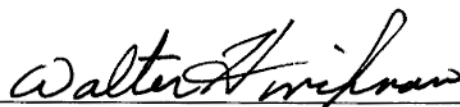
**1.0 TITLE: MAINTENANCE OF CUSTODY**

Description: The maintenance of custody defines who is responsible for sample integrity at any given time. The responsibility of sample custody changes during the sample's life in the laboratory. Sample custody is documented on bench logs, analysts logs and other QA documents while in the laboratory, from receipt until sample disposal.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 The following narrative describes the change of sample custody and responsibility during normal laboratory activities.
- 2.1.1 The Sample Custody Officer is responsible for sample receipt and placing samples in the correct cold storage facility.
  - 2.1.2 Sample custody changes when an extraction technician or analyst removes that sample from storage. Samples will remain in that person's custody until returned to storage.
  - 2.1.3 Once the sample arrives at the lab bench, the analyst or technician records all procedures completed on the sample in the proper logbooks.
  - 2.1.4 The remaining sample or extract is returned to the appropriate storage facility and custody returns to the Sample Custodian.
  - 2.1.5 Those samples requiring security are stored in a locked storage area, with the Sample Custodian or a designated analyst having access. Sample custody remains that person's responsibility at all times.

Approved By: \_\_\_\_\_

  
Walter Hinchman

QA Officer

Date: January, 2007

**1.0 TITLE: E-MAILING FINAL ANALYTICAL DATA**

Description: An e-mail of the original data is the preferred method of relaying client information. The following procedures outline the steps taken when e-mailing client information.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Analytical Data Verification**

- 2.1.1 Verify the analytical folder disseminated to the various analysts; match the final report generated by the report writer.
- 2.1.2 Verify the analytical folder was correctly placed in the appropriate master file located in the Bin World under the day the reports are due.
- 2.1.3 Verify the final analytical reports match the requested methods of analysis on the COC.
- 2.1.4 Verify that all analytical data and final reports are included in the master file.
  - 2.1.4.1 If the master file is not complete, place the analytical folders with their reports standing up in the master file and place them in the Bin World under the day the reports are due.
  - 2.1.4.2 If at the end of the day, folders have not been completed, then place the master file in the appropriate secondary bin (e.g. delayed data, needs inorganic results, etc.).
- 2.1.5 If clients have "Client Specific SOPs" these SOPs need to be checked for special reporting instructions before removing the master file from the due date bin to the secondary bins. These special reporting instructions include such things as: "e-mail results as soon as possible," etc.
- 2.1.6 If a master file is complete, it is ready for e-mailing. The master file is then removed from the Bin World and placed in the "Reports to be E-mailed" bin next to the scanner until it is e-mailed.

**2.2 E-mailing**

- 2.2.1 Prior to e-mailing, the entire report is scanned and saved in a sub-directory by month using the work order and job name to identify the file.
- 2.2.2 A permanent record is completed daily which lists all reports e-mailed during

that day. This record includes the following:

- Alpha's W.O. number,
- The test being reported, and
- Noting if the final data reported was a completed or partial data report.

2.2.3 Items to be e-mailed include the following:

- Final reports, and,
- Client's original COC. If the client's original COC is amended, then the amended COC is included with the e-mail.

2.2.4 Send the e-mail to the person listed under the "Report Attention" column located on the COC.

2.2.5 Refer to the client specific SOP for any special instructions, such as e-mailing to a group of people.

2.2.6 Once the e-mail process has been initiated, place the reports and COC back into the master file and place this file into the Awaiting Confirmation bin next to the scanner.

2.2.7 After the e-mail confirmation has been issued, stating the e-mail transmission was successful, the confirmation is placed in the master file as part of its permanent record.

2.2.8 Once the reports have been e-mailed, the master file is then forwarded to one of the following areas:

- Client specific QC,
- Electronic Data Deliverables(EDDs),
- Invoicing.

If the folder is not complete, place the folder back in Bin World in one of the secondary bins associated with its deficiency.

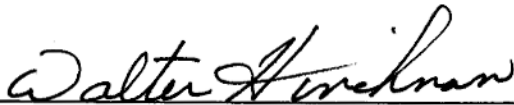


2.2.9 Once the daily e-mailing requirements has been completed, the list of e-mailed client reports are entered into the LIM System under

- Work Order, and
- Fax Update

Additionally, the daily permanent record listing all reports e-mailed during that day is archived in a 3-ring binder labeled "E-mailed Report Binder," segregated by month.

Approved By: \_\_\_\_\_



Walter Hinchman  
QA Officer

Date: January, 2007

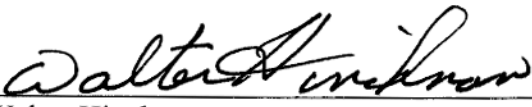
**1.0 TITLE: SAMPLE SCHEDULING AND DISCREPANCY REPORTING**

Description: The Sample Custody Officer and Supervisors communicate all sample integrity issues such as sample containers, preservation and holding times with the appropriate personnel to ensure samples are received and documented appropriately. The SCO will also help schedule sample receipt and analysis with the client and notify that person as to any analytical irregularities.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 The laboratory supervisors will coordinate sample scheduling with the appropriate personnel to maximize economy of effort with the number and types of analysis to be performed.
- 2.2 The laboratory supervisors will update the client as necessary throughout the analytical process as required by that client.
- 2.3 Irregularities with sample documentation, or problems encountered during sample analysis will be reported immediately to the laboratory director.
- 2.4 The director or analysts will document all discrepancies associated with the sample, before the problem is communicated to the client.
- 2.5 It is the responsibility of the laboratory director to communicate and resolve any out-of-control problems with a client.
- 2.6 In reporting data discrepancies, the client's sample identification is used both verbally and in correspondence.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

## 1.0 TITLE: WASTE DISPOSAL OPERATIONS

Description: Alpha Analytical, Inc. is committed to insuring to our customers and our employees that all wastes generated and/or accepted at the laboratory are disposed of in a proper and responsible manner.

## 2.0 CLASSIFICATION AND ORIGIN OF WASTE STREAMS

To work within the proper waste disposal program at Alpha, it is important to understand the different waste streams and their proper handling and disposal operations.

### 2.1 Types of Waste

2.1.1 **Non-Hazardous Waste:** Most materials, solids and liquids, that have not come in contact with solvents can be disposed of as non-hazardous wastes. These include unused soil and water samples, rinse waters, glass containers, etc.

2.1.2 **Hazardous Waste:** Those materials which are listed or characteristic wastes as defined in 40 CFR. Specifically, these include spent solvents, spent soil and water extracts, specific inorganic wastes, fuel samples and any material which comes into contact with large amounts (>10%) of any solvent.

### 2.2 Origin of Waste Stream

It is important to understand that a hazardous material is not a hazardous waste until its use is no longer viable within the laboratory. Only when a hazardous material is ready for disposal does it become a hazardous waste. Once it becomes a hazardous waste, it must be disposed of within the facility and can not be moved.

Tracking a soil sample for VOC analysis can better explain this concept. An aliquot of soil is removed from a brass sleeve and extracted into Methanol in a 40ml vial. This sample extract can be moved from the lab to the storage area on Freeport Way. The sample remains there for the 60-day retention period required by the client. Once the retention time is complete, the sample extract is ready for disposal and then becomes a hazardous waste. Other wastes, such as waste solvent from the SVOC extraction process has no benefits after the extraction process and therefore becomes a hazardous waste immediately. Because of this, the spent solvents from the SVOC lab must stay within the lab facility and can not be transported to the waste facility on Freeport.

It is important to understand that Alpha holds two EPA manifest sites: the lab facility on Glendale Avenue and the waste facility behind the lab on Freeport Way. All wastes generated within either facility must be manifested and disposed of from that facilities waste stream. If you have any questions regarding the status of a waste or

the proper placement of the waste, please contact Randy Gardner for information.

## **2.0 Waste Streams - Per Facility**

### **2.1 Laboratory Facility**

All hazardous waste that is generated in the laboratory facility, located on Glendale Avenue, must be temporarily stored in the waste area of the extraction lab or in the inorganic lab. The wastes that are located in the extraction lab are as follows:

2.1.1 Waste Solvent: All waste solvents, including Methylene chloride, Methanol, Hexane, Ether and any other rinse or spent solvents, are combined and manifested as F listed wastes.

2.1.1.1 Extracted Liquids: All extracted liquids for such analyses as SVOC and Pesticides.

2.1.1.2 Extracted Solids: All extracted solids, or any other solids that come in contact with solvents (i.e. sodium sulfate).

2.1.2 Wastes located in the inorganic laboratory include:

2.1.2.1 Metals Liquid Waste: Waste reagents from colormetric procedures (COD,  $\text{Cr}^{+6}$ , Sulfide, etc) and other inorganic procedures that produce wastes containing high levels of metals. Such wastes include silver, mercury, hexavalent chrome and other such metals.

### **2.2 Waste/Storage Laboratory**

All samples and sample extracts that are stored in the waste laboratory behind the lab facility (located on Freeport Way) are not classified as hazardous waste until they are ready to be disposed of. Again, all sample and sample extracts stored in the waste area have beneficial use and are stored there in case there is a need for a further analysis or a re-analysis for confirmation purposes. Alpha's general policy is to store these samples and their extracts for 60 days. After the 60-day storage period, the samples are ready to be disposed and therefore have no beneficial use, thus classifying themselves as hazardous waste. The hazardous wastes located in the waste/storage lab are as follows:

2.2.1 Waste Solvent: All solvent waste that has been decanted from the extracted samples. These are typically TPH and VOC soil samples that have been extracted and stored in a 40-ml closed-top vial. This waste solvent is primarily made up of Hexane and Methanol.

- 2.2.2 Extracted Solids: All extracted solids, or any other solids that come in contact with solvents (i.e. sodium sulfate). Once the solvents are decanted from the TPH or VOC extraction vials, these soils must be placed in the extracted soils container.

The non-hazardous wastes located in the waste/storage lab are as follows:

- 2.2.3 Non-extracted Solids: All non-extracted soils, such as duplicate soil containers and unused soil samples are disposed of as non-hazardous waste. These soils are stored in a 55-gallon drum and disposed of as Petroleum Contaminated Soils.
- 2.2.4 Non-extracted Liquids: All non-extracted liquids, such as duplicate containers and unused samples are treated as non-hazardous waste. Because liquids for TPH-Diesel (EPA Method 8015) use 1 ml of hexane for the extraction, these liquids are separated from the hexane and treated as non-hazardous. These liquids are collected and evaporated as described in the following section.

### **3.0 Waste Streams - Per Usage**

#### **3.1 Usage**

Following is a designation of all hazardous and non-hazardous wastes that Alpha deals with in conjunction with its laboratory functions. Also listed are the correct handling and disposal processes.

- 3.1.1 Waste Solvent: All spent solvent, both chlorinated and non-chlorinated must be disposed of as waste solvent. Several temporary waste containers are available throughout the laboratory. Once full, these containers must be emptied into the 55-gallon drum in the extraction or waste laboratory. The drum is self-contained. It is important that this drum remain under the fume hood at all times and that the drum is never tipped or rolled. Each time solvent is added to this drum, the level must be marked and dated. A weekly visual examination must be made of the drum to ensure that acidic or chlorinated solvents have not corroded the metal barrel. The lids must be kept on the barrels when not in use.
- 3.1.2 VOC Soil Extracts: Soils being analyzed for volatile organic (VOC) are extracted with methanol. The soils are extracted into methanol in 40 ml vials. After analysis, the extracts are moved to the waste facility for the required retention period. After the retention period, the vials are crushed with the glass crusher and the solid material (both glass and extracted soil) are disposed of as extracted soil. The solvent is screened off and disposed of as waste solvent.

- 3.1.3 VOC Waters: Those aqueous samples requiring VOC analysis are purged directly onto the instrument and are not solvent extracted with methanol. The purged water, along with the duplicate vials, are crushed with the glass crusher and the solid material (glass vials) are disposed of as non-extracted soil/solid. The remaining water is screened off and disposed of as non-extracted water.
- 3.1.4 TPH-E Soil Extracts: Soils being analyzed for TPH-E are extracted with Hexane/Acetone (90/10). The soils are extracted in 40 ml vials. After analyses, the extracts are moved to the waste facility for the required retention period. After the retention period, the vials are crushed with the glass crusher and the solid material (both glass and extracted soil) are disposed of as extracted soil. The solvent is screened off and disposed of as waste solvent.
- 3.1.5 TPH\_E Water Extracts: Those waters that require TPH/E analysis are extracted with Hexane in 40 ml vials. After analyses, the extracts are moved to the waste facility for the required retention period. After the retention period, the vials are crushed with the glass crusher. The water is separated from the solvent and disposed of as non-extracted water and the solvent is disposed of as solvent waste.
- 3.1.6 Extracted Soil for Semi-volatile Analyses (SVOC, and Pesticides etc): Soils requiring any semi-volatile analysis are solvent extracted with the Dionex ASE extraction unit. These solvent extracts are captured into 40 ml vials. The extracted soil is removed from the ASE extraction containers and are disposed of in the laboratory as extracted soils. Because the extracted soils from the ASE are not retained for further use, they become hazardous waste immediately and can not be moved to the waste facility. ALL ASE extracted soils are manifested from the laboratory facility. After analysis, the extracts and/or autovials are moved to the waste facility for retention. After retention, the vials are crushed with the glass crusher and the solid material is disposed of as extracted soil. The extract is disposed of as waste solvent.
- 3.1.7 Extracted Liquids for Semi-volatile Analysis (SVOC and Pesticides, etc): Liquids requiring any semi-volatile analysis are solvent extracted. The solvent extract is concentrated down and placed in the appropriate vials. The extracted water is disposed of in the laboratory as extracted water. Because the extracted waters are not retained for further use, they become hazardous waste immediately and can not be moved to the waste facility. ALL such extracted waters are manifested from the laboratory facility. After analysis, the extracts and/or autovials are moved to the waste facility for retention. After retention, the vials are crushed with the glass crusher and the solid material is disposed of as extracted soil and the extract is disposed of as waste solvent.

- 3.1.8 Auto-vialed Extracts: All samples that require a semi-volatile extraction are transferred into 2ml auto-vials. These auto-vials are then placed on the appropriate instrument and analyzed. After analysis, these vials are moved to the waste facility for retention. After the retention period, they are disposed of by crushing the vials with the small auto-vial crusher located under the waste fume hood. The solid material is disposed of as an extracted soil/solid. The solvent extract is disposed of as waste solvent.
- 3.1.9 Inorganic Laboratory Liquid Waste: All extracted and non-extracted waste water from the inorganic shop must be neutralized before being disposed of as a non-hazardous waste.
- 3.1.10 Inorganic Metals Waste: All reagents that contain hazardous levels of metals. Such wastes include many of the colormetric method wastes from the analysis of COD, Sulfide and  $\text{Cr}^{+6}$ . These wastes have high levels of Silver, Mercury and  $\text{Cr}^{+6}$ .

#### **4.0 Storage of Hazardous Waste**

- 4.1 One of the fundamental concepts of hazardous waste management is the 90-day limitation for accumulating and storing hazardous waste on-site before they must be transported off-site for treatment or disposal.

Whereas the 90-day limitation for on-site storage of hazardous waste is a requirement of HW management, it is important to know and understand the exceptions to this rule, namely the *Satellite Rule*. This exception is important in that it would be costly and inefficient for Alpha to accumulate its many different waste streams and to ship dispose of them every 90 days. However, compliance should be followed meticulously. Therefore it is imperative that the rules are understood.

The following is a quick reference summary of the accumulation and storage rules:

- 4.1.1 On-site storage of HW without a storage facility permit is limited to 90 days unless the generator qualifies for satellite rule accumulation.
- 4.1.2 The accumulation start date which begins the 90-day period and must be identified on the label of each container is different for large and small quantity generators. If more than 100 kgs (220 lbs or 27 g) of hazardous waste or 1 kg of extremely or acutely hazardous waste are generated at the entire facility, the 90-day period begins when the first drop hits the container. If not, the small quantity generator may accumulate up to these amounts before beginning the 90-day clock.

- 4.1.3 The *satellite rule* provides relief for generators who qualify by allowing accumulation of up to 55 gallons of hazardous waste (or 1 qt of extremely or acutely hazardous waste) or one year, whichever comes first, at or near the point of generation. This rule allows a reasonable accumulation period for smaller volume waste streams as they are usually generated in the laboratory.
- 4.1.4 Any facility which stores hazardous wastes over the 90-day period (unless the satellite rule applies) must obtain a permit or variance from the NDEP.

4.2 Accumulation Exemptions: (40 CFR 262.34)

- 4.2.1 A generator who generates greater than 100 kg but less than 1000 kg of HW in a calendar month may accumulate hazardous waste on-site for 180 days or less without a permit or without having interim status provide that:
  - 1) The quantity of waste never exceeds 6000 kg
  - 2) That requirements of subpart I of 265, except 265.176 are in compliance. These requirements are:
    - a) The condition of the containers are intact;
    - b) Stored wastes are compatible;
    - c) The management of the containers is prudent (i.e., container must be kept closed, and handled with care);
    - d) The containers must be inspected at least weekly;
    - e) The containers are clearly labeled and the accumulation dates are legible;
    - f) At least one employee must be available at all times to respond to an emergency at the facility;
    - g) The following information must be posted near the telephone:
      - 1) Name and phone number of emergency coordinator;
      - 2) Location of fire extinguishers, spill control material and fire alarms; and,
      - 3) Telephone number of local fire department unless the facility has a direct line through the fire alarm.



- 3) Alpha must ensure that all employees are thoroughly familiar with proper waste handling and emergency procedures, relevant to their responsibilities during normal facility operations and emergencies.
- 4) Immediate emergency response levels are:

FIRE: Call the fire department and/or attempt to extinguish it.

SPILL: In the event of a spill, contain the flow of HW to the extent possible, and as soon as is practicable, clean up the HW and any contaminated materials or soil.

OTHER: In the event of a fire, explosion, or other release which could threaten human health outside the facility or when the generator has knowledge that a spill has reached surface water, the SO must immediately notify the National Response Center (800/424-8802). The report must include the following information:

- 1) Name, address, and USEPA ID generator's number;
- 2) Date, time and type of incident (spill or fire, etc);
- 3) Quantity and type of HW involved;
- 4) Extent of injured; and
- 5) Estimated quantity and disposition of recovered materials if any.

If the generator meets the above criteria, and must transport his waste, or offer his waste for transport, over a distance of 200 miles or more for off-site treatment, storage or disposal may accumulate HW on-site for 270 days or less without a permit or without having interim status.

## **5.0 Disposal of Empty Containers**

- 5.1 Containers which are empty and no longer needed must be disposed of properly. Container disposal shall be as directed by 40 CFR 261.7 "Residues of hazardous waste in empty containers." Containers which have held acute hazardous materials as defined in 40 CFR 261.31, 261.32, or 261.33 require special handling. Please contact Randy Gardner for further information and assistance. To assist you in determining if an empty container is regulated, here are some further guidelines.

A container shall be considered "empty" if all the following conditions exist (for this section, a container shall be considered to be a primary container or an inner liner):

- 5.1.1 The container contained none of the chemicals that are listed in 40 CFR 261.33(e),
- 5.1.2 All chemicals have been removed that can be removed using practices commonly employed to remove materials from that type of container (i.e. pouring, pumping, aspirating, etc.),
- 5.1.3 There is less than one inch of residue left in the bottom of the container,
- 5.1.4 There is less than 3% by weight of residue left in the container (0.3% for >110 gal. containers),

The container shall be considered empty only if the container has been triple rinsed *or* cleaned by another method that has been shown in the scientific literature to achieve equivalent removal. If the container has not been cleaned as stated above, the container shall become hazardous waste.

Once a container has been declared "empty" by the above criteria, it can be placed in the normal refuse.

## **6.0 Waste Area Mechanical Operation**

All waste area technicians must have training in the operation of the jaw crusher, pump operation, evaporator operation and storage procedures before being allowed access to the waste area. Following is listing of the equipment in the waste laboratory and their standard operating procedures.

### **6.1 Waste water evaporator and holding/settling tank operation**

- 6.1.1 The waste water evaporator is designed for the evaporation of non-hazardous wastewater. Utmost care must be given to ensure that hazardous waste, solvents or pure fuel products are not introduced to either the holding tank or the evaporator. Wastewater may be introduced into the evaporator by either discharging directly into the evaporator or by transferring wastewater from the holding tank to the evaporator. The holding tank is designed to prevent sediment transfer directly into the evaporator. In essence it is a settling tank. Once evaporator wastewater has been pumped into the settling tank, a period of no less than twenty four hours of undisturbed settling time is required before the second transfer is initiated from the settling tank to the evaporator. Wastewater directly dumped into the evaporator must be free of sediment or very carefully decanted to eliminate sediment buildup of the evaporation unit. Waste streams that can be evaporated include:

- Mass spectrometer rinse water,
- Client water samples,
- Neutralized inorganic wastewater
- TPH-E water extracts (decanting the hexane solvent layer),
- HPLC waste, except acetonitrile waste waters; and,
- Non-extracted lab waters.

6.1.2 The evaporator requires routine maintenance in an effort to keep it as clean as possible and to preclude the build up of heavy metals and salts. Over time this type of scaling can be harmful to the heating coils. On a three-month frequency, the evaporator water should be cooled, then pumped out and disposed of as non-hazardous wastewater. The evaporator should be thoroughly cleaned including wire brushing scale deposits off the heating coils and sides of the evaporator and thorough checks should be made for cracking or unusual wear. Once a year, the combustion chambers' air to fuel mixture should be checked and adjusted according to the owner's manual specifications. The full maintenance procedure is found in section 4.3 of the owners manual.

6.1.3 Extreme care should be given when transferring wastewater from the crusher to the settling tank and on to the evaporator. This operation may require two people to prevent possible over-spillage.

## 6.2 Jaw Crusher Operation

6.2.1 The jaw crusher is used for the crushing of 40mL VOA vials and 1 liter bottles. DO NOT crush auto vials or screw cap vials smaller than 20 ml VOA's in the jaw crusher. All crushed glass collected in the lower strainer bin must be triple rinsed with water before being dumped into the trash container.

6.2.2 The jaw crusher is a very powerful piece of equipment. The following safety elements should be followed when operating the jaw crusher:

- 6.2.2.1 Never operate the crusher without the fan belt and fly wheel shrouds bolted in place. Use common sense when in operation;
- 6.2.2.2 Do not put your hands near any moving parts;
- 6.2.2.3 Always ensure no one is near the crusher when the machine is being tuned on;
- 6.2.2.4 Always use an extender such as a stick or a bar to free glass

stuck in the loading chute;

- 6.2.2.5 Always wear safety glasses;
- 6.2.2.6 Always wear cotton or leather gloves;
- 6.2.2.7 Never over fill the hopper. VOA vials will clog the hopper jaws and not feed properly;
- 6.2.2.8 The hopper lid must be installed to prevent glass from flying out of the hopper when crushing liter and smaller bottles;
- 6.2.2.9 Remove all liter and large bottle lids prior to crushing;
- 6.2.2.10 Bottles must be fed into the hopper one bottle at a time with the neck of the bottle towards the jaws;
- 6.2.2.11 All glass collected in the lower strainer bin must be triple rinsed before dumping into the trash container;
- 6.2.2.12 Always know what is being placed in the crusher;
- 6.2.2.13 Never leave uncrushed material in the hopper;
- 6.2.2.14 Keep the crusher and surrounding area clean and free of litter;
- 6.2.2.15 Clean all spills immediately;
- 6.2.2.16 Keep the white crusher “roll-around-bins” as clean and sediment free as possible;
- 6.2.2.17 Avoid vacuuming sediment in the pick-up hose when transferring water from the white crusher-bins to the settling tank; and,
- 6.2.2.18 It is mandatory that the crusher’s moving shaft be greased after each crushing episode to prevent the jaws from freezing up.

### 6.3 Auto vial crusher operations

- 6.3.1 The autovial crusher is used to crush auto vials, and 2 and 8 ml screw cap

vials. This operation is performed under the waste hood to capture solvent vapors. Avoid over filling the hopper and always ensure the discharge hose is securely placed in the solvent waste container under the crusher. The crushed glass is disposed of as extracted soil.

## **7.0 TPH-E Disposal Procedures**

7.1 There are four primary waste streams used in the disposal of TPH samples. They are non-hazardous wastewater, non-hazardous solids, hazardous solids, and solvent waste. The following standard operating procedures describe the disposal practices for each TPH-E waste stream.

### **7.2 Wastewater**

The following TPH-e waste streams are bulked together and disposed of as Non-hazardous wastewater. These streams include:

7.2.1 Unused Client Water Samples: Unused client water samples are packed in boxes labeled TPH WASTEWATER with the sample receipt date. These boxes are transferred for storage to the waste laboratory. After the required retention, the samples are disposed of as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-4B (duplicate samples)  
Cold Storage Sample Removal: 2 weeks from date of sample receipt  
Cold Storage Removal frequency: Weekly  
Total Storage Period: 60 days from date of sample receipt  
Waste stream: Non-hazardous wastewater

Note: Water sample are bulked into the non-hazardous wastewater stream. The containers are crushed. Once crushed, the glass shards are triple rinsed with water and discarded in the trash bin.

7.2.2 Extracted Water: All extracted water is packed in boxes and labeled TPH EXTRACTED WATER with the disposal date. These boxes are transferred for storage to the waste laboratory and placed on shelves marked TPH EXTRACTIONS.

Cold Storage: Refrigerator EXT-37A  
Cold Storage Sample Removal: Monthly  
Total Storage Period: 60 days from date of sample receipt  
Waste Stream: Non-hazardous wastewater

Note: The extracted water is decanted from the vials and the water fraction is bulked with the non-hazardous wastewater and the hexane solvent is bulked with the waste solvent.

- 7.2.3 TOC Waste: All TOC waste water is treated as non-hazardous wastewater. TOC wastewater is transferred to the waste room for storage, then disposed of as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-4B (duplicate samples)  
Cold Storage Sample Removal: 2 weeks from date of sample receipt  
Cold Storage Removal frequency: Weekly  
Total Storage Period: 60 days from date of sample receipt  
Waste stream: Non-hazardous wastewater

### 7.3 Hazardous Solids

The following TPH waste streams are bulked and disposed of as hazardous solids. Hazardous solid waste stream includes:

- 7.3.1 Extracted Soils: Soil extracts are packed in boxes labeled TPH SOIL EXTRACTS with the disposal date. Soil extracts are transferred for storage to the waste laboratory. These boxes are stored on the shelves marked TPH EXTRACTATIONS.

Cold Storage: Refrigerator EXT-37A  
Cold Storage Sample Removal: Monthly  
Total Storage Period: 60 days from date of sample receipt  
Waste Stream: Hazardous solid and Solvent waste

Note: The hexane is decanted and bulked with the solvent waste. The remaining soil extract and glass containers are crushed together, collected and bulk packed with the hazardous solid waste.

### 7.4 Non-hazardous Solids

The following TPH-E waste streams are bulked and disposed of as non-hazardous solids. Non-hazardous solid waste stream includes:

- 7.4.1 Unused Client Soil Samples: Unused client soil samples older than 14 days are transferred for storage to the waste laboratory. The samples are stored in order of date received and are disposed of in accordance to their disposal dates.

Cold Storage: Refrigerator EXT-37A  
Cold Storage Sample Removal: 14-days  
Total Storage Period: 60 days from date of sample receipt  
Waste Stream: Non-hazardous solid

## 7.5 Pure Fuel Product Samples

- 7.5.1 Pure product sample are removed for storage to the waste laboraory in sealed paint cans. They are labeled with their disposal date and placed on the Pure Product shelves in the storage area, according to the date of disposal.

Cold Storage: Refrigerator SAR-2B

Cold Storage Sample Removal: Monthly

Total Storage Period: 60 Days from date of sample receipt

Waste Stream: Waste Solvent

## 8.0 VOC Waste Disposal Procedures

- 8.1 There are four primary waste streams used in the disposal of VOC samples. They are non-hazardous wastewater, non-hazardous solids, hazardous solids, and waste solvent. The following SOP describes the disposal practices for VOC samples and their associated wastes.

### 8.2 Wastewater

The following VOC subwaste streams are bulked together and disposed of as Non-hazardous Wastewater. These waste streams include the following:

- 8.2.1 Unused Client Water Samples: Unused Client water samples older than 14 days are removed from the refrigerator and packed in boxes labeled VOC NON-EXTRACTED WATER with the date of sample receipt. The VOA rack labels indicating client name and date are removed, and attached to the inside lid of the storage box. These boxes are transferred to the waste area and placed in the storage racks designated for VOC samples. After a period of 60 days from the date of sample receipt, samples are discarded as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-12A

Cold Storage Sample Removal: 2 weeks from date of sample receipt

Cold Storage Removal Frequency: Daily

Total Storage Period: 60 days from date of sample receipt

Waste Stream: non-hazardous wastewater

Note: Water samples are bulked into the non-hazardous wastewater stream. The VOA vials are crushed. Once crushed, these glass shards are triple rinsed with water and discarded in the trash bin.

- 8.2.2 Mass Spectrometer Rinse Water: The mass spectrometer rinse water is collected from each instrument daily. Rinse water is transferred to the waste room for disposal as non-hazardous wastewater.

Removal Frequency: Daily  
Waste Stream: non-hazardous wastewater

- 8.2.3 Client Drinking Water Samples: VOC drinking waters are removed from the refrigerator every 30 days. Samples are packed into boxes labeled VOC DRINKING WATER with the date of disposal. These boxes are transferred to the sample storage area and retained for 60 days from date of sample receipt. Samples are disposed of as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-10A  
Cold Storage Sample Removal: 30 Days from date of sample receipt  
Cold Storage Removal Frequency: 30 Days  
Total Sample Storage Period: 60 Days from date of sample receipt  
Waste Stream: non-hazardous wastewater

Note: Water samples are bulked into the non-hazardous wastewater waste stream. The VOA vials are crushed. Once crushed, these glass shards are triple rinsed with water and discarded in the trash.

### 8.3 Non-hazardous Solids

The following VOC waste streams are bulked together and disposed of as non-hazardous solids. These waste stream include the following:

- 8.3.1 Unused Client Soil Samples: Unused Client soil samples older than 14 days are removed from the refrigerator and placed into green plastic tote boxes. Tote boxes are labeled with the date of sample receipt. These tote boxes are transferred to the waste area and placed on shelves. Sample are retained for a period of 60 days from the date of sample receipt. Samples are discarded as non-hazardous solids.

Cold Storage :Refrigerator SAR-11A  
Cold Storage Sample Removal: Date of analysis or 2 weeks from date of sample receipt  
Cold Storage Removal Frequency: Daily  
Total Storage Period: 60 Days from date of sample receipt  
Waste Stream: non-hazardous solids

Note: Soil samples and their glass jars are bulked into the non-hazardous solids waste stream.

### 8.4 Hazardous Solids

The following VOC waste streams are bulked together and disposed of as hazardous solids. These waste streams include the following:



- 8.4.1 Extracted Soil/Solid Samples: VOC extracts older than 30 days are removed from the refrigerator monthly. The extracts are packed into boxes labeled MEOH SOIL EXTRACTS with the date of disposal. The boxes of extracts are transferred to the sample storage area and placed on designated racks in the waste room. After a storage period of 60 days from date of sample receipt, the extracts are discarded as hazardous solids.

Cold Storage: Refrigerator EXT-36A

Cold Storage Sample Removal: 30 Days from date of sample receipt

Cold Storage Removal Frequency: 30 Days

Total Sample Storage Period: 60 Days from date of sample receipt

Waste Stream: Hazardous Solids and Waste Solvent

Note: The methanol is decanted and bulked with the solvent waste. The remaining soil extract and glass containers are crushed together, collected and bulk packed with the hazardous solids.

- 8.4.2 Methanolic Air Extracts: Methanol air extracts older than 60 days are discarded every 3 months. The methanol and carbon material in the vials are crushed in the autovial crusher and the resulting glass shards are disposed of as non-hazardous solids.

Cold Storage: Refrigerator SAR-1A

Cold Storage Sample Removal: 60 Days from date of sample receipt

Cold Storage Removal Frequency: 90 Days

Total Sample Storage Period: 60 Day from date of sample receipt

Waste Stream: non-hazardous solids

## 8.5 VOC Solvent Waste

The following VOC subwaste streams are bulked together and disposed of as Solvent Waste and non-hazardous solids. These subwaste streams include the following:

- 8.5.1 Solvents and Fuel Products: Client samples containing pure products older than 14 days are removed from the refrigerator monthly. Pure product samples are placed in labeled paint cans. Samples are transferred to the sample storage area and placed on VOC storage racks. After a period of 60 days from the date of sample receipt, samples are discarded as waste solvent.

Cold Storage: Refrigerator SAR-2A

Cold Storage Sample Removal: 14 Days from date of sample receipt

Cold Storage removal Frequency: Monthly

Total Storage Period: 60 Days from date of sample receipt

Waste Stream: Waste Solvent

Note: Pure product samples are decanted and bulked with the solvent waste stream. The glass containers are crushed. Once crushed, these glass shards are triple rinsed with water and discarded in the trash bin.

## **9.0 Semi-volatile/Wet Lab Waste Disposal Procedure**

- 9.1 There are six primary waste streams used in the disposal of Semi-Volatile wastes extracts and spent samples. They are non-hazardous wastewater, hazardous wastewater, waste solvent, non-hazardous solids, and hazardous solids. The following standard operation procedures describe the disposal practices for Semi-Volatile samples and their associated wastes.

A special note concerning all extracts from the semi-volatile extraction laboratory. All extracts can be moved and stored in all parts of both the laboratory and the storage area. This is due to the fact that they have a beneficial use within the laboratory (for analysis, both present and future) and are not hazardous wastes until the time that they are ready for disposal.

All solvent-extracted soils and waters, and all waste solvent generated within the extraction laboratory must stay within the extraction laboratory where they were generated. This is due to the fact that these materials do not have any further beneficial use and are ready for disposal.

### **9.2 Non-hazardous Wastewater**

The following semi-volatile waste streams are bulked together and disposed of as non-hazardous wastewater. These waste streams include the following:

- 9.2.1 Unused Client Water Samples: Unused client water samples older than 14 days are removed from the refrigerator and moved to the waste laboratory. The samples are stored on racks according to their disposal date and disposed of as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-3A

Cold Storage Sample Removal: 14Days from date of sample receipt

Cold Storage Removal Frequency: Daily

Total Storage period: 60 Days from date of sample receipt

Waste Stream: non-hazardous wastewater

- 9.2.2 Unused TCLP Extracts: Unused TCLP extracts older than 14 days are removed from the refrigerator and moved to the waste laboratory. The samples are stored on racks according to their disposal date and disposed of as non-hazardous wastewater.

Cold Storage: Refrigerator SAR-3A

Cold Storage Sample Removal: 14Days from date of sample receipt  
Cold Storage Removal Frequency: Daily  
Total Storage period: 60 Days from date of sample receipt  
Waste Stream: non-hazardous wastewater

### 9.3 Hazardous Wastewater

The following semi-volatile waste stream is bulked together and disposed of as hazardous wastewater. This waste stream includes the following:

- 9.3.1 Solvent Extracted Water Samples: Spent Water samples extracted with >10% solvent are manifested and disposed of in the extraction laboratory under the waste hood.

Solvent Extracted Water Removal: As needed  
Waste Stream: Hazardous Wastewater

### 9.4 Hazardous Solids

The following semi-volatile waste stream is bulked together and disposed of as hazardous solids. This waste stream includes the following:

- 9.4.1 Extracted Soil Samples: All solvent extracted soils from the ASE extractor must be manifested and disposed of in the extraction laboratory under the waste hood.

ASE Extracted Soil Removal: As needed  
Waste Stream: Hazardous Soil

- 9.4.2 Saturated Extraction Supplies (Sodium Sulfate/Filters, etc):

All solvent extracted soils, filter paper, extraction discs and sodium sulfate from semi-volatile extractions where the materials have come into contact with solvents must be manifested and disposed of in the extraction laboratory under the waste hood.

Removal: As needed  
Waste Stream: hazardous Soil

### 9.5 Non-hazardous Solids

The following semi-volatile extractable waste stream is bulked together and disposed of as non-hazardous solids. This waste stream includes the following:

- 9.5.1 Unused Client Samples: Unused Client soil samples older than 14 days are

removed from the refrigerator and moved to the waste laboratory. These soil containers are placed on racks according to the sample disposal date.

Cold Storage: Refrigerator SAR-3A  
Cold Storage Sample Removal: 14 days from date of sample receipt  
Cold Storage Removal Frequency: Daily  
Total Storage Period: 60 days from date of sample receipt  
Waste Stream: Non-hazardous Solids

Note: Soil samples are bulked into the non-hazardous solids waste stream. The wide mouth glass jars are crushed. Once crushed, the glass shards are disposed of along with the soil as non-hazardous soils.

#### 9.6 Waste Solvent (From the Extraction Laboratory)

The following waste streams are bulked together and disposed of as waste solvent. It is again important to emphasize that all waste solvent generated within the extraction lab must be manifested and disposed of from the extraction lab. Those solvents decanted from the stored samples and autovials in the waste laboratory on Freeport must be manifested and disposed of from the Freeport facility. These bwaste streams (from the extraction lab) include the following:

9.6.1 Glassware Rinse Solvent: Glassware solvent rinse, ASE waste solvent, waste solvent from standard preparation, and excess waste solvent from extractions are bulked in the extraction laboratory.

9.6.2 Solvent Extracts: ASE generated semi-volatile solvent extracts older than one month are removed from refrigerator EXT 31B. Extracts are placed into boxes labeled SEMI-VOLATILE EXTRACTS with the sample disposal date. These boxes are placed on racks in the waste laboratory and after a period of 60 days from date of sample receipt, the extracts are discarded as waste solvent.

Cold Storage: Refrigerator EXT-31B  
Cold Storage Sample Removal: Monthly  
Cold Storage Removal Frequency: Monthly  
Total Storage Period: 60 days from date of sample receipt  
Waste Stream: Waste solvent.

9.6.3 Semi-Volatile Auto Vials: Once semi-volatile analyses are completed, all auto vials are removed from their respective refrigerators. Autovials are packed into paint cans labeled SEMI-VOLATILE AUTO VIALS with a disposal date. The cans are transferred to the waste laboratory and placed on their appropriate rack and stored for 60 days before disposal.

Cold Storage Autovials: Refrigerator EXT-35B  
Total Storage Period: 90 Days  
Waste Stream: Waste Solvent, Hazardous Solids

Note: Autovials are crushed while collecting the waste solvent for disposal.  
The crushed glass is then bulked and disposed with the hazardous solids.

## 10.0 Inorganic Lab Waste Disposal Procedure

- 10.1 Inorganic Reagents (From the Inorganic Laboratory): The following waste streams are bulked together and disposed of as D-listed metals waste. This waste stream includes such metals as silver, mercury and chrome. The common wastes that are disposed of in this stream are those waste reagents/extracts associated with COD,  $\text{Cr}^{+6}$ , Sulfide and other such inorganic analyses.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: REVIEW OF REQUESTS, TENDERS AND CONTRACTS**

Description: Often times new work is associated with a request, tender and/or a contract. This may be a long drawn out and complicated process. Occasionally, new work requires no contract, and samples simply shows up unexpectedly at the sample receiving area. This SOP describes the general procedures for the review or requests, tenders and contracts and the maintenance of those contracts.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Review Items**

The purpose of this review is to establish that we possess the resources necessary for the performance of the contract or project. Items that affect the review of requests, tenders and contracts for acceptance of new work include the following:

- a) are the requested methods of analysis adequately defined, documented and understood;
- b) do we have the appropriate facilities and resources to meet the contract requirements to include:
  - i. do we have the required physical space;
  - ii. do we have the personnel with the skills and expertise;
  - iii. do we have the appropriate information and technical resources;
  - iv. what are the number of samples to be analyzed, and over what time period will they arrive;
  - v. can these samples be analyzed without overly stressing the current staff;
  - vi. are their performance evaluation (PE) samples available for the requested method of analysis; and if not are their QC samples available with know values in order to determine uncertainties of measurement, detection limits, confidence limits, or other essential quality control requirements;
- c) can we provide the specified methods of analysis to include:
  - vii. is the work in a state or under an agency which we have approval;
  - viii. are the requested methods of analysis, methods we are approved for;

- ix. are the target analytes, compounds we are approved for and analyze on a normal basis;
- d) can we provide the specified data quality objectives to include:
  - x. the requested reporting limits or LOQs;
  - xi. the requested accuracy and precision;
  - xii. the requested turn around times; and
  - xiii. are we capable of analyzing the constituents in the requested matrix.

This is not a comprehensive list of questions to be determined, but is a general starting point for all new projects. In addition, these do not answer the business questions regarding payment, pricing etc which are always a factor when evaluating new work.

## 2.2 Review Result

After the initial request of review is completed, the client is informed of the review results, to include:

- a) any potential conflict with ongoing projects;
- b) any deficiencies in the request for analysis;
- c) any lack of appropriate accreditation status; or
- d) the inability to complete the client's work.

## 2.3 Resolution of Differences

Any differences between the requested contract and our capabilities must be resolved before any work commences. The contract, if one exists, must be acceptable to both the laboratory and client.

Note: NELAC defines a contract as any written or oral agreement when providing a client with environmental testing services.

## 2.4 Documentation of Review

- 2.4.1 Records of reviews, including any significant contract changes are retained by Alpha. Most often these are oral contracts and those changes are documented in the "Client Information" section of the Omega LIM system.

2.4.2 These records may also include ongoing discussions with a client relating to the client's requirements or the results of the work during the period of execution of the project or contract.

2.4.3 For the review of routine and other simple tasks, the date and the initials of the person, in the laboratory, responsible for carrying out the contracted work is considered adequate.

For repetitive routine work, the review is made only once at the beginning of the contract or on-going work under a general agreement with the client, provided that the client's requirements remain unchanged.

For new, complex or advanced environmental testing tasks, a more comprehensive record is maintained.

2.4.4 This contract review also includes any work that is subcontracted.

## 2.5 Deviations from the Contract

It is Alpha's policy to inform the client of deviations from the contract.

## 2.6 Contract Amendments

If a contract needs to be amended after work has commenced, the same contract review process is repeated and any amendments are communicated to all affected personnel.

Suspension of accreditation, revocation of accreditation, or voluntary withdrawal of accreditation is reported to the client.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007



# **Appendix D**

## **Document Control Plan**

**1.0 TITLE: DOCUMENT CONTROL PLAN ( DCP)**

**2.0 STANDARD OPERATING PROCEDURE**

2.1 The opening statement to our QA manual explains the purpose and objectives of our quality assurance system. These statements are fundamental principals upon which we have built our QA plan and upon which policy decisions are made. The QA manual devotes much attention to fulfilling these objectives such as:

- Providing a uniform basis for analytical generation and reporting;
- Assisting in the early recognition of deficiencies which affect the quality of data; and,
- Requiring sufficient documentation to verify the quality of data submitted.

The quality assurance system is only as good as the record keeping system. If historical data can not be maintained and retrieved, and the construction of those historical records are in doubt, than a house-of-cards has been created, and the entire system is brought into question and the data is dubious at best.

**2.2 Control of Records**

It is a basic policy of Alpha to provide accurate, precise, complete, and representative determinations, and to document sufficient QA/QC of the analytical procedures. The procedure designed to document and maintain these as historical records is the Document Control Plan. This plan has established a set of SOPs written specifically for maintaining uniformity and consistency in support of these goals.

**2.3 Record Keeping System and Design**

The record keeping system is designed for the identification, collection, indexing, accessing, filing, storing, maintenance and disposal of quality and technical records. These records include reports from internal audits, and management reviews as well as records of corrective and preventive actions.

2.3.1 A major component in the design of the record keeping system is a procedure which allows the historical reconstruction of the laboratory activities that produced the analytical data. The sample history is easily reconstructed through this documentation system. Such records include:

- a) **records identifying the person** involved in sampling, sample receipt, sample preparation, calibration and analysis;

- b) **information relating to** laboratory equipment, analytical test methods, and related laboratory activities such as sample receipt, sample preparation or data verification;

2.3.2 The record keeping system is designed to facilitate the retrieval of working files and archiving records for inspection and verification purposes. This is accomplished by the following description.

2.3.2.1 All document entries are initialed by the responsible staff member. The reason for the signature or initials is indicated on all quantitation reports and QC reports by using a stamp indicating the checking of analytical data. This stamp describes the basic activities, and the person responsible for those activities, initials it. The stamp is as follows:

- Result,
- Peer/QC,
- Report, and
- Final

2.3.2.2 All generated data, except those that are generated by automated data collection systems, are recorded directly, promptly and legibly in permanent ink.

2.3.2.3 Entries in records are not to be obliterated by methods such as erasures, overwritten files or other markings. All record keeping corrections are made by a single line marked through the error. The individual making the correction will initial and date the correction. This criteria applies to electronic records as well, to avoid the loss or change of original data.

## 2.4 Records Management and Storage

2.4.1 All records, certificates and reports are safely stored and held secure and in confidence to the client.

2.4.2 All records are stored in an environment to prevent damage or deterioration and prevent record loss. Records are stored as follows:

- all records are scanned as pdf files and are electronically stored and filed.; and
- hard copy data is stored off-site with a company specializing in record management.

- 2.4.3 All records are retained for a minimum of five years.
- 2.4.4 Records stored or generated by computers are duplicated with either hard copies or write-protected backups.
- All electronic data is stored and supported by the hardware and software necessary for the retrieval, see Appendix F for details.
- 2.4.5 The information necessary for the historical reconstruction of data is maintained with these files. This includes, original observations, (e.g. quantitation reports), calculations and derived data, calibration records and a copy of the test report.
- 2.4.6 The Document Control Filing System SOP, found in this appendix, describes the identification scheme used to archive and retrieve hard copy data records.
- 2.4.7 Electronic data is stored and archived in a manner to facilitate retrieval and security. Appendix F, Software Quality Assurance Plan (SQAOP), has two SOP which describes this process, SOP F.3, Data Collection and Storage, and SOP F.7, Data Archiving.
- 2.4.8 Alpha has established a record management system for control of laboratory notebooks, instrument logbooks, standard logbooks, and other records for data reduction, validation storage and reporting.
- 2.4.9 Access to archival information is documented with an access log. Archival records are protected off-site, with a company specializing in record management, against fire, theft, loss and environmental deterioration, vermin and, in the case of electronic records, electronic or magnetic sources.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: DOCUMENT CONTROL FILING SYSTEM**

Description: The following procedure is a system designed for identification, collection, indexing, accessing, filing, storing, maintenance and disposal of quality and technical records.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 Our goal is to keep one year of the current past files in our main filing room. All older files are to be sent to Iron Mountain storage.

2.2 Access Log

2.2.1 Access to archival information is documented with an access log. Archival records are protected off-site, with a company specializing in record management, against fire, theft, loss and environmental deterioration, vermin and, in the case of electronic records, electronic or magnetic sources.

2.2.2 The access log also known as the "File Check-out and Return" log book maintained in the main file room.

Any person who retrieves a file is required to sign the access log. When the file is returned, the person returning the file is also required to sign the log to indicate the file was returned.

There are only two people allowed to re-file: Document Control Officer, and Assistant Document Control Officer.

We have specific, designated areas in our main file room for:

- Files that need to be bound,
- Returned files from cabinets,
- New folders ready to be filed in the cabinets.

2.3 Data Retention of Laboratory Support Activities

A description of our data retention policies and procedures is found in SOP D.1, Document Control Plan. The following list describes the activities that are documented and retained.

- a) all original raw data, used for calibrations, samples and quality control measures, including analysts' work sheets and data output records (chromatograms and other instrument response readout records);
- b) the analytical method SOP used for a specific test method to include a

description of the specific computational steps used to translate parametric observations into a reportable analytical value;

- c) copies of final reports;
- d) archived SOPs;
- e) correspondence relating to laboratory activities for a specific project;
- f) all corrective action reports, audits and audit responses;
- g) proficiency test results and raw data; and
- h) results of data review, verification, and cross checking procedures.

#### 2.4 Data Retention of Analytical Records

The essential information associated with data analysis, such as chromatograms, tabular printouts, computer files, analytical notebooks, and run logs are also retained. The specific information retained is as follows:

- a) laboratory sample ID;
- b) date of analysis and time of analysis, if the holding time is 72 hours or less, or when time critical steps are included in the analysis;
- c) instrument identification and instrument operating conditions/parameters;
- d) analysis type;
- e) all manual calculations, e.g., manual integrations;
- f) analyst's or operators's initials/signature;
- g) sample preparation including cleanup, separation protocols, sample IDs, volumes, weights, instrument printouts, meter readings, calculations, reagents;
- h) sample analysis;
- i) standard and reagent origin, receipt, preparation, and use;
- j) calibration criteria, frequency and acceptance criteria;

- k) data and statistical calculations, review, confirmation, interpretation, assessment and reporting conventions;
- l) quality control protocols and assessment;
- m) electronic data, software documentation and verification, software and hardware audits, backups; and
- n) method performance criteria including expected quality control requirements.

## 2.5 Data Retention of Administrative Records

The following administrative records are also retained:

- a) personnel qualifications, experience and training records;
- b) records of demonstration of capability for each analyst; and
- c) a log of names, initial and signatures for all individuals who are responsible for signing or initialing any laboratory record.

## 2.6 Collating Client Folders

The retention of data records mentioned above are retained in a number of locations. A large portion of the data associated directly with the final analytical data results is stored in the client file.

2.6.1 The following information is stored, maintained and retained in the client files as follows:

- Computer generated chain of custody,
- Client chain of custody,
- Sub-contract lab chain of custody (if present),
- Internal chain of custody (if present)
- Work order information (if present),
- Sample Receipt checklist,
- Sample Receipt checklist fax confirmation,
- Final Alpha reports,
- Alpha QA/QC (if present),
- Copy of EDD report (if present),
- Final sub-contract lab reports (if present),
- Sub-contract lab QA/QC (if present),
- Alpha invoice-always make sure than an invoice is present,
- Sub-contract lab invoices (if present),
- Raw Data,

- Final report raw data. The initials of the analyst indicating final data should be indicated somewhere on all of the sheets
- Screen Reports, re-runs, etc. This is indicated on the top sheet
- Air bills,
- Correspondence, and
- Report of e-mail (pdf file) confirmation.

2.6.2 If there were amendments made to the work order, put the amended COC on top of the previous COCs. Put amended final reports on top of other lab reports.

2.7 Files are scanned as pdf files and hard copy data is archived and sent to Iron Mountain Storage. Our account number is: NR007.

We receive a print out, upon request, from Iron Mountain indicating the box number and a description of the contents of the box, both of which we assign.

All data can be accessed from the computer as an electronic copy scanned in as a pdf file.

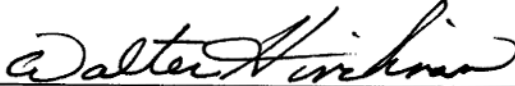
2.8 Assigning Box Numbers for Archiving Data

We assign each box a prefix and a number which helps identify the types of files located within that box.

AG = Accounting Files  
CF = Regular Customer Files  
DC = Data Control Files  
SP = Special Files  
QC = QC Files

2.6 A bar code system is used to identify archived data. We assign a box number and description to each box. A bar code sticker is affixed to the box and a records Transmittal Sheet (both provided by Iron Mountain) to record the bar code number is documented. A photocopy of the Records Transmittal Sheet is maintained for our records.

2.7 To locate a particular file, review the Iron Mountain Archives (Master File) and complete a request form which is faxed to Iron Mountain. The requested file is typically retrieved within a day.

Approved By:   
Walter Hinchman  
QA Officer

Date: January, 2007



**Table D.2-1**

<b>REPORT NUMBER</b>	<b>NAME</b>	<b>DATE CHECKED-OUT</b>	<b>DATE RETURNED</b>

**1.0 TITLE: LABORATORY SAMPLE DOCUMENT FLOW**

Description: This flow diagram is to assist all personnel in understanding document production flow from sample receipt through analysis to the final report.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Sample Receipt - Sample Custodian**

Documents: Chain-of-Custody, sample receipt checklist, air bills/shipping manifests, sample receipt log, and Field Sampling Book.

**2.2 Sample Storage - Sample Custodian**

Documents: Refrigerator Temperature Log and evidentiary COC records.

**2.3 Sample Preparations - Extraction Chemist**

Documents: Semi-volatile extraction logbook, volatile sample extraction logbook, metals digestion logbook, standards reagent logbook, and balance logbooks.

**2.4 Sample Analysis - Analyst**

Documents: Instrument sequence logs, maintenance logs, quantitation reports, chromatograms, calculations, draft reports, instrument document control logbook, calibration reports, tunes, corrective action reports, and QC data reports.

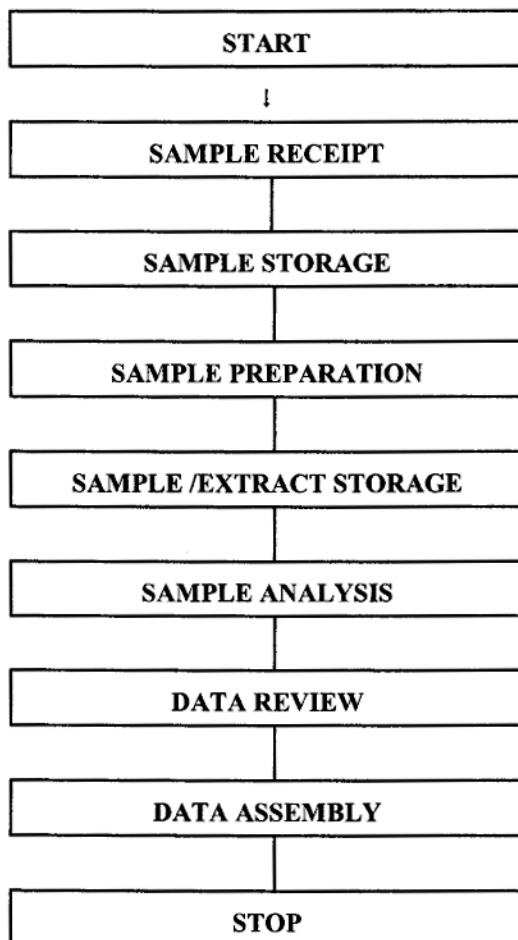
**2.5 Data Review - Lab Director/QA officer.**

Documents: All previous records, plus: final reports, archived SOPs, correspondence relating to laboratory activities, corrective action reports, proficiency test results and associated raw data and results of data review, verification and cross checking procedures.

**2.6 Assembly - Document Control Officer**

Documents: All

**FIGURE D.3 -1**  
**DOCUMENT FLOW PROGRAM**



Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: PROCEDURE FOR THE PREPARATION, REVIEW, APPROVAL, REVISION AND DISTRIBUTION OF SOP'S AND OTHER TECHNICAL DOCUMENTS.**

Description: It is of fundamental importance an a primary objective of our QA manual that SOP's and other technical documents follow strict adherence to established guidelines for their preparation, review, approval, revision and distribution in maintaining uniformity and consistency of laboratory activities and the documentation that these procedures were being followed historically.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 This procedure is intended to ensure that all records required under the National Environmental Laboratory Conference (NELAC), Chapter 5 Systems, are retained, as well as establishing the procedures for the control and documentation through a document control system.

2.2 This procedure ensures that all standard operating procedures (SOPs), manuals, or other technical documents clearly indicates the time period for which the procedure or document was in force.

2.3 SOP formats are modeled after suggestions from the National Environmental Laboratory Conference (NELAC). In addition, the USEPA, Quality Assurance Management Staff (QAMMS), has written guidelines and specifications for elements of a QA plan and recommended formats to be followed and, specifies how plans should be reviewed and approved.

**2.4 Document Control**

All quality system documents such as the QA Manual, SOPs, Quality Assurance Project Plans (QAPP's) and other technical documents are prepared using a document control format placed in the upper right-hand corner of each document page. This document control number uniquely identifies all quality system documents. The document control number includes the following information:

- Laboratory Name,
- Section Number,
- Revision Number,
- Date (of revision), and
- Page number and total pages of the document.

Document control is a key element for referencing and archiving active and historical technical procedures.

**2.5 Elements of a QA Manual and QAPP**

Each of the sixteen items listed below must be considered for inclusion in each QAPP and are the minimum requirements for a QA plan:

- 1) Title page with provision for approval signature;
- 2) Table of contents;
- 3) Project/plan description;
- 4) Project/plan organizations and responsibility;
- 5) QA objectives for measurements data in terms of precision, accuracy, completeness, representatives and comparability;
- 6) Sampling procedures;
- 7) Sampling custody;
- 8) Calibration procedures and frequency;
- 9) Analytical procedures;
- 10) Data reduction, validation and reporting;
- 11) Internal quality control checks and frequency;
- 12) Performance and system audits and frequency;
- 13) Preventive maintenance procedures and schedules;
- 14) Specific routine procedures to be used to assess data precision, accuracy and completeness of specific measurement parameters involved;
- 15) Corrective actions; and
- 16) Quality assurance reports to management.

## 2.6 Standard Operating Procedure (SOPs)

2.6.1 Analytical SOP's are written for each accredited test method. These are maintained in our Procedure Manual. The Procedure Manual consists of copies of the referenced test method as well as the in-house written analytical SOP. In cases where the published test method is ambiguous or provides insufficient detail, these cases are described in the SOPs as Clarification Boxes. The following elements are addressed and documented in each analytical SOP. They are as follows:

- 1) Identification of Test Method;
- 2) Applicable Matrix or Matrices;
- 3) Method Detection Limit;
- 4) Scope and Application;
- 5) Summary of Method;
- 6) Definitions;
- 7) Interferences;
- 8) Safety;
- 9) Equipment and Supplies;
- 10) Reagents and Standards;
- 11) Sample Collection, Preservation, and Storage;
- 12) Quality Control;
- 13) Calibration and Standardization;
- 14) Procedure;
- 15) Data Analysis and Calculations;

- 16) Method Performance;
- 17) Pollution Prevention;
- 18) Data Assessment and Acceptance Criteria of QC Measures;
- 19) Corrective Actions for Out-of-Control Data;
- 20) Contingencies for Handling Out-of-Control or Unacceptable Data;  
and
- 21) Waste Management.

2.6.2 Non-analytical SOPs are written primarily for those laboratory activities conducive to standardization and procedural formalization. They do not necessarily need to address all 21 elements required for an analytical SOP as stated above.

## 2.7 Responsibilities

### 2.7.1 Standard Operating Procedures

2.7.1.1 It is the responsibility of the Laboratory Director, QA Officer, Supervisors, and all affected personnel to discuss the need for SOP's and other technical documents.

2.7.1.2 It is the responsibility of the Laboratory Director, Laboratory Manager and QA Officer to ensure that all personnel, either technical or non-technical staff, are implementing the tasks described in an SOP, and/or are made individually aware that changes to an SOP has occurred.

2.7.1.3 SOP's and other technical documents will continue to be written revised and distributed as laboratory activities change with time. A copy (paper or electronic) of the updated SOP is made available in close proximity to the work area of the affected personnel.

### 2.7.2 QA Manual

2.7.2.1 The QA Officer is responsible for the preparation of a written QA Manual that involves all phases of environmental test measurements.

The QA Officer must ensure that the QA Manual contains procedures to document and report sufficient information for the assurance of precise, accurate and complete data generation.

2.7.2.2 It is the responsibility of the Laboratory Director, Laboratory Manager and QA Officer to ensure that all personnel, either technical or non-technical staff, are implementing the tasks described in the QA Manual, and/or are made individually aware that changes to the manual has occurred.

2.7.2.3 The QA Officer is responsible for the annual review of the QA Manual. However, the QA Manual may be revised more often and these revisions to the QA Manual are documented through the distribution of an Addendum to the QA Manual or the distribution of a new complete revision of the QA Manual.

2.7.2.4 The QA Officer is responsible for ensuring a copy (paper or electronic) of the QA Manual is made available in close proximity to the work area of the affected personnel.

### 2.7.3 QAPP's

2.7.3.1 Project Directors working in close coordination with the QA Officer have a responsibility to see that a written QA Project Plan is prepared for projects requesting site/project specific plans. The elements of a QAPP are typically identified from the QA Plan and written as a document independent from the QA Manual.

## 2.8 Plan Review, Approval and Distribution

### 2.8.1 Review and Approval

2.8.1.1 The review and approval of quality systems documents are conducted by the Laboratory Director, Laboratory Manager, supervisors and/or the QA Officer.

2.8.1.2 Changes to documents are reviewed and approved by the same function as that performed by the original review.

2.8.1.3 The reviewing person has the authority to access any of the quality system documents and may use any additional pertinent background information upon which to base the review and approval process.

Note: If practical, the altered document or new text may be identified in the document or appropriate attachments.

2.8.1.4 Completion of reviews and approvals are shown by signatures on the title page of the QA manual and/or a signature block at the end of the SOP.

2.8.1.5 The driving force behind revisions of the QA Manual or other technical documents are typically due to:

- 1) Changing regulations regarding certification;

- 2) Updates or revisions to existing methods of analysis;
- 3) The promulgations of new methods and/or lab procedures;
- 4) Expansions or contractions of analytical capabilities and/or services;
- 5) Changes in general laboratory policies or procedures; and
- 6) Recommendations and/or findings, resulting from on- site laboratory inspections from external and internal audits.

## 2.8.2 Distribution

2.8.2.1 The QA Officer is responsible for the distribution of the QA plan and associated SOP's. The QA plan will be numbered or inventoried to monitor the distribution of this publication and for the purpose of updating those distributed documents to the affected organizations.

2.8.2.2 The QA Officer is responsible for ensuring only authorized editions of the technical documents, to include the QA Manual and Analytical SOPs, are available at all locations where operations essential to the effective functioning of the laboratory are performed.

2.8.2.3 The QA Officer is responsible for promptly removing invalid or obsolete documents to ensure against their unintended use.

2.8.2.4 The QA Officer is responsible for retaining for either legal or knowledge preservation purposes, e.g., for their historical value, obsolete documents.

2.8.2.5 The QA Officer is responsible for maintaining a master list identifying the current status and distribution of documents in the quality system to preclude the use of invalid and/or obsolete documents, see Fig D.4-1.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007



**QA Manual  
Document Control  
Inventory Log**

**Alpha Analytical, Inc.**  
**Document: QA Manual**  
**Rev: 15.0**  
**Date: January, 2007**

<b>Name</b>	<b>Distribution Number</b>	<b>Date Signed Out</b>	<b>Initial</b>		<b>Date Signed In</b>	<b>Initial</b>
	1					
	2					
	3					
	4					
	5					
	6					
	7					
	8					
	9					
	10					
	11					
	12					
	13					
	14					
	15					

Fig D.4-1

**1.0 TITLE: LABORATORY TRAINING PROGRAM**

**2.0 STANDARD OPERATING PROCEDURE**

2.1 Scope and Application

2.1.1 Scope

This SOP outlines requirements for initial and continuing demonstration of staff proficiency in the methods or procedures which they perform. The ultimate goal of Alpha's Training Program is to provide comprehensive, effective, consistent and documented training for employees to ensure reliable and efficient job performance.

2.1.2 Application

Many factors determine the correctness and reliability of environmental test methods performed by Alpha. These factors include contributions from such things as:

- a) human factors;
- b) accommodation and environmental conditions;
- c) environmental test methods and method validation;
- d) equipment;
- e) measurement traceability;
- f) sampling; and
- g) the handling of samples.

The total contribution of these factors differ considerably between methods of analysis. Alpha takes these factors into account when developing analytical methods of analysis, the selection of the equipment and the training and the qualification of the personnel associated with individual methods of analysis. Therefore, our training program is customized to the individuals needs and the degree of difficulty of a particular procedure.

2.1.3 Definitions

Required reading- Reading of documents, procedures or publications identified by the Laboratory Director, QA Officer, Supervisor or Trainer as necessary to fulfill the duties of the position.

Method/SOP Training- Used to train analysts to perform analyses without the supervision of the Laboratory Director or another qualified analyst. Also used to train any employee to a task implemented through an approved SOP.  
Seminars- Classroom type training either led by a company representative or

analytical/supply vendor.

Work Cells - A work cell is considered to be all individuals who see a sample through the complete process of preparation, extraction and analysis.

## 2.2 Summary

- 2.2.1 Several types of training tools are used for staff training, including: 1) Required reading, 2) Method/SOP training, 3) Seminars, 4) Vendor supplied training, 5) Individual mentoring by a senior staff member, 6) MDL/IDC studies, 7) Performance Evaluation (PE) samples and 8) Multi-Media Interactive Computer Presentation.
- 2.2.2 Training files will be maintained by each employee for the current year and historical records are maintained by the Training Coordinator. These files will contain all associated training documents.
- 2.2.3 Safety, when not the topic of the training session, will be stressed during the applicable training sessions.
- 2.2.4 All training conducted by Alpha will be documented.

## 2.3 Responsibilities

### 2.3.1 Management Responsibilities

It is Alpha's policy that the Laboratory Director, Laboratory Manager, and QA Officer, is responsible for ensuring all staff members are trained accordingly. This is to include initial and on-going or continuing training. Management is responsible for ensuring that all staff members are implementing the tasks or procedures described in the QA Manual and analytical and extraction SOPs. It is managements responsibility to ensure that individual staff members are made aware of changes to an SOP if and when they occur. It is managements responsibility to ensure a copy (paper or electronic) of the updated SOP is available in close proximity to the work area.

### 2.3.2 Training Coordinator Responsibilities

The Training Coordinator is responsible for the implementation of the Training Program. The Training Coordinator will remind Trainers to conduct training for each staff member. It is the responsibility of the Training Coordinator to ensure that all historical staff training records are retained. It is the responsibility of the Training Coordinator to ensure that each staff member new and old are assigned the appropriate Trainers. It is the responsibility of the Training Coordinator to ensure that new staff members

have been given a laboratory orientation to include required documentation found in the Training SOP.

### 2.3.3 Trainer Responsibilities

Each staff member is placed under the direct supervision of a senior staff member upon entering employment at Alpha. The senior staff member will conduct an orientation meeting at which time a training guide listing those training requirements will be discussed. The trainer will provide the new staff member with a training log along with this training program document which will list their training on an annual basis. It is the responsibility of the senior staff member to ensure that assigned staff members under their guidance are given adequate training to perform their duties.

Trainers are generally assigned as follows:

QA Officer will oversee general QA procedures for all staff members.

LIMS Administrator will oversee all LIMS and PC training.

GC/MS Supervisor will oversee all GC/MS analysts.

GC/ECD/HPLC Supervisors will oversee all GC/ECD/HPLC analysts.

GC/FID Supervisors will oversee all GC/FID analysts/extraction chemists.

VOC Prep Supervisor will oversee all VOC extraction chemists.

SV Prep Supervisors will oversee all SV extraction chemists.

Sample Custodian will oversee all personnel associated with these duties.

Safety Officer will oversee safety training for all employees.

### 2.3.4 Trainee Responsibility

It is the responsibility of all Trainees to ensure that they have been adequately trained in their duties as prescribed in the various SOPs. It is the responsibility of the Trainees to ensure that all training, to include initial and continuing training, has been properly documented by the Trainer and Training Coordinator.

## 2.4 Annual Training Requirements

- 2.4.1 Each staff member who performs sample preparation, analytical procedures or is a technical staff member of a work cell must demonstrate his/her ability to successfully execute each method. In order to qualify to perform a given analytical method, those staff member must, at a minimum, successfully process an IDC study. After initial qualification on a given method, staff proficiency must be demonstrated on a continuing basis. At least annually each staff member who performs a method must successfully produce a continuing demonstration of capability study.

#### 2.4.2 Laboratory Ethics/Fraud Prevention and Data Integrity Program

Training courses in ethical and legal responsibilities including the potential punishment and penalties for improper, unethical or illegal actions are conducted and documented on an annual basis.

#### 2.4.3 Laboratory Ethics/Fraud Prevention and Manual Integration Program

Training is conducted annually for all analysts regarding manual integration. This training outlines the use of manual integrations, how manual integrations are to be produced, documentation of manual integrations, analysts responsibilities and the potential penalty for improper, and/or unethical use of manual integrations.

### 2.5 Procedure

#### 2.5.1 Seminars

Seminars are used to train employees on general topics such as health and safety. Seminars are generally conducted by the Laboratory Director, QA Officer or a Laboratory Supervisor.

Seminar topics and course objectives will be approved by the Laboratory Director, QA Officer or Supervisors.

Seminar attendance will be assigned by the Laboratory Director, QA Officer or Supervisors, and notices will be sent to all affected employees.

Upon completion of the seminar, the individual conducting the seminar will ensure the seminar is documented on the individual training records of all attendees.

#### 2.5.2 Required Reading

Required reading is used to convey important information to the staff. Required reading is of two types: 1) reading which does not require attendance at a seminar or "hands on" experience, and 2) reading which is part of mandatory attendance at a seminar. Typical required reading would be such things as : 1) EPA regulatory information, 2) company policies, 3) technical articles, and 4) SOPs which the employee performs. The trainer will document these requirements.

#### 2.5.3 Method/SOP Training

This form of training is used for new employees, training of analysts for new

methods and SOPs, and on-going training. The training is documented on their individual training records.

#### 2.5.4 Vendor Training

Employees may be trained by a Vendor representative either at our facility or off-site. If this training is conducted at our facility, the training is documented on their individual training record.

### 2.6 Training

#### 2.6.1 General

Each new employee is given a personal laboratory notebook. This notebook is a blank standard notebook for employees to use as they deem necessary. Alpha encourages employees to write notes in this logbook which will help in any way in the understanding of required material. This logbook is a notebook to be referenced, when necessary.

In addition, each employee is given a Personal Training Logbook. This is a standardized logbook used by all employees to document laboratory wide required training. General topics to be included in a Personal Training Logbook would include:

- Dates and descriptions of training,
- Detailed information regarding the training topics discussed,
- Changes to a method/SOP or general laboratory practice or policy,
- Audit deficiencies, and
- Documentation of annual training.

All training will be assigned by the Laboratory Director, QA Officer or Supervisor. All new employees will be under the direct supervision of a senior staff member for their initial and continuing training. SOP training for non-analysts shall be under the direct supervision of a senior staff member.

#### 2.6.2 Initial Demonstration of Proficiency

A new analyst/extraction chemist is one who has not met the training requirements for a method or any analyst/extraction chemist who has not performed the method for greater than 1 year. Before a new analyst/extraction chemist can perform work independently on a method, the following steps should be completed:

- The analyst/extraction chemist will read all pertinent QA and method

SOPs, EPA methods, manual or other documents as assigned by the trainer.

- The analyst will perform a minimum of 2-4 analytical sequences under the direct supervision of an experienced analyst.
- The extraction chemist will perform a minimum of 2-4 extraction batches under the direct supervision of an experienced extraction chemist.
- The analyst will complete the outlined training tasks, including an MDL/IDC study. See QAM, Vol II, Appendix E.8 for additional details.

### 2.6.3 Continued Demonstration of Proficiency

Each analyst/extraction chemist who has met the initial training criteria for a method must demonstrate and document continued proficiency by at least one of the following once per year:

- Acceptable performance of a blind sample.
- Another demonstration of capability, i.e. in-house PE.
- Successful analysis of a blind performance sample on a similar test method using the same technology (e.g., GC/MS Volatiles by purge and trap for methods by either 524.2, 624 or 5035/8260).
- At least 4 consecutive laboratory control samples with acceptable levels of precision and accuracy.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

## Training/Continuing Education Program

### Statement of Training Responsibilities:

I have received, read, understood and will implement all relevant items described in the QA Manual, Revision No. 15.0 and all relevant addendums.

**Signature:** \_\_\_\_\_

**Employee:** \_\_\_\_\_

**Years in Service:** \_\_\_\_\_

**Department:** \_\_\_\_\_

**Date:** \_\_\_\_\_

**Issued:** \_\_\_\_\_

**Supervisor:** \_\_\_\_\_

#	Training Code	Description	Training Time	Date Completed	Employee Initial	Trainer Initial
1	RR/SEM	<b>QAM Vol. I, Section 3.0</b> General Statement of Policy				
2	RR/SEM	<b>QAM Vol. I, Section 4.0</b> Organization and Responsibility				
3	RR/SEM	<b>QAM Vol. I, Section 5.0</b> Quality Assurance Objectives				
4	RR/SEM	<b>QAM Vol. I, Section 6.0</b> Sampling Procedures				
5	RR/SEM	<b>QAM Vol. I, Section 7.0</b> Sample Custody				
6	RR/SEM	<b>QAM Vol. I, Section 8.0</b> Analytical Procedures				
7	RR/SEM	<b>QAM Vol. I, Section 9.0</b> Calibration Procedures and Frequency				
8	RR/SEM	<b>QAM Vol. I, Section 10.0</b> Instrument Maintenance				
9	RR/SEM	<b>QAM Vol. I, Section 11.0</b> Quality Control Checks				
10	RR/SEM	<b>QAM Vol. I, Section 12.0</b> Data Reduction, Validation and Reporting				
11	RR/SEM	<b>QAM Vol. I, Section 13.0</b> Corrective Actions				
12	RR/SEM	<b>QAM Vol. I, Section 14.0</b> Performance and System Audits				
13	RR/SEM	<b>QAM Vol. I, Section 15.0</b> Quality Assurance Reports				
14	RR/SEM	<b>QAM Vol. I, Section 16.0</b> Laboratory Reports				



Continuing Education Program

Employee: \_\_\_\_\_

Issued: \_\_\_\_\_

#	Training Code	Description	Training Time	Date Completed	Employee Initial	Trainer Initial
1						
2						
3						
4						
5						
6						
7						
8						
9						
10						
11						
12						
13						
14						
15						
16						
17						
18						
19						
20						
21						
22						
23						
24						
25						

Continuing Education Program

Employee: \_\_\_\_\_

Issued: \_\_\_\_\_

#	Training Code	Description	Training Time	Date Completed	Employee Initial	Trainer Initial
26						
27						
28						
29						
30						
31						
32						
33						
34						
35						
36						
37						
38						
39						
40						
41						
42						
43						
44						
45						
46						
47						
48						
49						
50						

Continuing Education Program

Employee: \_\_\_\_\_

Issued: \_\_\_\_\_

#	Training Code	Description	Training Time	Date Completed	Employee Initial	Trainer Initial
51						
52						
53						
54						
55						
56						
57						
58						
59						
60						
61						
62						
63						
64						
65						
66						
67						
68						
69						
70						
71						
72						
73						
74						
75						

Continuing Education Program

Employee: \_\_\_\_\_

Issued: \_\_\_\_\_

#	Training Code	Description	Training Time	Date Completed	Employee Initial	Trainer Initial
76						
77						
78						
79						
80						
81						
82						
83						
84						
85						
86						
87						
88						
89						
90						
91						
92						
93						
94						
95						
96						
97						
98						
99						
100						

**Certification  
of  
Initial/Continued Proficiency Requirement**

Alpha Analytical, Inc.  
255 Glendale Ave., Ste 21  
Sparks, NV 89431

I the undersigned, **CERTIFY** that:

1) I have read, understood, and agree to perform the most current version of the test method, standard operating procedure and to keep current the initial and continued proficiency requirements for the following:

Method	SOP			Initial
	Doc No	Revision	Date Issued	
EPA Method 524.2	E.20			
EPA Method 608/8081A	E.30			
EPA Method 8082	E.31			
EPA Method 624/8260B	E.33			
EPA Method 625/8270C	E.34			
EPA Method 8270C SIMs	E.35			
EPA Method 8260B SIMs	E.36			
EPA Method 8015B-DRO/DHS-LUFT/NW-TPHdx	E.37			
EPA Method 8015B-GRO/DHS-LUFT/NW-TPHgx	E.38			
EPA Method 1311 TCLP	E.50			
EPA Method 1312 SPLP	E.51			
DHS Method (WET) Waste Extraction Test	E.52			
EPA Method 3545 (ASE) Accelerated Solvent Extraction	E.55			
EPA Method 3510 Separatory Funnel	E.56			
Organic Acids	E.64			
EPA Method 300.0/9056	E.65			
EPA Method 314.0	E.66			
EPA Method 415.1/9060	E.67			
Methane/Ethane	E.68			

\_\_\_\_\_  
Name

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

**Certification  
of  
Initial/Continued Proficiency Requirement**

Alpha Analytical, Inc.  
255 Glendale Ave., Ste 21  
Sparks, NV 89431

I the undersigned, **CERTIFY** that:

1) I have read, understood, and agree to perform the most current version of the test method, standard operating procedure and to keep current the initial and continued proficiency requirements for the following:

Method	SOP			Initial
	Doc No	Revision	Date Issued	
EPA Method 200.8/6020 (Metals)	E.60			
EPA Method 3015 Microwave Digestion	E.70			
EPA Method 3051 Microwave Digestion	E.71			
EPA Method 200.2/3010 Block Digestion	E.72			
EPA Method 120.1/9050A/SM2510B (Conductivity)	E.75			
EPA Method 150.1/9040B/9045C (pH)	E.76			
EPA Method 350.3 (Ammonia) 351.4 (TKN	E.77			
EPA Method 180.1 (Turbidity)	E.78			
EPA Method 160.1/SM2540C (TDS)	E.80			
EPA Method 160.2/SM2540D (TSS	E.80			
EPA Method 160.3/SM2540B (TS)	E.80			
ASTM 2216 (Percent Dry Weight and Percent Moisture)	E.81			
EPA Method 1664A (Oil and Grease)	E.82			
EPA Method 310.1/SM2320B (Alkalinity)	E.85			
EPA Method 305.1/SM2310B (Acidity)	E.85			
EPA Method 7196A/SM 3500-Cr D	E.90			
EPA Method 376.2/SM4500-S D (Sulfide)	E.92			
EPA Method 330.5/SM4500-Cl G (Residual Chlorine)	E.93			
EPA Method 410.4/SM 5520-D (COD)	E.94			
EPA Method 365.2/Hach8190 (Total Phosphorus)	E.95			
Standard Method SM3500-Fe D	E.96			

\_\_\_\_\_  
Name

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

**Certification  
of  
Education/Technical Background**

Alpha Analytical, Inc.  
255 Glendale Ave., Ste 21  
Sparks, NV 89431

I the undersigned, **CERTIFY** that:

- 1) The educational and/or training requirements as specified in Alpha's QAM have been meet.
- 2) The representation of my college degree and/or additional training is correct.

\_\_\_\_\_  
Degree

\_\_\_\_\_  
School of Degree

\_\_\_\_\_  
Date

\_\_\_\_\_  
Name

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

## DATA



## SUMMARY DATA

[illegible]

## SUMMARY DATA

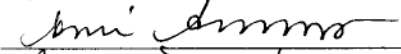

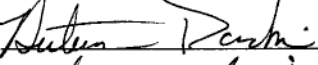

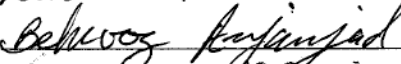

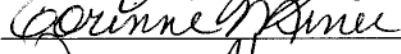

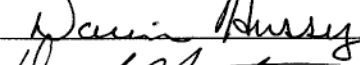



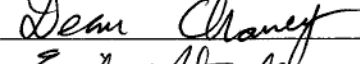

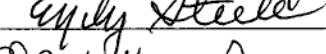



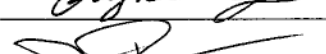









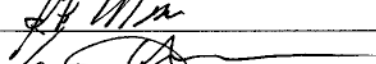



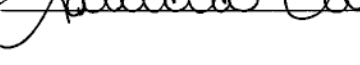



[illegible]

**1.0 TITLE: EMPLOYEE SIGNATURE/INITIAL LOG**

Description: An employee signature and initial log is recorded and maintained by Alpha in order to identify personnel from their initials or signatures on laboratory documents. The list contains the analyst's typed names, initials, written signatures, and written initials.

**TABLE D.6 - 1**

**LABORATORY SIGNATURE LIST**

<u>Name</u>	<u>Initials</u>	<u>Signature</u>	<u>Initials</u>
Ami Awano	AA		
Autumn Durkin	AD		
Behrooz Aryajinejad	BA		
Corinne Miner	CM		
Darin Hussey	DH		
David Maestas	DM		
Dean Chaney	DC		
Emily Steele	ES		
Elizabeth Sauvageau	ES		
Guanzhen Ji	GJ		
Joseph Blaylock	JB		
Jennifer Webster	JW		
Jeff Yoshimoto	JY		
John C Vasquez	JV		
Joseph T Simonsen	JT		
Joseph Mauro	JM		
Jason Herrmann	JH		
Latricia Edrosa	LE		

<u>Name</u>	<u>Initials</u>	<u>Signature</u>	<u>Initials</u>
Michael Aseltine	MA	<i>Michael Aseltine</i>	<i>MA</i>
Melanie Wickham	MW	<i>Melanie Wickham</i>	<i>MW</i>
Megan Leslie	ML	<i>Megan Leslie</i>	<i>ML</i>
Matthew Weikel	MW	<i>Matthew Weikel</i>	<i>MW</i>
Rickey L Overton	RO	<i>Rickey L Overton</i>	<i>RO</i>
Randy Gardner	RG	<i>Randy Gardner</i>	<i>RG</i>
Roger L. Scholl	RS	<i>Roger Scholl</i>	<i>RS</i>
Reyna Vallejo	RV	<i>Reyna Vallejo</i>	<i>RV</i>
Tammy Brace	TB	<i>Tammy Brace</i>	<i>TB</i>
Tara Dickinson	TD	<i>Tara Dickinson</i>	<i>TD</i>
Tasha Pascal	TP	<i>Tasha Pascal</i>	<i>TP</i>
Tom Wickham	TW	<i>Tom Wickham</i>	<i>TW</i>
Tara Manzini	TM	<i>Tara Manzini</i>	<i>TM</i>
Tracy Degiovanni	TD	<i>Tracy Degiovanni</i>	<i>TD</i>
Walter Hinchman	WH	<i>Walter Hinchman</i>	<i>WH</i>
Wei Wu	WW	<i>Wei Wu</i>	<i>WW</i>
Zhen Jin	ZJ	<i>Zhen Jin</i>	<i>ZJ</i>

Approved By: *Walter Hinchman* Date: January, 2007  
 Walter Hinchman  
 QA Officer

**1.0 TITLE: INSTRUMENT SEQUENCE LOGBOOK RECORD KEEPING  
PROCEDURE**

Description: Quality control can only be proven by the documentation that it exists. QC data, therefore, is fundamental in the operation, implementation and maintenance of a QA plan. The need for a simple yet efficient procedure for this activity is paramount in reconstructing historical analytical records.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 For each instrument there is an associated analytical data control record keeping file system. This record keeping file system is uniquely associated with a particular instrument for the sole purpose of reconstructing historical analytical records.
- 2.2 Each analytical run made on an instrument is partially or completely documented in the Instrument Sequence Logbook. The following is a description of how to maintain and document these analytical runs.
- 2.3 At the end of the analytical sequence a photocopy of the instrument sequence logbook is placed at the beginning of the analytical quantitation reports. This serves as an index to the data which follows.
- 2.4 Analytical runs are placed in chronological order as they were analyzed on the analytical sequence log.
- 2.5 Final data, which is placed in the client file, need not be photocopied. However, data such as CV's, tunes, blanks and other QC data should be included in the logbook with the analytical sequence data.
- 2.6 The only exception to this policy is the final data which is placed into the client file.
- 2.7 The reconstruction of an analytical sequence(s) is easily accomplished by the use of this record keeping system and the associated client files.
- 2.8 This record keeping system also serves as a secondary backup to the original instrument logbook.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: INSTRUMENT SEQUENCE AND MAINTENANCE LOGBOOK**

Description: All analytical instruments have two separate log books: 1) the Instrument Sequence Logbook; 2) and the Maintenance Logbook. The log books are annotated by the analyst to track and monitor all activities associated with that particular instrument. Complete documentation is a mandatory criteria that all personnel abide by

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Instrument Sequence Logbook**

This is not a complete list of all the types of activities which can be documented in the log book. However, the following is a list of the minimum information which should be annotated on a daily basis.

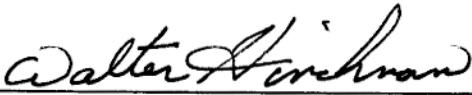
- o Date;
- o Instrument conditions, if they have been changed from the previous days conditions;
- o Minor instrument adjustments such as changing the multiplier etc.;
- o Full descriptions of each analytical run including: sample/client identification, amount injected or purged, analytical run# or file ID;
- o Initials on each page; and
- o If the instrument is operating other than normal, state conditions.

**2.2 Instrument Maintenance Logbook**

Any type and description of the maintenance performed on an instrument should be annotated in this book. There are no required information; however, common sense should indicate the types of entries that should be placed into this logbook.

**2.3** All log book entries are documented with the date and the initials of the person making the entry.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

## **1.0 TITLE: ANALYTICAL BALANCE LOGBOOK**

Description: Balances are the primary analytical test instrument for several gravimetric methods of analysis and are crucial instruments in support of various other analytical methods. In either case, final quantitative results are dependent on their accuracy and the need to track, monitor, and document their status is important to maintaining data integrity.

## **2.0 STANDARD OPERATING PROCEDURE**

The following is a description of practices and procedures used to monitor analytical balance accuracy and how this information is documented.

### **2.1 Daily Monitoring and Accuracy Checks**

2.1.1 Purpose - to ensure quantitative results produced by the analytical balances are accurate.

2.1.2 Frequency - Daily or before use with S - class weights.

2.1.3 Procedure - Daily balance checks, using two S class weights chosen to bracket the anticipated target weight, are recorded in the balance logbook.

2.1.4 Acceptance Criteria - The results of the daily accuracy check must be within the general performance criteria of  $\pm 0.1\%$  of the targeted weight value or  $\pm 0.5\text{mg}$ , whichever is larger, unless method-specified guidance exists.

2.1.5 Corrective Action - If the results are not within the specified guidelines:

- i. the balance is removed from service until repaired; or
- ii. the balance can be used if a correction factor can be reliably established and maintained.

### **2.2 Biannual Scheduled Maintenance and Traceability Check**

2.2.1 Purpose - required to ensure quantitative results are accurate using second source balance weights to verify accuracy.

2.2.2 Frequency - twice a year (biannually).

2.2.3 Procedure

2.2.3.1 A analytical balance service company is contracted to provide biannual preventative and scheduled maintenance. This service call

includes cleaning, lubricating, and adjusting all balances to the original manufactures specifications and testing for errors. In addition, built in weights are cleaned and tested for errors.

2.2.3.2 After balances have been serviced, they are calibrated and checked against NIST traceable weights to verify accuracy. The calibration also includes a linearity and cornerload check.

2.2.3.3 Balances are then certified for use with letters certifying accuracy.

2.2.3.4 All records of repair and maintenance activities, including the documentation of certificates of accuracy, are kept by a lab technician responsible for these duties.

2.2.4 Acceptance Criteria - Manufacturers specification.

2.2.5 Corrective Action - If the results are not within the specified guidelines:

- i. the balance is removed from service until repaired; or
- ii. the balance can be used if a correction factor can be reliably established and maintained.

2.3 S Class Weights.

2.3.1 S class weights used for daily balance accuracy check are certified at a minimum every 5 years for accuracy.

2.3.2 These weights are sent to a company certified under ISO/IEC 17025-1999 ANSI/NCSL Z540-1-1994 program to perform this procedure.

2.3.2 They use reference standards traceable to NIST using NIST IR 6969 SOP 4 Double Substitution Weighing Design Protocol.



**TABLE D.9-1**  
**GENERAL GUIDELINES FOR BALANCE WEIGHT ACCEPTABILITY**

Balance Model	Location	Alpha ID	Sensitivity	True Weight	Range of Acceptability	Criteria
Mettler AB-204S	organic ext lab	1A	0.0001g ( $\pm$ 0.1mg)	0.0050 g 5.00 g	0.0052 - 0.0048 g 5.005 - 4.995 g	4 % * 0.1 %
Mettler PM-300	organic ext lab	2A	0.01 g ( $\pm$ 10mg)	0.500 g 200.0 g	0.49 - 0.51 g 199.8 - 200.2 g	2 % ** 0.1%
Mettler PB602S	volatile prep lab	3A	0.0001g ( $\pm$ 0.1mg)	5.00 g 50.0 g	5.005 - 4.995 g 50.05 - 49.95 g	0.1 % 0.1 %
Mettler AJ-100	volatile prep lab	4A	0.0001g ( $\pm$ 0.1mg)	0.0050 g 5.00 g	0.0052 - 0.0048 g 5.005 - 4.995 g	4 % * 0.1 %
Mettler AB-204S	inorganic wet chem	5A	0.0001g ( $\pm$ 0.1mg)	5.00 g 50.0 g	5.005 - 4.995 g 50.05 - 49.95 g	0.1 % 0.1 %
Mettler AB-204S	inorganic Metals	6A	0.0001g ( $\pm$ 0.1mg)	5.00 g 50.0 g	5.005 - 4.995 g 50.05 - 49.95 g	0.1 % 0.1 %
Mettler AB-204S	inorganic gravimetric	7A	0.0001g ( $\pm$ 0.1mg)	5.00 g 50.0 g	5.005 - 4.995 g 50.05 - 49.95 g	0.1 % 0.1 %
Sartorius BP-310S	TPH-D ext lab	8A	0.001g ( $\pm$ 1mg)	0.500 g 50.0 g	0.5005 - 0.4995 g 50.05 - 49.95 g	0.1 % 0.1 %
Mettler AB-204S	TPH-D ext lab	9A	0.0001g ( $\pm$ 0.1mg)	0.500 g 50.0 g	0.5005 - 0.4995 g 50.05 - 49.95 g	0.1 % 0.1 %

Note: \* Criteria is to 0.1% or 0.5mg.

Note: \*\* Criteria check is at its limit of sensitivity.

Approved By: \_\_\_\_\_

*Walter Hinchman*  
Walter Hinchman  
QA Officer

Date: January, 2007

## **1.0 TITLE: SAMPLE EXTRACTION LOG**

Description: All sample preparations are documented in the Sample Extraction Logbook (Figure D.10-1). Any changes in the protocol during preparation must be authorized by the Laboratory Director.

## **2.0 STANDARD OPERATING PROCEDURE**

2.1 The following information is recorded by the LIM System:

- o Batch No.;
- o Matrix;
- o Analysis;
- o Date;
- o Page;
- o Extraction method;
- o Cleanup procedure;
- o Solvent;
- o Initials of extraction chemist;
- o Crucible number if applicable;
- o AAI Lab ID No.;
- o Client ID;
- o Sample weight information;
- o Remarks on odor or color, etc.;
- o Information on surrogates or spike solution added; and
- o Information on any specific difficulties ( i.e. emulsions) with sample.

2.2 The extraction or digestion procedure used is referenced, including detailed extraction and sample preparation methodology with adequate information to repeat the procedure at a later date.

- o 3510 - Separatory funnel extraction;
- o 3545 - Accelerated solvent extraction;
- o 3550 - Sonication extraction;
- o 5035 - Closed system purge-and-trap of soils;
- o 200.2 - SDWA metals block digestion;
- o 3015 - Microwave assisted digestion of waters; and
- o 3051 - Microwave assisted digestion of soils.

2.3 The sample cleanup procedure used is referenced such as:

- o (AW) Sulfuric acid cleanup (SW 3665A);
- o (SG) Silica gel cleanup (SW 3630C);
- o (SGM) Modified silica gel cleanup (NWTPH-dx);
- o (Al) Alumina cleanup (SW3610B);

- (Al-TPH) Alumin cleanup for TPH) (SW3611B); and,
  - (Ba/Ag/H) Perchlorate pre-treatment cleanup (314.0 x-up).
- 2.4 At the end of an extraction, the extracted material is labeled with the AAI sample number, solvent, date, initial and analytical method.
- 2.5 If an error is made, a single line is drawn through the information, dated and initialed.
- 2.6 LIMS Procedure for starting a Preparation Batch
- 2.6.1 Single click left button of mouse on “sample prep”;
  - 2.6.2 Click left button of mouse on “Add” in the upper right corner of the screen;
  - 2.6.3 Type in the method number followed by an underline and P or B (no spaces) in the space provided , designated “Prep Code” ie 515\_P or use select key;
  - 2.6.4 Choose the name of the technician, who will be extracting these samples, by clicking on the downward pointing arrow adjacent “technician”;
  - 2.6.5 If all samples available are to be extracted click on “User Select”. If samples are not to be extracted deleted them from the user select screen.
  - 2.6.6 Type in the necessary number of matrix spike samples in this column by typing the client ID, followed by the client number, and MS, ie 07101540-01 AMS;
  - 2.6.7 Click on “Reagents/Spikes” positioned left of center at the top of the screen;
- 2.7 After all spikes and surrogates are recorded click “main” at the top left of the screen to return to the main page of this prep Batch;
- 2.8 Choose an “end date” for when this batch will be ready for analysis and enter “end date”
- 2.9 Exit the file by using the icon in the upper left hand corner. Upon exiting this file the computer will ask to “complete the status of this Prep Batch report,” yes or no. Once this has been completed, the Prep Batch and all associated samples are annotated in the LIMS system.

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

Technician: Behrooz Aryanejad  
 Prep Batch 16151 Prep Code: 3510 B  
 Prep Start Date: 11/16/2006 10:44:41 AM  
 Prep End Date: 11/16/2006 5:05:23 PM

#	Laboratory Sample ID	Bottle #	Client Sample ID	Matrix	Initial pH	Final pH	Base pH	Acid pH	Int wt (g)	Int vol (mL)	Spike Added (mL)	Fin Vol (mL)	factor
1	MBLK-16151	1			6.0		13.0	1.9		500.0		20.0	0.04
2	LCS-16151	1			6.0		13.0	1.9		500.0		20.0	0.04
3	LCSD-16151	1			6.0		13.0	1.9		500.0		20.0	0.04
4	SEM06111620-01A	1	S200611-0870 - Plant Effluent	Aqueous	6.0		13.0	1.6		500.0		20.0	0.04
5	AAI06102243-05A	1	Composite 1-4 WP028 (SVOCs)	Aqueous	6.0		13.0	1.6		500.0		20.0	0.04
6	AAI06102243-05ADUP	1	Composite 1-4 WP028 (SVOCs)	Aqueous	6.0		13.0	1.6		500.0		20.0	0.04
7	QTLY QC CHK	1	Composite 1-4 LT#021606 PT#38069		6.0		13.0	1.6		500.0		20.0	0.04
8	QTLY QC CHKDUP	1	Composite 1-4 LT#021606 PT#38069		6.0		13.0	1.6		500.0		20.0	0.04
9	SEM06111620-01AMS	1	S200611-0870 - Plant Effluent	Aqueous	6.0		13.0	1.6		500.0		20.0	0.04

11/16/06  
 ef

#	Reagent ID	Reagent Name
1	NanoPure	NanoPure Water
2	Dichloromethane	Dichloromethane : Lot #CQ376
3	HCL (10N)	Hydrochloric Acid LOT:45003
4	Na2SO4	BAKED SODIUM SULFATE : EL060602-01
5	NaOH (10N)	NaOH EL060526-01

#	Spike ID	Spike Name and Concentration	AmtAdd (uL)
1	MS060929-01	8270 SURROG ACID: 200 UG/ML B/N: 100	500
2	MS060126-02	NSI 8270 SPIKE 100 UG/ML	500
3	MS060913-01	8270 SPIKE BENZID 100 UG/ML	500

**1.0 TITLE: ANNUAL THERMOMETER CALIBRATION PROCEDURE**

References: Method SM250B: Determination of Temperature Standard Methods for the Examination of Water and Wastewater, 20<sup>th</sup> Edition, 1998.

**2.0 Standard Operating Procedure:**

- 2.1 Temperature readings are used to monitor the thermal preservation of all samples received at the laboratory and for environmental samples stored in the laboratory. In addition, thermometers are used in a variety of other analytical techniques in the laboratory such as in the calculation of alkalinity, in studies of saturation and stability with respect to calcium carbonate, etc.
- 2.2 For laboratory operations which report temperature, and for reference thermometers, temperature measurements should be made preferably with a spirit-filled or digital thermometer. The spirit-filled thermometer should have a scale marked for every 0.1 to 0.2°C, with markings etched for glass capillary thermometers.
- 2.3 Glass thermometers should be used according to the type of thermometer.
  - 2.3.1 Partial immersion thermometers - are immersed in the fluid to the level (mm) marked on the stem or to the inscribed ring.
  - 2.3.2 Total immersion thermometers - are immersed in the fluid to the level of the temperature being measured. For example, if you are checking a temperature of 35° C, the 35° C increment of the thermometer stem and mercury level will be just below or above the top of the fluid.
  - 2.3.3 A complete immersion thermometer - would be completely immersed in the fluid.
- 2.4 Laboratory thermometers are calibrated annually against a National Institute of Standards and Technology (NIST) reference thermometer or NIST traceable thermometer.
- 2.5 Thermometers are assigned a unique identifier when calibrated and placed into service. This identifier is an internal sequential number assigned by QA and is used to identify the thermometers for calibration and temperature monitoring logs.
- 2.6 Quality Control
  - 2.6.1 Laboratory thermometers are calibrated annually. A calibration factor is assigned with each calibration event.
  - 2.6.2 In order for the NIST reference thermometer to be used for the annual

calibration of thermometers it should meet the following criteria:

- The reference thermometer should have divisions of 0.1 to 0.2° C and
- have listed NIST calibration points at 0°C, 100°C and 180°C.

## 2.7 Calibration Procedure

### 2.7.1 Hot Water Bath Thermometers

#### 2.7.1.1 Partial emersion reference thermometer

When calibrating thermometers for general use above room temperature, a standard water bath is heated to approximately 100°C or at the normal operational temperature of the batch. Place the (partial emersion) reference thermometer and the thermometers to be calibrated in the water bath and adjust them to their appropriate level in the fluid. Compare the values of the working thermometers against that of the reference thermometer at the temperature of interest.

### 2.7.2 Oven Thermometers

#### 2.7.2.1 Partial emersion reference thermometer

When calibrating thermometers for general use in an oven, a standard water bath is heated with vegetable oil to approximately 104°C or 180°C. Place the (partial emersion) reference thermometer and the thermometers to be calibrated in the water (oil) bath and adjust them to their appropriate level in the fluid. Compare the values of the working thermometers against that of the reference thermometer at the temperature of interest.

#### 2.7.2.2 Total emersion reference thermometer

When calibrating thermometers for use in ovens, heat an oven to approximately 104°C or 180°C. Place the (total emersion) reference thermometer and the thermometers to be calibrated in the oven thus the reference thermometer is totally emersed. Compare the values of the working thermometers against that of the reference thermometer at the temperature of interest.

## 2.7.3 Refrigerator Thermometers

### 2.7.3.1 Partial emersion reference thermometer

When calibrating thermometers for general use in a refrigerator, a standard water bath is chilled with ice to approximately 4°C. Place the (partial emersion) reference thermometer and the thermometers to be calibrated in the water bath and adjust them to their appropriate level in the fluid. Compare the values of the working thermometers against that of the reference thermometer at the temperature of interest.

### 2.7.3.2 Total emersion reference thermometer

Thermometers are calibrated at approximately 4°C when calibrating thermometers for use in sample refrigerators. Place the (total emersion) reference thermometer and the thermometers to be calibrated in the refrigerator thus the reference thermometer is totally emersed. Compare the values of the working thermometers against that of the reference thermometer at the temperature of interest.

## 2.7.4 Complete the information on the calibration form.

## 2.7.5 Identify each thermometer with the assigned unique identification number. Indicate the date of calibration, temperature of normal use, calibration factor, serial numbers of the reference thermometer and working thermometer, and analyst performing the calibration. The calibration factor is the difference between the working thermometer and the reference thermometer at the point of interest. It is the number of degrees that have to be added or subtracted from the measured temperature to get the true temperature.

## 2.8 Calibration

### 2.8.1 Calibrations factor, CF

$$CF = RT - WT$$

RT = Temperature of the reference thermometer

WT = Temperature of the working thermometer

### 2.8.2 If the calibration factor is positive, that amount is added to the measured temperature to get the true temperature.

### 2.8.3 If the calibration factor is negative, that amount is subtracted from the measured temperature to get the true temperature.

2.8.4 If the calibration factor is greater than  $\pm 2^{\circ}\text{C}$  that particular thermometer should be discarded and taken out of service and replaced with a newly calibrated thermometer.

2.9 Documentation

2.9.1 Calibrations Form - This form, Figure D.11-1, is used to document the calibration of laboratory thermometers. A copy of this form is maintained by the QA department.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer



**Thermometer Calibration Form**  
 Figure D.11-1

Thermometer ID: \_\_\_\_\_  
 Thermometer SN: \_\_\_\_\_

Type: (total/partial emersion or stem)  
 Placed In Service: \_\_\_\_\_

Calibration Data						
Date of calibration						
Ref. thermometer SN						
Accuracy of ref. therm. at temperature of interest.						
Calibration Data	Ref	Measured	Ref	Measured	Ref	Measured
Temp # 1						
Temp # 2						
Temp # 3						
Ave Temp						
CF (can not be less than the increments of the ref therm)						
Analyst						

Comments \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_

**Title: IR Thermometer Standard Operating Procedure**

**1.0 Identification of the Test Method**

IR Thermometer Standard Operating Procedure

Reference: Users manual for the Linear C-1000 IR thermometer

**2.0 Applicable Matrix or Matrices**

This method is applicable to the determination of temperature, using an IR gun, on essentially any type of material container.

**3.0 Method Detection Limit**

The IR gun has a temperature sensitivity of  $\pm 1^{\circ}\text{C}$  or F.

**4.0 Scope and Application**

Thermal energy is radiated by all objects and a portion of this thermal energy is radiated as infrared energy. An IR gun is a temperature measuring device which consists of a hand-held probe with a lens to focus the infrared energy onto a detector. Once the infrared energy is detected it is converted by a microprocessor to temperature and is displayed.

The unique benefits of measuring temperature with this type of IR probe are as follows:

- a) Measurements are made without physical contact,
- b) Measurements are made rapidly,
- c) Heat is not removed from the measured object so the accuracy of the readings are assured,
- d) Objects that are contaminated can be easily measured.

**5.0 Summary of Method**

An Infrared (IR) thermometer is used to determine and record sample temperature.

**6.0 Definitions**

6.1 Emissivity - The ratio of energy emission of an object being measured to that of a black object at the same temperature.

6.2 IR Gun Functions

6.2.1 ARROWS - Used in Sample or Emissivity mode to increase or decrease the emissivity. Depressing either key will adjust emissivity in steps of 0.01 for

six steps then switch to an accelerated mode for rapid adjustment.

- 6.2.2 AVERAGE - Used to display the average temperature of readings taken in the Sample mode.
- 6.2.3 °C and °F - Used to change the display units
- 6.2.4 CONTINUAL - Used to enter Continual mode for constant monitoring of the target.
- 6.2.5 EMISSIVITY - Used to enter Emissivity mode.
- 6.2.6 MAXIMUM - Used to display the maximum temperature reading obtained in either the Sample or Continual mode.
- 6.2.7 MINIMUM - Used to display the minimum temperature obtained in either the Sample or Continual mode.
- 6.2.8 SAMPLE - Used to enter sample mode and to take an individual reading of the target.

## **7.0 Interferences**

- 7.1 Temperature measurements made at a distance greater than 3 times the sample size may affect the temperature reading. Measurements can be made at virtually any distance from the bottle. However, as the distance from the bottle is increased, the diameter of the measured area increases proportionally. When the 3:1 distance to size ratio is breached and the distance is greater than 3 times the sample size, than background temperature will start interfering with the determination of the sample temperature.
- 7.2 Measurements must be made perpendicular to the object being measured, as non-perpendicular measurements will slightly increase the measured area and the 3:1 distance to size ratio may not be maintained and could affect the displayed temperature.
- 7.3 Avoid contact of the IR gun cone with warm objects for more than a few seconds, as this could cause the signal to drift several degrees.
- 7.4 Prolonged readings of large, hot sources are also to be avoided.

## **8.0 Safety**

The IR gun is used to measure sample containers which may be contaminated. Sample and sample containers need to be treated with caution. See appropriate SOPs for details. Also,

see Alpha's Laboratory Safety/Hazardous Communications Manual and Chemical Hygiene Plan for additional information and details.

## **9.0 Equipment and Supplies**

9.1 Linear C-1000 Infrared Thermometer

9.2 9 volt alkaline battery

## **10.0 Reagents and Standards**

Not applicable.

## **11.0 Sample Collection, Preservation, and Storage**

Not applicable. See the individual methods of analysis.

## **12.0 Quality Control**

See D.14, Standard Operating Procedure for Annual Thermometer Calibration

## **13.0 Calibration and Standardization**

See D.14, Standard Operating Procedure for Annual Thermometer Calibration

## **14.0 Procedure**

### **14.1 Maintenance**

A 9 volt alkaline battery is used to power the IR gun. When the battery needs replacing, a "Lo" appears on the LCD display. The battery is located in the probe. To gain access to the battery, first slide the probe off of the display unit. Lightly press down on the arrow on the battery cover while sliding the battery cover in the direction of the arrow. Replace the battery. To replace the cover, simply slide it back into the probe.

### **14.2 Modes of operation:**

- a) Sample, and
- b) Continual.

For our purpose the IR gun should be placed in the Sample Mode. In this mode, a sample measurement is taken each time the SAMPLE key is pressed. The measurement remains on the LED display until another sample measurement is

made or another mode is selected.

A running average is calculated of all the samples taken. The average temperature can be displayed by pressing the AVERAGE key. Pressing either the MAXIMUM or MINIMUM key will display the maximum or minimum temperature measured in any mode since the unit was turned on.

DO NOT RECORD THIS AS THE SAMPLE TEMPERATURE!

#### 14.3 Measuring Temperature

14.3.1 To measure temperature, simply hold the probe perpendicular to the object being measured at a distance not greater than three times the size of the object and press "SAMPLE." It is important that this 3:1 ratio is maintained for accurate temperature measurements.

#### 14.4 Calculating Distance

14.4.1 To measure an 8 inch sample bottle, hold the probe perpendicular to the bottle within 24 inches of the bottle.

#### 14.5 Emissivity Adjustment

14.5.1 An IR gun temperature measuring instrument reports temperature by adjusting the measured temperature against the emissivity of the object being measured.

Emissivity is the ratio of energy emission of an object being measured to that of a black object at the same temperature which produces an emissivity constant. This ratio is unique to the surface material being measured. Therefore, IR guns measurements are adjusted by multiplying the measured IR temperature by this emissivity ratio to correct for differences in energy emissions.

Once the emissivity function on the IR gun is adjusted, the temperature reading on the IR gun display has then been corrected for errors associated with surface emission.

##### 14.5.2 Adjusting emissivity

14.5.2.1 If the emissivity is known, enter the EMISSIVITY mode button and adjust the emissivity ratio using the ARROW key.

14.5.2.2 If the emissivity is unknown, it can be determined as follows:

- 14.5.2.2.1 Place a black sticker on the object to be measured.
- 14.5.2.2.2 Enter the Emissivity mode and adjust the emissivity to 1.0.
- 14.5.2.2.3 Save the new emissivity by exiting the Emissivity mode by pressing the EMISSIVITY key.
- 14.5.2.2.4 Measure the temperature on the sticker by pressing SAMPLE key and note the temperature.
- 14.5.2.2.5 Next measure an area next to the sticker by pressing the SAMPLE key.
- 14.5.2.2.6 If the reading is the same as the previous one, the emissivity is calibrated for the surface.
- 14.5.2.2.7 If it two temperatures are not the same, keep the probe pointed at the area next to the sticker. Press the down ARROW key. ( An emissivity value will briefly appear on the display and then the temperature compensated with the new emissivity will remain. Note the reading.
- 14.5.2.2.8 If it is not the same temperature as obtained on the sticker, press the down ARROW key until the measurements matches that of the black sticker. At this point the emissivity is calibrated for the surface.

Note: If a black sticker cannot be used, flat black paint can be used. Paint a small area on the object to be measured and follow the procedure just described.

**Emissivity Table**  
**Table E.14-1**

<b>Material</b>	<b>Emissivity (%)</b>
Glass	92
Glass, frosted	96
Plastic	99
Ice	97
Water	98
Wood	85
Rubber	95
Quartz	93

#### 14.6 Procedure

14.6.1 Turn the IR gun “ON” using the switch located on the left side of the IR gun.

14.6.2 Place the IR gun to the SAMPLE mode.

14.6.3 Enter the emissivity.

14.6.4 Point the IR gun perpendicular to the sample at a distance no greater than 3 times the size of the object being measured.

14.6.5 Press SAMPLE and note the measured temperature for recording. Take the average of three consecutive measurements when recording temperature. DO NOT USE THE AVERAGE function key for this procedure.

Note: If the IR gun has a correction factor to adjust for errors against the NIST calibrated thermometer, remember to adjust for this before recording the final temperature.

14.6.6 Turn the IR gun “OFF” using the switch located on the left side of the IR gun.

#### 15.0 Calculations

Not applicable.

Note: Temperature is reported in °C for most applications

**16.0 Method Performance**

Not applicable.

**17.0 Pollution Prevention**

No solvents, acids or bases are used with this procedure. The sample containers which are being measured by the IR gun may be contaminated and need to be treated with caution. See appropriate SOPs for details. Also, see Alpha's Laboratory Safety/Hazardous Communications Manual and Chemical Hygiene Plan for additional information and details.

**18.0 Data Assessment and acceptance criteria for quality control measures**

Not applicable.

**19.0 Corrective Actions for Out-of-Control Data**

Not applicable

**20.0 Contingencies for Handling Out-of-Control or Unacceptable Data**

20.1 Failed Instrument

See instrument maintenance and/or use another calibrated thermometer.

**21.0 Waste Management**

Reference Alpha Analytical's Sample Waste SOP

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer



## **1.0 TITLE: TEMPERATURE LOG**

Description: The following is a description of practices and procedures used to monitor refrigerator and oven temperatures and how this information is documented.

## **2.0 STANDARD OPERATING PROCEDURE (Thermometers)**

2.1 All thermometers or thermistors used to monitor temperatures are calibrated against an NBS/NIST calibrated or traceable thermometer on an annual basis.

### **2.2 Ovens (Spirit Filled Thermometers)**

2.2.1 Each oven is assigned a calibrated thermometer and is identified using the same identification scheme as the oven. If possible, this identification number is written on the thermometer to avoid the switching of thermometers. If not the thermometer is identified by its serial number.

2.2.2 Oven thermometers are generally full immersion, spirit filled thermometers placed in a sand filled beaker to buffer oven temperature variations.

2.2.3 Placing these thermometers in a sand bath helps monitor actual temperature variations which may occur in environmental samples. Secondly this practice facilitates the reading of the thermometer and eliminates the temperature probe from coming into direct contact with a non-temperature equilibrated surface.

2.2.4 The sample custodian or other delegated person is responsible for monitoring and recording oven temperatures. This information is recorded on the Temperature Log.

### **2.3 Refrigerators (Steel Stemmed Thermometers)**

2.3.1 Steel stemmed digital display refrigerator thermometers are particularly useful for easy reading. Each refrigerator thermometer is assigned a calibrated thermometer and is identified using the same identification scheme as the associated refrigerator. If possible, this identification number is written on the thermometer to avoid the switching of thermometers. If not the thermometer is identified by its serial number.

2.3.2 The refrigerator thermometer stems are placed through the septum of a 40 ml VOA vial or a 1 liter amber container filled with water in order to buffer refrigerator temperature variations.

2.3.3 This practice helps monitor actual temperature variations which may occur in environmental samples. Secondly this practice facilitates the reading of the

thermometer and eliminates the temperature probe from coming into contact with a non-temperature equilibrated surface.

- 2.3.4 The sample custodian or other delegated person is responsible for monitoring and recording refrigerator temperatures. This information is recorded 7 days a week on the Temperature Log.

## 2.4 Refrigerators (Wireless Thermometers)

- 2.4.1 Electronic thermometers or thermistors configured with a wireless data logger are particularly useful for the continuous monitoring of refrigerators. Each refrigerator with a wireless data recorder is assigned a calibrated thermistor and is identified using the same identification scheme as the associated refrigerator. This identification number is written on the data logger for positive identification.
  - 2.4.2 The refrigerator thermistors are placed through the septum of a 40 ml VOA vial or a 1 liter amber container filled with water in order to buffer refrigerator temperature variations.
  - 2.4.3 This practice helps monitor actual temperature variations which may occur in environmental samples. Secondly this eliminates the thermistor from coming into contact with a non-temperature equilibrated surface.
  - 2.4.4 The sample custodian or other delegated person is responsible for monitoring and recording refrigerator and oven temperatures. This information is recorded daily 7 days a week in the Temperature Log book.
- 2.5 Temperature readings are recorded directly from the spirit filled or steel stemmed thermometers without making any thermometer corrections. However, thermometer calibration correction factors are written on the Temperature Log and maintained by the responsible person.
  - 2.6 Alpha has established temperature range criteria for all ovens, refrigerators and freezers containing samples, sample extracts and standards.
  - 2.7 The person taking temperature recordings will make all refrigerator, or oven recordings at approximately the same time each day to avoid additional variability in temperature recordings.
  - 2.8 This person can make minor temperature adjustment in order to compensate seasonal fluctuations, sample mass to be cooled, or any other contributing factor. All adjustments are recorded in the comments section of the Temperature Log.

- 2.9 The Laboratory Director or Laboratory Manager must be informed of refrigerators or ovens with temperatures out side of their designated range, wildly fluctuating temperatures or abnormalities which might affect the sample integrity. All corrective actions and maintenance will be recorded on its log record.
- 2.10 The SCO is responsible for maintaining historical records.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

# Temperature Log

Oven/Refrigerator No: \_\_\_\_\_

NBS Certified Thermometer No: \_\_\_\_\_

Correction Factor: \_\_\_\_\_

TDS Oven Temperature  $180 \pm 2^{\circ}\text{C}$

TSS Oven Temperature  $104 \pm 1^{\circ}\text{C}$

TS Oven Temperature  $104 \pm 1^{\circ}\text{C}$

Refrigerator Temperature  $0.1 - 6^{\circ}\text{C}$

Freezer Temperature  $-15 \pm 5^{\circ}\text{C}$

	Monday			Tuesday			Wednesday			Thursday			Friday			Saturday			Sunday		
Week	Date	Temp	Init	Date	Temp	Init	Date	Temp	Init	Date	Temp	Init	Date	Temp	Init	Date	Temp	Init	Date	Temp	Init
1																					
2																					
3																					
4																					
5																					
6																					
7																					
8																					
9																					
10																					
11																					
12																					
13																					
14																					
15																					
16																					
17																					
18																					
19																					
20																					
21																					
22																					
23																					
24																					
25																					
26																					

Comments: \_\_\_\_\_

**1.0 TITLE: PROCEDURE FOR COLD STORAGE TEMPERATURE EXCURSIONS**

Description: Cold storage units require daily temperature monitoring to ensure samples and extracts are kept cool. Samples and extracts are refrigerated to minimize sample degradation and as a good laboratory procedure.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 These guidelines describe the necessary steps to follow in the event of a temperature excursion or other abnormality associated with a cold storage unit.

2.2 Refrigerator temperatures should be kept within an ideal temperature range of 2°C to 6°C. A refrigerator's cold setting is adjusted in order to maintain its temperature in an acceptable range. Freezer temperatures should be kept within an ideal temperature range of -20°C to -10°C and ovens should be maintained according to their particular use.

2.3 If the temperature continuously drifts outside the prescribed range of acceptability the following steps should be taken:

2.3.1 Notify the Laboratory Director and/or the QA Officer immediately;

2.3.2 Monitor the refrigerator or oven for a minimum of three consecutive days with a thermograph or wireless thermister to establish a real-time temperature baseline;

Note: A temperature excursion may be an isolated occurrence, therefore constant monitoring is required to document a more precise determination of a problem.

2.3.3 If the temperature excursions persist for three consecutive days then the samples or extracts located in that refrigerator or oven should be moved to a back-up and the out-of-control unit repaired or replaced; and,

2.3.4 If the Laboratory Director or QA Officer deems it necessary to circumvent the three day period at any time, they may do so.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

## **1.0 TITLE: REFRIGERATOR DOCUMENT CONTROL**

Description: Refrigerator logs and the maintenance of these historical records is an important part in the overall continuity of sample integrity. Records are kept and maintained to ensure there is no breach in sample integrity or the process of documenting this.

## **2.0 STANDARD OPERATING PROCEDURE**

2.1 Each refrigerator is assigned a one-time unique identification number to distinguish it from any other refrigerator in use by the laboratory. This identification number has the added feature in identifying refrigerators temporally.

2.2 Refrigerators/freezers are labeled with an identification number in the following format:

XXXNNZ

Where;

XXX represents a 3-letter prefix unique to each type of refrigerator/freezer

Such as;

SAR refers to sample refrigerator  
SAF refers to sample freezer  
STR refers to a standards refrigerator  
STF refers to a standards freezer  
EXT refers to an extraction refrigerator/freezer

And;

NN represents a two digit number identifying it from another refrigerator/freezer with the same prefix. Alpha uses the following sets of numbers (NN):

1-19	refers to sample refrigerators/freezers
20-40	refers to standard refrigerators/freezers
31-50	refers to extract refrigerators/freezers

And:

Z represents the refrigerator or freezer itself during a particular period of time.

For example:

If the VOC refrigerator, ID SAR-1A, happened to be replaced for whatever reason the ID for the new refrigerator would be SAR-1B. By using this scheme, refrigerators which are replaced can be temporarily tracked. This is particularly useful when tracking samples throughout the lab. Samples go in a particular refrigerator and therefore the confusion of which refrigerator and how it was operating during a particular time period can easily be traced.

- 2.3 Refrigerator adjustments and replacements are all annotated on the specific temperature log and are kept as historical record. Tables D.15-1-through 3 lists all of our refrigerators, their particular identification and the types of samples or standards which are stored in them.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**Refrigerator Identification Table**  
**Table D.15-1A**

<b>Refrigerator Name</b>	<b>Use</b>
SAR-1B	VOC water samples post analysis
SAR-2C	Contaminated VOC/TPH samples (in paint cans)
SAR-3A	8270/8081/8082 and general SV samples prior to extraction
SAR-4B	TPH/TOC water samples post analysis and non-extracted waters
SAR-5D	Laboratory prepared VOC trip and field blanks
SAF-6B	Air samples and methanolic air extracats
SAR-7A	TPH-E/TOC samples prior to analysis
SAR-8A	VOC water samples prior to screening and prepared sequences awaiting analysis
SAR-9B	Inorganic samples post analysis
SAR-10A	524.2 sample prior to analysis and storage after analysis
SAR-11A	TPH/VOC soil samples prior to analysis and storage after analysis
SAR-12A	VOC water samples after screening and prior to analysis and Alcohol waters
SAR-13A	Inorganic samples
SAR-14A	Department of Defense (DOD) samples
SAR-15B	Incoming sample (Sample Receipt)
SAR-16A	Dissolved Gases
SAR-17A	Organic Acid samples



**Refrigerator Identification Table**  
**Table D.15-2**

<b>Refrigerator Name</b>	<b>Use</b>
STR-20A	TPH extractable standards
STF-21B	VOC methanolic standards
STR-22A	TOC/Anions standards
STR-23B	Metals/Inorganic standards
STR-24C	VOC standards
STR-25B	8270 standards
STF-26C	VOC methanolic standards
STR-27C	Alcohols standards
STR-28A	ECD standards
STF-29C	524.2 standards
STR-30A	TPH standards
STR-32A	PEST standards
STR-33A	PCB standards
STR-34A	Metals standards

**Refrigerator Identification Table**  
**Table D15-3**

<b>Refrigerator Name</b>	<b>Use</b>
EXT-31B	SV extracts
EXT-32B	
EXT-33C	ECD extracts
EXT-34B	SV extracts (autovials)
EXT-35B	SV extracts
EXT-36B	MeOH VOC soil extracts
EXT-37A	TPH extracts
EXT-38A	SV extracts requiring internal C-O-C procedures
EXT-39A	

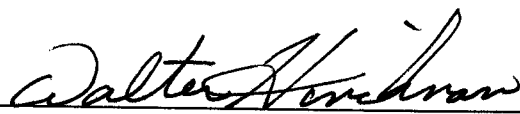
**1.0 TITLE: MECHANICAL VOLUMETRIC DISPENSING DEVICE (MVDD) PROCEDURE**

References: ASTM E542 Standard Practice for Calibration of Volumetric Apparatus, and E969 Standard Specifications for Volumetric Pipettes.

**2.0 STANDARD OPERATING PROCEDURE**

- 2.1 Mechanical volumetric dispensing devices are assigned a unique identifier when purchased and placed into service. This identifier is used to identify the dispensing device for monitoring accuracy records. Mechanical volumetric dispensing devices are checked quarterly for accuracy against a Class A volumetric cylinder or a certified syringe.
- 2.2 Mechanical volumetric dispensing devices are checked quarterly for accuracy at a single volume against a Class A volumetric cylinder or a certified syringe.
- 2.3 If accuracy measurements are greater than  $\pm 3\%$  of the expected value, then the volumetric dispensing device must be taken out of service and either sent back to the manufacturer for recalibration or replaced.
- 2.4 Dispensing devices are labeled with an identification number and checked for accuracy. This information is documented on the Mechanical Volumetric Dispensing Device (MVDD) Accuracy Form.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

**MVDD Acceptance Limits**  
**Table D.16-1**

ID	Brand Name	Range	Check Volume	Accuracy
MVDD-1	VWR	0-10mL	4mL	3.88-4.12mL
MVDD-2	Optifix	10-100mL	100mL	97-103mL
MVDD-3	Optifix	10-50mL	30mL	29.1-30.9mL
MVDD-4				
MVDD-5	Brinkman	0-2.5mL	2mL	1.94-2.06mL
MVDD-6	Brinkman	0-2.5mL	1ml	0.97-1.03mL
MVDD-7	LabMax	0-2.5mL	0.09ml	0.087-0.0927ml
MVDD-8	VWR	0-2.5mL	0.09mL	0.087-0.0927ml
MVDD-9	Optifix	0-50mL	20mL	19.4-20.6mL
MVDD-10	VWR	0-10mL	1 mL	0.97-1.03mL
MVDD-11	Brinkman	0-10mL	1 mL	0.97-1.03mL
MVDD-12				
MVDD-13	Optifix	0-50mL	20mL	19.4-20.6mL
MVDD-14	Brinkman	0.2.5 mL	2 mL	1.94-2.06mL
MVDD-15	Brinkman	0-2.5mL	1mL	0.97-1.03mL
MVDD-16	VWR	0-10mL	8mL	7.76-8.24mL

Alpha Analytical, Inc.  
Section No.: D.16  
Revision No.: 15.0  
Date: January, 2007  
Page: 3 of 3

**MVDD**  
**Quarterly Accuracy Check**  
**Table D.16-2**

Brand \_\_\_\_\_  
Placed in Service \_\_\_\_\_  
MVDD # \_\_\_\_\_  
Year \_\_\_\_\_

Expected Volume \_\_\_\_\_  
Accuracy Limits \_\_\_\_\_

Month	1	2	3	Average	Acceptable	Non-Acceptable	Initials	Date
Jan								
April								
July								
Oct								

# **Appendix E**

## **Analytical and Extraction Support Procedures**

**1.0 TITLE: Analytical and Extraction SOPs**

- 1.1 The QA manual and associated SOPs have been written to provide the basis for QA and QC in both field sampling and laboratory analysis. Data generated by Alpha could potentially be used to support litigation; therefore, documentation of laboratory procedures is essential to maintain a defensible audit trail for the generation of sample results. One part of this trail of documentation is the need and a compilation of Standard Operating Procedures to assure the various activities were performed and documented with uniformity and consistency.

Not all activities have written SOPs, however, procedures which are performed on a routine basis that may have a significant impact regarding the Data Quality Objectives have SOPs.

Approved By: Walter Hinchman Date: January 2007  
Walter Hinchman  
QA Officer

**1.0 Title: Dishwasher Steam Scrubber Operation**

Description: To prevent cross-contamination or carry-over, all organic extraction glassware is thoroughly washed in a Labconco Dishwasher steam scrubber.

**2.0 Standard Operating Procedure:**

- 2.1 As soon as possible, glassware that has come in contact with samples or standards should be rinsed with methylene chloride or the solvent last used in the glassware.
- 2.2 Soak the glassware in hot water with a non phosphate detergent such as Extran-300, to loosen and float particles.
- 2.3 Manually scrub the glassware with a pad or brush.
- 2.4 Hot water rinse the glassware to flush away any floated particulates.
- 2.5 Dishwasher Settings:
  - 2.5.1 Steam action on;
  - 2.5.2 Power dry heat;
  - 2.5.3 Fill detergent container with appropriate amount of detergent,
  - 2.5.4 Close the detergent container and fill the extra containers as needed,
  - 2.5.5 Set to appropriate cycle: i.e. plastic, glass 1, glass 2, rinse, etc.,
  - 2.5.6 Push start switch to begin cycle; and,
  - 2.5.7 After the dishwasher stops, inspect and separate glassware which visibly does not look clean and repeat the procedure again.
- 2.6 Once the glassware has been cleaned and visually inspected it is oven dried at 104 °C for a minimum of 4 hours.

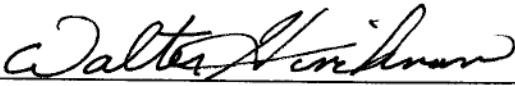
Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 Title: Manual Glassware Cleaning**

Description: Certain glassware is incompatible with cleaning in the dishwasher steam scrubber; therefore, the following procedure will be followed.

**2.0 Standard Operating Procedure**

- 2.1 Remove surface residuals immediately after use. As soon as possible, glassware that has come in contact with sample and standards should be rinsed with methylene chloride or the same solvent used prior.
- 2.2 Soak the glassware in hot water with a non phosphate detergent such as Extran-300, to loosen and float particles.
- 2.3 Manually scrub the glassware with a pad or brush.
- 2.4 Hot water rinse the glassware to flush away any floated particulates.
- 2.5 After washing, inspect and separate glassware which visibly does not look clean and repeat the procedure again.
- 2.6 Once the glassware has been cleaned and visually inspected it is oven dried at 104 °C for a minimum of 4 hours.

Approved By:  Date: January, 2007  
Walter Hinchman  
QA Officer



**1.0 Title: Sample Container Cleaning Procedures**

Description: Cleaning procedures are practiced to minimize contamination from the containers in which they are stored.

**2.0 Standard Operating Procedure**

**2.1 Policy**

It is Alpha's policy to not reuse any sampling bottles or containers to collect environmental field samples. All containers purchased by Alpha are purchased from manufacturers which follow the minimum EPA Level I Protocol prescribed cleaning protocols as outlined below.

**2.2 Sample Bottle Material**

2.2.1 If the analysis to be determined are organic in nature, the container should be made of borosilicate glass.

2.2.2 If the analytes are inorganic, the container should be polyethylene.

2.2.3 When both organic and inorganic substances are expected to be present, separate sample aliquots should be taken.

**2.3 Cleaning Protocols**

Commercially cleaned containers are used if cleaning procedures comply with EPA procedures as outlined below.

2.3.1 The procedures for cleaning glass and polyethylene containers and their caps for EPA level I protocol are as follows:

2.3.1.1 Cleaning procedure A for extractable organics:

- a) Scrub and wash bottles, teflon liner, and caps in laboratory grade, non-phosphate detergent;
- b) Rinse three times with tap water;
- c) Rinse with 1:1 nitric acid;
- d) Rinse three times with ASTM type 1 organic free water;
- e) Rinse with pesticide grade methylene chloride; and
- f) Oven dry at 125 degrees Centigrade, allow to cool to room temperature in an enclosed contaminant-free environment.

Note: Procedure A is used for amber and clear glass, wide mouth jars and amber Boston rounds.

#### 2.3.1.2 Procedure B for purgeable volatile organics:

- a) Wash glass vial in hot tap water using laboratory grade non-phosphate detergent;
- b) Rinse three times with distilled water;
- c) Rinse three times with ASTM type 1 deionized water;
- d) Oven dry vials at 104 Degrees centigrade for one hour;
- e) Allow vials to cool to room temperature in an enclosed contaminant-free environment; and
- f) Seal 40 ml vials with septa and cap.

Note: Procedure B is used for 40 ml glass vials.

#### 2.3.1.3 Procedure C for inorganic metals:

- a) Wash polyethylene bottles in hot tap water with laboratory grade non-phosphate detergent;
- b) Rinse with 1:1 nitric acid;
- c) Invert and air dry in contaminant-free environment; and
- d) Cap bottle.

Note: Procedure C is used for High Density Polyethylene (HDPE) bottles.

#### 2.3.1.4 Teflon liners

- a) Wash with laboratory grade non-phosphate detergent;
- b) Rinse with distilled water;
- c) Rinse with acetone;
- d) Rinse with Hexane;
- e) Air dry;
- f) Place liners in cleaned caps;
- g) Heat to 104 Degrees Centigrade for two hours;
- h) Allow to cool in contaminant-free environment; and

Alpha Analytical, Inc.

Section No.: E.4

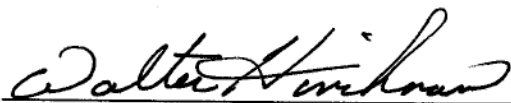
Revision No.: 15.0

Date: January, 2007

Page: 3 of 3

- i) Cap cleaned bottles.

Approved By:



Walter Hinchman  
QA Officer

Date: January, 2007

## **1.0 Title: Prevention of Sample Contamination**

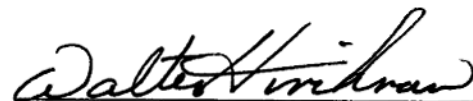
Description: The prevention of sample contamination or cross-contamination is an important element in maintaining sample integrity. Contamination can occur virtually anywhere, and therefore, strict adherence to standard operating procedures must be followed.

## **2.0 Standard Operating Procedure**

- 2.1 Prevention of sample contamination begins in the field when samples are collected. Samples must be collected according to the procedures described in the FSP or a project specific plan.
- 2.2 Alpha maintains a three month sequestered supply of sample containers. These containers are factory cleaned according to the procedures described by the USEPA. Alpha only buys, uses and distributes QC'd containers. Each case of sample jars contain a lot number which can be traced back to an analytical report certifying its cleanliness.
- 2.3 Samples are segregated from one another depending upon factors such as matrix or the types of analysis requested. Extracts are segregated from samples and standards in the hope of preventing cross-contamination or laboratory contamination. Additionally standards are segregated between types of analysis (e.g. volatile, semi-volatiles and metals) and maintained under refrigerated conditions.
- 2.4 Extraction technicians adhere to SOP's regarding cleaning of glassware, operations of the dishwasher, and other practices used in the laboratory. All glassware which is used during sample preparation, extraction or dilution, is rinsed three times with the solvent or acid used in the procedure prior to its use.
- 2.5 Once the glassware is used, it is scrubbed and washed according to the written SOP's, and oven dried at 104°C for four hours.
- 2.6 Syringes or other equipment which may come in contact with the sample or its extract are thoroughly rinsed before and after use.
- 2.7 Analytical instruments are oven-cycled baked or cleaned after highly contaminated samples have been analyzed. Prior to the analysis of additional samples, a method blank or instrument blank is analyzed for proof that a clean analytical instrument was used and did not contribute to environmental sample contamination caused by potential analytical "carry-over".
- 2.8 Alpha buys the best solvents and acids available for low-level environmental analysis. Solvents and acids are maintained in an environment free of potential cross-contamination from air-borne analytes.

- 2.9 Alpha has modified its HVAC system, such that, analytical instrument rooms are operating on separate systems than the extraction labs. In addition, the analytical rooms are under positive pressure in relation to the extraction labs; thereby pulling makeup air from the outside. Make up air vents are upwind from the extraction exhaust hoods.
- 2.10 Extraction personnel or staff members which have come from the extraction lab and need to enter the VOC analytical room, first must aerate themselves by walking outside prior to entering.
- 2.11 Water which is used in washing or the rinsing of VOC syringes or associated equipment is organic free produced from our water system.
- 2.12 HCl which is used in the preservation of many samples is analyzed prior to its use to prevent possible VOC contamination.
- 2.13 Glassware and syringes used in the preparation of standards are segregated for VOC, semi-volatile and metals use only, and may be method specific.
- 2.14 Alpha uses several types of blanks to monitor possible contamination either from the lab or sources other than environmental contamination. These blanks are analyzed on an as-need basis for many of them and on a regular frequency for the remainder. For a better understanding of the types of blanks analyzed, and how they are used to monitor contamination see the main body of the QA plan.

Approved By:



Walter Hinchman  
QA Officer

Date: January, 2007

## **Standards Preparation Procedures**

### **1.0 Identification of Test Method**

The preparation of standards and reagents are in support of the various analytical methods generally described below.

#### **1.1 Clean Water Act (CWA)**

“Guidelines Establishing Test Procedures for the Analysis of Pollutants”, U.S. Environmental Protection Agency, Code of Federal Regulations, 40 Part 136, as updated.

“Methods for Chemical Analysis of Water and Wastes”, EPA-600/4-79/020, U.S. Environmental Protection Agency, Environmental Monitoring and Support Laboratory, Office of Research and Development, Cincinnati, Ohio, 45268, Revised March 1983.

“Standard Methods for the Examination of Water and Wastewater”, American Public Health Association, American Water Works Association, and Water Environmental Federation, 20<sup>th</sup> Edition, 1998.

#### **1.2 Resource Conservation and Recovery Act (RCRA)**

“Test Methods for Evaluating Solid Waste, Physical/Chemical Methods”, U.S. Environmental Protection Agency, Office of Solid Waste and Emergency Response, SW-846, 3<sup>rd</sup> Edition, Final Update III, 1996.

#### **1.3 Safe Drinking Water Act (SDWA)**

“Methods for the Determination of Organic Compounds in Drinking Water”, EPA-600/4-88/039, U.S. Environmental Protection Agency, Environmental Monitoring Systems Laboratory, Office of Research and Development, Cincinnati, Ohio, 45268, Revised July 1991.

“Methods for the Determination of Inorganic Substances in Environmental Samples”, EPA-600/R-93/100, U.S. Environmental Protection Agency, Environmental Monitoring Systems Laboratory, Office of Research and Development, Cincinnati, Ohio, 45268, August 1993.

“Methods for the Determination of Metals in Environmental Samples, Supplement I”, EPA-600/R-94/111, U.S. Environmental Protection Agency, Environmental Monitoring Systems Laboratory, Office of Research and Development, Cincinnati, Ohio, 45268, May 1994.

## **2.0 Applicable Matrix or Matrices**

Not applicable

## **3.0 Method Detection Limit**

Not applicable

## **4.0 Scope and Application**

The preparation and maintenance of standards are essential elements for the determination of chemical analytes. The procedures for the preparation of all standards and subsequent dilutions must be consistent and standardized to minimize possible analytical errors.

This procedure includes all reportable target analytes for all methods. Many EPA standardized methods do not describe in full detail, the preparation of standards, compound stability, expiration dates, etc for the listed method analytes. However, each in-house method SOP describes these procedures, see individual method SOPs, section 10 for specific details.

This procedure is intended as a general guideline for the preparation and the documentation of standards and reagents and should be used in conjunction with the individual in-house method SOPs.

## **5.0 Summary of Method**

Pure product "Neat" standards are prepared by measuring an appropriate mass or weight of the target analyte into Class A glassware and diluting to volume with the appropriate solvent. The net weight gain is calculated, and the concentration is determined.

Commercial standards and/or primary dilution standards may be further diluted (secondary dilutions) to a pre-determined concentration, by measuring an appropriate volume of the concentrated standard and diluting with solvent to the appropriate final volume.

All measuring and weighing devices including balances, syringes, auto-pipettors, volumetric glassware and/or manual volumetric dispensing devices are certified to accuracy and precision as described by NELAP.

## **6.0 Definitions**

6.1 America Chemical Society (ACS) Grade Chemicals - Chemicals of the highest quality; which often exceeds the latest purity standards set by the American Chemical Society. ACS grade chemicals does not imply a standardized percent purity for all chemicals, but rather is a purity threshold established for individual compounds. ACS grade is the only universally accepted standard grade when evaluating chemicals.

- 6.2 Primary Dilution Standard - A solution prepared from the stock standard solution and diluted as needed to prepare a more dilute solution.
- 6.3 Reagent Grade Chemicals - Chemicals with purity that are typically equal to ACS grade. The grading of these types of chemicals are similar to that of ACS grade in that reagent grade does not imply a standardized percent purity for all chemicals; rather it is a purity threshold established for individual compounds and is typically less pure than ACS grade. This grade is suitable for analytical work and is more than adequate for general lab use.
- 6.4 Stock Standard Solution (SSS) - A concentrated solution containing method analytes prepared using reference material or purchased from a commercial source.

## **7.0 Interference**

- 7.1 There are several potential sources of contamination when preparing standards with this procedure.
  - 7.1.1 Contaminated Neat material. This is particularly a potential problem if the Neat material is < 98% pure product. Since it is not “pure” product, the remaining non-target material, may be a source of contamination and must be evaluated carefully.
  - 7.1.2 Contaminated diluting solvent.
  - 7.1.3 Contaminated standard preparation glassware to include, syringes, volumetric flasks, weigh boats, etc.

Contamination by carryover can occur whenever high-concentration and low-concentration standards are prepared in sequence. To reduce the potential for carryover, the standard preparation syringe and/or volumetric measuring containers should be rinsed between standard preparation with an appropriate solvent, typically the dilution solvent.

### **7.1.3.1 Syringe Rinsing**

Clean syringes by flushing the syringe barrel approximately 10 times using the standard dilution solvent before and/or after the preparation of individual standards.

### **7.1.3.2 Volumetric-ware Rinsing**

All volumetric glass and plastic ware must be scrupulously cleaned.



Clean as soon as possible after use by rinsing with the last solvent used. This should be followed by detergent washing with hot water, and rinse with tap water and reagent-free water. Drain the glassware and dry in an oven at 104°C for several hours or rinse with methanol for organics and/or 1:1 nitric acid for inorganics and drain. Store dry volumetric-ware in a clean environment.

Note: It is extremely important not to dry Class A glassware or plastic ware in ovens above this temperature, due to the potential change in internal volume from the heating and cooling cycles. Most Class A volumetric ware is certified to 1-2%, but is not warranted if heated above 104°C.

- 7.2 Methods for organic analysis may have potential interferences by phthalate esters.
  - 7.2.1 Common plastics contain varying amounts of phthalate esters and are easily extracted from these material during laboratory operations.
  - 7.2.2 Interference from phthalate esters are best minimized by eliminating contact with any plastic material by using only borosilicate glassware and checking all solvents and reagents for phthalate contamination.
- 7.3 Methods for inorganic analysis may have potential interferences as follows:
  - 7.3.1 Zinc, and other metals may leach from the walls of plastic volumetric glassware.
  - 7.3.2 Boron may leach from the walls of borosilicate glassware.
- 7.4 It is beyond the scope of this SOP to describe all potential sources of interference, but is a reminder to verify glassware, plasticware etc. are free from interferences when preparing standards and reagents.

## **8.0 Safety**

The toxicity or carcinogenicity of each reagent used in this method has not been precisely defined. However, each chemical compound should be treated as a potential health hazard. A reference file of material safety data sheets are available. See Alpha's Laboratory Safety/Hazardous Communications Manual and Chemical Hygiene Plan for additional information and details.

## **9.0 Equipment and Supplies**

### **9.1 Equipment**

- Analytical and top loading balances, see balance SOP for details
- Manual Volumetric Dispensing Devices, see MVDD SOP for details

### **9.2 Supplies**

- Syringes of various sizes
- Pipettes of various sizes
- Volumetric flasks of various sizes
- ACS grade solvents and acids

## **10.0 Reagents and Standards**

### **10.1 Reagents**

10.1.1 Acids - Ultra high-purity grade acids used in the preparation of standards must be used to avoid any potential contamination. Concentrated nitric and hydrochloric acids should be analyzed to determine the levels of impurity.

10.1.2 Reagent Water - Distilled or deionized water 17.8 Mohm or better, free of the analytes of interest. Water should contain particles no larger than 0.20  $\mu\text{m}$ . This is commonly referred to as ASTM type I water (ASTM D1193).

10.1.3 Solvents - Organic solvents used in the preparation of standards must be of pesticide grade or better.

### **10.2 Standards**

See the following tables for details.

## **11.0 Sample Collection, Preservation and Storage**

Not applicable.

**Stability of Standards**  
**Organic Methods of Analysis**  
**Table E.6-1**

Method	Expiration Stock Standard Solutions	Expiration Primary/Secondary Dilutions	Expiration Working Standards	Dilution Solvent
608/SM6630C/8081	Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Hexane, iso-octane, acetone, methanol
608/SM6630C/8082	Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Hexane, iso-octane, acetone, methanol
EPA-600/4-81/045	Method states: "...standards are non- labile and if maintained, can be stored indefinitely..."  Clarification: will use expiration dates as described in 608/SM6630C/8082 above.			Hexane, iso-octane, acetone, methanol
8015B-DRO	Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Same as SSS	Same as SSS	Hexane
8015B-GRO	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at -10 to -20°C with minimal head space.	Same as SSS	Prepare aqueous working standards daily with zero head space.	Methanol for SSS and dilutions. Organic free water for daily working standards.
624/8260B	Gases- stable for at least 1 week. Longer if drift is less than 20%. Liquids - stable for at least 6 months or manufacturer's expiration date or sooner if QC indicates a problem. Store at -10 to -20°C with minimal head space.	Same as SSS	Prepare aqueous working standards daily with zero head space.	Methanol for SSS and dilutions. Organic free water for daily working standards.
625/8270C	Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C in the dark	Methylene chloride or other appropriate solvent.
524.2	Gases - stable for at least 1 week Liquids - stable for at least 4 weeks, Should replace after 6 months Manufacturer's expiration date or sooner for all standards if comparison with check standards indicates a problem. Gases - Store at -10 to -20°C with minimal head space. Liquids - Store at 4°C with minimal head space.	Same as SSS	Prepare aqueous working standards daily with zero head space.	Methanol for SSS and dilutions. Organic free water for daily working standards.
Organic Acids	Replace after 6 months or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 6 months or Sooner if QC indicates a problem. Store at 4°C in the dark	Replace after 1 week or Sooner if QC indicates a problem. Store at 4°C in the dark	Reagent-free water

**Stability of Standards**  
**Inorganic Wet-chemistry Methods of Analysis**  
**Table E.6-2**

Method	Expiration Stock Standard Solutions	Expiration Primary/Secondary Dilutions	Expiration Working Standards	Dilution Solvent
120.1/SM2510B	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.  Note: SSS are purchased as satchets. These satchets contain a small volume of standard sealed in an aluminum bag. Once a satchet is opened and used, it is discarded and not re-used due to potential degradation from atmospheric CO <sub>2</sub> .	Same as SSS	Same as SSS	Reagent grade water.
150.1/SM4500H B/9040B/9045C	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	NA Same as SSS	No method specified criteria.  In-house criteria - Pour fresh working standards from SSS daily. Do not re-use due to potential degradation from atmospheric CO <sub>2</sub> .	NA
160.1/160.2/160.3	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Reagent grade water.
180.1/SM2130B	Method specifies 1 month if preparing in the lab. Does not specify commercial standards.  In-house commercial standards - AMCO-AEPA manufacturer specifies 1 year or sooner if comparison to QC indicates a problem	NA Same as SSS	Method specifies 1 week if preparing in the lab. Does not specify commercial standards.  In-house commercial standards - AMCO-AEPA manufacturer specifies 1 year or sooner if comparison to QC indicates a problem	NA
305.1/SM2310B	Method states a minim of 1 week for sodium carbonate used to normalize hydrochloric or sulfuric acid. Method has no other expiration date for standards or reagents.  In-house criteria for all other reagents Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	Same as SSS	Reagent grade water.

**Stability of Standards**  
**Inorganic Wet-chemistry Methods of Analysis**  
**Continued**  
**Table E.6-3**

Method	Expiration Stock Standard Solutions	Expiration Primary/Secondary Dilutions	Expiration Working Standards	Dilution Solvent
310.1/SM2320B	Method states a minim of 1 week for sodium carbonate used to normalize hydrochloric or sulfuric acid. Method has no other expiration date for standards or reagents.  In-house criteria for all other reagents Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	Same as SSS	Reagent grade water.
350.3	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Reagent grade water.
351.4/SM4500NH3	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Reagent grade water.
SM3500Fe D	One year or manufacturer expiration date prior to color development.	Same as SSS	Non-colored working standards are prepared fresh daily. Method specifies a 6 month criteria for color developed standards provided they are sealed and protected from light.  In-house criteria - Working color developed standards should be replaced after 1 month or sooner id QC indicates a problem.	Reagent grade water.
7196A/SM3500Cr D	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily.	Reagent grade water.
1664A	Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at room temperature.	Same as SSS	Same as SSS	Acetone

**Stability of Standards**  
**Inorganic Spectrophotometric/TOC Methods of Analysis**  
**Table E.6-4**

Method	Expiration Stock Standard Solutions	Expiration Primary/Secondary Dilutions	Expiration Working Standards	Dilution Solvent
330.5/SM4500Cl G	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Reagent grade water.
365.2/SM4500P	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily, but may be used up to 48 hours.	Reagent grade water.
376.2/SM4500S D	No method specified criteria.  In-house criteria - Replace after 1 month, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Reagent grade water.
410.4/SM5220D	Replace after 3 months if no visible microbial growth, or manufacturer's expiration date or sooner if QC indicates a problem Refrigerate	Same as SSS	Same as SSS	Reagent grade water
415.1/SM5310C	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C.	Same as SSS	Same as SSS	Reagent grade water.
SM3500Fe D	One year or manufacturer expiration date prior to color development.	Same as SSS	Non-colored working standards are prepared fresh daily. Method specifies a 6 month criteria for color developed standards provided they are sealed and protected from light.  In-house criteria - Working color developed standards should be replaced after 1 month or sooner id QC indicates a problem.	Reagent grade water.
7196A/SM3500Cr D	No method specified criteria.  In-house criteria - Replace after 6 months, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	No method specified criteria.  In-house criteria - Working standards should be prepared fresh daily.	Reagent grade water.

**Stability of Standards  
Inorganic Metals and Anion Methods of Analysis**

**Table E.6-5**

Method	Expiration Stock Standard Solutions	Expiration Primary/Secondary Dilutions	Expiration Working Standards	Dilution Solvent
200.8/6020	No method specified criteria.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	Same as SSS	Working CAL standards 2 weeks.  Not specified for non-CAL standards. In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem.	1-2% nitric acid and reagent grade water.
300.0/9056	Method states a minimum of 1 month when stored at 4°C.  In-house criteria - Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at 4°C.	Same as SSS	Working standards 1 week. Working standards containing Nitrate, Nitrite and/or O-phosphate should be prepared fresh daily, but may be used up to 48 hours.	Reagent grade water.
314.0/9058	Replace after 1 year, or Manufacturer's expiration date, or Sooner if QC indicates a problem. Store at room temperature.	Same as SSS	Same as SSS	Reagent grade water.

## **12.0 Quality Control**

### **12.1 Isolation of Standards**

12.2.1 Potential contamination and/or cross-contamination is a concern for all samples, sample extracts and standards. In order to minimize the potential damage from this type of problem, standards are stored and isolated by refrigeration independent from samples and sample extracts. Therefore, there are three types of refrigerators for the storage and isolation of these materials.

12.2.2 Standards are also isolated according to method of analysis. Typically, methods are designed to analyze a specific class of compounds or to analyze analytes of a common volatility range or other chemical characteristic. This provides a common separation of standards, samples and sample extracts by method of analysis.

12.2.3 Neat standards are additionally isolated from their common diluted standards of the same method. Neat standards, are stored and sealed in paint cans to further help minimize any potential problems due to cross-contamination.

## **13.0 Calibration and Standardization**

### **13.1 Standardization**

Standards made from neat material are a source of reference standard material used in the determination of various methods of analysis. If available, neat material is purchased as ACS grade and no purity correction is required. However, occasionally, ACS grade is <98 % pure or ACS grade is unavailable and the next best grade of neat material, Reagent Grade, is purchased. In these cases all standard material should be corrected for purity, when the purity is <98%.

## 14.0 Procedure

### 14.1 Identification of Standards, Solvents and Reagents

14.1.1 All reference and non-reference chemicals used in any of the methods of analysis and/or extraction procedures are given an in-house identification number for traceability.

14.1.2 Reagent identifications are segregated by type of method followed by a date, followed by a numerical identification for that day. For example:

FID070503-06 would indicate:

FID - identifies the section of the laboratory which uses GC/FID instruments and where the standard was made or reference material will be used,

070503 - indicates the date in which the standard was logged into the system or prepared, and

-06 - indicates the sixth standard or other reference material identified during that day.

### 14.2 Documentation of Reagents

14.2.1 After the reagents, such non-standard reference chemicals, have been given an identification they are further documented by entering their relevant information such as lot number, manufacturer, expiration date etc. into to our Solvent and Chemical Log.

### 14.3 Documentation of Reference Material

14.3.1 Reference material such as commercially prepared standard mixes and neat material of individual compounds must also be documented into Alpha's tracking system to insure proper identification and traceability of these materials.

14.3.2 Certificates of Analysis ( C of A)



- 14.3.2.1 These types of reference material are typically accompanied with a C of A, which describes the compound or mix of compounds, a lot number, and purity or absolute concentrations of the analytes from the manufacturer.
- 14.3.2.2 Upon receipt of the standards and C of As, the chemist or technician responsible for the preparation and traceability of that standard must date and initial all certificates. The standard is also given a standard identification, which follows that standard throughout its life in the laboratory.
- 14.3.2.3 If the reference material is received without the C of A, it is the responsibility of that analyst, or technician, to call the manufacturer and request this information. Often times this information can be downloaded from the manufacturer's web site.
- 14.3.2.4 Once the standard is logged in to the tracking system the C of A for that standard is then placed into the Reference Material Logbook, a 3-ring binder, and stored for historical reference.

#### 14.4 Preparation of Standards from Neat Material

- 14.4.1 Determine the concentration and volume of the secondary or working standard required. Typically, stock standards are prepared at a concentration of ten to a hundred times the concentration required for the final working standard to allow for additional dilutions.
- 14.4.2 Prepare stock standards in the diluting solvent, or acid.
- 14.4.3 If a 10 mL volumetric flask is used, place approximately 9.8 mL of solvent in the tared volumetric flask.
- 14.4.4 Allow the flask to stand, un-stoppered, for approximately 5-10 minutes or until all wetted surfaces have dried.
- 14.4.5 Weigh the flask to the nearest 0.1mg and record and/or re-tare the balance.
- 14.4.6 Liquid Neat Material

Using a dedicated syringe, disposable pipette, immediately add two or more drops of reference material to the flask, then re-weigh. The liquid must fall directly onto the solvent without contacting the neck of the flask.

#### 14.4.7 Solid Neat Material

Using a disposable weigh boat or weigh paper, measure an approximate weight of the neat material and add directly into the tared flask.

14.4.8 Re-weigh the volumetric flask and record the net gain in weight.

14.4.9 Dilute to volume with the diluting solvent, then mix by inverting the flask several times.

14.4.10 Record all data in the Analytical Standard Log book.

14.4.11 Transfer the stock standard solution into a teflon-lined screw cap vial or other appropriate long-term storage container.

### 15.0 Calculations

#### 15.1 Calculation of Concentration

When secondary or working standards are prepared, final standard concentrations are determined as follow:

15.1.1 Determine the concentration and volume of the diluted standard required,

15.1.2 Calculate the amount of stock standard needed using the following equation:

$$(V_f)(C_f) = (V_i)(C_i)$$

Where:

$V_f$  = Volume of final secondary/working standard,

$C_f$  = Concentration of final secondary/working standard,

$V_i$  = Volume of initial primary/stock standard, and

$C_i$  = Concentration of initial primary/stock standard.

15.1.3 Add the appropriate solvent and dilute to volume, stopper, then mix by inverting the flask several times, and

15.1.4 Record all required information in the Analytical Standard Log book.

#### 15.2 Calculation of Concentration Corrected for Purity

- 15.2.1 When standards are less than 98% pure, concentrations should be adjusted by calculating the correct concentration. For example if a neat standard were prepared at a target concentration of 1000 ug/mL and the standard was 96% pure the actual standard concentration would be:

$$1000 \text{ ug/mL} \times (0.96) \text{ or } 960 \text{ ug/mL.}$$

**16.0 Method Performance**

Not applicable.

**17.0 Pollution Prevention**

Minor amounts of solvents and/or acids are used with this procedure as the diluting reagent. The only other chemicals used in this procedure are the occasional use of neat materials and/or commercially prepared standard mixes used in the preparation of standard dilutions. All are used in extremely small amounts and pose no threat to the environment.

All expired standards and/or reagents must be disposed of properly as described in our Waste Disposal SOP.

**18.0 Data Assessment and Acceptance Criteria for Quality Control Measures**

Not applicable.

**19.0 Corrective Actions for Out-of-Control Data**

Not applicable.

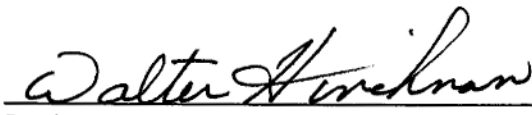
**20.0 Contingencies for Handling Out-of-Control or Unacceptable Data**

Not applicable.

**21.0 Waste Management**

Reference Alpha Analytical's Sample Waste SOP.

Approved By: \_\_\_\_\_

  
Walter Hinchman

QA Officer

Date: January, 2007

Alpha Analytical, Inc.  
255 Glendale Avenue  
Suite 21  
Sparks, NV 89431  
(775) 355-1044

**Standard/Reagent/Inventory Log  
&  
Standard Preparation Log Book**

Book Number: \_\_\_\_\_

Date Start: \_\_\_\_\_

Date End: \_\_\_\_\_

## STANDARD / REAGENT LOGIN / PREPARATION LOG

[illegible]



[illegible][illegible]

## **1.0 Title: Storage Blank Procedures**

Description: VOC samples may be contaminated in a number of different ways. Several types of blanks are used to monitor various sources of potential contamination. Storage blanks are used to monitor possible VOC contamination contributed by the refrigerator or other samples stored in the refrigerator.

## **2.0 Standard Operating Procedure**

2.1 Storage blanks are used to monitor possible VOC contamination. This potential for cross-contamination may occur by the out gassing of highly contaminated samples through the container septum or sample containers that are contaminated on the outside of container and are contaminating the storage space.

VOC samples may also be contaminated by the refrigerant (Freon) used in refrigerators or air conditioning units or by a number of other means.

2.2 Storage blanks are prepared and placed in sample containers and treated as a sample. Storage blanks are not, however, transported to the sample site or exposed to any sampling activities.

2.3 It is the responsibility of the assigned person to prepare, store, and monitor storage blanks for potential contamination.

### **2.4 Refrigerator Storage Blank Preparation (Aqueous Refrigerators)**

2.4.1 The refrigerator monitor prepares refrigerator storage blanks, by filling 40mL VOA vials with organic free water.

2.4.2 When making refrigerator storage blanks an additional blank is prepared as a QC check blank. This blank is prepared and analyzed to assure the original storage blanks are free from VOC contamination. The results of this analysis is documented in the Refrigerator Log Book.

### **2.5 Refrigerator Storage Blank Preparation (Soil Refrigerators)**

2.5.1 The refrigerator monitor prepares refrigerator storage blanks, by filling 40mL VOA vials with 8 mLs of methanol, mimicking the volume of methanol used in the soil extraction procedure.

2.5.2 When making refrigerator storage blanks an additional blank is prepared as a QC check blank. This blank is prepared and analyzed to assure the original storage blanks are free from VOC contamination. The results of this analysis is documented in the Refrigerator Log Book.

### **2.6 Quarterly Storage Blanks**



2.6.1 Twelve refrigerator storage blanks are prepared at the beginning of the calendar year and placed in the designated refrigerators.

2.6.2 During the first week of each new quarter one of the three vials designated for that quarter is analyzed for VOCs.

## 2.7 Semi-monthly Water Storage Blanks

2.7.1 In addition to the quarterly storage blanks, three storage blanks are prepared at the beginning of each 2-week interval and placed in the designated VOC sample refrigerators. At the end of that two week period, one of the three storage blanks is analyzed for VOCs.

## 2.8 Analytical Procedure

Storage blanks are analyzed for VOCs by Method 8260. Analytical results are documented in the Refrigerator Log Book.

## 2.9 Procedure for Non-contaminated Storage Blanks

2.9.1 If VOCs are not found in a storage blank, then it can be determined the refrigerator has been clean of potential air-borne contaminants and no cross-contamination could have occurred for that time period.

2.9.2 If the semi-monthly storage blank indicates no contamination after the two week storage period, another set of three storage blanks is prepared and placed in the designated refrigerator.

The two remaining original vials are also kept as a backup and purged from the refrigerator on the subsequent 2-week interval.

2.9.3 If the semi-monthly storage blank and quarterly blanks indicates no contamination at the end of the year, another set of 12 quarterly storage blanks is prepared and placed in the designated refrigerator.

The remaining original quarterly blanks are purged from the refrigerator.

## 2.10 Procedure for Contaminated Storage Blanks

2.10.1 If the semi-monthly storage blank indicates possible contamination, the second of the three semi-monthly storage blank vials is analyzed to confirm the presence of the contaminant.

2.10.2 If the second semi-monthly storage blank is confirmed positive, then a third stage of confirmation is performed.

2.10.3 This third stage consists of analyzing one of the quarterly storage blanks and potentially the activated carbon. See section 2.8 below.

2.10.4 If the quarterly storage blank analysis is confirmed positive, then it should be considered that all samples within that 2 week period may be suspect for that analyte and appropriate actions should be taken.

#### 2.10.5 Corrective Actions

2.10.5.1 Appropriate actions would be decontamination of the refrigerator or decommissioning of the refrigerator.

2.10.5.2 If the refrigerator contained activated charcoal, it should be removed as a protection against additional cross-contamination while cleaning the refrigerator. A tray of new activated carbon is then placed back into the refrigerator.

2.10.5.3 All samples stored in the refrigerator which has been confirmed positive must be isolated to prevent any additional cross-contamination.

2.10.5.4 All reported sample data, who's sample was stored in the contaminated refrigerator during the associated time period, must be re-evaluated for the identified contaminant.

These samples require a case-by-case decision whether to submit an amended report to the client.

#### 2.11 Housekeeping Procedures for VOC Refrigerators

2.11.1 A tray of activated carbon is placed in all VOC refrigerators to absorb any potential VOC contaminants within the refrigerator.

2.11.2 This carbon should be replaced on an annual basis and the date of replacement noted on the refrigerator log.

2.11.3 If it was determined, by the analysis of the storage blanks, cross-contamination may have occurred, then the activated charcoal may be analyzed as another confirmation.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**TABLE E.7-1**  
**TABLE OF STORAGE BLANKS AND FREQUENCY**

<b>VOC Refrigerator</b>	<b>Bimonthly Analysis</b>	<b>Quarterly Analysis</b>
SAR-1B	Every other Friday	Every 3 Months
SAR-5D	Every other Friday	Every 3 Months
SAR-8A	Every other Friday	Every 3 Months
SAR-14A	Every other Friday	Every 3 Months
SAR-10A	Every other Friday	Every 3 Months
SAR-11A	Every other Friday	Every 3 Months
SAR-12A	Every other Friday	Every 3 Months

**1.0 Title:** A Practical Application Guide for Performing a Determination of Capabilities (DOC) and Method Detection Limit (MDL) Study

**2.0 References:** Definition and Procedure for the Determination of the Method Detection Limit, 40 CFR, Part 136, Appendix B, Revision 1.11.

National Environmental Laboratory Accreditation Conference, Quality Systems, Chapter 5, Revision 16, 2002 and Appendix C, 2003.

Department of Defense Quality Systems Manual for Environmental Laboratories, Version 3, January 2006.

### **3.0 Demonstration of Capability**

#### **3.1 Initial Demonstration of Capability Study**

An Initial Demonstration of Capability (IDC) is performed prior to using test methods or when a significant change in instrument, test methods or personnel have been made. Demonstration of Capability (DOC) does not test the performance in real world samples, but in available clean matrix ( a sample matrix in which no target analytes or interferences are present at concentrations that would impact the results of a specific test method). In addition, for analytes which do not lend themselves to spiking, e.g, TDS, the DOC may be performed using quality control samples.

1) Clarification: "Significant change" refers to any change in personnel, instrument, test method, or sample matrix that potentially impacts the precision, accuracy, sensitivity and selectivity of the data ( for example, a change in detector, column type, sample matrix, or a significant method revision). All new analysts, regardless of experience on that instrument in another laboratory, shall complete a demonstration of capability.

#### **3.2 Continuing Demonstration of Capability**

Once the IDC has been performed a continuing demonstration of capability is performed annually and the documentation of continued proficiency by at least one of the following once per year:

- i. acceptable performance of a blind sample (single blind to the analyst),

Note: successful analysis of a blind performance sample on a similar test method using the same technology, e.g., GC/MS volatiles by purge and trap for Methods 524.2, 624 or 5030/8260, would require documentation for one of the test methods.

- ii. a new IDC study, or a study from a similar test method,
- iii. at least four consecutive laboratory control samples with acceptable levels of

precision and accuracy, or

- iv. if i-iii cannot be performed, analysis of authentic samples with results statistically indistinguishable from those obtained by another trained analyst.

Note: Continuing demonstration of capabilities is continuously monitored during the normal course of sample analysis through the use of LCS data evaluation. Laboratory control samples are typically spiked and monitored at the same spike concentration and windows of acceptability as used for the IDC study. Therefore, analytical batch LCS data may also be used to satisfy this requirement.

### 3.3 Work Cells

A work cell is considered to be all individuals who see a sample through the complete process of preparation, extraction and analysis.

#### 3.3.1 Individual Demonstration of Capability within the Work Cell

This work cell or “group” as a unit must meet the DOC criteria and this DOC study must be documented for all individuals within that work cell. Even though the work cell operates as a “team,” the demonstration of capability at each individual step in the sequence, as performed by each individual team member, remains of utmost importance, e.g., each team member must demonstrate capability in his/her area of responsibility in the sequence.

2) Clarification: It is not the intent to require each combination/permutation of work cell members to demonstrate group capability since DOC is for the individual only. Even though the work cell operates as a “team,” the demonstration of capability at each individual step in the sequence, as performed by each individual analyst/team member, remains of utmost importance. For example, if multiple individuals contribute to a single analytical result (e.g., perform preparation, extraction, and analysis) and that result meets appropriate acceptance criteria, then all individuals have demonstrated capability.

Work cells can not be defined as a group within that team who perform the same step in the same process (for example, extractions for Method 8270), represented by one extraction chemist who has demonstrated capability for that step.

#### 3.3.2 New Members within a Work Cell

3.3.2.1 When a work cell is used, and a member of a work cell changes, the new employee must work with experienced personnel in that area of the work cell where they are employed, e.g., a new analyst should work with an experienced analysts for initial training.

3.3.2.2 This new work cell must demonstrate acceptable performance through acceptable continuing performance checks such as laboratory control samples. This performance is documented and the preparation batch following the change in personnel must not result in the failure of any batch acceptance criteria, e.g., method blank and laboratory control sample, or the demonstration of capability must be repeated. In addition, if the entire work cell is changed or replaced, the work cell must perform a new complete DOC study.

### 3.4 DOC Study Training Documentation

The DOC performance of individual analysts as well as all members of the work cell are documented through the use of the DOC Summary Data Form and individual training records.

#### 3.4.1 Certification Statement

The DOC certification statement, Figure E.8-1, is used to document the completion of each demonstration of capability. A copy of the certification statement is retained in the personal training files of each affected employee.

#### 3.4.2 Supporting Data

Data applicable to the study does not need to be attached to the form, but is retained and available.

### 3.5 Continuing DOC Training Documentation

Technical training of each member of the technical staff is kept up-to-date (on-going) by requiring each member to have on record (training file) a certificate that they have read, understood, and is using the latest version of QA Manual and SOPs which relate to his/her job responsibilities.

3.5.1 The Certification of Initial/Continued Proficiency Requirements are used to document the completion of each demonstration of capability. A copy of the certification statement is retained in the personnel training records of each affected employee.

#### 3.5.2 Supporting Data

All associated supporting data, such as quantitation reports etc., necessary to reproduce the analytical results summarized in the Certification Statement is also retained.

### 3.6 DOC and Continuing DOC Study Organization

- 3.6.1 Data is peer reviewed, to verify the analytical method requirements associated with the study have been achieved. This is typically documented using the Internal Data Quality Audit Checklist, see QAM, Vol I, Section 14, Audits.
- 3.6.2 Data packets should be prepared with the analytical sequence log, followed by the sample data in chronological order. Batch QC summary sheets and extraction logs should also accompany this data packet. This helps organize the data for peer review.
- 3.6.3 A copy of the standards preparation logbook clearly documenting the source of the standards, and/or a brief summary of the stock standard solution, secondary solution, working standards, etc. should also be placed in a separate document packet.

Note: A typical level IV data package is more than adequate for complete study documentation.

### 3.7 DOC Study Procedure

The following guidelines are used when completing a initial demonstration of capability or continuing demonstration of capability study.

- 3.7.1 A Quality Control (QC) sample is obtained from an outside source or the QC sample may be prepared using stock standards that are independently prepared from those used in the instrument calibration.
- 3.7.2 The analytes are spiked into a volume of clean matrix sufficient to prepare four aliquots at the method specified concentration or at a concentration 1-4 times the limit of quantitation.

3) Clarification: The limit of quantitation is synonymous with the reporting limit. This would suggest that the DOC study should be analyzed at a concentration of 1-4 times the reporting limit; while the NELAC standards also allow the use of LCS samples which are typically spiked at 5-10 times the reporting limit to be used. Therefore, a concentration of 1-10 times the reporting limit is acceptable.

- 3.7.3 At least four aliquots are prepared and analyzed according to the test method either concurrently or over a period of days.
- 3.7.4 Using all the results, calculate the mean recovery and the standard deviations of the population sample (n-1) for each parameter of interest.

4) Clarification: The average recovery (not individual recoveries) is compared to the accuracy limits that are either method specified or determined by the laboratory. For most cases this will be the laboratory established LCS window of acceptability.

3.7.5 Compare the information from above to the corresponding acceptance criteria for precision and accuracy in the applicable test method or in laboratory-generated acceptance criteria (if there are not established mandatory criteria). If all parameters meet the acceptance criteria, the analysis of samples may begin. If any one of the parameters does not meet the acceptance criteria, the performance is unacceptable for that parameter and the test needs to be repeated.

3.7.6 When one or more of the target compounds fail the acceptance criteria then the analyst should proceed according to the following:

3.7.6.1 Beginning with (3.7.3) above, repeat the test for all parameters that failed to meet criteria.

3.7.6.2 Repeated failure, however, generally confirms a problem with the measurement system. If this occurs, locate and correct the source of the problem and repeat the test for all compounds of interest beginning with (3.7.3) above.

#### **4.0 Test Method Evaluation**

##### **4.1 Evaluation for Standardized Methods**

For all environmental organic and inorganic analytical methods of analysis, the method must be evaluated for those parameters that adversely affect data quality. These parameters are things such as method detection limit, reporting limit, accuracy and precision.

##### **4.2 Evaluation for Non-standardized Methods**

As part of method development, and to ensure continuous quality of data, QC criteria must be proposed and established that is consistent with similar methods or technology. At a minimum these QC requirements must address the following:

- Calibration,
- Contamination,
- Precision and accuracy,
- Interference, and
- Analyte identification.

##### **4.3 Limit of Detection (LOD)**



- 4.3.1 The LOD is defined as an estimate of the minimum amount of a substance that an analytical process can reliably detect. An LOD is analyte and matrix specific and may be laboratory dependent.

Note: The limit of detection was formerly know as the Method Detection Limit (MDL).

- 4.3.2 One way to establish a LOD, is to perform a MDL study defined as, the minimum concentration of a substance that can be measured and reported with 99% confidence that the analyte concentration is greater than zero and is determined from analysis of a sample in a given matrix containing the analyte.

A properly conducted LOD study requires that all sample processing steps be included in the determination of the limit of detection.

#### 4.3.3 LOD Study Frequency

- 4.3.3.1 Appendix D.1.2.1 of the System 5 NELAC standards states the following:

*An LOD study is NOT REQUIRED for a test method when the test results are not reported outside of the calibration range.*

5) Clarification: DOD requires a LOD study be conducted and determined regardless of the NELAC standards.

- 4.3.3.2 NELAC also requires the LOD study be conducted by the protocol in the mandated test method or applicable regulation, e.g., 40 CFR, Part B, Appendix B.

6) Clarification: Alpha follows the procedures as described in 40 CFR, Part 136, Appendix B, Definition and Procedure for the Determination of the Method Detection Limit.

#### 4.3.4 LOD Study Procedure

- 4.3.4.1 The LOD study is not conducted using real world samples, but using an available clean matrix (a sample matrix in which no target analytes or interferences are present at concentrations that would impact the results of a specific test method). This is the same matrix as used for the preparation of the method blank or laboratory control sample, e.g., reagent grade water for water matrices, or ottowa sand, sodium sulfate or teflon chips for soil matrices.

#### 4.3.4.2 Cleanup Procedures

LOD studies are generated using the preparatory and cleanup procedures routinely used on sample.

#### 4.3.4.3 LOD Study Frequency

An LOD study is performed initially for all compounds of interest in each test method. An LOD study is determined each time there is a change in the test method that affects how the test is performed or when a significant change in instrumentation occurs that affects the sensitivity of the analysis.

Note: See the clarification box for “significant change” as it applies to both the DOC and LOD studies.

#### 4.3.4.4 Number of Replicates

It is recommended to analyze a minimum of ten replicates containing the analytes of interest at the selected concentrations over the course of several days.

7) Clarification: As stated in 40 CFR 136B, the MDL study shall be determined using a minimum of seven replicates. In addition, several regulatory agencies require that if more than seven replicates are processed, data cannot be excluded, unless exclusion is supported with sound, documented technically based justification (e.g., outlier test). The state of California does not agree with this and will not allow exclusion of any data regardless of reason, (e.g., outlier tests are not allowed) and/or number of replicates.

#### 4.3.4.5 Spike Source

A Quality Control (QC) standard is obtained from a source independent from the source used in the instrument calibration. If a completely independent source cannot be obtained, then as a last resort, the QC sample may be prepared using stock standards that are independently prepared from those used in the instrument calibration.

8) Clarification: Since the LOD study samples must be prepared from a source independent from the calibration source; it may be wise to use the same independent second source for the Initial Calibration Verification (ICV), DOC study and LOD study.

#### 4.3.4.6 Spike Concentration

Determine the LOD spike concentration that corresponds to the following criteria:

- an instrument signal/noise ratio within the range of 2.5 to 5.0,
- the region of the standard curve where there is a significant change in sensitivity, i.e., a break in the slope of the curve.

4.3.4.7 Calculate the Standard Deviation (SD) of each of the analytes.

4.3.4.8 Using the Student t-Test, determine the LOD for each analyte as follows:

$$\text{LOD} = (t_{n-1}) (\text{SD}),$$

where ( $t_{n-1}$ ) is the student's t value for a 99% confidence level and a standard deviation estimate with n-1 degrees of freedom, Table E.8-2.

4.3.4.9 Calculate the LOD for each analyte and matrix.

4.3.4.10 LOD Verification Check

Analyze an LOD Verification Check standard immediately following the LOD study. If an annual LOD study is not performed, LOD verification checks should be performed quarterly. The LOD should be evaluated as follows:

- The LOD Verification Check standard should be analyzed at approximately 2 times the current LOD. The concentration of the verification standard should be no greater than 3 times the LOD for single analyte tests and 1-4 times the LOD for multiple analyte tests.
- The LOD Verification Check standard is acceptable if the analyte is detected using method specified criteria. If the method has no established LOD criteria, the check standard must produce a signal that is at least 3 times the background noise.
- If the LOD Verification Check standard fails, additional LOD verification checks may be performed at a higher level to set the LOD higher, or the LOD study must be reconducted.

4.3.5 LOD Evaluation

4.3.5.1 If the calculated LOD is higher than the spike concentration, then the study should be repeated at a higher concentration.

4.3.5.2 Compare calculated LOD results to ensure they are at or below the Limit of Quantitation (LOQ).

9) Clarification: Method 8000B requires the LOQ be established at or above the lowest calibration point established during the initial calibration curve. This should be further refined and supported by establishing LOQs that have calculated LODs at or below this limit.

4.3.5.3 Compare the ratio between the mean recovered concentration and the calculated LOD. This ratio should be between 1 to 5 for reagent water matrices and 1 to 10 for other matrices. If the calculated LOD is less than 1/10 of the spike concentration, then the study should be repeated at a lower concentration.

4.3.5.4 LOD From Multiple Instrument.

4.3.5.4.1 If multiple LOD results are generated from multiple instruments with identical configurations, then the highest LOD among those should be used in reporting data from all of those instruments. If a lower LOD is reported for specific samples, then the samples must have been run on that specific instrument on which the lower LOD was generated.

4.3.5.4.2 If multiple instruments with identical configurations are used, then conduct a LOD study on at least one of the instruments and confirm the attainability of that LOD on all instruments by using an LOD verification check sample. The LOD verification check standard must be performed quarterly on every instrument.

4.3.6 LOD Study Documentation

4.3.6.1 All associated supporting data, such as quantitation reports etc., necessary to reproduce the analytical results summarized in the DOC Certification Statement is also retained.

4.4 Limit of Quantitation (LOQ)

4.4.1 The LOQ is defined as the minimum level, concentration or quantity of a target analyte that can be reported with a specified degree of confidence.

Note: The limit of quantitation was formerly know as the Practical Quantitation Limit (PQL), Minimum Reporting Limit (MRL) or simply the reporting limit.

4.4.2 The lowest calibration point must be at or below the established LOQ, but may be the same.

4.4.3 The LOQ is determined for each analyte at a concentration no less than 3 times the LOD.

4.4.3.1 If a client requires a reporting limit below the established LOQ, method modification is required or the client will be required to accept the established LOQ as the lowest technically valid value that can be provided.

4.4.3.2 If analytes are reported below the established LOQ, they should be flagged as follows:

- If the analyte is reported (detect or non-detect) down to one-half of the LOD, the data is flagged with a J.
- If the analyte is reported as a non-detect down to the LOD, the data is flagged with a U.
- If the analyte is detected between the LOD and LOQ, the data is flagged with a J.

4.4.4 LOQ Verification Check

Analyze a LOQ check standard for each analyte at a concentration of 1-2 times the LOQ. The LOQ study is not required for components or methods for which spiking solutions are not available or inappropriate, e.g. pH.

4.4.5 LOQ Evaluation

If the recovery of the LOQ verification check standard is within established test method criteria, or client data quality objectives for accuracy, than the LOQ is acceptable.

4.4.6 LOQ Frequency - The LOQ is verified annually.

## 5.0 Outlier Test

Even though CA NELAP does not allow the use of the Dixon Outlier test it is included for reference.

5.1 Often in a series of measurements, one or more of the results will differ greatly from the other values. Theoretically, no data results should be rejected, because it may indicate either a faulty technique that casts doubt on all results or the presence of a true variant in the distribution. In practice, reject the result of any analysis in which a known error has occurred (data rejection must be supported with proof of the

error). In environmental analysis, extremely high and low concentrations of contaminants may indicate the existence of areas with problems or areas with no contamination, so they should not be rejected arbitrarily.

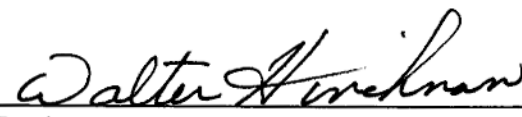
- 5.2 If a set of data is ordered from low to high, and the average and standard deviation are calculated, then suspected high or low outliers can be tested by the following procedure. First calculate the statistic T:

$$T = (X_H - X_{AV})/S \text{ for a high value, or}$$

$$T = (X_{AV} - X_L)/S \text{ for a low value.}$$

- 5.3 Compare the value of T to the Outlier Statistics Table E.8-3 at the 1% level of significance. If the calculated T is larger than the table value for the number of measurements, n, then the  $X_H$  or  $X_L$  is an outlier at that level of significance.

Approved By: \_\_\_\_\_

  
Walter Hinchman  
QA Officer

Date: January, 2007

**Demonstration of Capability  
Certification Statement**

Date: \_\_\_\_\_  
Alpha Analytical, Inc.  
255 Glendale Ave Ste21  
Sparks, NV 89431-5778

Analyst Name: \_\_\_\_\_  
Matrix: \_\_\_\_\_  
Method #: \_\_\_\_\_  
SOP #/Rev #: \_\_\_\_\_

Parameters: MDL/IDC

We, the undersigned, CERTIFY that:

- 1) The analyst identified above, using the cited test method(s), which is in use at this facility for the analyses of samples under the National Environmental Laboratory Accreditation Program, have met the Demonstration of Capability.
- 2) The test method(s) was performed by the analyst identified on this certification.
- 3) A copy of the test method(s) and the laboratory-specific SOPs are available for all personnel on-site.
- 4) The data associated with the demonstration capability are true, accurate, complete and self-explanatory (1).
- 5) All raw data (including a copy of this certification form) necessary to reconstruct and validate these analyses have been retained at the facility, and that the associated information is well organized and available for review by authorized assessors.

\_\_\_\_\_  
Analyst

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

\_\_\_\_\_  
Quality Assurance Officer

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

This certification form must be completed each time a demonstration of capability study is completed.

- (1) True: Consistent with supporting data.  
Accurate: Based on good laboratory practices consistent with sound scientific principles/practices.  
Complete: Includes the results of all supporting performance testing.  
Self-explanatory: Data properly labeled and stored so that the results are clear and require no addition explanation.

**Figure E.8-1**

<b>TABLE E.8-1 TABLE OF STUDENT'S <math>t</math> VALUES AT THE 99 PERCENT CONFIDENCE LEVEL</b>		
NUMBER OF REPLICATES	DEGREES OF FREEDOM ( $n-1$ )	$t_{n-1, .99}$
4	3	4.541
5	4	3.747
6	5	3.365
7	6	3.143
8	7	2.998
9	8	2.896
10	9	2.821
11	10	2.764
12	11	2.718
13	12	2.681
14	13	2.650
15	14	2.624
16	15	2.602
17	16	2.583
18	17	2.567
19	18	2.552
20	19	2.539



# Outlier Statistics Table

Table E.8-2

Number of Measurements n	Critical Value 1%
3	1.15
4	1.49
5	1.75
6	1.94
7	2.10
8	2.22
9	2.32
10	2.41
12	2.55
14	2.66
15	2.71
16	2.75
18	2.82
20	2.88
30	3.10

**TABLE E.8-3**  
**STATISTICAL CALCULATIONS**

STATISTIC	SYMBOL	FORMULA	DEFINITION	USES
Mean	$\bar{x}$	$\frac{\sum_{i=1}^n X_i}{n}$	Measure of central tendency	Determine average value of measurements
Standard Deviation	SD	$\left( \frac{\sum (x_i - \bar{X})^2}{(n - 1)} \right)^{\frac{1}{2}}$	Measure of relative scatter of the data	Calculating variation of measurements
Relative Standard Deviation	RSD	$\left( \frac{s}{\bar{x}} \right) \times 100$	Relative standard deviation, adjusts for magnitude of observations	Assess precision for replicate results
Percent Difference	% D	$\frac{x_i - x_2}{x_1} \times 100$	Measure of the difference of 2 observations	Assess accuracy
Relative Percent Difference	RPD	$\left( \frac{(x_1 - x_2)}{\left( \frac{x_1 + x_2}{2} \right)} \right) \times 100$	Measure of the variability that adjusts for the magnitude of observations	Assess total and analytical precision of duplicate measurements
Percent Recovery	% R	$\left( \frac{x_{measured}}{x_{true}} \right) \times 100$	Recovery of spiked compound in pure matrix	Assess accuracy
Percent Recovery	% R	$\frac{\text{Value of Spiked Sample} - \text{Value of Unspiked Sample}}{\text{Value of Added Spike}} \times 100$	Recovery of spiked compound in sample matrix	Assess matrix effects and total precision

x = Observation (concentration) n = Number of observations

## **1.0 POLICY FOR MANUAL AND AUTOMATED INTEGRATION PROCEDURES**

### **2.0 Manual Integration Policy**

This procedure is intended to give guidelines to all analysts in determining the most appropriate manual or automated integration routines. There are no mandated rules that can be applied to the practice of integration. However, it is Alpha's policy to produce analytical data using the automated and manual integration practices, in a manner that is:

- Non-arbitrary (meaning standards, control samples, and client samples are all integrated using consistent integration practices).
- Rational (data can be backed up with the reason for a particular integration practice).

It is Alpha's policy to document manual integrations as follows:

- Manual integrations should only be integrated one time. However, occasionally a second manual integration is required. These types of integrations are documented with the normal description justifying the reason for the manual integration. When more than two manual integrations are used on any one analyte it must be supported with a description of why three integrations were required in addition to the primary reason for the manual integration to begin with. If more than three manual integrations are warranted, the integration must be initialed by the analysts direct supervisor.
- Manual integrations of all data in support of the DOD IR Program and the State of Arizona must be performed in such a manner as to produce a complete audit trail. Audit trails are most easily produced by printing a "before" and "after" chromatogram and initialing and dating the printouts.

This procedure does not imply that an additional electronic data file needs to be kept; however, it does imply the final data file be fully supported electronically. In addition, the "before" data file chromatogram needs to be printed in such a manner to visually verify the manual manipulation of peaks and/or a notation of rationale. This does not imply each and every peak be printed out on individual pages, but it is a strong suggestion that a "clear picture speaks a thousand words." The principal purpose of printing a "before" and "after" chromatogram is to develop and track a complete audit trail of all manual integrations.

Alpha's other general policy directives for manual integration are as follows:

- This policy is applicable to all data produced from environmental analyses performed by GC, GC/MS, HPLC, IC or any other instrument capable of producing a chromatogram used to report quantitative data.
- This policy is not intended to teach chromatography, but rather provide guidance when chromatographic difficulties present itself.

- This policy is independent of any software.
- This policy is consistent with the EPA's procedures and methods for environmental analyses, good laboratory practice and Alpha's ethics policy conducive to producing consistent, scientifically valid and defensible data for use by any end user.

### **3.0 Summary**

The goal of analytical chromatography is to separate sample constituents within a reasonable time. Baseline resolution of each target analyte from co-extracted materials provides the best quantitative results but is not always possible to achieve.

Some analytical procedures list analytes that may not all be resolved from one another. Therefore, while each of these methods is suitable for the listed compounds, they may not be suitable to measure the entire list in a single analysis. In addition, some methods include compounds that are isomers or closely related compounds which are well-known as co-eluting or are not completely separable. In these instances, the results should be reported as the sum of the two (or more) compounds.

This policy is written as a guide to help aid analysts in determining if there is acceptable baseline resolution and peak shape and that appropriate integration routines are being used. It is also the purpose of this policy to give analysts guidance in how to consistently integrate; produce accurate quantification, and how to handle data in situations where accurate quantification is difficult, such as coeluting isomeric pairs. The importance of the analysts' experience in performing these types of analyses to the ultimate success of the methods can not be overemphasized.

It is not the intent of this policy to provide guidelines for all possible case scenarios when less than perfect integration is achievable. However, this policy is written such that, when doubt exists regarding integration, then the analyst must use the philosophy of consistent integration techniques that are also compatible with the guidelines of Alpha's Ethics Policy.

This policy is also written as a philosophical approach to help aid analysts' decision making procedures in the integration of chromatograms.

### **4.0 Consistent Vs. Inconsistent Integration**

All calibration standards and calibration verification standards should be integrated consistent with the other standards in the curve based on the integration technique the analyst has chosen. Regardless of concentration, calibration standards should have basically the same chromatography and can all be integrated in the same fashion.

If this is not the case, the analyst must find the problem. Is there column overloading at the high concentrations? Does the separation between compounds become worse as the

concentration increases? Is the low standard response so small that it does not produce a three to one signal to noise ratio? The analyst needs to evaluate whether the instrument is linear throughout the whole curve and if the column capacity is sufficient for the amount of material being analyzed. The calibration range may need to be shortened, a different column used, or less injection volume.

Generally, samples should be integrated in a manner consistent with the calibration standards. Because of matrix effects, interferences, and compounds at varying concentrations, it is not always possible nor appropriate to integrate samples exactly the way the calibration samples are integrated. In calibration standards, each compound is usually at the same concentration as every other compound in the standard. In a sample, each compound will probably have a different concentration. Therefore, an integration technique that was accurate for a calibration standard may not be accurate for a sample. The analysts' experience, detailed examination, and good judgement is required in these cases.

## **5.0 Manual Integration Documentation**

Computer generated automatic integrations must be supported by sound integration routines and parameters. These integration routines are documented and supported by the method file saved in support of the final analytical data. When these computer generated integration routines need to be adjusted; then the manual integrations must be based upon industry accepted chromatographic integration practices.

Manual integrations performed on all Initial Calibrations must be documented. Documentation must include a before and after chromatogram or reconstructed ion chromatogram. In addition, the analyst must document these IC manual integrations with a chromatographically valid reason to support the manual integration.

In summary, there are two situations which automatically trigger the printing of a before and after chromatogram when manual integrations have been performed, and three situations which trigger the proper documentation of these chromatograms to produce a complete audit trail:

- a) All initial calibrations (before and after chromatograms plus documentation);
- b) All sample and batch QC data associated with DOD and Arizona samples (before and after chromatograms plus documentation); and,\
- c) When more than one manual integration has been performed on a single analyte (proper documentation only).

Total Petroleum Hydrocarbon methods, require an integration routine encompassing an extremely wide retention time window. For these types of compounds, (toxaphene, technical chlordane, and PCBs) manual integration is essentially required with each chromatographic run as the software is incapable of making these types of integration decisions. Therefore, the documentation of manual integrations with before and after chromatograms and the

justification of those manual integrations are not required; rather good chromatographic judgment is required by the analyst.

Reasons to perform manual integrations are documented using the mnemonic scheme described as follows:

Mce	<b>M</b> anual integration due to <b>co</b> -eluting (target and/or non-target analyte)
Mpnfi	<b>M</b> anual integration <b>peak not</b> fully integrated (typically due to peak tailing, small peaks near the baseline, etc.).
Mpoi	<b>M</b> anual integration <b>peak over</b> integrated (typically due to negative peak, or a dip in the baseline, etc.)
Mmp	<b>M</b> anual integration <b>missed peak</b> or peak simply not integrated.
Mipi	<b>M</b> anual integration <b>incorrect peak</b> integration.
Mrts	<b>M</b> anual integration <b>retention time</b> shift
Mcsm	<b>M</b> anual integration <b>complex sample matrix</b>
Mcd	<b>M</b> anual integration <b>chromatographic degradation</b> (creates a situation where the software routine, instrument needs to undergo maintenance).
Mad	<b>M</b> anual integration <b>analyst discretion</b>

## 6.0 Defining a Peak and Acceptable Peak Shape

A peak is defined as a pictorial representation of the response of the compound where the response of the compound is at least three times the background noise. Background noise can be from the instrument electronics, solvent, sample matrix, mobile phase, gases, column bleed, etc. Ideal peak shape is a bell curve or Gaussian. The perfect peak should be symmetrical with a round top. The compound signal should take a few seconds to rise and fall. The rise and fall of a signal too quickly would generally be indicative of a noise spike.

Due to detectors, matrix and compound polarity, peaks may become broadened, have shoulders or tailing. These are all acceptable as long as it is not indicative of co-elution. Some methods have specific criteria as to how much of these imperfections are allowable. Judgement is necessary as to what is excessive. An analyst should base the judgement of current chromatography on historical instrument performance.

## 7.0 Baseline Resolution and Ideal Peak Shape

The best case chromatography is an ideally shaped peak residing on a flat baseline completely resolved from any other peak or interference. Baseline integration is a line drawn from each endpoint of the peak, baseline to baseline, including the entire area under the curve of the peak (Figure E.9-1). This is appropriate when the quantification will be based on peak area.

from each endpoint of the peak, baseline to baseline, including the entire area under the curve of the peak (Figure E.9-1). This is appropriate when the quantification will be based on peak area.

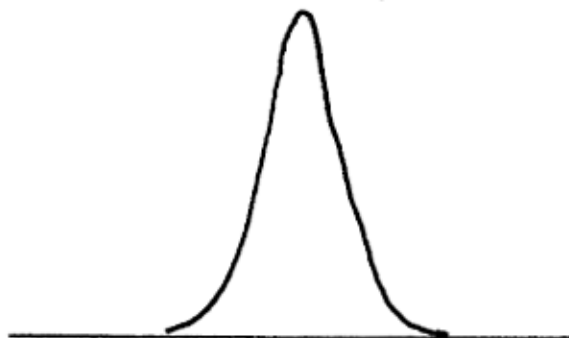


Figure E.9-1

## 8.0 Integration Techniques

### 8.1 Irregular Peak Shape

When there is an irregular peak shape first ensure that it is not excessive and indicative of instrument problems or column overload. Then ensure this is not due to any co-elution problems from either target compounds or the matrix. If neither of the above is the reason for the irregular peak shape, then the general guideline is to integrate the entire peak (Figure E.9-2).

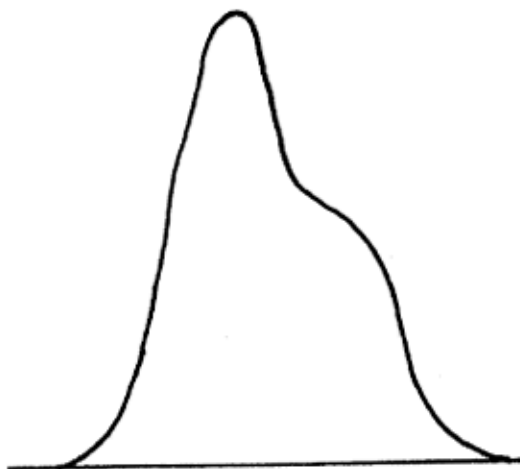


Figure E.9-2

## 8.2 Less than 100% Baseline Resolution

### 8.2.1 Dropping a Perpendicular

This integration technique is used when two compounds partially co-elute and there is a discernable valley between the two peaks. A perpendicular line is drawn from the valley between the two peaks to the baseline.

This technique assumes that both compounds have symmetrical peak shape and equal detector response, such as in the case of isomeric pairs. If this assumption is correct, then the area contributed from one compound as compared to the other compound will be the same and the quantification will be equivalent to 100% baseline resolved. Usually this is not the case, which means the detector has different sensitivity, peak shape and amount of resolution or co-elution must be carefully evaluated before using this technique.

Several methods state that acceptable resolution is achieved for structural isomers if the height of the valley between the two isomer peaks is less than 25% of the sum of the two peak heights. Otherwise, structural isomers are identified as isomeric pairs. The method formula for acceptable resolution is  $V < 0.25(P_1 + P_2)$ . Where V is the height of the valley between the peaks; P1 is the left peak and P2 is the right Peak (Figure E.9-3).

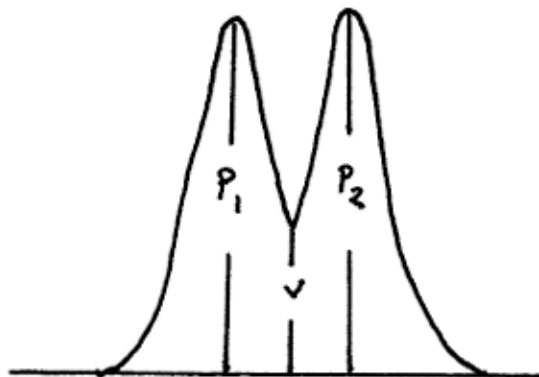


Figure E.9-3.

A second approach used by many methods in determining adequate resolution between two peaks is defined by the equation:



$$R = \frac{t}{w}$$

where  $t$  is the difference in elution times between the two peaks and  $w$  is the average peak width, at the baseline of the two peaks. Resolution is adequately resolved when  $R > 1.0$  using this formula (Figure E.9-4).

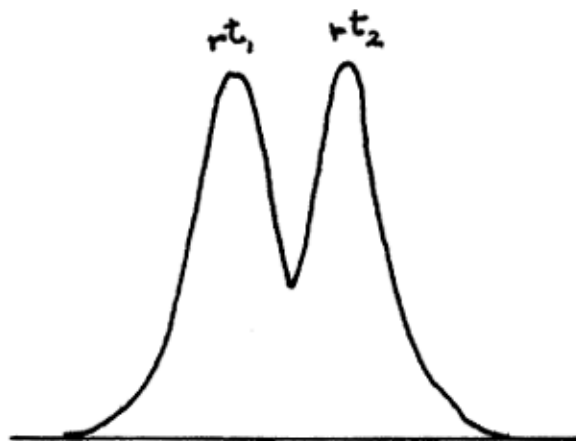


Figure E.9-4

While structural isomers represent the best conditions under which to integrate peaks by dropping a perpendicular, this criteria can be applied to all co-eluting peaks. If in the analyst's judgement adequate resolution cannot be achieved than this technique should not be used. As an alternative, either integrate the two peaks together and report them with a footnote as co-eluting peaks or do further method development to achieve better separation.

- 8.3 In GC/MS methods where chromatograms of a single quantitation ion are used, more unique ions that have adequate intensity should be considered for cases of close compound elution.
- 8.4 Valley to Valley and Baseline to Valley Integration

The valley to valley technique involves drawing the integration line from the valley between two peaks to another valley between two peaks.

Baseline to valley integration is drawing the line from the baseline at one end of the peak to the valley between two peaks at the other end of the peak being integrated.

The weakness of these two techniques occurs with the quantitation of samples that have a calibration produced where peak integration was produced on a flat baseline

as compared to samples which do not exhibit the same flat baseline attribute, and thus may under quantitate or over quantitate actual sample concentration.

The following examples attempt to explain the weakness of this integration technique.

Example 1. Two peaks have similar detector response such that standards would be integrated by dropping a perpendicular where less than 100% baseline resolution occurs (Figure E.9-5).

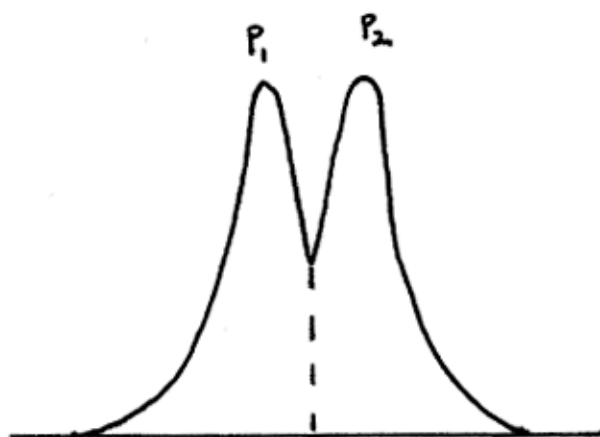


Figure E.9-5.

Situation A: In this situation peak #1 (P1) has a greater signal intensity or response than peak #2 (P2) (Figure E.9-6.).

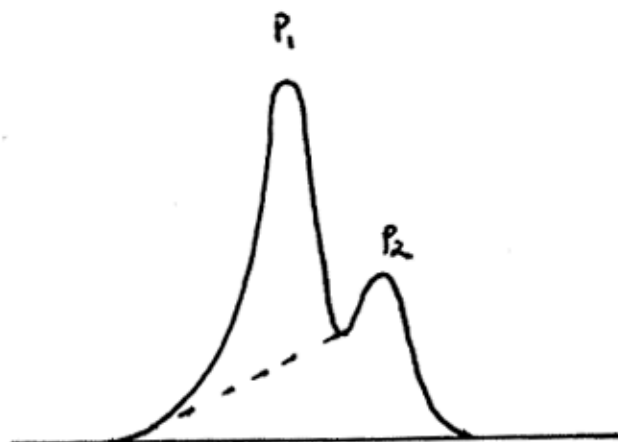


Figure E.9-6.

If P1 is integrated baseline to valley, then P1 will be under quantitated and P2 will be over quantitated. The percentage error associated with P1 is less than the error associated with P2. This may be significant if P2 is at or slightly above the reporting limit where the error could be as much as or greater than 100% or a factor of 2 of the actual concentration. It may be appropriate to drop a vertical (Figure E.9-7) or integrate valley to baseline (Figure E.9-8).

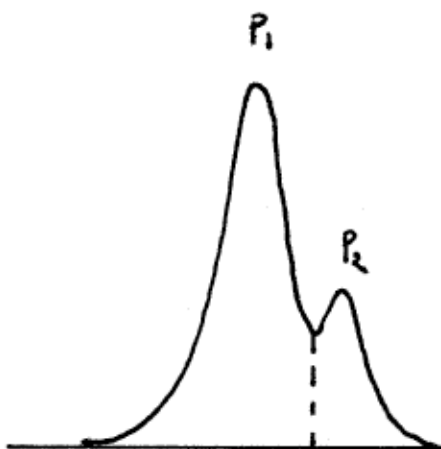


Figure E.9-7.

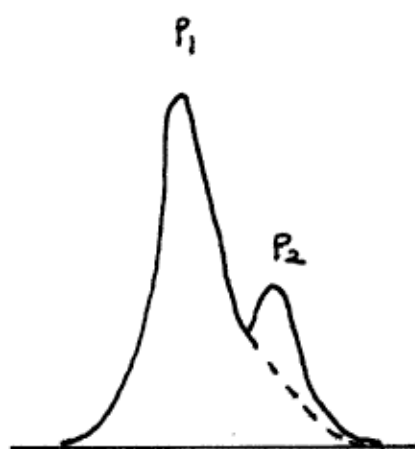


Figure E.9-8

The weakness of integrating valley to baseline (Figure E.9-8) are the same problems as discussed in the following section, peak skimming; namely, excluding area from P2 thought to be contributed by the larger peak. Normally peak skimming is not a recommended practice since peak tailing is very inconsistent. Therefore, it may be most appropriate to drop a vertical and integrate.

Example 2. Two peaks have similar detector responses such that standards would be integrated by dropping a perpendicular where less than 100% resolution occurs. The same situation as indicated in Figure E.9-5.

Situation B: In this situation P2 has a greater signal response than P1 (Figure E.9-9).

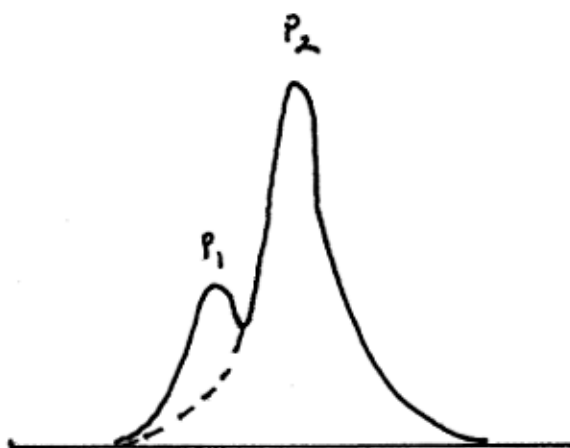


Figure E.9-9

If P1 is integrated baseline to valley (Figure E.9-9), then the same problems associated with Figure E.9-8 exists. Area is being excluded from P1 thought to be contributed by P2. Secondly the error associated with P1 will be significant if it is at or slightly above the reporting limit.

If P1 is integrated valley to baseline (Figure E.9-10) then P1 will be over quantitated and P2 will be under quantitated and it may be more appropriate to drop a vertical.

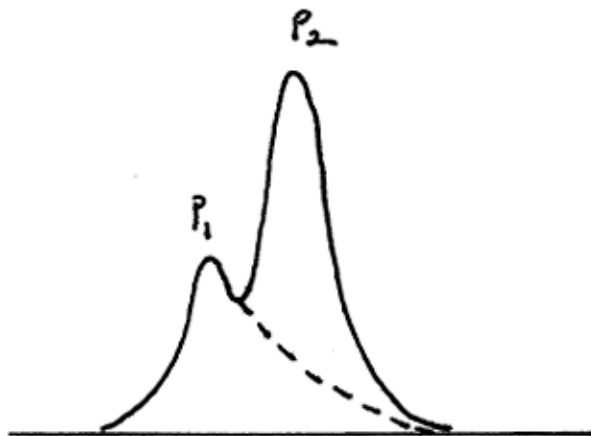


Figure E.9-10.

- Example 3. The two peaks have widely differing detector responses such that the standards would be integrated by dropping a perpendicular where less than 100% baseline resolution occurs (Figure E.9-11.).

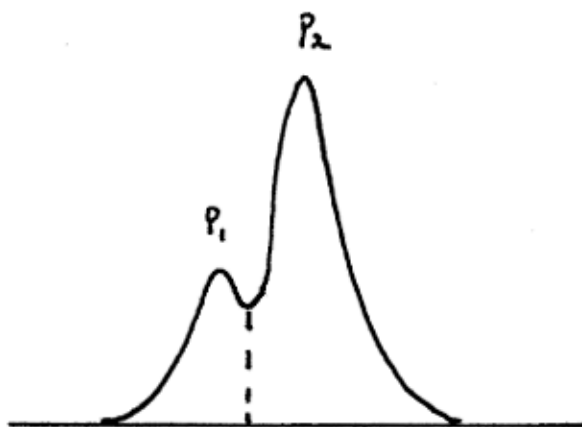


Figure E.9-11.

- Situation C: The integration scenarios detailed in example #1 and example #2 also apply to this situation. It is up to the analysts best judgment in determining the best integration techniques in order to minimize the potential quantitation errors.
- Example 4. The two peaks have similar or widely varying detector responses such that the standard would be integrated by dropping a perpendicular where less than 100% baseline resolution occurs.
- Situation D: In this situation either P<sub>1</sub> or P<sub>2</sub> as integrated in the standard is not found in the sample. Regardless of integration techniques used the target analyte will be under quantitated due to the area in the standard that was less than 100% baseline resolved. In this situation it is important to understand the significance of the error associated with the situation and overall impact of the data.

Because of these limitations, valley to valley and baseline to valley integration techniques are recommended only for instances where the baseline is shifting or matrix interferences are inherent in the analysis.

To help determine where the baseline is in a calibration standard, analyze a method blank. This should give a good indication of how the baseline is shifting throughout the run due to inherent system characteristics. When the calibration standards are analyzed, draw the baseline in the same general shape and slope as it appears on the method blank. Do not forget to take into account the additional baseline rise or shifting that the compound is causing.

#### 8.5 Peak Skimming

Peak skimming is used when one peak elutes on the tail of another, larger peak (Figure E.9-12). Generally, the analyst would draw the integration line that defines where and how long the tail of the larger peak is. This excludes area from the smaller peak thought to be contributed by the larger peak. Normally peak skimming is not a recommended practice since peak tailing is very inconsistent. The analyst should try to achieve better resolution; however, this is not always practical.

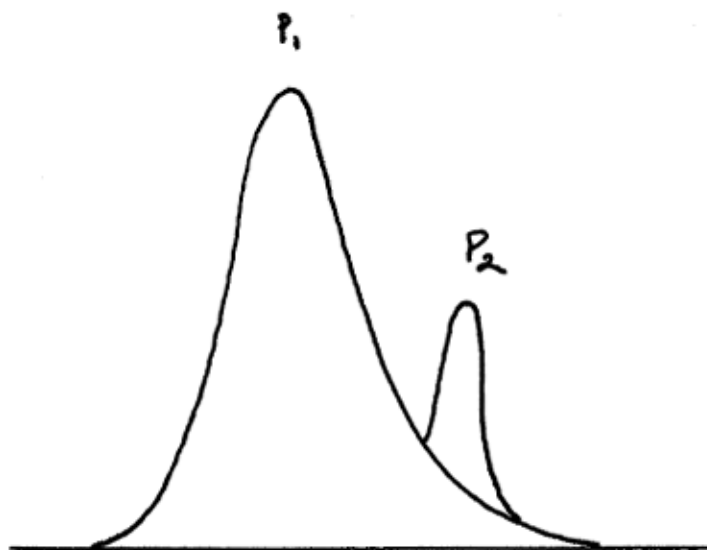


Figure E.9-12.

## 9.0 Inappropriate and Unacceptable Integration Practices and Techniques

9.1 The following scenarios constitute an ethics and fraud violation when integrating analytical data:

- a) No manual integration will be performed exclusively on QC samples in order to meet QC acceptance criteria. Manual integration is acceptable as long as consistency rules are applied to all standards and samples, and there is a reason why it is necessary.
- b) Under no circumstances is peak shaving allowed for the sole purpose for obtaining QC acceptance.

### 9.2 Peak Shaving

Peak shaving is integrating in such a manner that part of the peak response is excluded. Omitting portions of peaks or drawing the integration further into the peak is an unacceptable practice.

### 9.3 Adding Area Under the Baseline

This involves creating a peak from area under the baseline or adding to the response of a peak by taking additional area from under the baseline.

In analyses where system behavior or matrix effects make it difficult to determine where the baseline is, some discretion is to be used to avoid adding or subtracting area which is not due to the compound's response.

### 9.4 Inadequate Resolution


Adding area from another peak often occurs with partially or completely co-eluting compounds with poor resolution. Even if the analyst knows a compound is present, as in the case of standards, and the irregular peak shape is due to this compound, the analyst must not artificially create a separation between two peaks which do not have adequate resolution. As mentioned earlier, either integrate the two peaks together and report them with a footnote as co-eluting peaks or do further method development to achieve better separation.

In a sample with matrix interference, either integrate the compound and its' interferent together, or if a discernable valley exists or shoulder exists, drop a perpendicular.

95      Negative Spikes or Peaks

A negative spike or peak occurs when matrix interferences cause a sudden drop then increases in the baseline. In general the analyst should improve chromatography so that all target compounds are resolved from the negative spike. For some analyses this is very difficult. This problem should be corrected by getting the analyte to move away from this region then integrate the peak normally. If this occurs in a sample but not in the standards, integrate as best as possible.

Approved By:



Walter Hinchman  
QA Officer

Date: January, 2007



**Laboratory Ethics/Fraud Prevention  
Program  
for  
Manual Integration**

The following scenarios constitute an ethics and fraud violation when integrating analytical data:

- a) No manual integration will be performed exclusively on QC samples in order to meet QC acceptance criteria. Manual integration is acceptable as long as consistency rules are applied to all standards and samples, and there is a reason why the manual integration is necessary.
- b) Under no circumstance is peak shaving allowed for the sole purpose for obtaining QC acceptance.
- c) Under no circumstance may an analyst add area under the baseline of a peak, in an effort to create additional peak area.

These are specific examples of fraudulent and unethical manual integration practices, and are not intended to be a comprehensive list; rather this list is intended to give the analyst an idea of the types of unethical behavior that will not be tolerated.

I the undersigned, CERTIFY, that:

I have read, acknowledged and understand the personal ethical and legal responsibilities including the potential punishments and penalties for improper, unethical or illegal actions regarding manual integrations.

\_\_\_\_\_  
Employee Name (Print)

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

## **1.0 Title: Preparation of Reagent Grade Water**

Description: The preparation of reagent grade water is an essential element in the success of our laboratory. Reagent grade water is used in many critical and non-critical areas of the laboratory during the production and analysis of environmental samples. The use of reagent grade water in the critical areas of the laboratories requires this water to be of such quality as to not interfere with the analysis of target analytes at the method reporting limits.

References: Method 1080 (Reagent Water): Standard Methods for the Examination of Water and Wastewater, 20<sup>th</sup> edition, 1998.

American Society for Testing and Material, ASTM, Volume 11.01, Section DH93-91, 1992.

## **2.0 Standard Operating Procedure**

### **2.1 Introduction**

2.1.1 One of the most important aspects of chemical and biological analysis is the preparation of reagent grade water. The quality of water required is related directly to the analysis being made. Reagent grade water is used for several critical areas in the final production of analytical data. Some of these critical areas include such things as:

- Dilution of chemical reagents,
- Dilution of samples requiring direct analysis,
- Analytical instrument rinse water, and
- Preparation of Quality Control samples such as
  - Method Blanks,
  - Laboratory Control Samples,
  - Trip Blanks,
  - Field Blanks,
  - Equipment/Rinsate Blanks, etc.

2.1.2 ASTM Type I ultra-pure water is produced at Alpha using the Barnstead/Thermolyne combination reverse osmosis deionization water system. This system uses the ROPure Infinity reverse osmosis system with a 100 L reservoir as pretreatment for the NANOpure Infinity deionization unit.

- 2.1.3 The ROpure Infinity reverse osmosis system is a fully automated system providing up to 600 liters per day of ASTM Type I water when used in combination with the NANOpure Infinity deionization unit.

## 2.2 Definition

- 2.2.1 Adsorption - a process used to remove chlorine and organic impurities. It is accomplished typically with granular activated carbon. In general, organic adsorption efficiency is inversely proportional to solubility and may be inadequate for the removal of low-molecular-weight, polar compounds.

- 2.2.2 ASTM Type Water - water that has the following water quality characteristics:

	Type I	Type II	Type III	Type IV
Electrical conductivity, Max $\mu\text{S}/\text{cm}$ at 25°C	0.056	1.0	0.25	5.0
Electrical resistivity, Min M $\Omega$ at 25°C	18.0	1.0	4.0	0.2
pH at 25°C	*	*	*	5.8-8.0
Total Organic Carbon (TOC) Max, $\mu\text{g}/\text{L}$	100	50	200	no limit
Sodium, Max, $\mu\text{g}/\text{L}$	1	5	10	50
Chlorides, Max, $\mu\text{g}/\text{L}$	1	5	10	50
Total silica, Max, $\mu\text{g}/\text{L}$	3	3	500	no limit

Note: \* The measurement of pH in Type I, II and III reagent water has been eliminated from this specification because these grades of water do not contain constituents in sufficient quantity to significantly alter the pH.

- 2.2.3 Ion exchange - a process in which water is prepared by passing feed-water through a mixed-bed ion exchanger, consisting of a strong anion and strong cation resins mixed together.
- 2.2.4 Reagent water - water with no detectable concentrations of the compounds or elements to be analyzed at the detection limit (reporting limit) of the method of analysis and free of substances that may interfere with those analytical methods.
- 2.2.5 Reverse osmosis - a process in which water is forced under pressure through a semipermeable membrane removing a portion of dissolved constituents and suspended impurities.

## 2.3 System Description

### 2.3.1 Reverse Osmosis

2.3.1.1 The ROPure Infinity reverse osmosis system uses a thin micro-porous surface that rejects impurities, but allows water to pass through. The membrane rejects the following:

<u>Constituent</u>	<u>Rejection Rate</u>
• Monovalent ions	90-95%
• Polyvalent Ions	95-99%
• Inorganic solids	85-95%
• Organic solids	>99% (with a mol. weight > 300)
• Dissolved gasses	0%
• Microorganisms	>99% (bacteria and viruses)

This system is capable of producing up to 30 liters per hour of reagent grade water. This system is also microprocessor controlled which can allow automatic operation up to 24 hours a day. The RO membranes are automatically flushed for 10 minutes every 4 hours, eliminating contaminant buildup on the membranes.

#### 2.3.1.2 Reject Water

A large percentage (50-98%) of the feed water does not pass through the membrane but flows across the membrane surface, constantly cleaning the membrane and carrying the inorganic and organic solids to drain.

#### 2.3.1.3 Feed Water Pressure

The purity of the product water depends on the purity of the feed water. During the reverse osmosis process feed water pressure affects both the quantity and purity of reverse osmosis product water. Lower feed water pressure causes lower product flow rate and lower product purity. Feed water pressure should operate in the range of 30 - 100 psi.

#### 2.3.1.4 RO Pretreatment System

The ROPure Infinity uses a thin film composite membrane to produce reagent grade water. The thin film membrane may be damaged by free chlorine; therefore, a pretreatment system using activated carbon is used to remove the free chlorine.

#### 2.3.1.5 Feed Water Temperature

RO membrane performance is based on feed water temperature of 25°C (77°F). For every 1°C below 25°C product water quantity is reduced 3%. Therefore, a mixing valve (hot and cold) is used to maximize water production at a feed water temperature of 25°C/77°F.

#### 2.3.2 Storage Reservoir

2.3.1 The reverse osmosis system requires storage to eliminate back pressure on the RO membrane and also provides a large quantity of water on demand. This system uses a 100 L reservoir with an automatic shut-off float valve designed for automatic hands free water production.

#### 2.3.3 Deionization

2.3.3.1 The NANOpure Infinity system is designed to deliver up to 1.5 L/min of deionized water with resistivities of up to 18.3 MΩ/cm and less than 1ppb organics.

This system uses 4 distinct steps to produce final product water. They are a pretreatment adsorption cartridge, mixed bed deionization cartridge a finish grade organic adsorption cartridge and a final membrane filtration.

##### 2.3.3.2 Pre-deionization Adsorption Cartridge

An pre-deionization adsorption cartridge made of activated carbon and a unique macroporous resin is used to remove organics, chlorine, colloids, and some bacteria from the feed-water. This cartridge is used to extend the life of the deionization cartridge. The use of two different carbons promotes the removal of both large molecular weight and smaller volatile organics, providing for lower overall TOC values in the final product water.

##### 2.3.3.3 Mixed Bed Deionization Cartridge

Ions are removed from water as the water passes through the ion exchange resin beds. In the generated form, cation resin contains hydrogen ions on its surface which are exchanged for positively charged ions. Anion resin contains hydroxide ions on its surface which are exchanged with negatively charged ions. The final product of these two exchanges form water molecules.

#### 2.3.3.4 Membrane Filtration

Final product water is filtered using an in-line 0.2 micron membrane filter to remove bacteria or particles that may have passed through the other cartridges.

### 2.4 Cartridge and Membrane Installation

#### 2.4.1 ROpure Infinity Reverse Osmosis System

2.4.1.1 To replace the cartridges or membranes first disconnect the unit from the power supply. Open the front door and disengage the cartridge hold-down bracket by pulling it out and up.

##### 2.4.1.2 Prefilter Installation

The prefilter is installed to remove particulates from the feed-water which could cause damage to the RO membrane. Install the prefilter cartridge as follows:

- 2.4.1.2.1 Remove a new prefilter (PN D9004) and wet the o-rings on both end caps.
- 2.4.1.2.2 Insert the upper end cap into the upper farthest left position of the two cartridge end cap sockets until it bottoms out in the connector.
- 2.4.1.2.3 Lower the prefilter and insert the lower end cap into the lower socket until it is firmly seated.

##### 2.4.1.3 Pretreatment Carbon Cartridge

The pretreatment carbon cartridge is designed to remove chlorine from the feed-water which could cause damage to the RO membrane. Install the pretreatment carbon cartridge as follows:

- 2.4.1.3.1 Remove a new pretreatment carbon cartridge (PN D9005) and wet the o-rings on both end caps.
- 2.4.1.3.2 Insert the upper end cap into the upper position immediately right of the prefilter until it bottoms out in the connector.
- 2.4.1.3.3 Lower the carbon cartridge and insert the lower end cap into the lower socket until it is firmly seated.

- 2.4.1.3.4 With both the prefilter and pretreatment carbon cartridges installed, reposition the cartridge hold-down bracket.

#### 2.4.1.4 Membrane Protection System (MPS) Installation

The MPS is a clear plastic bag containing organophosphate powder. This powder is combined with water and pumped through the RO system to eliminate scale buildup on the membrane surface. After feed-water flows through the prefilter and pretreatment carbon cartridges, a pump injects 4-8 ppm of the powder-water mix into the water flowing through the unit before it reaches the RO membranes. Install the MPS cartridge as follows:

- 2.4.1.4.1 Remove the lid on the new MPS cartridge (PN CM900X1) and add approximately 1L of deionized water and agitate.
- 2.4.1.4.2 Close the lid, and locate and reattach the MPS bag using the quick disconnect fitting.
- 2.4.1.4.3 Carefully place the bag upside down in the MPS holder.

#### 2.4.1.5 Membrane Installation

This is a 2 membrane system used for the production of reverse-osmosis water. The membranes are installed as follows:

- 2.4.1.5.1 Place the first membrane cartridge in the right rear of the system cabinet. Ensure that the center connector (product water) points to the left and the offset connector (reject water) points to the rear of the unit.
- 2.4.1.5.2 Locate the tubing labeled Product and attach to the appropriate connector. Locate the Reject tubing and attach this tubing to the Reject connector.
- 2.4.1.5.3 Place the second membrane in the front of the first membrane and install as mentioned above.

#### 2.4.2 NANOpure Infinity Deionization System

- 2.4.2.1 To replace the cartridges first disconnect the unit from the power supply. Open the front door and disengage the cartridge hold-down bracket by pulling it out and up.

#### 2.4.2.2 Cartridge Installation

- 2.4.2.2.1 Install the cartridges in the following order from left to right:

Position 1)	Pretreatment	PN D50251
Position 2)	Ultrapure	PN D50253
Position 3)	Ultrapure	PN D50253
Position 4)	Organic Free	PN D50252

A kit consisting of all four cartridges can be purchased using part number PN D50254.

- 2.4.2.2.2 Wet the o-rings with water on both cartridge nipples.
- 2.4.2.2.3 Install the appropriate cartridge in the correct position by pressing the top cartridge end cap into the upper socket until it bottoms out.
- 2.4.2.2.4 Lower the cartridge and insert the lower end cap into the lower socket until it is firmly seated.
- 2.4.2.2.5 Replace the cartridge hold-down bracket.

#### 2.5 System Operation

For complete system operation see the operating manual.

Approved By: \_\_\_\_\_

*Walter Hinchman*  
Walter Hinchman  
QA Officer

Date: January, 2007



**1.0 TITLE: SUP-SAMPLING PROCEDURES AND SAMPLE COMPOSITING**

**Description:** When samples are received in greater volumes or numbers than are required for final analyses, laboratory sample mass reduction is necessary. There are two primary approaches for preparing samples for analysis: 1) sub-sampling the original discrete sample in order to prepare the sample for final analysis; and, 2) compositing multiple discrete samples by sub-sampling each individual sample and physically combining the individual sub-samples into a single final composite sample to be analyzed. This procedure describes the necessary steps to sub-sample discrete individual samples during this process in order to ensure that a representative sample is used for analysis.

**2.0 REFERENCES**

- 2.1 Method EPA/600/R-03/027: Guidance for Obtaining Representative Laboratory Analytical Subsamples from Particulate Laboratory Samples, USEPA, Office of Research and Development, November, 2003.
- 2.2 Method ASTM D 6323- 98: Standard Guide for Laboratory Subsampling of Media Related to Waste Management Activities, ASTM Standards, Vol II.04, 1998.

**3.0 STANDARD OPERATING PROCEDURE**

**3.1 Introduction**

Most analytical methods require a sample preparation procedure prior to the final analysis. These sample preparation procedures sometimes require the extraction chemist to sub-sample (i.e. take a smaller aliquot) the original grab sample. When sub-sampling is necessary, the goals of EPA's Guidance for Obtaining Representative Laboratory Analytical Subsamples from Particulate Laboratory Samples and ASTM's Standard Guide for Laboratory Subsampling of Media related to Waste Management Activities should be met.

As these documents note, representativeness is a matter of scale. Fortunately for water and virtually all soil samples received for environmental analysis, the laboratory subsampling devices greatly exceed the minimum interior diameter of 3 times the largest particles, once extraneous material has been removed from the samples.

For soil samples, multiphase samples and samples containing extraneous material, sub-sampling can be the major source of error in the measurement process, so great care must be used in applying these procedures to obtain a representative sample.

**3.2 Multiphase Samples**

- 3.2.1 Inspect the sample and determine if the sample consists of a single matrix or if multiple matrices are encapsulated in the single discrete grab sample.
- 3.2.2 It is Alpha's policy to analyze and report single discrete sample matrix and to not prepare and analyze multiphase samples as a single discrete sample. Because each matrix must be analyzed separately, a determination of which matrix or matrices to be analyzed must be made. If multiple matrices are included perform the following procedure:
- Scrutinize the chain-of-custody and determine if the client has provided further guidance.
  - If no guidance is given on the COC, the client should be called to clarify exactly which matrix or matrices the client wishes to have analyzed. This information must be documented in the client file.
  - If a particular matrix is requested, the sample is decanted into the various matrices and the requested matrix is isolated for final sample preparation.

### 3.3 Water Subsampling Procedures (excluding VOCs)

- 3.3.1 Discrete water samples are typically collected in 1-L glass or plastic containers. Samples are visually inspected to determine if there are extraneous materials that should not be included as an integral part of the sample matrix. Extraneous materials may include such things as twigs, leaves, worms, etc which obviously are not an integral part of the matrix to be extracted/digested and analyzed. The extraneous materials are to be removed from samples before subsampling and preparation for analysis.
- 3.3.2 Inspect the sample and determine if filtering is required. Colloidal suspensions typically associated with high dissolved solids often times interfere with the organic solvent extraction procedures and may need to be filtered prior to the extraction. Refer to the particular analytical and sample preparation procedures for specific filtering requirements. The only universal exception which prohibits the filtering of any sort is oil and grease, sulfides, and ammonia analysis.
- 3.3.3 Once the sample matrix has been inspected, and prepared, the entire grab sample should be vigorously shaken in an effort to completely homogenize the sample constituents.
- 3.3.4 The sample is now ready for sub-sampling. This is most easily accomplished by decanting the required volume in a volumetric glass cylinder.

3.3.4.1 Once samples have been sub-sampled into a secondary container, they must never be poured back into the original sampling container, thus preventing any possibility of contamination.

3.3.4.2 If samples have been decanted back into their original container, they should be considered compromised and not used in any further preparation or analysis.

3.3.4.3 Sample volumes are then recorded and the sample preparation procedure may commence.

#### 3.4 Soil Subsampling Procedures (excluding VOCs)

3.4.1 Discrete soil samples are typically collected in 250 ml wide mouth glass containers or occasionally in brass sleeves. Samples are visually inspected to determine if there are extraneous material that should not be included as an integral part to the sample matrix. Extraneous material may include such things as twigs, leaves, rocks, etc which obviously are not an integral part of the matrix to be extracted/digested and analyzed. The extraneous materials are to be removed from samples before subsampling and preparation for analysis.

3.4.2 Inspect the sample and determine if the sample consists of a single solid matrix or if the matrix is a sludge.

##### 3.4.3 Sludge Subsampling Procedures

3.4.3.1 Inspect the sample and determine if the sample consists of a single matrix or if multiple matrices are encapsulated in the single discrete grab sample.

3.4.3.2 If the sample is defined as a sludge (i.e. solid matrix that has the viscosity properties of a liquid), then the following procedures are followed:

- Scrutinize the chain-of-custody and determine if the client has provided further guidance.
- If no guidance is given on the COC, the client should be called to clarify exactly which matrix or matrices the client wishes to have analyzed. This information must be documented in the client file.

Note: It is Alpha's policy to analyze and report single discrete sample matrix and to not prepare and analyze

multiphase samples as a single discrete sample.

- If the supernate is requested, the liquid matrix is decanted and isolated for final sample preparation.
- If the sludge is requested, the supernate is decanted from the sludge and the sludge is isolated for final sample preparation.

#### 3.4.4 Sub-sampling Wide Mouth Sample Containers

3.4.4.1 Once the sample matrix has been inspected, and isolated, the entire grab sample should be stirred with a clean, non-reactive spatula, or glass stirring rod, in an effort to completely homogenize the sample constituents. Samples should be homogenized until the texture and color appear to be uniform.

3.4.4.2 Sub-sampling is most easily accomplished by scooping, with a clean spatula, a portion of the well mixed solid matrix into a receiving container.

3.4.4.3 Once samples have been sub-sampled into a secondary container, they must never be placed back into the original sampling container, thus preventing any possibility of contamination.

3.4.4.4 If samples have been decanted back into their original container, they should be considered compromised and not used in any further preparation or analysis

3.4.4.5 Sample weights are then recorded and the sample preparation procedure may commence.

#### 3.4.5 Sub-sampling Brass Sleeves

3.4.5.1 Sample collection using brass sleeves involves pounding the brass sleeve into the ground or collecting the sample from a split spoon sample coring device. This collection procedure prohibits the sample preparation person from easily extruding the sample from the brass sleeve without enormous effort or possible contaminating the sample. Therefore, the sample can not be completely homogenized prior to sample preparation. However, the sample is sub-sampled as follows:

- Open the sample container and discard the first 0.25 to 0.5 inch of material. This material should not be used due to possible contamination during the sampling event and contact with the sample lid.

- Proceed with weighing the required sample amount needed to perform the extraction/digestion.
- If a higher degree of homogenization is required then weigh a portion of material approximately 2-3 times the amount required for the sample preparation.
- Perform this same procedure on the opposite side of the brass sleeve and combine the two sub-samples.
- Stir with a clean, non-reactive spatula, or glass stirring rod, in an effort to completely homogenize the sample constituents. Samples should be homogenized until the texture and color appear to be uniform.
- Proceed with weighing the required sample amount needed to perform the extraction/digestion.

### 3.5 Sample Compositing Procedures

- 3.5.1 Compositing is a method of combining several samples of a specific sample matrix for a single chemical analysis. The single chemical analysis of a composite sample results in an averaging of the concentrations of its individual component samples.
- 3.5.2 Composite sampling and analysis can substantially reduce analytical costs by reducing the number of required analysis. A composite sample is typically produced by the physical mixing of individual samples and subsequently analyzing this as a single sample. By selecting the appropriate composite sample size and retesting individual samples, the composite sample may reveal the same information as would otherwise require many more analyses.
- 3.5.3 Prepare composite samples using equal volumes or weights of each single discrete sample. Sub-sampling of individual discrete samples should be completed as described below.
- 3.5.4 Mix the composite sample thoroughly by stirring with a clean, non-reactive spatula, or glass stirring rod, in an effort to completely homogenize the sample constituents. Samples should be homogenized until the texture and color appear to be uniform. For water samples, vigorously shake the final sample composite or stir to homogenize the resultant sample.
- 3.5.5 Ideally, when preparing a composite sample, each sub-sample comprising the composite sample should be of the same weight or volume. This would limit

the error associated with the compositing procedure to the same final analytical error expressed on the final analytical report as significant figures.

Secondly, analytical methods of analysis vary greatly in significant figures and concentration units. Therefore, to decrease the systemic error produced through sample compositing a higher degree of homogenization is required.

Soil samples inherently have a larger degree of potential sampling error associated with them when compared to liquid samples due in most part to particle size distribution. This is most easily overcome by sub-sampling a larger portion of material, approximately 2-5 times the amount required for the sample preparation, thereby decreasing the overall error. Liquid samples are simply proportioned mathematically to the overall volume required.

There are no universal rules applied to the error margin involved with sample compositing due to the complexity of the overall analytical procedures etc. Therefore, a general guideline to minimize the potential sub-sampling error for soils would be to weigh a sample aliquot to  $\pm 5\%$  of the final target weight. Conversely a general guideline for liquids of  $\pm 2\%$  sub-sampling error should be used when preparing composite samples.

Note: Volumetric cylinders have a volumetric error of  $\pm 2\%$  (Class A) and are acceptable to use when preparing liquids for compositing with the exception of VOCs.

- 3.5.6 The single most important compositing rule is NEVER mix the entire discrete samples into a single composite sample. Sample weights and volumes will probably not be the same, thereby introducing biased analytical results, and secondly, no additional sample is left to analyze the individuals as discrete samples.

**SVOC Soil Compositing Table**  
**Targeting a Overall Weight**  
**Two Times Extraction Requirement**  
**Table E.11-1**

Number of composite samples	Sub-sample aliquot ( $\pm 5\%$ error)	Target weight of individual sub-samples	Final overall Target weight	Weight required for analysis
5	3.8 - 4.2	4.0 g	20 g	10 g
4	4.8 - 5.3	5.0 g	20 g	10 g
3	6.3 - 7.0	6.7 g	20 g	10 g
2	9.5 - 10.5 g	10 g	20 g	10 g

**SVOC Soil Compositing Table**  
**Targeting a Overall Weight**  
**Five Times Extraction Requirement**  
**Table E.11-2**

Number of composite samples	Sub-sample aliquot ( $\pm 5\%$ error)	Target weight of individual sub-samples	Final overall Target weight	Weight required for analysis
5	9.5 - 10.1	10.0 g	50 g	10 g
4	11.9 - 13.1	12.5 g	50 g	10 g
3	15.9 - 17.5	16.7 g	50 g	10 g
2	23.8 - 26.3	25 g	50 g	10 g

**SVOC Water Compositing Table**  
**Table E.11-3**

Number of composite samples	Sub-sample aliquot ( $\pm 2\%$ error)	Target volume of individual sub-samples	Final overall Target volume	Volume required for analysis
5	98 - 102 ml	100 ml	500 ml	500 ml
4	122 - 128 ml	125 ml	500 ml	500 ml
3	163 - 170 ml	167 ml	500 ml	500 ml
2	245 - 255 ml	250 ml	500 ml	500 ml

### 3.6 VOC Preparation

The preparation of VOCs are of great concern when sub-sampling or preparing samples for compositing. The compositing of VOCs is not recommended because the compositing process itself provides a mechanism for the contaminants to escape into the atmosphere. Consequently, composite samples may underestimate the amount of VOCs actually present in the sample.

If VOC samples are to be sub-sampled or composited, then the following SOP is appropriate, with the following caveats:

- 3.6.1 Soil samples must not be mixed or shaken to homogenize the matrix at any point with the exception of shaking or vortexing the sample once it has been extracted with methanol.
- 3.6.2 VOC sub-sampling or compositing must take precedence over all other analysis, to minimize their potential loss into the atmosphere, as well as minimizing the potential for contamination by lab solvents which are also VOC analytes.

- 3.6.3 Once the sample container has been opened, speed is of paramount importance, DO NOT DELAY in any aspect of preparing the sample for final analysis. Because of the fragile nature of these compounds, samples should be subsampled mathematically to the overall volume or weight required for the analysis.
- 3.6.4 As discussed previously, there are no universal rules applied to the error margin involved with sample compositing due to the complexity of the overall analytical procedures etc. Therefore, a general guideline for soils of  $\pm 5\%$  sub-sampling error of the final weight should be used when preparing composite samples. Conversely a general guideline for VOC liquids of  $\pm 2\%$  sub-sampling error should be used when preparing composite samples.

**VOC Soil Compositing Table**  
**Table E.11-4**

Number of composite samples	Sub-sample aliquot ( $\pm 5\%$ error)	Target weight of individual sub-samples	Weight required for analysis
5	3.8 - 4.2	4.0 g	20 g
4	4.8 - 5.3	5.0 g	20 g
3	6.3 - 7.0	6.67g	20 g
2	9.5 - 10.5	10 g	20 g

**VOC Water Compositing Table**  
**Table E.11-5**

Number of composite samples	Sub-sample aliquot ( $\pm 2\%$ error)	Target volume of individual sub-samples	Volume required for analysis
5	8.4 - 8.8	8.6 ml	43 ml
4	10.5 - 11.0	10.8 ml	43 ml
3	14.0 - 14.6	14.3 ml	43 ml
2	21.1 - 21.9	21.5 ml	43 ml

Approved by: Walter Hinchman

Walter Hinchman  
QA Officer

Date: January, 2007



## **A Practical Application Guide for Performing Instrument Calibration, Calibration Model Determination and Calibration Verification**

### **1.0 Constructing a Calibration Curve**

1.1 The most commonly constructed calibration curve used in analytical chemistry is a two variable calibration curve. The two variables are as follows:

- a) Independent - the one we set (e.g. concentration - X axis), and
- b) Dependent - the one we measure (e.g. response - Y axis).

In reality both variables are somewhat considered independent. In developing a calibration curve for sample quantitation, the primary goal is to develop a quantitation model to relate the two variables.

1.2 The approach used to develop a calibration model is to:

- a) Prepare a series of known analyte standards,
- b) Hold all other factors constant (e.g. sample injection volume),
- c) Measure the response, and
- d) Develop a calibration model (calibration curve).

actually In all reality it must be stressed that the analyte response after calibration may rely on a number of factors that may affect sample quantitation such as:

- a) matrix,
- b) interfering analytes,
- c) random errors,
- d) sample preparation,
- e) sample calibration, etc.

Therefore the relationship between the analyte and its response is a function of the entire method.

1.3 Most methods have a range in which the relationship between the response and the analyte concentration may be determined (i.e. range of calibration). There are four primary limiting points in any calibration curve. They are as follows:

- a) Limit of Quantitation (LOQ)- concentration point above which an analyte is detected and may be reported with a defined manageable error.
  - SW846 requires Reporting Limits (RL) to be established at or above the lowest calibration point in a calibration curve.

- The LOQ and RL are used by our laboratory synonymously.
  - NELAP requires the LOQ/RL be established no lower than three times the MDL.
- b) Limit of Detection (LOD) - concentration point above which an analyte is detected with no concern of actual associated error.
- This may or may not be the same as the MDL.
  - This is typically a concentration point below the lowest calibration point which may or may not be included in the calibration curve.
- c) Method Detection Limit (MDL) - 40 CFR Part 136 Appendix A states the following: "The Method Detection Limit (MDL) is the minimum concentration of a substance that can be measured and reported with 99% confidence that the analyte concentration is greater than zero."

The MDL is a statistical determination which may or may not be at a concentration that can be measured. This is not factored into the MDL study but is clearly mis-stated in the overall definition.

Some regulatory agencies are beginning to recognize this fact by requiring laboratories to verify the calculated MDL. This is accomplished by analyzing a standard at two to four times the MDL. If the MDL verification standard has a signal to noise ratio of greater than three than the MDL is recognized as valid. If not the MDL study needs to be re-conducted at a higher concentration.

- d) Limit of Linearity (LOL) - the uppermost concentration point above which an analyte does not exhibit the proven linear relationship developed for lower concentration point standards.
- SW846 would define this as the upper calibration point in the established calibration curve, regardless of calibration model used for calibration.
  - Other methods (300.0 and 200.8) would define this as a concentration point above which the linear relationship does not exist at a defined limit of acceptability. For example, if a aluminum standard at a concentration of 1000 mg/L produced a quantitated value of 1256 mg/L it would not be considered linear at that particular concentration. The defined limit of acceptability for these methods are  $\pm 10\%$  and the method would require aluminum to have a quantitated value no larger than 1100 mg/L to be acceptable.

In reviewing the four limits of detection it should become obvious that the Range of Calibration (ROC) is defined by the LOQ/RL and the LOL. This is stated in our analytical SOPs as follows:

Calibration standards must cover the working range of the instrument with the low level standard at or below the reporting limit. In order to produce acceptable sample results, the response of the instrument must be within the working range established by the initial calibration.

## 2.0 Calibration Models

### 2.1 Types of Calibrations

There are four primary models for calibration curves that can be considered for instrument calibration, Linear, Quadratic, Power and Non-Linear. We are interested in only two of the four.

- a) Linear  $Y = a + bX + \epsilon$
- b) Quadratic  $Y = a + bX + c(X*X) + \epsilon$

Where:

Y denotes a instrument response  
X denotes the known concentration value of a reference standard  
a,b,c denotes coefficients to be determined, and  
 $\epsilon$  denotes a measurement error also known as residual.

Note: For our application,  $\epsilon$  will be ignored.

### 2.2 Special Calibration Cases

There are a couple of special cases in which case no calibration is required as suggested in the individual method and instrument manufacturer. They are as follows:

- Free and total residual chlorine by Method 330.5, and
- Sulfide by Method 376.2.

These two methods require no multipoint calibration because they have a proven linear response. That is:  $a = 0$ , and  $b = 1$ .

2) Clarification: It is the intent of SW-846 to prioritize the use of IC mathematical models. The first is to begin with the simplest approach, the linear model through zero, and progressing through other options until the calibration acceptance criteria are met. However, it is also acceptable to use *priori knowledge* of the detector response to choose the calibration model.

## 2.3 Linear Calibration Models

2.3.1 The linear calibration model is widely applied to many instrument calibration procedures because it has several advantages over the more complicated quadratic model. A couple of these advantages are as follows:

- The computation of coefficients and standard deviations is easy
- The correction for bias is easy
- There is often a theoretical basis for the model (i.e. a stoichiometric relationship exists between color intensity and concentration).

2.3.2 Linear calibrations forced through zero as defined in SW846 Method 8000B

2.3.2.1 The linear model  $Y = a + bX + \epsilon$  is often written as  $Y = mX + b$  where:

Y	denotes the instrument response
X	denotes the known value of a reference standard
m	denotes the slope of the curve, and
b	denotes the y intercept.

Therefore, if  $b = 0$ , the equation can be rewritten as:

$$Y = mX$$

and the average calibration used to generate a calibration curve.

If the RSD calculated from the RF of each analyte is generally ( $\leq 15\%$  for GC/MS methods and  $\leq 20\%$  for GC methods), then the response of the instrument is considered linear and the average response factor can be used to determine sample results (e.g., linear forced through zero).

Note: Linearity through zero is a statistical assumption and is not a rationale for reporting results below the calibration range demonstrated by the analysis of standards. It may be tempting to exclude the y intercept, b, from the model because a zero on the x-axis should lead to a zero response on the y-axis. However, the correct procedure is to test the

model for the significance of the y intercept term.

That is to say, test the linear calibration model through zero with low level standards quantitated against itself and/or method blanks to understand the significance of the quantitation as it approaches zero.

### 2.3.3 Linear calibration not forced through zero as defined by SW846 Method 8000B

2.3.3.1 The linear model is most commonly written as  $Y = mX + b$  where:

Y	denotes the instrument response
X	denotes the known value of a reference standard
m	denotes the slope of the curve, and
b	denotes the y intercept.

In this case b is not equal to 0.

2.3.3.2 If the RSD of the RF is generally greater than 15% for GC/MS methods or 20% for GC methods, then linearity through the origin cannot be assumed. However, a linear calibration not forced through zero can be generated by using a linear regression model. The linear regression calculation will generate a weighted correlation coefficient (r) that is a measure of the “goodness of fit.” In order to be used for quantitative purposes, r must be greater than or equal to 0.99.

### 2.3.4 Quadratic calibration models not forced through zero as defined by SW846 Method 8000B

2.3.4.1 The quadratic calibration model is written as

$$Y = a + bX + c(X*X)$$

2.3.4.2 The statistical considerations in developing a nonlinear calibration model requires more data than the linear model. A minimum of five standards for a linear (first order) calibration model is required, a quadratic (second order) model requires six standards and a third order polynomial requires seven standards. The “goodness of fit” of curve fitting equations is evaluated by calculating the weighted Coefficient of Determination COD. In order to be an acceptable nonlinear calibration, the COD must be  $\geq 0.99$ .

2) Clarification: DoD requires the correlation coefficient to be  $\geq 0.995$ . The method and NELAP only require a ( r ) value of 0.99.

3) Clarification: Alpha's calibration policy restricts the use of polynomials higher than the second order for calibration (e.g. cubic etc.).

2.3.4.3 Some of the disadvantages of the use of quadratic and higher order polynomials are:

- They require more reference standards to capture the region of curvature and define the range of calibration,
- The correction for bias is more complicated than for the linear model and may not be well understood,

Note: A plot of the data is always recommended, but is not sufficient for identifying the correct model for the calibration curve. Instrument response may not appear to be non-linear over a large calibration range. Calibration models must be justified by the RSD of the RF or the CF.

In addition, test the quadratic calibration model with low level standards quantitated against itself and/or method blanks to understand the significance of the quantitation as it approaches zero and/or the reporting limit.

Non-linear calibrations can not force the line through the origin, i.e., do not set the intercept as 0, and do not include the origin as a calibration point.

### **3.0 Initial Calibration**

3.1 Instrument calibration is intended to eliminate or reduce bias in an instrument's readings over a selected range for all concentration values within that designated calibration range. Reference standards with known values for selected concentration points are measured.

Initial calibration points for the individual target analytes are prepared and analyzed using a minimum of three points for 600 series organic methods and a minimum of five points for 8000 series organic methods to construct a calibration curve. An internal calibration procedure is used for GC/MS methods of analysis and an external calibration procedure is generally used for GC methods of analysis. Calibration standards are prepared at series of concentration points. Some compounds are calibrated at staggered limits and are compound specific (e.g. MTBE, acetone etc.).

3.2 Once the calibration standards have been analyzed, a functional relationship is established between the values of the standards and the corresponding measurements.

There are two varying situations that occur with calibrations:

### 3.2.1 External Calibration

The instrument quantitates sample data in the same units as the reference standards (e.g. external calibration). For example, GC/FID instruments measure and quantitate sample data for Total Petroleum Hydrocarbon (TPH) in mg/L units. These are the same units used for instrument calibration, mg/L or ug/ml.

### 3.2.2 Internal Calibration

The instrument converts the sample data to response ratios and amount ratios during sample quantitation (e.g. internal calibration). For example, GC/MS instruments measure and quantitate sample data for TPH units in mg/L. However, these instruments use internal standards to ratio the sample response to the response of the associated internal standard to produce a response ratio. These ratios are used to develop the Response Factors (RF) to quantitate data before converting back to concentration units as displayed on a final quantitation report.

The calibration method is the same for both situations and requires following the same basic steps:

- Selecting reference standards with known concentration values to cover the range of interest.
- Measurements of the reference standards with the instrument to be calibrated.
- Determining the functional relationship between the measured and known values of the reference standards (construction of a calibration curve).
- Correction of all sample measurements by the inverse of the calibration curve.

## 3.3 Response Factors (RF)

Instruments using an internal calibration procedure use response factors to evaluate initial calibrations. If a linear calibration forced through zero model is used, then the average RF can be easily used to quantitate the sample data. Response factors are determined as follows:

$$RF = \frac{A_s \times C_{is}}{A_{is} \times C_s}$$

Where:

$A_s$  = Peak area of the analyte  
 $A_{is}$  = Peak area of the internal standard  
 $C_s$  = Concentration of the analyte in ug/L  
 $C_{is}$  = Concentration of the internal standard in ug/L

### 3.3 Initial Calibration Verification

Immediately after the initial calibration has been established, the calibration must be verified by the analysis of the ICV.

4) Clarification: The ICV using a second source standard is not required by any of the established methods. However, NELAP has established the use of an ICV, but has not established acceptance criteria. Since the ICV is from a second source standard, additional variability is inherent in the quantitation of this standard. Therefore, the ICV should generally meet the CV criteria.

## 4.0 What can go wrong with the calibration procedure

There are several circumstances where the calibration curve will not reduce or eliminate instrument bias as intended by the analyst. A critical review of the calibration data should expose such problems.

- 4.1 Poor instrument precision or unsuspected day-to-day effects (e.g. room temperature variances over the course of the day) may result in standard deviations that are large enough to jeopardize the calibration. The best strategy is to estimate the instruments' precision and determine if it is acceptable for the precision required.
- 4.2 Outliers in the calibration data can seriously distort the calibration curve, particularly if the outliers lie near one of the endpoints of the calibration range.

4.2.1 If the RSD for one or more analytes exceed 15% or 20%, then the following steps are recommended but not required:

- 1) If the RSD appears to be associated with a single standard, that one standard may be re-analyzed and the RSD recalculated. Replacing the standard may be necessary in some cases.
- 2) Narrow the calibration range by eliminating one or more of the calibration standards producing a narrower calibration range.

Note: Changes to the upper end of the calibration range will affect the need to dilute samples above the range, while changes to the lower end will affect the overall sensitivity of the method. Hence, narrowing the calibration range by changing the concentration of the lowest standard will, by definition, change the method quantitation limit.



5) Clarification: If a particular calibration model can be achieved by either replacing an aberrant standard, or using a narrow calibration range, then the decision must be clearly documented. If a standard is replaced, then the entire standard and all associated RFs associated with the standard must also be replaced.

- 4.2.2 It is possible for different analysts to produce instrument measurements with variances that differ between analysts on the same instrument. This is not usually a problem with GC and/or GC/MS instrumentation. Small differences among analysts can be accepted as part of the imprecision of the measurement process, but large systematic differences among analysts require resolution.

These differences of imprecision can usually be traced back to systematic problems such as the preparation of standards etc.

- 4.3 Once established, the calibration procedure relies on the instrument continuing to respond in the same way over time. If the system drifts or takes unpredictable excursions, the calibrated values may not be properly corrected for bias, and may further degrade the accuracy of the measurements. To assure that future measurements are being produced correctly the calibration curve is verified on a regular frequency with calibration verification standards.

## **5.0 Calibration Model Validation**

- 5.1 A primary consideration for employing any calibration model is determining a procedure to identify and validate the model used for the calibration curve. The statistical criteria used to validate the calibration model has already been discussed. These criteria are generally established and stated in the individual methods of analysis. The SW846 Method 8000B criteria has been referenced in this SOP because it is the most stringent set of established criteria used by the EPA.
- 5.2 However what is not discussed in most analytical methods of analysis is the criteria to validate the software used with the calibration model used to quantitate sample data.

The in-house established criteria is as follows:

Data acquisition and quantitation software must display sample data in a format that would allow the analyst to verify the sample quantitation (concentration) report against the initial calibration regardless of calibration model.

- 5.3 Alpha uses two primary software packages for data acquisition:
- a) Agilent Technologies, Chemstation, and
  - b) Dionex, AI-450 software.

## 5.4 Chemstation Validation Procedure

### 5.4.1 SVOC (Internal Calibration Procedure)

The table presented below represents the quadratic equation  $[a(x*x) + bx + c]$  determined by the HP Chemstation for the compound 2,4-dinitrophenol. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

This compound was quantitated against a quadratic equation displayed at the bottom of the compound's calibration curve. Since this is an internal standard calibration procedure, Response Ratios are used to verify the Amount Ratios.

The response ratios are simply determined by:

$$\frac{\text{Response of the analyte}}{\text{Response of the associated internal standard}}$$

This value can be used to verify the concentration on the quantitation report.

The concentration value displayed on the quantitation report can be converted to a Amount Ratio simply by:

$$\frac{\text{Concentration of the analyte on the quantitation report}}{\text{Concentration of the associated internal standard}}$$

If these two values agree than the calibration model and software quantitation has been verified.

This same decision-tree-procedure is used regardless of calibration model.

2,4-Dinitrophenol	Quadratic equation $R = [.0399 * (a*a)] + (.181*a) - .0449$							
Std Conct.	0.5 ng/ul	0.75 ng/ul	1 ng/ul	1.5 ng/ul	2 ng/ul	2.5 ng/ul	3.0 ng/ul	4.0 ng/ul
Injection Vol	20 ul	20 ul	20 ul	20 ul	20 ul	20 ul	20 ul	20 ul
mass on column	10 ng	15 ng	20 ng	30 ng	40 ng	50 ng	60 ng	80 ng
Amt Ratio Cont (mass)/Cont IS	0.250	0.375	0.500	0.750	1.00	1.25	1.50	2.00
Response Std	4353	13038	22055	43054	66929	92647	130286	158290
Response IS (Acenaphthene)	452336	425018	418961	411835	406809	380772	381953	338524
Response Ratio Rsp(std)/Rsp(IS)	0.00962	0.0306	0.0526	0.1045	0.1645	0.2431	0.3411	0.4675
R= (manual calculation) (e.g., a = .25, .375 etc) (Theoretical Response Ratio calculated to best fit quadratic curve)	0.00284	0.0285	0.0556	0.1133	0.1760	0.1820	0.2275	0.4767
concentration on quantitation report	11.34 ng	15.39 ng	19.46 ng	28.53 ng	38.21 ng	49.92 ng	63.22 ng	78.89 ng
Amt Ratio calculated from quantitation report (e.g. 11.34/40 = 0.2835 )	0.2835	0.38475	0.4865	0.71325	0.95525	1.248	1.5805	1.97225
Response Ratio calculated from Amt Ration from the Qt report inserted into the quadratric equation.	0.00962	0.0306	0.0526	0.1045	0.1645	0.2431	0.3408	0.4673
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes

## 5.4.2 VOC

### 5.4.2.1 Software validation of a quadratic calibration model (Internal Calibration Procedure)

The table presented below represents the quadratic equation  $[a(x*x) + bx + c)]$  determined by the HP Chemstation for the compound Chloroethane. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

This same decision-tree-procedure is used regardless of calibration model as described below.

Chloroethane	Quadratic equation $R = [-0.0101 * (a*a)] + 0.152*a - 0.000109$							
Std Conct.	0.25 ug/L	0.50 ug/L	1.0 ug/L	2.0 ug/L	4.0 ug/L	8.0 ug/L	16 ug/L	
Amt Ratio "a" Cont (mass)/Cont IS	0.0250	0.05	0.1	0.2	0.4	0.8	1.6	
Response Std	2616	4975	8787	12919	40853	72905	138320	
Response IS	617903	622915	633385	626190	597915	622710	644054	
Response Ratio Rsp(std)/Rsp(IS)	0.004	0.008	0.014	0.021	0.068	0.117	0.215	
"R" determined from quadratic using "a"	0.004	0.007	0.015	0.030	0.059	0.115	0.217	
concentration on quantitation report ( ug/L)	0.29	0.53	0.93	1.38	4.65	8.16	15.82	
Amt Ratio calculated from quantitation report (e.g. 0.29/10 = 0.029 )	0.029	0.053	0.093	0.138	0.465	0.816	1.582	
Response Ratio "R" calculated from Amt Ration from the Qt report inserted into the quadratic equation.	0.004	0.008	0.014	0.21	0.068	0.117	0.215	
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	

#### 5.4.2.2 Software validation of a linear regression calibration model (Internal Calibration Procedure)

The table presented below represents the linear regression equation ( $aX + b$ ) determined by the HP Chemstation for the compound 1,1,2,2-Tetrachloroethane. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

1,1,2,2-tetrachloroethane	Linear Regression equation $R = [0.492*a) - 0.00356$							
Std Conct.	0.25 ug/L	0.50 ug/L	1.0 ug/L	2.0 ug/L	4.0 ug/L	8.0 ug/L	16 ug/L	32 ug/L
Amt Ratio "a" Cont (mass)/Cont IS	0.0250	0.05	0.1	0.2	0.4	0.8	1.6	3.2
Response Std	2477	5280	11718	21909	48065	87603	189127	402829
Response IS	242481	246435	253070	248662	237135	245818	248731	260591
Response Ratio Rsp(std)/Rsp(IS)	0.010	0.021	0.046	0.088	0.203	0.356	0.760	1.55
"R" determined from linear equation using "a"	0.009	0.021	0.046	0.095	0.193	0.390	0.784	1.57
concentration on quantitation report ( ug/L)	0.28	0.51	1.01	1.86	4.19	7.31	15.51	31.47
Amt Ratio calculated from quantitation report (e.g. $0.28/10 = 0.028$ )	0.028	0.051	0.101	0.186	0.419	0.731	1.551	3.147
Response Ratio "R" calculated from Amt Ration from the Qt report inserted into the quadratic equation.	0.010	0.022	0.046	0.088	0.203	0.356	0.760	1.54
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes

5.4.2.3 Software validation of a linear through zero (average RF) calibration model  
(Internal Calibration Procedure)

The table presented below represents the linear equation ( $aX$ ) determined by the HP Chemstation for the compound Toluene. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

Toluene	Linear Average RF equation $R = (3.35 \cdot a)$							
Std Conct.	0.25 ug/L	0.50 ug/L	1.0 ug/L	2.0 ug/L	4.0 ug/L	8.0 ug/L	16 ug/L	32 ug/L
Amt Ratio "a" Cont (mass)/Cont IS	0.0250	0.05	0.1	0.2	0.4	0.8	1.6	3.2
Response Std	19818	41005	85866	161992	338446	621027	1407233	2869298
Response IS	242481	246435	253070	248662	237135	245818	248731	260591
Response Ratio Rsp(std)/Rsp(IS)	0.082	0.166	0.339	0.651	1.43	2.53	5.66	11.01
"R" determined from linear equation using "a"	0.084	0.168	0.335	0.670	1.34	2.68	5.36	10.72
concentration on quantitation report ( ug/L)	0.24	0.50	1.01	1.94	4.26	7.54	16.88	32.86
Amt Ratio calculated from quantitation report (e.g. $0.24/10 = 0.024$ )	0.024	0.05	0.101	0.194	0.426	0.754	1.688	3.286
Response Ratio "R" calculated from Amt Ration from the Qt report inserted into the quadratic equation.	0.080	0.168	0.338	0.650	1.43	2.53	5.65	11.01
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes

## 5.5 Dionex Validation Procedure

### 5.5.1 Software validation of a quadratic calibration model (External Calibration Procedure)

The table presented below represents the quadratic equation  $[a(x^2) + bx + c]$  determined by the Dionex software for the compound alpha-BHC. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

This same decision-tree-procedure is used regardless of calibration model as described below.

$\alpha$ -BHC	Quadratic equation Amount (concentration) = $[5.746319 \times 10^{-15} * (a^2a)] + (1.510110 \times 10^{-7} * a) + (-0.0010)$									
Std Conct ( ug/ml)	0.0005	0.001	0.03	0.05	0.06	0.09	0.12	0.15	0.18	0.20
$\alpha$ -BHC response	9307	8683	203684	348143	393666	588711	793844	941483	1143017	1280581
Conct. From QT report in ug/L	0.0004	0.0003	0.0300	0.0522	0.0593	0.0899	0.1225	0.1462	0.1791	0.2018
Calluated Conct from equation	0.0004	0.0003	0.0300	0.0522	0.0593	0.0899	0.1225	0.1462	0.1791	0.2018
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes

### 5.5.2 Software validation of a Linear calibration model (External Calibration Procedure)

The table presented below represents the linear equation  $[(ax + b)]$  determined by th Dionex software for the compound alpha-BHC. This equation has been used to verify and validate the quantitation determined for this compound as tabulated below.

$\alpha$ -BHC	Linear equation Amount (concentration) = $(1.557 \times 10^{-7} * a) + (0)$									
Std Conct ( ug/ml)	0.0005	0.001	0.03	0.05	0.06	0.09	0.12	0.15	0.18	0.20
$\alpha$ -BHC response	9307	8683	203684	348143	393666	588711	793844	941483	1143017	1280581
Conct. From QT report in ug/L	0.0014	0.0014	0.0317	0.0542	0.0613	0.0917	0.1236	0.1466	0.1780	0.1994
Calluated Conct from equation	0.0014	0.0014	0.0317	0.0542	0.0613	0.0917	0.1236	0.1466	0.1780	0.1994
Verified	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes	Yes

## 5.6 Chemstation Metals Validation Procedure

- 5.6.1 Sample concentrations may be verified by the use of the Counts Per Second (CPS) information displayed on the calibration and final sample data quantitation reports. This data used in conjunction with the initial calibration equation report may be used to verify the calculated concentrations on the final sample quantitation report.

Example:

Scandium initial calibration equation using the weighted linear regression equation is as follows: (This changes for each new initial calibration, ie. daily).

$$Y = (1.642E+0 * X) + 4.195E-2$$

Therefore:

$$X = \frac{Y - (4.195E-2)}{1.642E+0}$$

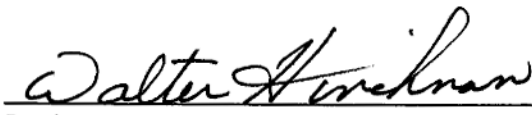
The sample concentration may be verified as follows:

Y = Sample counts for Scandium is 84.9575

Therefore:

X = 51.70 ug/L which equals the concentration on the sample quantitation report.

Approved By: \_\_\_\_\_

  
Walter Hinchman

QA Officer

Date: January, 2007



# **Appendix F**

## **Software Quality Assurance Plan**

## 1.0 SOFTWARE QUALITY ASSURANCE PLAN (SQAP)

The generation, compilation, and reporting of electronic data is a critical component of Alpha's laboratory operations. In order to generate data of known and acceptable quality, the quality assurance procedures and quality control practices for electronic data systems have been developed to be comparable in sophistication and intent to other elements of Alpha's QA program. Sections 8.1 through 8.11 of EPA Document 2185-Good Automated Laboratory Practices has been used as the foundation for the development and implementation of Alpha's SQAP.

Alpha has developed and implemented a Software Quality Assurance Plan (SQAP) that addresses requirements and responsibilities for QA activities related to the LIMS, local software, instrumentation software, and other systems related to the generation, compilation, reporting of data, or other supporting information. It is the responsibility of the LIMS Administrator to implement the SQAP and to ensure that all software and QA functions are being properly documented. The LIMS Administrator is also a part of the QA Team and reports on a regular basis to the QA Officer. The SQAP describes policies and practices for the development, modification, maintenance, archival, and use of computer software.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

## **1.0 TITLE: COMPUTER SOFTWARE OPERATIONS**

Description: This standard operating procedure outlines and defines the techniques, operations, maintenance, and security when dealing with any of the various electronic data collection, processing, and archival devices present in Alpha's facilities.

## **2.0 STANDARD OPERATING PROCEDURE**

2.1 In general the computer department is responsible for all software loading, upgrades, coding changes, debugging and hardware/software retirements.

The following general policies apply to these functions:

- i. Only authorized, trained personnel are allowed to perform these functions;
- ii. No unauthorized software is to be loaded on any company computer;
- iii. Only authorized, trained personnel are allowed to use company connections to the network or internet; and,
- iv. No hardware, software, or raw data is to be removed from the laboratory without written authorization.

### **2.2 Definitions**

Alpha - A specific MSAccess2003 database containing laboratory data.

Khemia - Software program utilizing MSAccess2003 security formats for the Omega and Alpha databases.

LIMS - Laboratory Information Management System: a general term used to describe the entire in-house network.

Network - The interconnection of various PCs and servers.

Omega - a specific MSAccess2003 database containing templates for queries, reports and forms. This term is also used as a general grouping of the interconnected Khemia, Omega, and Alpha database structure.

PC - The individual computers interfaced into the network.

Server - A computer used to handle data that is jointly utilized by various PCs.

## 2.3 Safety

### 2.3.1 General

When dealing with any of the equipment in this facility, assume that hazardous conditions are present in the form of mechanical, electrical, or chemical situations. Complete a new assessment of the potential hazards of any equipment prior to undertaking any interactions with this equipment. This assessment should be made each time the equipment has been out of sight. There should also be a continual scanning of the equipment during use to ensure that no new situations have occurred or been overlooked. If any hazardous conditions are present, appropriate precautions should be undertaken to limit the user's interactions with these conditions.

### 2.3.2 Mechanical Hazards

#### 2.3.2.1 Temperature

Most of the equipment present in this facility contains zones which are used to heat or cool. Either of these conditions can cause severe burns with only momentary contact. If it is necessary to work around these zones without bringing them to room temperature, the use of protective devices should be used to limit the possibility of contacts.

#### 2.3.2.2 Pinching and Crushing

Various portions of the equipment in this facility use automated robotics. Due to automated movements of these devices severe damage can occur. Shielding should be used whenever it is necessary to work around these devices without disabling the automated processes. Extreme caution is to be taken to ensure that the automated processes are not accidentally started.

#### 2.3.2.3 Weight

Training has been given to all employees regarding the appropriate techniques to use when moving or lifting heavy equipment. These should be followed without exception.

#### 2.3.2.4 Ergonomics

Much of the equipment is located in areas that are not easily accessible for maintenance. If it is not possible to disconnect and move the equipment to a more accessible area, extreme care should be taken to limit the possibility of injury due to falls, strains, or space constraints. Also, care should be taken to ensure weight limits are not exceeded that may be present when climbing on or around the equipment, especially when dealing with tables and ladders.

#### 2.3.2.5 Dust

Due to the area in which we are located, dust is quickly accumulated in all of the facility's equipment. This dust creates a potential for blinding or ingestion. Before working on any piece of equipment it is recommended that the dust be removed with a vacuum. If a pressurized container is to be used to remove the dust, extreme care should be taken that items are not propelled into the nose, mouth, or eyes.

### 2.3.3 Electrical Hazards

2.3.3.1 Dangerous currents are present in most of the equipment in this facility. Extreme care should be taken to ensure that the operator does not come into contact with any source of electricity. Whenever dealing with electrical areas, it is strongly recommended that all sources of current be removed. When systems have been unplugged from their current source, care should be taken that they are not accidentally reconnected. Lockout tags or plastic ties can be used to alert the need for the source to remain disconnected.

Current can still exist after the source has been disconnected! Allow ample time for capacitors to discharge prior to maintenance.

2.3.3.2 Static electricity is an ever present consideration. The discharge of static through the equipment components can cause failures. It is strongly recommended that static wrist bands be worn whenever electronic components are being maintained.

2.3.3.3 Liquids of various types are used throughout our facility. Many of these are excellent conductors of electricity. Do not work on electrical equipment that is wet without removing any potential source of current.

#### 2.3.4 Chemical Hazards

2.3.4.1 This facility utilizes many chemicals that are of an extreme hazardous nature. Read the MSDS for each chemical present in the work area prior to starting work. Especially note any potential situations that might affect the work area such as flammability and contact thresholds.

2.3.4.2 When dealing with any cleaning solutions, be aware of their potential effects on the surrounding equipment. Many of the cleaning solutions are potential contaminants in our testing procedures.

#### 2.4 Software Documentation

##### 2.4.1 Software

Software development activities are conducted in a traceable, planned, and orderly manner such that all major activities such as, coding changes and debugging can be documented.

##### 2.4.2 Software Documentation

Alpha documents changes to software in a logbook entitled Computer Software, and in several three ring binders entitled “ Software Documenting/Coding Modifications”.

###### 2.4.2.1 Computer Software Logbook

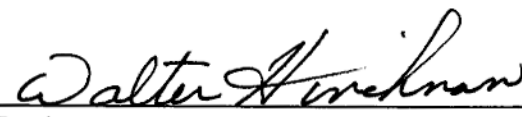
This bound logbook contains dated numerical journal entries for each software change made in the laboratory. Journal entries are things such as initial software loading, debugging, modifications, baseline testing and retirements. Typically, the PC or software will be noted in a header above the changes made to that specific PC or software. These entries are numbered to correspond to documentation (if present) in the Software Inventory/Coding Modification Logbook. These logbook entries are dated and initialed by the person responsible for these activities.

###### 2.4.2.2 Software Inventory/Coding Modification Logbook

These three ring logbooks contain records documenting all controlled software programs and their revisions used by Alpha. Each software product has its own logbook. Logbook entries common to each logbook and the Computer Software Logbook are linked by the use of the controlled numerical journal entry. These documents are a

more descriptive entry of the software activity than those found in the computer Software Logbook. Therefore, changes made to each PC documented in the Computer Software Logbook does not need this descriptive journal entry, rather it tracks the software by the use of a common controlled numerical journal entry back to the Software Inventory/Coding Modification Logbook which contains the descriptive logbook journal entries.

Approved By: \_\_\_\_\_

  
Walter Hinchman

QA Officer

Date: January, 2007

**1.0 TITLE: DATA COLLECTION AND STORAGE**

Description: This procedure is used to standardize the naming of subdirectories used to collect and store computer data for consistency throughout the laboratory.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Data Collection**

All data collection, interpretation, and corrections are made in a manner consistent with the practices and protocols outlined in the various Standard Operating Procedures governing these areas.

**2.2 HP Chemstation / Enviroquant**

**2.2.1 Data File Nomenclature**

YYMMDD## where:

YY - the YEAR that the data file was created.

MM - the MONTH that the data file was created.

DD - the DAY that the data file was created.

## - a sequential NUMBER indicating the order in which the data file was analyzed and resets with each sequence to 01.

Thus a data file that was created on 01FEB07, and was the third in that sequence would be : 07020103.

**2.2.2 Data File Storage**

Each PC has its own harddrive (C: or D:) location for the storage of the Chemstation/Enviroquant files. Within this harddrive, all of the Chemstation/Enviroquant files are stored under the subdirectory HPCHEM or MSDCHEM.

Within this subdirectory, the METHODS, SEQUENCES, and DATA, have their own subdirectories that exist in a subdirectory named after the specific instrument (e.g., C:\HPCHEM\MS05\DATA\).

Sequences should be named the same as the data files, but without the sequential number indicating order (e.g., 07201.S).

DATA should be stored in its own data subdirectory that is named the same as the sequence within which it was collected. This additional subdirectory exists within the DATA subdirectory and helps to create "packets" of related



data that are easier to locate, archive, and retrieve (e.g., C:\HPCHEM\MS05\DATA\070201\07020113.D)

### 2.3 Dionex

#### Data File Nomenclature

Data files are stored locally under the DX\Data subdirectory. Each day a new subdirectory is created to include that day's files. This subdirectory is named using the month, day, and year (MMDDYY). Each data file is named using an alpha character and a continuous numerical string. The alpha character denotes which detector (when multiple detectors of the same type are present) the data file was created from. (e.g. C:\DX\Data\021907\A#####).

### 2.4 Data File Storage

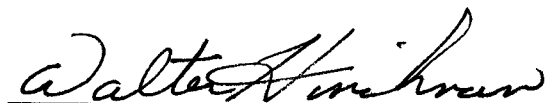
Each month's data subdirectories are compiled into a single subdirectory named for the month and year (e.g., Feb 07).

Each month, the previous month's data is archived on dual DVDs, one of which is kept offsite.

As space permits, previous data is kept on each PC, with the oldest data being removed first.

Each day, the entire, non-monthly archived, contents of each system's data collection software is copied onto the main server. Each subsequent day's archive is stored separately in folders on the server. Mid-month, the previous month's archives are deleted.

Approved By: \_\_\_\_\_



Walter Hinchman  
QA Officer

Date: January, 2007

## 1.0 TITLE: DATA FILE UPLOADING

Description: All data produced by analytical instruments must be formatted in a way the Omega software can parse-out information that is critical to the final compilation, reduction and reporting of data.

## 2.0 DATA FILE DOWNLOADING TO OMEGA FORMATS (Chemstation Data)

- 2.1 For correct interfacing between the Chemstation/ Enviroquant text file and the Omega database, the following information must be present in specific areas on the quantitation report. This interface needs to have the specified information in the exact position noted below. Each field must contain the exact number of characters. Use blank spaces for empty characters. If too few or too many spaces are present, the file will not upload. Any information after the sample ID on the sample line is ignored as far as Omega is concerned.
- 2.2 The positions are those as seen when in the sequence table or when editing the file information from the quantitation screen.

Field	Example	Quant report Line	Start Position	Length
Sample Type	SAMP	Sample	1	5
Sample ID	05021422-23AMSD	Sample	6	16
VOC or BNA Batch ID	MS06W0625A	Misc	1	10
VOC or BNA Test Name	VOC_W	Misc	11	6
VOC or BNA PQL Multiplier	1.000	Misc	17	6
TPH Batch ID	MS06W0625B	Misc	23	10
TPH Test Name	TPH/P_W	Misc	33	8
TPH PQL Multiplier	1.000	Misc	41	5
Alcohol Batch ID	08460	Misc	1	6
Alcohol Test Name	Alcohol_W	Misc	7	10
Alcohol Multiplier	1.000	Misc	17	6

If there are no TPH batches associated with the data file, these fields can be left blank or used for other information.

2.2.1 <u>Field</u>	<u>Explanations</u>
Sample Type	use <b>SAMP</b> for all client samples use <b>MBLK</b> for all Method Blanks (not needed for cleanout blanks) use <b>DUP</b> for all sample Duplicates (not fortified/spiked) use <b>LCS</b> for all Laboratory Control Samples and Laboratory Fortified Blanks use <b>LCS D</b> for all LCS Duplicates and LFB Duplicates use <b>MS</b> for all Matrix Spikes use <b>MSD</b> for all MS Duplicates
Sample ID	Client ID & Sample ID & Fraction (must be identical to that listed on our work-order sheet.) (e.g. GMT07020143-06A) (e.g. GMT07020143-06ADUP) (e.g. GMT07020143-06AMS) (e.g. GMT07020143-06AMSD)

TPH or VOC or BNA or Alcohol:

Batch ID

Instrument number and Matrix and Month and Day and Letter of the batch associated with the VOC samples (e.g. MS09W0214B) or prep batch ID associated with SEMIVOA samples. (e.g. 12345)

Test Name

- PNA\_SIM\_S
- PNA\_SIM\_W
- TPH/P\_W
- TPH/P\_S
- TPH/P\_A
- VOC\_W
- VOC\_S
- VOC\_A
- BNA\_W
- BNA\_S
- ALCOHOL\_W
- ALCOHOL\_S

Must be identical to that listed on our work-order sheet.

#### PQL Multiplier

The amount by which the report's standard reporting limit needs to be raised.  
Note that this is not the instrument multiplier.

- 2.3 When all of the reportable runs have been reviewed, run a DoList selecting MACRO and then OMEGA. This will dump the selected files onto the floppy in the A: drive, in a text format that is readable by Omega. This floppy should then be included with the hard copies of the QC data.
  - 2.3.1 Include the associated sample with the MS/MSD on the QC floppy (as the LIMS system needs to reference the MS/MSD to its parent sample).
  - 2.3.2 The sample associated with the MS/MSD must be logged-in for the associated analytical test.

### 3.0 DATA FILE DOWNLOADING TO OMEGA FORMATS (Dionex Data)

#### 3.1 GC/ECD & HPLC General QA Summary Sheets and Data Uploads

- a) In the Batch program, click FILE; Build Batch Schedule File.
- b) Select the appropriate sub-directory.

NOTE: It needs to be of the form c:\dx\data\021407 for the worksheet to identify the file names.

- c) Select the data files in the following order: (i.e, Method 8081, data upload and QA summary)

NOTE: If one of the files in the list was not analyzed or is not available to select, you must use a similar file as a place holder - for example, select the initial "CV 0.1ug/L" in place of the final CV if the final CV is not available.

-0.0025ug/L  
-0.005ug/L  
-0.01ug/L  
-0.025ug/L  
-0.05ug/L  
-0.1ug/L

- 0.2ug/L
- CV
- MBLK
- LCS
- MS
- MS (pick any two that were acceptable)
- SAMPLES (up to twenty-six in one batch)

d) Click FILE; Save as and save the new schedule as BATCH.sch. (usually this will be the default name)

e) Click OPTIONS; Processing Options and select "Reprocess using methods embedded in data files"

None of the boxes to print, save after reprocess, or update files should be checked. You are merely copying this data to excel.

f) Click OPTIONS; Export Options and choose \*.csv as the export file suffix. Name the export file like A:\030214. The suffix will be picked up automatically.

g) Go into Fields>> and mark the following:

- Sample Name
- Inj. Time
- Data File
- Retention Time
- Concentration
- Area

h) Click "OK" and return to the main Batch window.

Note: Make sure that you have a floppy in your A drive.

i) Click CONTROL; Start.

j) Take the floppy with your data and put it into the A drive of the office computer or GC/ECD#3 computer. Both have Excel and Omega.

k) Open Excel and click FILE; Open. The QA summaries are in c:\omega\excel and have the suffix xls. Highlight the appropriate spread-sheet (e.g., 8081qa.xls) and double click or click on "OK" to open it.

l) Go to row 51 on the worksheet and leave your cursor on the SAMPLE NAME cell. When you are ready to paste your data, you won't need to worry

about positioning the cursor.

- m) Click FILE; Open again. Find your data file (suffix is csv) in the A:\ directory. Open the data file.
- n) Check the file names to make sure they are named with the appropriate format - with four spaces used for TYPE, one space as a placeholder, three spaces either left blank or used for the three letter client code, and then the sample name.
- o) Check that all SAMP samples have the "A" at the end.

MBLK \_ \_ \_ \_ MB-1181  
LCS \_ \_ \_ \_ LCS-1181  
LCSD \_ \_ \_ \_ LCSD-1181  
MS \_ \_ \_ GMT03021401-01MS  
SAMP \_ GMT03021401-01A

- p) Make any corrections necessary.
- q) Check to make sure that you don't have any duplicate SAMP samples.
- r) Check to ensure the CV, LCS, and MS samples have appropriate values for the analytes, and that all samples have appropriate surrogate recoveries.
- s) Highlight the cells which contain *analyte* information for the samples

NOTE: If you need to make any further changes to the data, now is the time.

- t) Highlight the numbers of the rows containing data and click EDIT; Copy.
- u) Minimize ( '-' in the top right corner of the smaller window) your data file. You should see the worksheet, with the cursor on SAMPLE NAME at row 51. Click EDIT; Paste.
- v) You should be able to now click FILE; Print and then hit 'Enter' to print the QC Summary sheets.

The only times that you will need to change information in the worksheets is when your spike amount in an LCS or MS is not at its customary value.

- w) You need to hit 'Page Down' until you get to the QA Summary worksheet and change the Spike Amt. value under the appropriate sample to the new value for each analyte.

- x) If the type of calibration has changed (example: from 8-pt Quadratic to 8-pt Linear, Force zero) You need to hit 'Page Down' until you get to the Calibration Factor /Continuing Calibration worksheet and type in the new calibration type.
- y) If one or more standards in the initial calibration is not at its customary level, 'Page Down' until you get to the Calibration Factor /Continuing Calibration worksheet and type in the new concentration in the appropriate cell(s). You do not need to change the calculations for the calibration factor, these change automatically. However, the value must be numerical (typing '0.025ug/L' won't work).
- z) Once the QC Summary sheets are printed, you can hit 'Page up' until you reach the top of the worksheet. You will notice that only samples show up. You shouldn't have to do anything except write down the range of cells you wish to export into Omega.
- aa) Save your worksheet and close Excel ('X' in the top right hand corner)

3.2 If data is to be imported into Omega continue

- a) Double click on the Omega Icon.
- b) Open DATA ENTRY TABLE
- c) To add a new record into the database use the ADD button.
- d) Fill in the following fields:

Instrument ID: (e.g. GCECD#1)

Run Start Date: \_\_\_\_\_

Analyst: \_\_\_\_\_

- e) Next, import the Excel spreadsheet into the database using the DATA IMPORT button.
- f) File Import information:
  - Specification: Omega Excel File
  - Check the button to import the values as "FINAL RESULTS"
  - Then press "OK"
- g) Select the file to upload (e.g. C:\OMEGA\EXCEL\8081QA.XLS)
- h) You will be prompted to enter the range of files for import. (example A1:L9)

The database will check the header, upload the data, and update the prep batch files.

- i) Now go into VIEW DATA to check the imported results. Hit the CALC SEQUENCE button.
- j) Toggle off any analytes not required.

NOTE: If any sample has an actual hit, now is the time to *make sure* that it has calculated correctly and that it has the proper number of significant figures, and if it hasn't, now is the time to change it.

- k) Print out a copy of the Omega batch QC report from the Clients Reports. Check this against the original data for any errors or omissions.
- l) Close Omega and Excel and you're FINISHED!

#### **4.0 HARD COPY, NON-AUTOMATED, AND NON-EXCEL DATA**

These types of data are stored in the work order folders and are manually entered into Omega by the report writing department.

#### **5.0 Excel Electronic Diskette Deliverables (EDDs) for Uploading to Omega**

5.1 These formats need to be in a special Excel structure with the following column architecture:

- BatchID: Batch grouping from the Subcontractor
- DF: The amount by which the PQL is to be raised to give the true reporting limit (RL)
- PrepDate: The date that the sample was prepared/extracted for analysis
- FileID: The Subcontractor's unique sample/test identifier
- SampType: see Sample Type in Data File Downloading SOP F.4.
- TestCode: The test code that is listed on our in-house work order
- PQL: The Practical Quantitation Limit (regular, undiluted reporting limit)
- SampID: The Alpha sample ID that is listed on our in-house work order
- AnalDate: The date that the sample was analyzed (and time, if available).



- Units: The units in which the results are listed
- Compound #1  
Compound #2  
Compound ...

These compound names need to be exactly the same as those listed in our test codes and are to go across the columns. In the rows for each column are to be listed the results for that specific analyte in a numeric format. Null spaces or zeros are to be used for non detect (ND) values. The system does not accept text characters (e.g.: ND or < or B).

Any column names not recognized by Omega are ignored.

- 5.2 EDDs are stored on the L: drive of the server (\\Curly) within the EDD subdirectory and located in specific locations based upon the subcontractor and the date received. (e.g., L:\Elec\_DD\sem\_imports\070214).

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: ELECTRONIC DISKETTE DELIVERABLES (EDDs)**

**Description:** Many of our clients have a need to incorporate their data into a variety of databases for a multitude of end uses. In an effort to help our clients decrease the need for manually entering the hard copy results received from our facility, we provide a variety of electronic data formats.

There are generally no industry standard formats that are required by our clients; therefore, Alpha maintains the ability to create and structure a specific format for each client. These formats range from a simple text listing of the analytes and their results, through a typical excel format of the full sample and QC data package, to a six file package of the laboratory EDF (GeoTracker) result tables.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 EDD Format Specification**

Clients who request EDDs should provide a specific format and nomenclature for their data. Specific coding has been added to Omega to insure consistency in the EDD creation.

**2.2 Data Inclusions**

Some clients require QC in their EDDs, while others do not. If required, QC is entered by the reporting department prior to EDD creation.

**2.3 EDD Creation**

The following description briefly describes the essential elements for the creation of an electronic data deliverable.

2.3.1 Choose the EDD REPORTS option from the main categories in omega.

2.3.2 Under WORK-ORDER, enter the desired work-order.

- If QC is required use the buttons at the bottom of the form to double check for errors or omissions prior to creating the EDD.

2.3.3 Click the OK button.

2.3.4 Use the EDD DATE UPDATE button to update the date that the EDD is emailed.

2.3.4.1 Enter the work-order number and then the ENTER key.

2.3.4.2 Repeat until all work-orders have been entered.

2.3.4.3 Click CLOSE to exit.

2.3.5 The unfinished files are located in the L:\Elec\_DD\<ClientID>\ subdirectory.

2.3.6 Open each Excel file and reformat the column widths, margins, and page settings.

2.3.6.1 Widths = size to fit (autofit).

2.3.6.2 Margins - top (0.5), bottom (0.25), left (0.25) and right (0.25).

2.3.6.3 Page settings (depends on the client format):

- Landscape,
- Fit to 1 page wide.

2.3.7 Check the number of rows against the number of expected line items (analytes x samples).

2.3.8 Spot check results with excel entries.

2.3.9 Save each Excel file in an Excel 97/5.0 file format.

2.3.10 Email the files to the address listed in L:\Elec\_DD\EDD\_LIST.XLS. Include a copy of the Email in the data folder.

2.3.11 Place the completed work-order folders in the scanning department file bins next to the outside door in the fax room.

Approved By: Walter Hinchman Date January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: MSAccess2003 DATABASES**

Description: The Access databases are used to correlate, format and store the various data from our client's requests. These include (but are not limited to) chain-of-custodies, bottle orders, quotes, test preparation, analytes, sample prep, analysis, reports, QC, and EDDs. The Omega database consists of the templates, formats, and structures, for the various tables, forms, and reports. The Alpha database contains all of the data utilized by the Omega database.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Omega Database**

2.1.1 The Omega database should be renewed on each working day through the double-clicking of the UPDATE icon located on the desktop of each PC. By updating on a daily basis, all of the various PCs will be utilizing the most recent changes to these tables, forms, and reports.

2.1.2 No one other than those authorized should make any coding changes to the databases. The structural integrity of the Omega database is so interwoven, that seemingly minor changes can have massive effects.

Unless specifically requested, only the LIMS administrator is authorized to make any changes.

2.1.3 Request forms are available if changes are desired to the Omega database. If approved, these changes will be incorporated in the next version of the update (see F.6.2.1.1 above).

2.1.4 Any changes to this database are only saved on the local PC. These changes must then be copied over to the server. Extreme care must be taken to insure that a copy of the most recent version is present locally prior to making any changes, otherwise it is possible to overwrite previous changes with an older version.

**2.2 Alpha Database**

2.2.1 This database contains the actual laboratory generated data. This data consists of not only analytical data, but also all of the information contained within substructures, such as laboratory personnel, analyte names, tests, client information, etc.. This database exists on the server and is therefore accessible to all areas of the laboratory simultaneously.

2.2.2 Due to the intricate interlacing of the information in this database, changes to any portion affects the entire database. Personnel must only utilize those

areas of this database in which they have been trained, and then assure to the best of their ability that the information they enter is correct. Since this information is utilized by all of the laboratory, any changes made to information already present may adversely affect other areas of the laboratory. Any changes made to information already present in the system should be closely examined for those other laboratory areas affected, and the personnel in those areas notified as to the changes.

- 2.2.3 Since this is a shared database, only one current copy exists. Any data deleted from this database is not retrievable! Only those personnel trained and authorized should perform any data deletions, and then only after careful scrutiny.

A copy of the Alpha database is made by the LIMS Administrator daily. If major problems occur, the copy can be restored, however this will remove any data added or changed since the copy was created.

- 2.2.4 Due to the shared status of this database, data corruptions occasionally occur for various reasons. These corruptions can be limited through the simple process of returning to the main menu as soon as possible whenever the current task or set of tasks is completed. If the system is not being used for several minutes, then return to the main menu. This action will release resources that might be needed by another area of the laboratory as well as prevent those resources from being corrupted in case of local PC malfunction. The system should never be allowed to stand idle for extended periods with any screen other than the main page present on the PC monitor.

## 2.3 Database Corruption

Database repair should only be undertaken by the LIMS Administrator or a designated person.

### 2.3.1 Repair

- 2.3.1.1 Copy the Alpha.mdb database from S:\Alpha.mdb to the local drive. (This is done to increase the speed of the repair process).

Caution: Be certain that all of the PCs have closed Omega prior to the copy. If any of the PCs are active in the database then the repair process will truncate the tables that were active, causing ALL of the data in those tables to be lost!

- 2.3.1.2 Open MSAccess without any open databases.

2.3.1.3 Under TOOLS click UTILITIES, then REPAIR and choose the Alpha.mdb database that was copied to the local drive.

2.3.1.4 After the repair is completed, COMPACT the database.

2.3.1.5 If any ERRORS have occurred, then repeat the process using the daily archived database.

- All data and changes entered into the database that day will have been lost and will need to be redone.
- Sometimes, compacting a corrupted database will allow a repair to proceed without errors.
- Sometimes, decompiling will fix a corrupted database.

2.3.1.6 After compacting the repaired database, copy the file back to S:\Alpha.mdb.

## 2.3.2 Compact

2.3.2.1 Open the MSAccess without any open databases.

2.3.2.2 Copy the database from S:\ to the local drive.

- This will increase the speed of the compaction.

Caution: Be certain that the pathways are correct. Multiple copies of the same database from different time periods are present throughout the network.

2.3.2.3 Under FILE click UTILITIES, then COMPACT and choose the correct database on the local drive.

2.3.2.4 Compact the file into itself (i.e. - save the file under the same name and location).

2.3.2.5 Copy the file back into the S:\ directory, overwriting the original.

Approved By: Walter Hinchman Date January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: DATA ARCHIVING**

Description: Laboratory data is stored and archived for possible future use in a traceable, planned and orderly manner to facilitate access, ease of use, and security.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 Analytical Instruments**

On a daily basis, in-house software automatically copies all instrument data to a backup server. These daily backups are kept until after the monthly archiving to DVDs are completed. In case of data loss, (at the PC due to various reasons,) the original data can be restored.

On a monthly basis, the system administrator will archive any data files greater than one month old. These files are transferred to dual DVDs for both local and offsite storage. The archived data also remains on the individual PCs as hard drive room permits. When hard drive room is necessary, the oldest archived files are removed first.

2.1.1 In preparation for archiving, the system administrator compiles the various monthly data, method, and sequence files to create a monthly archival subdirectory on each analytical PC.

2.1.1.1 Within the Archive subdirectory, create a new folder called by the month and year of the data (e.g. FEB07).

2.1.1.2 Move the data and sequences into this new folder.

2.1.1.3 Drag (copy) the methods into this new folder.

2.1.2 This archive subdirectory is then copied across the network to the server via the in-house archiving software.

2.1.3 The monthly archive subdirectories are then copied onto DVDs.

2.1.3.1 Copy the archived data from the backup server to the archive PC.

2.1.3.2 Insert a new DVD into the R-RW device (LIMS Administrator's archive PC).

2.1.3.3 Click on the Creator Classic icon.

2.1.3.4 Enlarge the subdirectories in the upper screen such that the desired monthly folder is present on the right side.

2.1.3.5 Drag (copy) the monthly folder down to the right side of the lower screen. If multiple instruments, drag all of the desired instrument folders, not the monthly. Each DVD holds 6.4 Gb. The amount of the selected files is listed at the bottom of the screen. Do not exceed the 6.4 Gb limit.

2.1.3.6 Double click on the UNTITLED on the left bottom screen, and name the DVD with the month and year.

2.1.3.7 Click on the DISC tab at the bottom left, and then on the red RECORD button.

2.1.3.8 When the DVD is finished, the DVD will eject.

2.1.3.9 Insert a new DVD and restart at 1.3.7 for the duplicate DVD.

2.1.4 Each DVD is labeled with the month and year

2.1.4.1 Use only felt pens (Sharpie) to write on the DVDs.

2.1.4.2 Label each DVD and jewel box with "RAW DATA," and the month and year.

2.1.5 Store the in-house copies in the DVD bins, and send the duplicates to offsite storage.

2.1.6 Data retrieval is accomplished by copying the desired files from the DVD back onto the archive PC and changing the properties (removing the read only check mark after double right clicking on the file or files). The files are then moved to the backup server where they can be uploaded onto the instrument PC.

2.1.7 Due to various circumstances, the data on the new DVDs may be corrupt. To ensure that a good copy is present, the data from the DVDs is copied to another PC and random files are checked for data integrity.

## 2.2 Omega

The Omega databases are backed up daily by the in-house archive software in the same manner as the instrument data.

The Omega databases are backed up monthly by the LIMS Administrator for offsite archiving.



2.3 Administration Files

The financial files (BUSINESSWORKS) are backed up nightly, weekly, and monthly with nightly and monthly archived files stored offsite.

2.4 Server

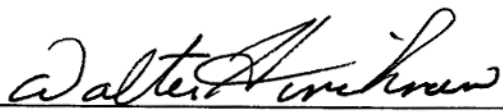
The entire server contents are backed up nightly, weekly, and monthly with monthly archives stored offsite.

2.5 Accessing Archived Data

2.5.1 Occasionally, there is a need to review data that is no longer on the local instrument PC that has been archived.

2.5.2 The LIMS Administrator maintains the original instrument data archived onto DVDs. In the event access to a DVD containing analytical data is required, that access is controlled by the LIMS Administrator.

2.5.3 Upon request, a copy of the required data is made on a CD from the original DVD. The original DVD never leaves the LIMS Administrator's controlled area.

Approved By:  Date January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: PC/SERVER INTEGRITY AND SOFTWARE VALIDATION**

Description: PC/Server hardware and software must be used in an environment that promotes data integrity. All software used for compilation, reduction and reporting of data is validated and verified prior to use.

**2.0 STANDARD OPERATING PROCEDURE**

**2.1 PC/Server Integrity**

**2.1.1 Software Dos and Don'ts**

- All personnel should only use those programs in which they have been trained or have prior experience.
- No unauthorized or unlicensed software is to be loaded on any company computer.
- Only authorized, trained personnel are allowed to use company connections to the network or internet.
- No hardware, software, or raw data is to be removed from the main offices without written authorization.
- No unauthorized removal or modification of any software programs, functions, or coding.
- Document all error messages or undocumented features (bugs).
- No unauthorized modifications to PC settings.

**2.1.2 Hardware Dos and Don'ts**

- Only trained, authorized personnel should attempt any hardware installations, repairs, or modifications.
- No unauthorized hardware is to be attached to any PC or to the LIMS network.

**2.2 Software Validation**

- 2.2.1** Software validation and verification activities are performed to ensure that the software adequately and correctly performs all intended functions, and that the software does not perform any unintended functions. Software

validation and verification are performed jointly by the computer QA unit and the end PC user.

2.2.2 To evaluate the technical adequacy of software, testing activities are performed to assure that the software produces correct results for the test case. Test case results are then compared to results from alternate methods, such as manual computation of results or use of another program to process the same data file.

2.2.3 Once software has been validated, then continuing maintenance of the baseline software is produced during periods of use. As deemed appropriate, this may require removal of latent errors, or adoptions to changes in the operating environment. On a schedule determined by the LIMS administrator, baseline systems are checked for corruption.

### 2.3 Security

The following security controls are established to permit authorized access and to prevent unauthorized access to software systems and data files:

2.3.1 All personnel should only use their own individual accounts when logging onto the PCs, server, or Omega.

2.3.2 All personnel should log off of the system whenever they are away from their station for extended periods.

2.3.2.1 In place of logging out, the use of password secured screen savers is adequate for normal working hours.

2.3.2.2 Instruments may have their own log-in identifications with limited server and LIMS access.

2.3.3 No unauthorized personnel are to use an account not their own to access secured portions of the system.

Approved By:  Date: January, 2007  
Walter Hinchman  
QA Officer

**1.0 TITLE: SAMPLE LOG IN (Computer Entry)**

Description: The following SOP is established to maintain uniformity, clarity and consistency when inputting information into the lab database upon sample receipt. This computerized Sample Log In SOP is implemented with all the provision established in the main Sample Log In SOP.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 To enter Omega from the main computer menu, double click on the Omega icon. When the log-in prompt appears, type access name and click the "O.K." button. This will take you to the main Omega II menu. From this menu, double click on "Work Orders."

2.2 To add a work order, double click the "Add" key in the lower left-hand corner of the screen. Amend the work order number to correspond with the correct sequence:

SCO: 1-19  
Assistant SCO: 20-39  
Assistant SCO: 40-59  
Assistant SCO: 60-79

2.3 In the "Client ID" box, enter the client's three letter code, then double click the "Client Information" field. The company name, contact person, address, phone and FAX numbers will appear automatically.

Amend the "TAT" (turn-around-time) box as needed for TAT's shorter than ten days. Add an "Order name" if applicable. Click the box and type the name. Add "Report Attention", COC #, sampled by, cooler temp. and comments.

2.4 Double click the "Received" box. A calendar will appear with today's date selected. Double click the proper date. The computer will automatically enter the correct date. Double click the "Date Due" box. The computer will automatically count ahead the numbers of days selected in the TAT box (excluding weekends).

Add any additional comments in the "Comments" box by clicking once in the box, then start typing. Input information such as:

- Real ice present or not,
- Samples frozen or not,
- Security seals present or not,
- Rush Turn Around Time (TAT) sample,
- California samples, etc.

in the comments section. If the client would like to give a purchase order number,

click the "Invoice Info" button at the top of the screen and enter the Purchase Order number.

## 2.5 Sample Log-In Procedure

2.5.1 Click once in the "Client Sample ID" box. Type in the ID requested by the client.

2.5.2 TAT will automatically appear.

2.5.3 Enter the sample "Collection Date." This should be written on the sample label - otherwise, ask the client. Double space and enter the time sampled in military time.

2.5.4 Enter the sample "Matrix." This can be typed by hand, or by clicking the down arrow adjacent to the matrix box and then selecting the proper matrix (soil, aqueous, etc.). Enter the bottle (container) type the list provided.

2.5.5 Enter the number of containers which belong to the sample currently being entered.

2.5.6 For drinking water, some clients will give Public Water System numbers. This number can be typed in the "PWS/DWS" box in the field data page. A Sample Site ID and System ID can also be added at the client's request in the proper box.

2.5.7 Add any extra comments.

2.6 Click once in the "Test Group" box. Enter the proper test, then click the down arrow next to the box to view the list of possible choices. Tests are segregated by state and test method.

2.6.1 For each *new* sample, click the "Add Sample" and repeat the sample log-in process, starting with a new client ID. Many samples will have Trip Blanks included. Treat the Trip Blank as a separate sample. For Client Sample ID type "Trip Blank." The TAT will be the same as for the corresponding samples. The collection date is listed as the earliest sample date present in the work order. Trip Blanks are not analyzed unless requested by the client. Therefore, under Test Groups type "Hold". This will tell the analysts that the sample should not be run.

2.6.2 When all of the relevant tests are entered, click the "out-the-door" button at the lower right-hand corner of the screen ( picture of an open door with an arrow pointing into it).

- 2.7 To print the Chain of Custody, click the WO COC button at the top of the screen.
- 2.8 To print the analyte lists, click on the PRINT TEST Button at the top of the screen. Enter the sample ID (e.g. 02) of the sample with the test parameters that are to be printed. Use a "\*" to print all of the test codes, unless a specific test is desired for printing.
- 2.8.1 These analyte lists are to be included in each of the appropriate test file folders (e.g. the VOC\_W list goes into the VOC file folder).

Approved By: Walter Hinchman  
Walter Hinchman  
QA Officer

Date: January, 2007

**1.0 TITLE: SAMPLE PREPARATION OMEGA SOP**

Description: The following SOP is established to maintain uniformity, clarity and consistency when inputting sample preparation data. This computerized sample preparation SOP is implemented with all the provisions established in the main Sample Preparation SOP.

**2.0 STANDARD OPERATING PROCEDURE**

2.1 From the Windows desktop, Click on the OMEGA Icon

2.2.1 Every morning, before opening Omega, the UPGRADE icon should be clicked to ensure that the newest version of the Omega software is being used.

2.1.2 The Alpha screen will appear, press enter.

2.1.3 Click on the SAMPLE PREP button.

2.2 To add samples to a previously existing batch:

2.2.1 Scroll down the left list and double click on the desired prep batch.

2.2.2 Click on the SAMPLE ID column and arrow down to the bottom of the list

2.2.3 Enter the sample ID in the first blank line (Only the numbers followed by the letter A).

Note: This can be accomplished by either typing the sample ID or using the user select option.

2.2.3.1 Make sure that the sample ID matches that on the work-order (e.g. : 07021401-01A).

2.2.3.2 If a Client ID and Client Sample ID do not appear after moving out of that cell, then double check the sample ID. If it still does not appear, hand enter the correct information.

2.2.3.3 For MS, MSD, and DUP samples use the parent sample ID followed by the type of QC (e.g. : 07021401-01AMS or 07021401-01AMSD or 07021401-01ADUP).

2.2.3.4 There should be no more than twenty (20) samples in a batch (not including QC).

- 2.2.4 Fill in the sample volumes and/or weights used for the extraction.
- 2.3 To create a new batch
  - 2.3.1 Click on the ADD button,
  - 2.3.2 Choose the correct PREP CODE,
  - 2.3.3 Enter the TECHNICIAN'S name,
  - 2.3.4 Start adding samples (see F.10.2.2 above)
    - 2.3.4.1 The system automatically adds a MBLK and LCS/LCSD set. If the LCSD is not required, Click on the far left of the LCSD row to highlight the entire line. Use the DELETE key to remove the entry.
    - 2.3.4.2 A MS/MSD set will need to be added to each batch.
- 2.4 Click on the REAGENTS/SPIKES tab to enter the surrogate spike data
  - 2.4.1 Enter the IDs from the spike mix containers. If the ID is not present on the list, also enter the spike name in the next cell.
  - 2.4.2 Enter the correct sample type with which the spike is associated.
    - 2.4.2.1 If the same volume and spike mix is used, then:
    - 2.4.2.2 Use SAMP with those spike mixes which are added to everything (e.g. Internal Standards and Surrogates).
    - 2.4.2.3 Use LCS for the mix associated with all of the LCS, LCSD, MS, and MSD QC.
    - 2.4.2.4 If different volumes or mixes are used, add a separate line for each combination.
  - 2.4.3 Enter the correct volume of spike added to each individual sample.
- 2.5 Click on the VIEW button, to return to the sample list.
- 2.6 Click in the first SPK ADDED cell and begin to add the spike information.
  - 2.6.1 Enter the line number(s) from the associated reagent/spike table.
    - 2.6.1.1 For example, if the LCS was spiked with three different spikes (IS,



Surrogate, and Spike) that corresponded to the first three lines of the spike table, then the line numbers would be 1,2,3.

- 2.6.1.2 Either spaces or commas may be used as delineators between the different mixes.
- 2.6.2 If the same reagent is used throughout the entire batch, no notation is necessary for each sample.
- 2.6.3 If different reagents are used, notate the reagent line using alpha characters (A,B,C,etc.)
- 2.7 Any deviations from the normal methodology are to be entered in the comments line of the associated sample.
  - 2.7.1 These deviations include such things as different initial or final volumes, spike problems, filtration problems, emulsions, spilled sample, etc.
  - 2.7.2 If the samples are aqueous, measure the sample pH and add this value to the pH column.
- 2.8 Check all fields before printing.
  - 2.8.1 Before printing, double click on the 2<sup>nd</sup> prep date line to complete the status of the preparation batch.
  - 2.8.2 After all of the associated information is entered, click on the PRINT button.
  - 2.8.3 Review all of the information on the print out. If everything is accurate initial and date the batch report.
  - 2.8.4 Include a copy of the hard copy batch report in each of the associated work-order folders.
  - 2.8.5 Place the original preparation batch report in the appropriate extraction batch binder.

Approved By: Walter Hinchman Date: January, 2007  
Walter Hinchman  
QA Officer

# **Appendix G**

## **Laboratory Ethics/Fraud Prevention and Data Integrity Program**

# **Laboratory Ethics/Fraud Prevention and Data Integrity Program**

I the undersigned, CERTIFY, that:

I have read, acknowledged and understand the personal ethical and legal responsibilities including the potential punishments and penalties for improper, unethical or illegal actions.

This includes, data integrity and/or data authentication issues such that the analytical process can be completely reviewed by recreating the paper trail.

\_\_\_\_\_  
Employee Name

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

\_\_\_\_\_  
Senior Management Name

\_\_\_\_\_  
Signature

\_\_\_\_\_  
Date

## **1.0 Laboratory Ethics, Fraud Prevention and Data Integrity Program**

- 1.1 The QA Program includes the components necessary to achieve acceptable data and assumes that personal behavior is ethical. Analytical methods, procedures and EPA programs do not set standards for ethical behavior in the laboratory and assumes that work being conducted in an analytical chemistry laboratory is of the highest integrity. This program is established to communicate to all employees what constitutes expected conduct, ethical behavior, unethical behavior and fraudulent behavior.
- 1.2 It is Alpha's policy to establish and maintain a Laboratory Ethics, Fraud Prevention and Data Integrity program. There are four required elements within our data integrity system. These are: 1) data integrity training, 2) signed data integrity documentation for all laboratory employees, 3) in-depth, periodic monitoring of data integrity, and 4) a data integrity SOP.

## **2.0 Definitions**

- 2.1 Data integrity - the act of conducting data analysis under a program that documents data analysis without any type of data manipulation, falsification or misrepresentation of the actual data results.
- 2.2 Ethical Behavior - behavior that conforms to accepted professional standards of conduct. Unethical behavior therefore is behavior not conforming to these standards of conduct.
- 2.3 Ethics - a set of moral principles or a code of right and wrong.
- 2.4 Fraud - an intentional act of deceit that may result in termination or legal prosecution.
- 2.5 Integrity - moral soundness, or honesty.
- 2.6 Laboratory Fraud - the deliberate falsification, with intent to conceal analytical or quality assurance results.

## **3.0 Objectives**

- 3.1 The objective of our Ethics, Fraud Prevention and Data Integrity Program is to prevent fraudulent or unethical employee behavior and to ensure ethics violations do not occur. The impact of unethical behavior and fraud could be devastating to our laboratory, our employees, as well as to the data users.
- 3.2 An effective and rigorous ethics program has been established in conjunction with our laboratory QA program to better ensure that employees act ethically and within

the bounds of these programs.

- 3.3 The objectives of the Ethics and Fraud Prevention program have been established, and are carried out by the annual discussion and documented training of the following critical elements:

3.3.1 Ethics education and discussion,

3.3.2 Awareness and prevention of unethical or fraudulent acts, and

3.3.3 Establishment of policies to prevent and handle problem situations.

- 3.4 Data integrity training is focused on the primary objective of creating an environment that fosters professional and scientific honesty.

#### **4.0 Laboratory Ethics/Fraud Prevention and Data Integrity Program**

##### **4.1 Examples of Inappropriate Practices**

The QAM is written as a set of policies and procedures that defines what laboratory personnel are required to do; however, the following policies are written to ensure that employees are educated as to what they are not allowed to do. The following table lists specific examples of unacceptable laboratory practices and our policy regarding each issue.

Unacceptable Laboratory Practice	Laboratory Policy
1. Dry Labbing  Making up data such as creating data for an analysis that was not performed or creating information that is not true.	Analytical results for all sample and Quality Control must be based on actual analysis that were performed. Documented data must match actual data (Data Integrity). Sampling information must be based on actual sampling events.
2. Time Traveling  Resetting the internal clock on an instrument to make it appear that a sample was analyzed within a specified holding time when in fact it was not. Alternatively, changing the actual time or recording a false time to make it appear that holding times were met, or changing the time for sample collection, extractions or other steps to make it appear that they were performed at the correct time when in fact they were not.	The recorded date and the time of collection, preparation or analysis must match the actual date and time that the action was performed. Samples exceeding holding times are reported in a footnote or case narrative.

Unacceptable Laboratory Practice	Laboratory Policy
<p>3. Improper Peak Integrations</p> <p>Artificially subtracting (peak shaving) or adding (peak enhancing) peak area to produce an erroneous area that forces data to meet specific QC criteria when in fact the criteria were not met.</p>	<p>Instrument peaks must be consistently integrated and reporting according to proper techniques, generally baseline to baseline, valley to valley or a combination of the two. Peak area cannot be subtracted or added to force the data to meet specified criteria. Preventative or corrective action must be taken on instrument data not meeting required criteria.</p>
<p>4. File Substitutions</p> <p>Substituting previously generated runs from a non-compliant calibration or QC run to make it appear that an acceptable run was performed when in fact it was not.</p>	<p>All data must be generated and reported for actual analysis performed. Reported dates and times for all analysis must match actual dates and times. Substitution of files is not permitted.</p>
<p>5. Alteration of Analytical Conditions</p> <p>Improperly altering analytical conditions, such as changing instrument conditions for sample analysis from that used for standard analyses. Also using different procedures to process standards data or standard concentrations other than those used for samples.</p>	<p>All sample analyses must be performed under the same conditions as those used for standard analyses. All standards data must be processed by the same procedures as those used for processing sample data.</p>
<p>6. Improper Calibration</p> <p>a) Performing more than two calibrations or altering a calibration by eliminating one or more points, until one analysis barely meets criteria, rather than taking needed corrective action after the second failed analysis, and not documenting or retaining data for the other unacceptable data.</p> <p>b) Using the incorrect initial calibration to make calibration verification data appear to be acceptable when in fact it was not acceptable when compared to the correct initial calibration.</p> <p>c) Randomly discarding points in the initial calibration to force the calibration to meet the acceptance criteria.</p>	<p>a) All calibration and QC data associated with sample analysis must be documented. Preventative or corrective action must be taken and documented if calibration and/or other QC criteria were not met.</p> <p>b) Acceptance of calibration verification data must be based in the correct initial calibration</p> <p>c) Calibration points can be rejected for inclusions in the calibration curve if a known or suspected error was made. When multiple target analytes are included in each calibration standard it may become necessary to discard selected upper or lower points for individual target analytes. Points can be discarded at the upper end of the curve if the linear range of the detector has been exceeded. For this case samples must be diluted that exceed the highest point of the calibration curve. Points can be discarded at the lower end of the curve if the detector is not producing a response. For these cases reporting limit must be adjusted accordingly.</p>

Unacceptable Laboratory Practice	Laboratory Policy
<p>7. Misrepresenting QC Samples and Spikes</p> <p>a) Adding SV surrogates after sample extraction rather than prior to sample extraction.</p> <p>b) Reporting post-extracted spikes or duplicates as re-extracted spikes and duplicates.</p> <p>c) Not preparing or analyzing method blanks and Laboratory Control Samples (LCS's) the same way that samples are prepared and analyzed in order to make it appear that method blank or LCS results are acceptable when in fact they may not be.</p>	<p>QC samples and spikes must be prepared, analyzed and reported according to appropriate procedures.</p> <p>a) Surrogates must be added prior to sample extraction.</p> <p>b) Post extracted spikes and duplicates must be reported as post extracted and must not be misrepresented as pre-extracted spikes and duplicates.</p> <p>c) Method blanks and LCS's must be prepared and analyzed the same way that samples are prepared and analyzed. QC results outside of acceptance criteria are reported with a footnote or a case narrative.</p>
<p>8. Deletions of Non-Compliant Data</p> <p>Intentional deletion or non-recording of non-compliant data to conceal the fact that analyses such as calibration or QC were non-compliant.</p>	<p>All data associated with sample analysis, including any out of control events or non-compliant data must be documented and retained. Preventative or corrective action must be taken and documented for non-compliant data.</p>
<p>9. Manipulation of Computer Software</p> <p>Unwanted manipulation of computer software to force calibration or QC data to meet criteria.</p>	<p>Computer manipulation is allowed only for warranted reasons and any manipulation should be minimal and traceable.</p>
<p>10. Concealment of a Known Problem</p> <p>Concealing a known analytical or sample problem from laboratory management. Concealing a known behavior or action from laboratory management.</p>	<p>Any knowledge of analytical or sample problems must be communicated to laboratory management and the client. Any knowledge of unethical behavior or actions must be fully communicated to laboratory management.</p>

- 4.2 Most individuals do not personally gain from committing an unethical act except to relieve some pressure they feel, whether it is real or perceived. Therefore, education and communication are key elements for our program.

## 5.0 Relevant Criminal Laws

### 5.1 Unethical Behavior

An unethical action becomes a fraudulent act when the law is violated. For example, it is unethical if an analyst changes the instrument clock to make samples appear to be analyzed within holding time, when in fact they were not. It is also unethical to manipulate instrument calibration or QC samples to make the calibration or QC analysis meet an acceptance limit, when in fact the actual data was not acceptable.

## 5.2 Fraudulent Acts (wire fraud / mail fraud)

It becomes a fraudulent act when the falsified data is faxed, mailed or e-mailed. Faxing or mailing false information is an example of wire fraud or mail fraud, respectively, and the person or organization that does so could be charged with wire fraud or mail fraud, as well as making false statements if the work was done under a government contract.

The following is a list of relevant criminal laws all of which can result in substantial fines and possible imprisonment.

- False Claims - 18 U.S.C § 287
- False Statements - 18 U.S.C § 1001
- Mail Fraud - 18 U.S.C § 1341
- Wire Fraud - 18 U.S.C § 1343
- Conspiracy - 18 U.S.C § 371
- Misprison ( concealment) of Felony - 18 U.S.C § 4

## 6.0 Possible Penalties for Environmental Crimes

### 6.1 Laboratories can face these types of legal action for breaking the law:

- Administrative action - punishment which can result in debarment or probation.
- Civil Action - punishment which can result in large fines.
- Criminal Action - punishment which can result in prison sentences for company officials.

### 6.2 Individual who commit an unethical act and/or break the law can face the following types of action:

- Disciplinary actions up to and including termination from the job.
- Civil actions - punishment which can result in large fines.
- Criminal action - punishment which can result in prison sentences and/or probation sentences.

## 7.0 Zero Tolerance Policy

### 7.1 Alpha Analytical has a zero tolerance policy on unethical activities or fraudulent behavior. Unethical behavior includes, but is not limited to, the **intentional** falsification of the following:



- Data or records,
- Professional credentials,
- Employment records,
- Sampling or sample handling records,
- Laboratory worksheets,
- Analytical logbooks,
- Instrument settings,
- Sample results, and
- Laboratory analytical reports.

## **8.0 Data Integrity**

8.1 One of our principal managerial guiding philosophies is to create and foster a work place culture that promotes professional and scientific honesty. Management acknowledge its support of these issues by:

- a) upholding the spirit and intent of all data integrity procedures, and
- b) effectively implementing the specific requirements of the procedures.

8.2 Laboratory analysis is a human endeavor which requires a vast amount of professional decision making. Often times these decisions are not black and white issues. Therefore, the documentation trail created during sample analysis is a critical data integrity element. Data integrity requires this paper trail to be complete in order for the data to be properly reviewed, evaluated and authenticated.

Analysts are encouraged to document non-perfect data results or instrument/sample abnormalities by a written narrative. These narratives, can be described as footnotes or other annotations that are written on data quantitation reports, corrective action reports, calibration summaries etc.

It is a laboratory policy to document clearly how all analytical results were obtained and to supply to the data user all relevant information.

8.3 Data integrity discussion items are the key sections found in the QA Manual, Volume I. Many of these items are expounded upon with additional detailed standard operating procedures typically found in QA Manul, Volume II. Data

integrity training covers many topics and is continuously trained upon. Key training topics include the following:

- Organizational structure of the laboratory, and the relationship with management to create honesty and full disclosure in all analytical reporting,
- How and when to report data integrity issues,
- Record keeping.

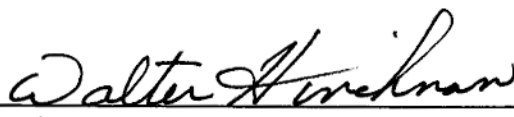
## **9.0 REPORTING**

- 9.1 It is Alpha's policy to assure personal and scientific confidentiality and to provide a receptive environment in which all employees may privately discuss ethical issues or report items of ethical concern, if unethical activities or fraudulent behavior is suspected by an co-worker,

## **10.0 INVESTIGATIONS**

- 10.1 If unethical activities or fraudulent behavior is discovered as a result of an internal audit or other discovery mechanism, a complete and comprehensive review is conducted by management.
- 10.2 The discovery of potential issues are handled in a confidential manner until such time as a follow up evaluation, or other appropriate actions have been completed and the issues clarified.
- 10.3 All investigations that result in finding of inappropriate activity is documented and will include any disciplinary actions involved, corrective actions taken, and appropriate notifications of clients. All documentation of these investigations and actions taken are maintained for a minimum of five years.

Approved By: \_\_\_\_\_

  
Walter Hinchman

QA Officer

Date: January, 2007

# **Appendix H**

## **State Certifications and Parameters of Analysis**

## State Certifications and Parameters of Analysis

**H.1** The individual parameters and/or methods certified by each of the various state agencies are kept on file by the QA Officer.

### **H.2 Home State Certification**

State	Certification Category	Status
Nevada Donald LaFara/Sara Rairick (775)-687-9491/9490 Certification # NV000162007A	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified

### **H.3 Additional State Certifications**

State	Certification Category	Status
Arizona Barbara Escobar/Gail Adams (602)-364-0720 Certification # AZ0467	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified
Arkansas Jeff Ruehr/Jane Hurley 501-682-0955 Certification # 88-0622	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified
California ELAP Aida Dente/George Kulasingam (510)-540-2800 Certification # 2019	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified
Idaho Steven Radakovich (208)-334-2235 Certification # NA	Waste Water Hazardous Waste Safe Drinking Water	No Program No Program Certified
Kansas Jack McKenzie/Aurora Shields (785)-296-1639/6198 Certification # E-10308	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified

## State Certifications and Parameters of Analysis

### Additional State Certifications

State	Certification Category	Status
Oregon Dr. Irene Ronning (503)-229-5505 Cert. # 10316718855	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified
Washington Mr. Stewart M. Lombard (360)-895-6178 Certification #C304	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified



#### H.4 NELAP Certification

Primary Accrediting Authority	Certification Category	Status
California NELAP Aida Dente/George Kulasingam (510)-540-2800 Certification # 01154CA	Waste Water Hazardous Waste Safe Drinking Water	Certified Certified Certified

#### 4.5 Department of Defense (DOD) Navy Certification

Accrediting Authority	Certification Category	Status
Navy (NFESC) Ms. Patricia Moreno (805)-982-1659 Certification # NFESC 413	Waste Water Hazardous Waste	Approved Approved